CIA-RDP86-00513R000101120012-3

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KADYMOV, L.G.; ALIYEV, G.K., prof., zesluzhennyy deyatel' nauki; GUSMAN, S.M., prof.; TESIER, Ya.Ye.

20 to 22

On the 70th anniversary of the Dahareridze No. 3 Glinicel Hespital. Azerb. med. zhur. 41 nc.1:84-85 Ja 464. (MiRA 17:12)

1. Glavnyy vrach klinicheskoy bol'nitsy No. 3 Emeri Exhaparidze, Baku (for Kadymov).

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AUTHOR	S:	And Abdullayev, or bu		ана са селото на село На селото на
TITLE:		A Note on the Influence of a Chlorine Admixture on the Influence Conductivity of Selenium (O vliyanii primesi khlora na teplopro=	1	
PERIO	DICAL:	Doklady AN SSSR, 1957, Vol. 116, Nr 4, pp. 598-600 (USSR).	i i	· ·
ABSTR	ACT:	the chamical composition and his students (reference 1, 2, 3, 4). In one by A. F. Ioffe and his students (reference 1, 2, 3, 4). In one production of selenium rectifiers admixtures of halogenes, in par- ticular chlorine, are used for the purpose of increasing the cur- tern passing through the semiconductor. The experiments showed the following results. During the electric formation and the further following results. During the electric formation and the further admixtures takes place, which modifies the electric and thermal admixtures takes place, which modifies the electric and thermal characteristics of the selenium layer and of the system as a whole. In the backward direction the voltage applied to the rectifier is localized almost entirely at the anode at the electron-hole transi- because of the formation of a great resistance. This causes		
Card	1/3	a temperature gradient along the semiconductor, The during mem a temperature gradient along the semiconductivity by a stationary mem mined the coefficient of thermal conductivity by a stationary mem		
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A Note on the Influence of a Chlorine Admixture on the Thermal 20-1-20/51 Conductivity of Selenium.

thod by means of a cylindrical set up, containing a sensitive semiconductor ring for the removal of lateral heat losses. A diagram illustrates the curves of the modification of heat conductivity of selenium and its dependence on the chlorine content from 20 to 22°C. The different curves are related to vitreous and crystallised se lenium. The course taken by these curves is independend of the degree of crystallization, but depends only on the admixtures. The heat conductivity decreases as far as 0,03% at an increase of the chlorine content, then it increases again and remains constant above a value of 0,5% . A similar dependence of the heat conductivity was found by the author in the case of iodine and bromine admixtures. The dependence of heat conductivity on the degree of crystallization is mainly determined by the conditions of the scattering of the phonons with increasing concentration of selenium the concentration of the admixtures is decreased and there with the mean free path of the phonons increases. By this, the frequency of their scattering decreases and the anharmonicity degree of the oscillations and therefore the heat conductivity increases. The authors here evaluate this influence of the modification of the free path and the numerical values, which were found, are given. On crystallisation the free

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		69396	
		sov/137-59-4-8423	
-	rom: Referativnyy zhurnal, Metallurg	iya, 1959, Nr 4, p 155 (USSR)	
24.77 DO AUTHORS:	Aliyev, G.M., Abdullayev, G.B.		
TITLE:	The Effect of the Admixture of Chlor Selenium $\sqrt{\gamma}$	ine on Electric Properties of ν	
PERIODICAL:	Izv.AS AzerbSSR, Ser. Fiz-tekhn. 1 k (Azerb. résumé)	him. n., 1958, Nr 4, pp 23 - 30	
ABSTRACT:	The authors investigated changes in emf \propto of Se depending on Cl concentr ture. Crystallization of a smelted out under pressure first at 130°C, t admixture up to 0.125% raise \bigcirc of S is attained at 0.125% an then \bigcirc de conductivity is preserved. The coef 85°C increases with elevated tempera	ration (0.0035 - 0.5%) and tempera- Se and SeCl _h mixture was carried hen at 200°C (40 minutes each). Cl e (up to 1,000 times) the maximum oreases with higher Cl amount. Hole ficient \propto within a range of 25 - ture. Electroconductivity in this	
Card 1/2	range of both pure and admixed Se in		
			an to the second

"APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000101120012-3 693% Sov/137-59-4-8423 The Effect of the Admixture of Chlorine on Electric Properties of Seleniun according to the exponential law. If the Cl amount increases, dissociatium work and coefficient ~ decrease, and concentration and effective mobility of charge carriers increase. It is concluded that admixtures of Cl cause the formation of additional energy levels in Se, which are arranged at the upper boundary of the filled-up zone. A.A.

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81505 sov/137-59-5-10665 24.7600 Translation from: Referativnyy zhurnal, Metallurgiya, 1959, Nr 5, p 172 (USSR) Aliyev, G.M., Abdulayev, G.B. AUTHORS: 21 On the Effect of Chlorine Admixtures on the Heat Conductivity TITLE: of Selenium N Tr. In-ta fiz. i matem. AS AzerbSSR, 1958, Vol 9, pp 20 - 26 PERIODICAL: (Azerb, résumé) The atuhor used the method of the stationary thermal field to ABSTRACT: investigate the effect of the admixture of 0.0035 - 1% Cl on the heat conductivity of Se. It was found that the heat conductivity minimum was attained with a 0.03% Cl concentration. After the Cl concentration was as high as 0.5%, the heat conductivity approached a constant value butdid not, however, attain its initial value. The course of the heat conductivity curve does qualitatively not depend on the Se recrystallization and is explained by the presence in Se mainly of phonon heat conductivity and by changes in the cross section of phonon scattering depending on the Cl concentration. A.L. Card 1/1

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	UTHORS:	Aliyev, G. M., Abdullayev, G. B. SOV/20-120-1-19/63	
•	TITLE:	The Temperature Dependence of the Thermal Conductivity of Selenium With Small Chlorine Additions (O temperaturnoy zavisi- mosti teploprovodnosti selena s primes'yu khlora)	,
	PERIODICAL:	Doklady Akademii nauk SSSR, 1958, Vol. 120, Nr 1, pp. 76 - 78 (USSR)	0
	ABSTRACT; Card 1/3	The present paper investigates the temperature dependence of the thermal conductivity of crystalline selenium with different additions of chlorine. The samples of different chlorine content were produced of a mixture of selenium tetrachloride and sele- nium (purity 99,996%). The amount of the chlorine contained in the selenium was determined argentometrically. The coeffi- cient of thermal conductivity was determined by means of the stationary method with a cylindrical apparatus. A diagram shows the temperature dependence of the coefficient of the thermal conductivity upon different chlorine contents. The coefficient of thermal conductivity decreases with rising temperature. Only in the case of pure samples there is a small deviation from the linearity. Another diagram shows the dependence of the	
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	are Dependence of the Thermal Conductivity SOV/20-120-1-19/63 With Small Chlorine Additions	
PRESENTED : SUBMITTED ;	and therefore reduces the coefficient of thermal conductivity. The deterioration of the Volt-Ampère characteristics of the solenium rectifiers as a consequence of a temperature rise partly is dependent on the decrease of the coefficient of thermal conductivity of solenium and therefore also on the decrease of the thermal scattering. Solenium with an addition of 0,0035% chlorine has its greatest thermal conductivity at 80° (which corresponds to the operational temperature of the selenium rectifiers). There are 4 figures and 13 references, allof which are Soviet. November 1, 1957, by A.F.Ioffe, Member, Academy of Sciences, USSR October 11, 1957	
	 SeleniumConductivity ConductorsTemperature factors SeleniumHeat transfer HeatConductivity Chlorine Properties Dry disk rectifiersAnalysis 	
Card 3/3		



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	(4) FILLER FLOUR ALFLACTAR Identy neur Ukrinskey JUL, Institut f Goalskittabekity 1. Optionakiy yventinnö Eruky parvoga vasooyurnoka bovantinnök from and antym vasooyurnoka bovantinnök from antym frakting antym var af antym antym frakting antym af antym af antym antym antym af antym af antym antym af antym af antym antym af antym af antym antym af antym af antym antym af anty	Muddi Inter are indicated by arrithmy running that four and the objection of the inf four and the second term of the inf four and the second of the second	Findomelectric and Optical Phanchens (and Chlorine Midmarghili, A.K. Infra-red Conduct Thin Lead Builide M. Infra-red Conducty Lead Builide M. Lead Funder The Lead Builide M. Lead Funder, Flow Motosleetric Froperius Phones J Muther M. Lisetron Exchange of 3 Autocreed Molecules
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بعدته تراجى 68263 SOV/81-59-10-34022 24.7600 Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 10, p 28 (USSR) Aliyev, G.M., Abdullayev, G.B. AUTHORS: On the Effect of the Admixture of Chlorine on the Heat Conductivity of TITLE: Selenium 1 Tr. In-ta fiz. matem. AS AzerbSSR, 1959, Vol 9, pp 20-26 (Azerbaydzhanian PERIODICAL: summary) The heat conductivity λ of amorphous and crystalline Se and the effect of ABSTRACT : a Cl admixture on it has been studied. The increase of λ at the transition from amorphous to crystalline Se is connected with the reduction of the quantity of defects in the lattice which are centers of scattering of phonons. The admixture of C1 to a certain percentage increases the efficient cross section of phonon scattering, which leads to the reduction of λ ; at a further increase of the Cl concentration due to recombination of the admixtures and the formation of neutral molecules, λ increases again. The ratio $\lambda_{cr}/\lambda_{am}$ for admixture-free samples is equal to 2, in the case of samples with Cl it is, independently from the Cl content, equal to 3. It is assumed that in the crystallization of Se the admixtures are displaced Card 1/2

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"APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000101120012-3 68263 SOV/81-59-10-34022 On the Effect of the Admixture of Chlorine on the Heat Conductivity of Selenium and are concentrated in the intercrystalline interlayer and affect their heat conductivity. V. Ostroborodova Card 2/2

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24.7700	28016 5/081/61/000/015/005/13 B101/B110	39
AUTHORS	Abdullayev, G. B., Aliyev, M. I., Bashshaliyev, A. A., Aliyev, G.M.	
TITLE:	Effect of halide impurities on the physical properties selenium	of
PERIODICAL	Referativnyy zhurnal. Khimiya, no. 15, 1961, 29-30, at 156196 (Sb. "Vopr. metallurgii i fiz. poluprovodnikov", AN SSSR, 1959, 80-88)	
crystallizat X-ray analys impurities a	withors studied the effect of halide impurities on the ion rate, electrical, thermal, and optical properties of his showed that at annealing temperatures from $60 - B0^{\circ}C$ accelerate Se crystallization. In the presence of I and	iodine Br, Se
at 80 ⁰ C. H	ystallize at 60 ⁰ C, while pure Se begins to crystallize of Calide impurities increase the electrical conductivity of Fred times. The dependence of the hole mobility on the	Se by
Card 1/2		κ,

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28016 s/081/61/000/015/005/139 Effect of halide impurities on the ... B101/B110 impurity concentrations shows a maximum. With rising temperature the hole mobility in pure Se and in Se with iodine impurities increases, while their concentration decreases. This phenomenon is explained by structural peculiarities of Se which is a polymer, and by the effect of the intercrystalline amorphous layers acting as potential barriers. On transition from the amorphous to the crystalline modification, thermal conductivity of Se increases from $3.13 \cdot 10^{-3}$ to $7.01 \cdot 10^{-3}$ cal/cm·sec·deg (25[°]C). In this case spec fic heat decreases. At 640 mp the forbidden-band width of the amorphous Se is 1.94 ev, that of crystalline Se (at 680 m/, is 1.83 ev. [Abstracter's note: Complete translation.] Х, Card 2/2

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1052/1101

The effect of gallium admixture on ...

ture range of $20 - 200^{\circ}$ C both on pure Se samples and on those with 0.25, 0.5, 1, 2, 3 and 4 weight \sharp Ga. It has been shown that with an increased Ga concentration the electric conductivity increases, reaches maximum, and then drops. At 4 \sharp Ga, added in the form of GaSe, Se changes metallic character of conductivity into semiconductor one. The electric conductivity of Se samples with a metallic Ga admixture increases with the temperature. The differential thermoelectromotive force was measured in the temperature range of $20 - 200^{\circ}$ C. At indoor temperature the thermoelectromotive force of Se is $914 \mu cv/degree$, and it drops rapidly with the increase of temperature. Samples with a Ga admixture have a hole type conductivity. GaSe and metallic Ga admixtures change essentially the course of the temperature force of Se with a GaSe and metallic Ga admixture increases essentially with the temperature. The hole mobility in Se with a GaSe admixture increases with the temperature, and in Se with a metallic Ga admixture it decreases up to 70°C and increases with a further increase of the temperature.

B. Turovskiy

[Abstracter's note: Complete translation]

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AUTHORS: Aliyev, B. D., Aliyev, G. M., Kerimov, I. G.

TITLE: Effect of some metallic impurities on electrical and thermal properties of hexagonal selenium

PERIODICAL: Referativny zhurnal, Fizika, no. 5, 1962, 29, abstrat 5E231 ("Izv. AN AzerbSSR Ser. fiz.-matem. i tekhn. n.", 1961, no. 4, 37 - 44; Azerb. summary)

TEXT: It is shown that Bi and Cd impurities up to a specific content (0.04% Bi and 0.125% Cd) reduce the thermal conductivity of Se, but raise it if their content is increased further. Bi, Cd, and Ga impurities raise the electrical conductivity of Se. Ga raises it to a higher degree than B. and Cd. Bi and Cd impurities reduce the thermo-emf of Se, whereas Ga raises it. The thermoemf of both pure and impurity-containing Se grows with temperature. The sign of the thermo-emf of both pure Se and Se containing Bi, Cd, and Ga impurities, is indicative of the hole mechanism of the carriers.

[Abstracter's note: Complete translation]

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ACCESSION NR: AP4027708	\$/0233/63/000/006i'0073/0078	
AUTHOR: Barkinkhoyev, Kh. G.; Aske	rov, Ch. M.; Allyev, G. M.	;
TITLE: The effect of a mercury admix	ture on the electric properties of	
selenium SOURCE: AN AzerbSSFI. Izvestiya. Se	riya fizmatem. i tekhn. nauk, no. θ_k	1
1963, 73-78 TOPIC TAGS: mercury, mercury vapor fusion factor, component suspension, a forces doner level, acceptor level	r, selenium, electric conductivity, dif- nolybdenum ampule, thermoelectromotive	•
tric properties of selenium was prompt subject published in literature. The sa	effect of mercury impurities on the elec- ed by the contradictory opinions on This imples involved in the test were molyb-	
denum ampules with selenium and mero samples were crystallized at 210C for the modelectromotive force were then m	cury. Following a special treatment, the 25 hours. The electric conductivity and	
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ACCESSION NR: AP4027708	•	
merifis. The same samples were used for measuring the thermoelectromotive force in relation to copper within an 8-10 degree gradient and 20-200 C temper- ative range. The experimental data reveal that the small concentrations of mercury atoms in the selenium tend to reduce its electrical conductivity. This can be explained by the assumption that the mercury atoms in the selenium pro- duce donor levels which increase with increasing impurities, intensifying their, compensation of the selenium acceptor levels. Such an effect of the impurities prior to the full compensation of the selenium acceptor levels, should lead to a reduced electric conductivity. The increasing temperature relationship of the concentration and the reduced mobility of the current carrier in selenium are natural from the point of view of the band theory. All the data published in literature indicate that the mobility increases and the concentration of the cur- rent carriers in selenium decreases with temperature. But this problem, on the whole, is still not very clear. Orig. art. has: 6 figures ASSOCIATION: AN AserbSSR SUBMITTED: 00 DATE ACQ: 17Apr64 ENCL: 00 SUB CODE: PH, CH NO REFSOV: 004 OTHER: 006		
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ASKEROV, Ch.M.; ALIYEV, G.M.; AKHUNDOVA, E.G. Heat conductivity, density, and microhardness of the system seleniumsulphur. Izv. AN Azerb. SSR. Ser. fiz.-tekh. i mat. natk no.1:83-89 '64. (MIRA 17:9)

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<u>L 7012-65</u> BAT (1)/EP&(a)-2/EAT (a)/EP?(a)-2/EBG(v)/EP?(A)/EAP(a)/EA -Po-5/Paul/Pi-10/Pa-1, ESD(ga) EDM/AN/AD	To the second second second second second second second with the second s
ACCESSION NR: AP4044630 8,0233/64/000/ AUTHORB: Aliyev, <u>G. M.;</u> Veliyev, M. I.	*0 2/0097/0100
TITLE: Effect of heat treatment on the thermal conduction	t vity of selen
SOURCE: AN AzerbSSR. Izvestiys. Seriya fiziko-tekhnid maticheskikh nauk, no. 2, 1964, 97-100	ch skikh i mate-
TOPIC TAGS: selenium, thermal conductivity, heat tree ABSTRACT: The selenium investigated was 99.9999% pure	and the second secon
ABSTRACT: The <u>Belenium</u> investigated what vacuum (10 ⁻⁴) amorphous by cooling selenium molten in vacuum (10 ⁻⁴) sangles were made and their thermal conductivities me to 300K. They were then annealed as follows: one at	as red from 77
483, 488 and A90K, and the other at 453K for 1, 4, 10 360 hours. Plots of λ against T (thermal conductivity temperature) were plotted after each annualing. All	y igainst the
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<u>L 6981-65</u> EWT(1)/EPA(8)-2/EWT(V)/EPF(n)-2/EWG(V)/EPII/EWP(q)/EWP(b)/EWA(3)	
Pe-5/Ps-4/Pt-10/Pu-4 BSD(gs)/RAEW(t) BDW/JD/AW ACCESSION NI: AP4044631 S/02:13/64/000/002/010	/0114	
AUTHORS: ALdinov, D. Sh. Aliyev, G. M.		
TITLE: On the thermal conductivity of selen we γ		
SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko tekhnicheskikh matematicheskikh nauk, nc. 2, 1954, 109-114	1	
TOPIC TAGS: selenium, thermal conductivity, glass property, perature dependence, ordered structure	:en-	
ABSTRACT: The authors first point out that the changes in th		
thermal conductivity occurring in glass-like materials in the	sof-	2.
tening interval, including glass-like selenium are debatable that there is no direct evidence to confirm the presence of a		
malous variation of the thermal conductivity of glasslike sel They have consequently investigated the thermal conductivity	enium.	
glass-like and crystalline selenium over a wide range of temp	Pra-	
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and a second	en e	







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CIA-RDP86-00513R000101120012-3



ACCESSION NR: AP4028423 $S/0181/64/006/004/1018/1022$ AUTHORS: Abdullayev, G. B.; Aliyev, G. M.; Barkinkhoyev, Kh. G.; Askerov, Ch. M.; Larionkina, L. S. TITLE: Electrical properties of crystallin.) and liquid selenium after deoxygenation SOURCE: Fizika tverdogo tela, v. 6, no. 4, 1964, 1018-1022 TOPIC TAGS: electric conductivity, selenium, deoxygenation, thermoelectromotive force, solid liquid study ABSTRACT: The authors measured the electrical conductivity and the thermoelectro- motive force of three samples of Se in the temperature interval 293-773K. The samples were characterized by the following impurity concentrations: 10^{-5} %, 10^{-4} %, and 10^{-5} % for the three samples, respectively. Measurements were made on all three samples before deoxygenation (ordinary Se) and on camples 1 and 3 after deoxygena- tion. Different jumps in conductivity were observed during fusion of all three samples of ordinary Se. The activation energy of electrical conductivity was found to be 2.05 ev for liquid Se of whis type. In the solid phase, the thermoelectro- motive force of sample 1 ordinary Se declined with increase in temperature. During Card $1/2$	i in a sé i r <u>ing per contractor de contractor p</u> er de la la l				
Larionkind, L. S. TITLE: Electrical properties of orystalling and liquid selenium after deoxygenation SOURCE: Fizika tverdogo tela, v. 6, no. 4, 1964, 1018-1022 TOPIC TAGS: electric conductivity, selenium, deoxygenation, thermoelectromotive force, solid liquid study ABSTRACT: The authors measured the electrical conductivity and the thermoelectro- motive force of three samples of Se in the temperature interval 293-773K. The samples were characterized by the following impurity concentrations: 10^{-5} %, 10^{-4} %, and 10^{-5} % for the three samples, respectively. Measurements were made on all three samples before deoxygenation (ordinary Se) and on samples 1 and 3 after deoxygenu- tion. Different jumps in conductivity were observed during fusion of all three samples of ordinary Se. The activation energy of electrical conductivity was found to be 2.05 ev for liquid Se of this type. In the solid phase, the thermoelectro- motive force of sample 1 ordinary Se declined with increase in temperature. During	ACCESSION NR: AP4028	8423	s/0181/6	4/006/004/1018/1022	
TITLE: Electrical properties of crystalling and liquid selenium after decoxygenation SOURCE: Fizika tverdogo tela, v. 6, no. 4, 1964, 1018-1022 TOPIC TAGS: electric conductivity, selenium, decoxygenation, thermoelectromotive force, solid liquid study ABSTRACT: The authors measured the electrical conductivity and the thermoelectro- motive force of three samples of Se in the temperature interval 293-773K. The samples were characterized by the following impurity concentrations: 10^{-3} %, 10^{-4} %, and 10^{-5} % for the three samples, respectively. Measurements were made on all three samples before decoxygenation (ordinary Se) and on samples 1 and 3 after decoxygena- tion. Different jumps in conductivity were observed during fusion of all three samples of ordinary Se. The activation energy of electrical conductivity was found to be 2.05 ev for liquid Se of this type. In the solid phase, the thermoelectro- motive force of sample 1 ordinary Se declined with increase in temperature. During	AUTHORS: Abdullayev Larionkina, L. S.	, G. B.; Aliyev, G. M	; Barkinkhoyev, Kh. G.;	Askerov, Ch. M.;	
TOPIC TAGS: electric conductivity, selenium, deoxygenation, thermoelectromotive force, solid liquid study ABSTRACT: The authors measured the electrical conductivity and the thermoelectro- motive force of three samples of Se in the temperature interval $293-7?3K$. The samples were characterized by the following impurity concentrations: 10^{-3} %, 10^{-4} %, and 10^{-5} % for the three samples, respectively. Measurements were made on all three samples before deoxygenation (ordinary Se) and on samples 1 and 3 after deoxygenu- tion. Different jumps in conductivity were observed during fusion of all three samples of ordinary Se. The activation energy of electrical conductivity was found to be 2.05 ev for liquid Se of this type. In the solid phase, the thermoelectro- motive force of sample 1 ordinary Se declined with increase in temperature. During		roperties of crystall	ing and liquid selenium	after deoxygenation	
force, solid liquid study ABSTRACT: The authors measured the electrical conductivity and the thermoelectro- motive force of three samples of Se in the temperature interval 293-773K. The samples were characterized by the following impurity concentrations: 10^{-3} %, 10^{-4} %, and 10^{-5} % for the three samples, respectively. Measurements were made on all three samples before deoxygenation (ordinary Se) and on samples 1 and 3 after deoxygenu- tion. Different jumps in conductivity were observed during fusion of all three samples of ordinary Se. The activation energy of electrical conductivity was found to be 2.05 ev for liquid Se of this type. In the solid phase, the thermoelectro- motive force of sample 1 ordinary Se declined with increase in temperature. During	SOURCE: Fizika tver	dogo tela, v. 6, no.	4, 1964, 1018-1022	1.	·
ABSTRACT: The authors measured the electrical conductivity and the thermoelectro- motive force of three samples of Se in the temperature interval 293-773K. The samples were characterized by the following impurity concentrations: 10^{-3} %, 10^{-4} %, and 10^{-5} % for the three samples, respectively. Measurements were made on all three samples before deoxygenation (ordinary Se) and on samples 1 and 3 after deoxygenu- tion. Different jumps in conductivity were observed during fusion of all three samples of ordinary Se. The activation energy of electrical conductivity was found to be 2.05 ev for liquid Se of this type. In the solid phase, the thermoelectro- motive force of sample 1 ordinary Se declined with increase in temperature. During	TOPIC TAGS: electri force, solid liquid	c conductivity, selen study	ium, deoxygenation, the	rmoelectromotive	
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to be 2.05 ev for liquid Se of this type. In the solid phase, the intracord puring motive force of sample 1 ordinary Se declined with increase in temperature. During	and 10 ⁻⁵ % for the th samples before deoxy tion. Different jum	ree samples, respecti genation (ordinary Se ups in conductivity we	vely. Measurements wer) and on samples 1 and) ore observed during fusi mergy of electrical con	after deoxygenur on of all three ductivity was found	
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	ACCESSION NR: AP4028423		
•	fusion the sign changed to negative, and in the liquid phase it increased absolute value. The thermoelectromotive force of samples 2 and 3 ordinary crystalline state increased with rise in temperature. During fusion it for (to zero), did not change sign, and increased again in the liquid state. deoxygenation, the conductivity at room temperature declined approximately factor of 100. No jumys were observed. The activation energy of the cond- in such liquid Se becaue 0.6 ev. The thermoelectromotive force of samples in the liquid state indicates n-type conductivity, increasing in absolute In crystalline Se of sample 3, no chermoelectromotive force was observed. observed in sample 1, but the value was shall and corresponded to hole con- "The authors express their thanks to Professor A. R. Regel' for his inter- work and for his valuable advict." Orig. art. has: 1 figure.	y Se in the ell sharply After y by a ductivity s 1 and 3 value. It was nductivity.	
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	SUBMITTED: 1952p63	ENCL: 00	
-,	SUB CODE,	THER: 011	
_	Cord 2/2	5. 	

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	ACCESSION NR: APLO39227		s/0249/64/020/	002/0027/0031		
•	AUTHORS: Abdinov, D. Sh.; Abdu	llayev, G. B.; Aliyev,	G. M.	•		
,	TITLE: The effect of antimony microhardness of selenium	admixture on density, h	eat conductivit;	y, and		
	SOURCE: AN AzerbSSR. Doklady*	, v. 20, no. 2, 1964, 2	27-31			
	TOPIC TACS: antimony, selenium	, recrystallization, se	lenium heat trea	1.tment		
	ABSTRACT: The effect of antimo was studied. The samples consi- various proportions. These pow evacuated to 10 ⁻⁴ mm Hg and sea oven at 850C for 8 hours and co were amorphous. The measuremen before they were replaced in th 180C for one hour and at 210C f relation between the physical p was studied. The variation of Cord 1/4	sted of antimony and se ders were poured into q led. In this state the oled to room temperatur ts of their heat conduc e ampules and allowed to or 60 hours. After eac roperties of every samp	lenium powders i uartz ampules wi samples were he re. At this stay stivity and dens: crystallize at h crystallization	nixed in mich were hated in an ge the samples ity were made 90, 130, and on period the mony content		
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ACCESSION NR: AP4039227 respect to antimony concentration at 20-22C is shown in Fig. 1 of the Enclosures, where the conductivity is seen to increase during the transition from the amorphous to the crystalline state. It decreased with the increase in antimony conten's to 0.125%, beyond which point it started rising. This behavior was explained by the hypothesis of V. N. Lange and A. R. Regel' (FTT, v. 1, no. 4, 1959) which states that small quantities of antimony distort the crystalline lattice of selenium, while larger amounts of antimony have the opposite effect. The variation in the microhardness, thermal conductivity, and density of crystalline selenium with respect to the antimony content is shown in Fig. 2 of the Enclosures. The microhardness minimum also occurred at 0.125% antimony content. In order to check the accuracy of the experimental results, the variation of selenium properties was calculated according to the formula derived by A. V. Ioffe and A. F. Ioffe ("DAN SSSR", 1954, v. 98, No. 5). The theoretical and experimental data correlated closely. Orig. art. has: 1 table, 2 figures, and 3 formulas. ASSOCIATION: Institut fisiki (Institute of Physics) ENCL: 02 DATE ACQ: 05Jun64 ' SUBMITTED: 19Jul63 OTHER: 002 NO REF SOV: 010 SUB CODE: SS. GC Card 2/4

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The results also indi to affect the EPR sig	ata are available to con cate that the presence of mal intensity. Orig. a	rt. has: 1 f.gure.	
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SUBMITTED: 16Ju164	ENCL: 00	SUB CODE: NP, 25	
BR REF SOVI 001	OTHER: 005		
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ALT HOR:	Veliver, N. A.S. Aliver, O. A	
TITLE	freet of sodium impurities on the electric conductivity of selenium	
nauk, n TOPIC I ABSTRAC positiv band v band v vas vai All sav ured w electr is not energy	AN AzerSSR. Isvestiya. Setiya fiziko-tekhnicheskikh i stematicheskikh AN AzerSSR. Isvestiya. Setiya fiziko-tekhnicheskikh i stematicheskikh i for 30 hours, in the salenium and gives rise to local donor state thout affecting their compensation in the lattice. the content of sodium thout affecting their compensation in the lattice. the content of sodium thout affecting their compensation in the lattice. the content of sodium the form 0.01 to 1 vtd. Samples were crystallized at 215 for 30 hours. Is for 30 hours. Is sod um influences the th a bridge in the temperature interval 25270C. The sod um influences the th a bridge in the temperature interval 25270C. The sod um influences the th a bridge in the temperature dependence of crystalline selenium, but active in the liquid state depending on the amount of supurivity the activation in the solid state fluctuates between 1.62 and 0.82ev The sodium also af-	
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L 3537-56 EPA(s)-2/EWT(m)/EPF(c)/ETC/EPF(n)-2/IWG(m)/EWP(t)/EKP(b) IJP(c) ACCESSION NR: AP5015450 RDW/JD/WW/JG UR/0249/65/021/003/0018/002	a		
AUTHORS: Abdullayev, G. B.; Abdinov, D. Sh.; Aliyev, G. M.	9		
TITLE: Effect of oxygen on transport phenomena in selenium of high purity $\frac{1}{5}$	P		
SOURCE: AN AzerbSSR. Doklady, v. 21, no. 3, 1965, 18-21			
TOPIC TAGE: selenium, selenium rectifier, thermal conductivity, electric conductivity, thermal emf, Hall effect, carrier density, Hall mobility		2011 - 2014 - 2014 - 2014	
ABSTRACT: The authors report results of investigations of the influence of artimony impurity, which effectively compensates the accepted action of oxygen, on the electric properties of crystalline and liquid selening and on the thermal conductivity of crystalline selent	n-	-	
V liquid selening and on the thermal conductivity deoxidation and after ium of purity 99.9999 per cent before and after deoxidation and after oxidation. The deoxidation was by the method of P. T. Kazyrev (FTT oxidation. The deoxidation was by the method of P. T. Kazyrev (FTT v. 1, 113, 1959). The procedure for measuring electric conductivity and the thermal conductivity as functions of the impurities and of	y		
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L 3537--66 ACCESSION HR: AP5015450 temperature was described earlier (FTT v. 4, 1018, 1964 and elsewhere) The Hall effect was measured with direct current by a compensation method in a magnetic field of 20,000 Oe. The article includes a table of the dependence of the electric conductivity, the thermal con-ductivity, the Hall density, and the Hall mobility prior to deoxidation, and also of the electric conductivity and thermal conductivity after deoxidation, as functions of the antimony content, and plots of the temperature dependence of the electric conductivity buffore and after deoxidation. The results show that the antimony has different effects on the electric and thermal conductivities before and after deoxidation, and varies with the antimony content. The jump in the conductivity occurring at the melting point also depends on the oxygen The results have a direct bearing on the fact that various content. mechanical properties of selenium rectifiers and photocells are governed principally by their oxygen content. Orig. art. has: 2 figures and 1 table. Institut fiziki AN AzerbSSR (Institute of Physics, AN ASSOCIATION: AzerbSSR) Card 2/3

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	"APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000101120012-3	3
	32952-00 EWT(m)/EWF(t)/ETI IJP(c) RDW/WW/JD/JG ACC NR: AP6017056 (N) SOURCE CODE: UR/0233/65/000/004/0074/0079 77	
1	(10)	
0	UTHOR: <u>Abdinov, D. Sh.; Aliyev, G. M.</u> RG: none	
T	ITLE: Effect of oxygen additions on the electrical properties of selenium	
	OURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-tekhnicheskikh i matematicheskikh nauk, o. 4, 1965, 74-79	
	OPIC TAGS: selenium, thermal (mf, Hall effect, activation energy, Hall mobility, urrent carrier, electric property	-
di to ca	BSTRACT: Measurements were made of the effects of <u>Sb</u> additions on the electrical con- uctivity σ of Se before and after deoxygenation, after oxygenation, as well as of the emperature function of the Hall effect in the solid and <u>liquid</u> States. The work was arried out to fill a gap in the literature. The antimony was added as Sb and Sb ₂ Se ₃ n amounts of 0.05, 0.1, 0.125, 0.25, 0.5, 0.75, 1, 2, and 5 wt %. For ordinary Se	
() ra tl P	prior to deoxygenation), the σ decreases with increasing content of Sb and at 0.5% it eaches a minimum; with further addition of Sb, it increases. The σ was found to be he same for Sb ₂ Se ₃ . At 20-220°C, the σ practically does not change. At the melting oint, the σ drops abruptly. After melting, (starting at 240°C), it rises exponential- y with temperature. Activation energies ΔE , calculated from the slope of the lg vs	
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vs $1/T$ curve, are	affected litt	lu by small	concentration	e: etanti	ng at 1% Sh	AF and	7	
dually increases	and at 5%, rea	ches 0.52 ev	. In melting	pure Se.	the concent	ration	of	
the current carri								
it continues to d in the solid and								
ing. Temperature								
From the Hall eff	ect and the th	ermal emf, i	s about the s	ame. At	room tempera	ture, t		
all mobility in	pure Se is equ						li-	
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erature data. T ture in the solid prows exponential	The mobility of state and in ly with temper	melting it 1 ature. In c	rops abruptly onclusion the	, but in authors	the liquid s thank Profes	state, i ssor G.	t B.	
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terature data. The solid cure in the solid prows exponential <u>bdullayev</u> for su for assistance in	he mobility of state and in ly with temper pervising the	melting it 2 ature. In c work and Ya. Hall effect	cops abruptly onclusion the N. Nasirov,	, but in authors R. Kh. Na has: 2	the liquid s thank Profes ni and V. B.	state, s ssor <u>G.</u> Antono gures.	t B.	
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cerature data. The solid grows exponential Abdullayev for suffor assistance in SUB CODE: 20/	he mobility of state and in ly with temper pervising the measuring the	melting it 2 ature. In c work and Ya. Hall effect	rops abruptly onclusion the <u>N. Nasirov,</u> , Orig. art.	, but in authors R. Kh. Na has: 2	the liquid s thank Profes ni and V. B. tables, 3 fi	state, s ssor <u>G.</u> Antono gures.	t B.	



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ACC NR: AT6001330 SOURCE CODE: UR/0000/65/000/000/0027/00	29
AUTHOR: <u>Aliyev, G. M.; Larionkina, L. S.;</u> Dzhalilov, N. Z.	
ORG: none ßt	
TITLE: The production of selenium single crystals	
SOURCE: <u>AN AzerbSSR. Institut fiziki</u> . Selen, tellur i ikh primenenjye (Selenium tellurium and their utilization). Baku, Izd-vo AN AzerbSSR, 1965, 27-29	,
TOPIC TAGS: selenium, single crystal growth, single crystal production, growth rate, pressure dependence, illumination, ultra high purity metal, heat treating nace	fur-
ABSTRACT: Methods of increasing the normally slow growth rate of selenium single	e
<u>crystals were studied</u> . The growth rate is slow owing to the closed chain-like structure of the amorphous selenium molecules. The single crystals were grown from a vapor in a vacuum and also under slight pressure of argon or helium. Three tubes made of Mo glass (50 cm high and 3.5 cm in diameter) were filled with powdered senium of 99.99999% purity to a height of about 6 cm and evacuated to 10^{-3} mm Hg pr sure; two of these were then filled with argon and helium respectively to a pres-	ruc-
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sure of 1 atm. All the tubes were then placed in a cylindrical furnace and heated to 260°C (a schematic of the apparatus is shown). After 8 days at 260°C, the tubes were quickly removed from the furnace. On the walls of all the tubes, as a result of the removal from the furnace, needlelike crystals grew away from the wall toward the interior and slightly downwards. After crystallization, the tube walls were covered with a red deposit in the case of helium and argon and with a gray deposit for the vacuum. For the vacuum-grown crystals the needles were short and cactuslike, while in argon and helium the growth was typified by a uniform density of needles of lengths varying from 0.5 to 1.5 cm; in helium the needles were sligh γ longer. An x-ray rotating pattern of a needlelike single crystal of selenium is shown. The increased growth rate of the selenium resulted in the longer crystals. The lack of data on the thermal, electrical and photoelectrical properties of selenium single crystals is attributed to the difficulties encountered in growing selenium crystals. The authors express their gratitude to <u>K. P. Hamedov</u> for the x-ray pattern. Orig. art. has: 3 figures.

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ORIG REF: 002/

OTH REF: 003

AUTHOR: Achundov	20		C) JI) DE: UK/0233/65/0.10	58	
ductivity of hex	gonal and liquid	and dysprosium imp selonium v7 iz-tekhn i matom n,	no. 6, 1965, 69-74	trical con-	
TOPIC TAGS: sul elicities cond ABSTRACT: In or and the jump in S, Cl and Dy or 99.999% <u>pure</u>)(i melting range wa itios. It is se centration of Cl ties can be stud place the volati strongly decrease	fur, chlorine, dys uction of the electrical control the of selenium n the hexagonal must s studied. Fig. en that Dy increas , the more slowly ied in the product le Cl impurities. es it. A study of	sprosium, semiconduc e influence of impu- nductivity σ of sel of brands B ₃ and B odification and in 1 shows the curve o sees σ by a factor o σ reaches a maximu- tion of selenium re- of the temperature of the temperature of	stor conductivity, s rities on the format enium on <u>melting</u> , fith (respectively 99.5 the liquid state ind f σ vs. the concentr f 10, and that the h m. This indicates f ctifiers and can be ases σ in Equston ependence of σ shows antially with the in evels in Se, and the	ion of current as offect of 99 and cluding the ration of impu- nigher the con- that Dy impuri- used to re- um, whereas S ad that o in mpurity concen-	
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ACC NR: AP6023950 the current carrier concentration and hence σ. The decrease in σ caused by S is apparently due to the fact that the acceptor action of oxygen is partly offset by sul- apparently due to the fact that the acceptor action of oxygen is partly offset by sul- fur impurities, which decrease the carrier concentration and hence σ. The jump in σ fur impurities is due to the presence of impurities in selenium which after melting become on melting is due to the presence of impurities in selenium which after melting become inactive. In conclusion, authors thank Prof. <u>G. A. Abdullayev</u> for his steady interest and useful suggestions. Orig. art. has: 5 figures, I table and 1 formula. Fig. 1. Electrical conductivity of selenium vs. concentration of Cl, S and Dy impurities: 1 - Set#Cl; 2 - Se + #S; 3 - 3e + #Dy.	سو وهد	L 04971-67
the current carrier concentration and hence σ . The decrease in σ caused by S is apparently due to the fact that the acceptor action of oxygen is partly offset by sul- fur impurities, which decrease the carrier concentration and hence σ . The jump in σ for melting is due to the presence of impurities in selenium which after melting become on melting. In conclusion, authors thank Prof. G. A. Abdullayev for his steady interest inactive. In conclusion, authors thank Prof. G. A. Abdullayev for his steady interest and useful suggestions. Urig. art. has: 5 figures, I table and 1 formula.		
Fig. 1. Electrical conductivity of selenium		the current carrier concentration and hence σ . The decrease in σ caused by S is apparently due to the fact that "the acceptor action of oxygen is partly" offset by sul- fur impurities, which decrease the carrier concentration and hence σ . The jump in σ
Fig. 1. Electrical conductivity of selenium		inactive. In conclusion, authors thank Froi. U. A. Authors in a formula.
10 ¹ 01 02 0.5 1 % CES, Dy		Fig. 1. Electrical conductivity of selenium vs. concentration of Cl, S and Dy impurities: 1 - Set#Cl; 2 - Se + #S; 3 - 3e + #Dy.
SUB CODE://,20 SUBM DATE: none/ ORIG REF: 008/ OTH REF: 009		

ORG: none TITLE: Effect of sodium on the heat conductivity and density of selerium SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-tekhnicheskikh i matematicheskikh nauk, no. 4, 1965, 84-89 TOPIC TAGS: selenium, heat conductivity, sodium ABSTRACT: Density (δ) and heat conductivity (λ) measurements were made on Se (99.9999% ABSTRACT: Density (δ) and heat conductivity (λ) measurements were made on Se (99.9999% ontaining 0.034, 0.17. 0.34, 0.85, 2, 3.4 at % Na. The temperature function purity) containing 0.034, 0.17. 0.34, after which the properties of both crystalline Se, λ has a minimum for 0.34 at % Na, after which the properties of both approach those of pure Se. The absolute values of a decrease with an increase in impu- approach those of pure Se. The absolute values of a decrease with an increase in impu- approach those of pure Se. The absolute values of a decrease with an increase in further increase to 0.85 at % decreases a to 0.068 g/cm ³ . Measurements of x-ray and further increase to 0.85 at % decreases a to 0.068 g/cm ³ . Measurements of x-ray and further increase of polycrystalline Se and the addition of Na confirm the assump- pyncnometric densities of polycrystalline Se and the addition of Na confirm the assump- tion that the Se has vacancies and pores. Orig. art. has: 3 formulas, 2 tables, 2	"APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000101120012-3	
AUTHOR: Veliyev, M. I.; Aliyev, G. M. ORG: none TITLE: Effect of sodium on the heat conductivity and density of selerium SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-tekhnicheskikh i matematicheskikh nauk, no. 4, 1965, 84-89 TOPIC TAGS: selenium, heat conductivity, sodium ABSTRACT: Density (δ) and heat conductivity (λ) measurements were made on Se (99.9999§ purity) containing 0.034, 0.17, 0.34, 0.85, 2, 3.4 at § Na. The temperature function of λ is expressed by $\lambda \sqrt{T^{\alpha}}$. For amorphous Se, λ has a maximum for 0.17 at § Na and for crystalline Se, λ has a minimum for 0.34 at § Na, after which the properties of both rity content. With the addition of up to 0.034 at § Na, a increases to 0.153 g/cm ³ ; rity content. With the addition of up to 0.034 at § Na, a increases to 0.153 g/cm ³ ; tion that the Se has vacancies and pores. Orig. art. has: 3 formulas, 2 tables, 2	L :32951-66 EVI(m)/EVP(t)/ETI IJF(c) RDW/JD/JG ACC NB: AP6017058 (N) SOURCE CODE: UR/0233/65/000/004/0084/0089	
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TOPIC TAGS: selenium, heat conductivity, sodium ABSTRACT: Density (δ) and heat conductivity (λ) measurements were made on Se (99.99998 ADSTRACT: Density (δ) and heat conductivity (λ) measurements were made on Se (99.99998 purity) containing 0.034, 0.17, 0.34, 0.85, 2, 3.4 at % Na. The temperature function of λ is expressed by $\lambda \sqrt{T^{\alpha}}$. For amorphous Se, λ has a maximum for 0.17 at % Na and for crystalline Se, λ has a minimum for 0.34 at % Na, after which the properties of both approach those of pure Se. The absolute values of α decrease with an increase in impu- approach those of pure Se. The absolute values of α decreases to 0.153 g/cm ³ ; rity content. With the addition of up to 0.034 at % Na, α increases to 0.153 g/cm ³ ; further increase to 0.85 at % decreases α to 0.068 g/cm ³ . Measurements of x-ray and further increase to 0.85 at % decreases α to 0.068 g/cm ³ . Measurements, 2 tables, 2	SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-tekhnicheskikh j matematicheskikh hauk,	
SUB CODE: 11, 20/ SUBM DATE: 10Mar65/ ORIG REF: 010/ OTH KEF: 002	TOPIC TAGS: selenium, heat conductivity, sodium ABSTRACT: Density (δ) and heat conductivity (λ) measurements were made on Se (99.999% purity) containing 0.034, 0.17, 0.34, 0.85, 2, 3.4 at % Na. The temperature function of λ is expressed by $\lambda \nabla T^{\alpha}$. For amorphous Se, λ has a maximum for 0.17 at % Na and for crystalline Se, λ has a minimum for 0.34 at % Na, after which the properties of both approach those of pure Se. The absolute values of α decrease with an increase in impu- approach those of pure Se. The absolute values of α decreases to 0.153 g/cm ³ ; rity content. With the addition of up to 0.068 g/cm ³ . Measurements of x-ray and further increase to 0.85 at % decreases α to 0.068 g/cm ³ . Measurements of x-ray and pyncnometric densities of polycrystalline Se and the addition of Na confirm the assump- pyncnometric densities of polycrystalline Se and the addition of Na confirm the assump- tion that the Se has vacancies and pores. Orig. art. has: 3 formulas, 2 tables, 2	

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	ACC NR. AP6011427 ECURCE CODE: UR/0020/66/167,004/0782/0784			
	AUTHOR: Aliver, G. M.; Abdinov, D. Sh.; Mekhtiyuva, S. I.			
	ORG: Institute of Physics, Acidemy of Sciences, AzerbSSR (Institut filliki Akademii)			
	TITLE: <u>Helenium</u> as a polymer semiconductor and the mechanism of its conductivity			
	SOURCE: AN SSSR. Doklady, v. 167, no. 4, 1966, 782-784			g
1000 C	TOPIC TAIS: selenium, polymer structure, semiconducter, matrial ductivity, thermoelectric power, Hall effect, liquid state, carrier dunsity, suctoic fundactivity			к. К
	ABSTRACT: In view of the fact that the mechanism of conductivity of melenium has not been fully explained and the experimental data contradictory, that the influence			
	of different impuritics, especially oxygen, on the electrical properties of selenium has not been clarified, nor has the melting of selenium and its liquid state been			
	studied, the authors present the results of a comprehensive investigation of the elec- tric conductivity, thermoelectric power, and Hall effect in solid and liquid selenium (from 20 to 450°), including the melting region. The experiments were made with very pure selenium type B_5 (99.9999)%) before and after removal of oxygen, and with dif-			
ĺ	ferent degrees of oxidation and with different amounts of oxygen-compensating impuri- ties (Sb ₂ Cd, Mn). The electric conductivity (σ) of solid and liquid selenium in-		$(2, \gamma)$	
	creases with the temperature exponentially, and experiences an abrupt decrease during melting. The carrier density is found to be independent of the temperature (~10 ¹⁵			
	cm^{-3}). The jumplike decrease in σ on melting is due both to the decrease in the	Z	•	
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L 07250-67 EWI(d)/EWI(m)/EWP(w)/EWP(v)/EWP(t)/EWP(k)/EWP(h)/EWP(h)/EWP(1) ACC NR: AF602891B IJP(c) J)/RH SOURCE CODE: UR/0233/66/0C0/001/COT7/0084 AUTHOR: Abdullayev, G. B.; Mekhtiyeva, S. I.; Abdinov, D. Sh.; Aliyev, C. M. AUTHOR: Abdullayev, G. B.; Mekhtiyeva, S. I.; Abdinov, D. Sh.; Aliyev, C. M. TITLE: New properties of high jurity selenium SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-takhnicheskikh i matematicheskikh nauk, so. 1, 1966, 77-84 TOFIC TAGS: selenium, chemical purity, oxidation, thermoelectric pover, heat conduc- tion, physical diffusion, activition energy, semiconductor conductivity tion, physical diffusion, activition energy, semiconductor conductivity stood, the authors have checked on the hypothesis that many of them are due to the presence of oxygen and oxygen complexes in the selenium. The authors have investigated presence of oxygen and oxygen complexes by and B ₅ , with purity 99.99:99 and 99.99999%) selenium of special high purity (grades D, and B ₅ , with purity 99.99:99 and 99.99999%) selenium of special high purity (grades D, and B ₅ , With purity 99.99:90 and 99.99999%) and measurements are indicated in earlier papers (FTT v. 6, 1020, 1954 and elsewhere). The parameters tested were the electric conductivity, the thernoelectric power, the The parameters after introducing impurities, and the effect of oxygen-compensating im- microhardness after introducing impurities, and the point of view that the oxygen im- purities (Cd, Sb, Ma, TI, Na, S). The measurement results are presented in graphic purities produced in selenium acceptor levels, whereas the addition of the impurities purities produced in selenium acceptor levels, whereas the addition of the impurities	"APPROVED FOR RELEASE: 03/20/2001 CIA-RDP86-00513R000101120012	2-3
	L 07250-67 EWT(d)/EWT(m)/EWP(w)/EWP(v)/EWP(t)/ETT/EWP(k)/EWP(h)/EWP(1)/COTT/0084 ACC NR: APG020918 IJP(c) J)/RH SOURCE CODE: UR/023/66/OCD/001/COTT/0084 AUTHOR: Abdullayev, G. B.; Mekhtiyeva, S. I.; Abdinov, D. Sh.; Aliyer, G. M. TITLE: New properties of high rurity selenium SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-takhnicheskikh i matematicheskikh nauk, solRCE: AN AzerbSSR. Izvestiya. Seriya fiziko-takhnicheskikh i matematicheskikh nauk, no. 1, 1966, 77-84 TOFIC TAGS: selenium, chemical purity, oxidation, thermoelectric power, heat conduc- tion, physical diffusion, activation energy, semiconductor conductivity tion, physical diffusion, activation energy, semiconductor conductivity tood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have checked on the hypothesis that many of them are due to the stood, the authors have investigate presence of oxygen and oxygen complexes, and the short. The methods for oxidation before and after de-oxidation, and also after oxidation. The methods for oxidation and measurements are indicated in earlier papers (FTT v. 6, 1020, 1954 and elsewhere) and measurements are indicated in earlier papers (FTT v. 6, 1020, 1954 and elsewhere) and measurements after introducing impurities, and the effect of oxygen.compensating im- microhardness after introducing impurities, and the effect of oxygen.compensating im- microhardness after introducing impurities, and the effect of oxygen.compensating im-	

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	which oxidize easily is equivalent to de-oxidation. The latter makes selenium closer to an intrinsic semiconductor. It is concluded that the p-conductivity of selenium, the fact that the thermal conductivity, the electric conductivity, the density, and the microhardness go through a minimum when impurities are introduced, the anomalously the microhardness go through a minimum when impurities are introduced, the anomalously large value of the scattering cross section, the stron decrease in the electric con- ductivity and thermoelectric power on melting, as well as other factors are connected with the presence of oxygen impurities and its complexes in the selenium. Evidence in favor of this conclusion is drawn from a comparison of numerous experimental data by others. The influence of oxygen on the rectifying properties of selenium is also discussed. Orig. art. has: 6 figures and 1 formula.	47		
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ACC NRI AF7002839 (A) SOURCE CODE: UR/0253/56/00004/0087/0090		
AUTHOR: Dzhalilov, N. Z.; Azizov, T. S.; Aliyev, G. M.		
ORG: none		
TITLE: Influence of electron bombardment on the electric conductivity of hexagonal selenium single crystals		
SOURCE: AN AzerbSSR. Izvestiya. Seriya fiziko-tekhnicheskikh i matematicheskikh nau no. 4. 1966. 87-90		
TOPIC TAGS: selenium, semiconductor single crystal, electric conductivity, electron bombardment. crystal defect, annealing		
ABSTRACT: The authors present the results of an investigation of the influence of electron bombardment on the electric conductivity of single crystals of hexagonal selenium grown from the vapor phase and from the melt. The resistance was measured with a dc bridge in conjunction with a mirror galvanometer. The bombardment and the measurement were at 300K, with the samples kept in darkness prior to the measurement to eliminate the effect of light on the conductivity. The bombardment was with 5-Me electrons from an accelerator, in pulses of 3 µsec length and a reputition frequency 400 cps. The results show that bombardment increases the conductivity from 4 x 10^{-5} (ohm-cm) ⁻¹ in darkness to 6.8 x 10^{-4} within a few minutes, and then gradually to 8 x 10^{-4} after ninety minutes. The increase in conductivity is due to defects in the structure produced by the electron bombardment and to the ionizing effect of the	Y	
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ACC NR: AP7002839 irradiation on the impurity atoms. The estimated maximum energy transferred to the selenium atom by bombarding electrons of energy 1 and 3 Mev is 62 or 729 ev respectively. The tests have shown that annealing of the sample after the bombardment rapidly returns the conductivity to its initial value. While the variation of the electric conductivity of selenium as a function of the bombarding electron flux density agrees with that of germanium, the behavior of the selenium after bombardment differs from that of germanium or silicon, in that no special annealing is necessary to revonve the radiation defects. Orig. art. has: 1 figure and 1 formula. SUB CODE: 20/ SUEM DATE: 00/ ORIG REF: 003/ OTH REF: 004.

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