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	10.9110	22247 R/008/60/000/005/006/014 A231/A126	
	AUTHOR:	Arcan, M.	
	TITLE:	Investigations into the bending of girders with rectangular cutouts	
	PERIODICAL:	Studii și Cercetări de Mecanică Aplicată, no. 5, 1960, 1175 - 1193	
	outs on the girder. Spe on the distr ture paper, The experime in the labor tii si econo Economy of (The author examines the pure bending of girders with an axial ngular cutouts in function of the influence of this row of cut- distribution of stresses and the ability of distortions of the cial attention is given to the influence of the row of cutouts ibution of the unitary stresses in front of the webs. In a fu- he will present the investigations on bending with cutting forces intal studies have been carried out by the photo-elastic method atories of the INCERC (Institutul de cercetări pentru construc- omia construcțiilor = Research Institute for Constructions and constructions) in Bucharest. Models, having some characteristics and is the height of the cutout, in which H is the height	X
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Investigations into the bending of ...

of the girder and α varies between 0.15 and 0.66. β H is the base of the cutout, in which β varies between 0.01 and 0.5. β 'H is the base of the web, β ' varying between 0.166 and 1.165. K = $\frac{\beta}{2}$ varies between 0.015 and 3.333. K! = $\frac{\beta}{2}$ varies between 0.25 and 4.6. Photoelastic tests for the determina-tion of isoclines have been made on plexiglass samples and for the determination of isochromates on Dinox F-110 samples. Isoclines have been established in white light and recorded by drawing. Isochromates have been recorded by photographing in sodium monochromate light. The device used for bending tests is shown. The following observations have been made on the behavior and participation in distortion of different zones of the girder: The bases of the girders have formed an isoclinal zone of 0° parameter, except disturbances around the corners. The webs have a similar trend, not depending on their absolute dimensions for the same $K' = \frac{\beta}{N}$ <u>s'</u> ratio. In case of K >1, the following categories of singular points could be observed: a) Internal singular points of the attractive type, one in every web, located in its center; b) Singular contour points of the repulsive type, one at every vertical side of the cutout. They indicate a change of the sign of stress on the contour and thus are neutral; c) Singular corner points located at every corner of the cutouts, being of II order. In case

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Investigations into the bonding of

of K'<1 two more singular points are formed, located at the two ends of the webs, on their vertical axes. They are singular internal points of the repulsive type. In case of their appearance, the singular internal point located in the center of the web is transformed into a repulsive-type point. In this case the singular corner points become attractive-type points. The study of isochromates confirms the observations made on isoclinal and isostatic tables. . The isochromatic tables present in the base-bands parallel to the girder axis the characteristics of the pure bending of the band. Tf the height of the cavities is relatively small, $\alpha < 0.33$, the behavior of . these bands is not disturbed and their parallelism is maintained behind the zones of the cutouts. In case of α >0.33, the influence of the cutouts is extended to the outer fibers. If K'>1, there appear on both sides of the cutouts zones of very small stresses along the height of the cutouts and along the length αH (Fig. 10 b). The singular internal points are in the neutral zones. On the basis of these data, the author was able to establish the type of distortion of cutout and web. The singular points located in the middle of the verticals which limit the cutout, correspond to some inflection points. The formation of a neutral zone having a 2a height is very The evaluation of the unitary normal forces in the vertical secimportant.

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.(3)

Investigations into the bending of ...

tion of the web axis shows a Navier type diagram modified by the appearance of some high neutral zones. In case of $K' \leq 1$, the unitary normal forces in the external fiber of the web's transversal section is superior to the unitary normal stress in the external fiber of the transversal section in front of the cutout, which is contrary to expectations. This is due to the fact that the inertia moment of the cutout section is smaller, so that the unitary normal force in the external fiber is greater than in the web's section. A calculation formula for the unitary normal force in the most stressed section of the web can easily be established. Starting with

$$f = \pm t(y - a),$$

in which $y \ge a > 0$, and t is a constant of proportionality, the author deduces

$$p = \pm \frac{M}{I_a - aS_a} (y - a),$$

in which M is the external moment, I_0 the inertia moment of the active section in relation to the neutral axis and S_0 twice the value of the static moment of one of the active section's half value. The expression (3) is a general form of the Navier's formula for the pure bending. In case of

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CIA-RDP86-00513R000101920014-3

R/008/62/013/002/006/009 14.4200 D272/D308AUTHOR: Arcan, M. TITLE: A method of designing rigidity diaphragms with a row of holes PERICDICAL: Studii și cercetări de mecanică aplicată, no. 2. 1962, 451 - 471 TEXT: The differential equation of the shearing forces in upright beams is established, and solutions are given for various types of loadings - uniformly distributed load (due to wind), triangular load with its maximum at the diaphragm extremity (due to seismic action) and concentrated loads at the diaphragm extremity. In all cases the conclusions were verified by photoelastic models. The method of calculation is based on the hypothesis that the zone of the row of holes can be replaced by a strip with reduced elasticity modulus, of holes can be replaced by a strip with reduced trasticity modulas, obtaining thus a beam without holes with an inhomogeneous structure. The results are given in the form of graphs. There are 10 figures. ASSOCIATION: Institutul de cercetări în construcții și economia con-strucțiilor, București (Institute of Construction Research and Construction Economics) Card 1/1

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R/016/62/007/004/001/002 E191/E135

AUTHORS: Arcan, M., and Nicolau, Ed. TITLE: Investigations of load carr

Investigations of load carrying building structures by means of plastic models

FERIODICAL: Revue de mécanique appliquée, v.7, no.4, 1962, 751-780.

The work of the INCERC with models made of plastics TEXT: for the stress analysis of building structures is reported. The diversity of the methods used is emphasised since every method is limited. Models of up to 1 m size are used in spite of differences of behaviour on account of a higher Poisson's ratio in plastics and the unsuitability for destructive testing. Model laws have been studied to determine the choice of materials and techniques and the interpretation of results. The conditions of similarity are obtained from the equilibrium of an elementary slab, the geometric relations between the deformations and displacements and the physical relations between the stresses and strains. The similarity factors are hereby determined only in part: the remaining factors being chosen from the test conditions, including space available and adequate deflections of the measuring devices. Card 1/4

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Investigations of load carrying ... R/016/62/007/004/001/002 E191/E135

The inexpensive investigation of many design variants is made possible by model techniques. Examples are discussed, including several arch constructions, a circus roof structure, a multi-arch factory roof and a restaurant structure. Distribution of load carrying in complex statically indeterminate structures is among the main features discoverable by models. Structural elements made of composite materials have been examined by model techniques including reinforced and pre-stressed concrete structural elements, pressure distributions when two bodies with a different Young's modulus are in contact (for instance, a foundation on soil), and the simplification of models by the replacement of cavities with a material of low modulus. For example, the neutral axis of reinforced concrete beams was examined on the assumption of an ideal bond. Reinforced concrete beams in the condition of incipient cracking, owing to a loss of bond in the section between the cracks, have been tested by model techniques. Reinforced models with artificial cracks were used, made from a material specially developed at the Institute with a high optical activity, a reduced edge effect, and a capacity to be bonded with adhesives Card 2/4

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to the reinforcement consisting of high tensile steel wire. Two types of crack were tested, namely with and without the capacity to transmit compressive stresses. For large crack heights it was found that the neutral axis coincides with the tip of the crack so that the crack does not propagate with an increase in the load. Generally, photoelastic models were found to behave in close agreement with actual full-scale structures. Pre-stressed complex structures, such as the dome of the Bucharest State Circus building, were model tested with strain gauges. Cavities can be replaced by a material with a low equivalent shear modulus. A wider range of materials is called for with Young's moduli between 20 and 40 000 kg/cm². Practical hints are given on: model making including pre-stressed models; method of prestressing with cables and its inspection by sound frequency measurement; strain and deflection measurements. 'There are 24 figures and 1 table.

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REAL PROPERTY IN -R/016/62/007/004/001/002 Investigations of load carrying... E191/E135 ASSOCIATION: Nauchno-issledovatel'skiy institut po stroitel'stvu i ekonomike stroitel'stva - INCERC -Bukharest (Scientific Research Institute of Building and Building Economics - INCERC - Bucharest) Card 4/4**建筑**和19月

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APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000101920014-3"

ARCAN, M.; NICOLAU, Ed.

Structural strength of buildings studied with plastic models. Studii core mee apl 13 no.3:621-649 '62.

1. Institutul de cercetari in constructii si economia constructiilor INCERC, Bucuresti. 2. Membru al Comitetului de redactie, "Studii si cercetari de mecanica aplicata" (for Arcan).

APPROVED FOR RELEASE: 06/05/2000

ARCAN, M.; NICOLAU, Ed.

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"Stresses and deformations in the machine parts and joints" by N.J.Prigorovskiy. Reviewed by M.Arcan and Ed.Nicolau. Studii cerc mec apl 13 no.3:807-808 '62.

1. Institutul de cercetari in constructii si economia constructiilor (INCERC), Bucuresti, si membru al Comitetului de redactie, "Studii si cercetari de mecanica aplicata" (for Arcan).

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Difficulties of diagnosis of cerebral tumors in children. Diagnostic value of the intracranial hypertension syndrome.(1.H.S.) Fediatria (Bacur.) 13 no.52359-395 S-0 164

1. Luorare efectuata in Serviciile de pediatrie, nourologie si neurochirurgie, Timisoara.

APPROVED FOR RELEASE: 06/05/2000

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ARCANIN, Blanka, ing.

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Pyrethrum, a plant protective. Kem ind 10 no.1:21-22 Ja '61.

1. Zavod za zastitu bilja, Zagreb.

APPROVED FOR RELEASE: 06/05/2000



POKROVSKIY, A.A.; ARCHAKOV, A.I.

Study of the effect of promazine on the solubilization of liver acetylesterase (3.1.1.6) as affected by carbon tetrachloride. Dokl. AN SSSR 160 no.5:1203-1205 F '65.

(MIRA 18:2)

1. Vtoroy Moskovskiy gosudarstvennyy meditsinskiy institut im. N.I. Pirogova. Submitted October 12, 1964.

APPROVED FOR RELEASE: 06/05/2000

PANCHENKO, L.F.; ARCHAKOV, A.I.

Activity of acid phosphatase (phosphomonoesterase-II) in the brain of rats in traumatic shock. Biul.eksp.biol. i med. 59 no.5:51-53 ¹65. (MIRA 18:11)

1. Kafedra biokhimii (sav. - chlen-korrespondent AMN SSSR prof. A.A.Pokrovskiy) II Moskovskogo gosudarstvennogo meditsinskogo instituta imeni N.I.Pirogova. Submitted July 12, 1964.

APPROVED FOR RELEASE: 06/05/2000



1918

ARCHAKOV, N. V., Cand Med Sci -- (diss) "Morphology of the -the Medial Cutaneous Brachial Nerve and Medial Cutaneous AntAbrachial Nerve in Woronezh, 1957. 19 pp (Voronezh State Medical Inst), 100 copies (KL, 48-57, 109) - 66 -

ALTERNATION OF THE ALTERNATION O

APPROVED FOR RELEASE: 06/05/2000

ARCHAKOV, Yu.I., Cand 'ech Sci -- (diss) "Study of the stability of steels applicable in machines and devices equipment of the petroleum and chemical industry in high temperatures and pressures of hydrogen." Len, 1958, 10 pp (Min of Higher Education USSR. Len Urder of Ladbr and Banner Tech Inst im Lensovet) 120 copies (KL, 29-58, 131) - 48 -

APPROVED FOR RELEASE: 06/05/2000

"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000101920014-3

 Translation from: Referativnyy zhurnal. Khimiya, 1959, Nr 9, p 59 (USSR) AUTHOR: <u>Archakov, Yu.I.</u> TITLE: <u>Hydrogen in Steel</u> PERIODICAL: Tr. Leningr. tekhnol. in-ta im. Lensoveta, 1958, Nr 46, pp 125 - 132 ABSTRACT: The solubility of hydrogen (SH) in some ferrite and austenite steels and the effect of the individual alloying additions (W, V, Ti and Cr) on SH at a pressure of 100 atm and a temperature of 500°C are investigated. An installation is described for saturating metals with hydrogen permitting the fast cooling of samples from high temperatures to room temperature without reducing the gas pressure. The results obtained are presented in graphs and tables, from which follows that small additions of W (1.61%) and V (0.45%) increase SH 5.5 times and twice, respectively, an admixture of 0.48% Ti practically does not affect SH, and the increase in the Cr content to 13% increases SH only insignificantly. In alloyed steels of the austenite class SH is by 4 times higher than in carbon steel of grade 20 and by ~ 6 times higher than in alloyed steels of the austenite class. With the aim of explaining the possibility of extending the law 		18-81-00	69712 sov/81-59-9-30466
 TTTLE: Hydrogen in Steel. PERIODICAL: Tr. Leningr. tekhnol. in-ta im. Lensoveta, 1958, Nr 46, pp 125 - 132 ABSTRACT: The solubility of hydrogen (SH) in some ferrite and austenite steels and the effect of the individual alloying additions (W, V, Ti and Cr) on SH at a pressure of 100 atm and a temperature of 500°C are investigated. An installation is described for saturating metals with hydrogen permitting the fast cooling of samples from high temperatures to room temperature without reducing the gas pressure. The results obtained are presented in graphs and tables, from which follows that small additions of W (1.61%) and V (0.45%) increase SH 5.5 times and twice, respectively, an admixture of 0.48% Ti practically does not affect SH, and the increase in the Cr content to 13% increases SH only insignificantly. In alloyed steels of the austenite class SH is by 4 times higher than in carbon steel of 		Translation f	rom: Referativnyy zhurnal. Khimiya, 1959, Nr 9, p 59 (USSR)
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ABSTRACT: The solubility of hydrogen (SH) in some ferrite and austenite steels and the effect of the individual alloying additions (W, V, Ti and Cr) on SH at a pressure of 100 atm and a temperature of 500° C are investigated. An installation is described for saturating metals with hydrogen permitting the fast cooling of samples from high temperatures to room temperature without reducing the gas pressure. The results obtained are presented in graphs and tables, from which follows that small additions of W (1.61%) and V (0.45%) increase SH 5.5 times and twice, respectively, an admixture of 0.48% Ti practically does not affect SH, and the increase in the Cr content to 13% increases SH only insignificantly. In alloyed steels of the austenite class SH is by 4 times higher than in carbon steel of		TITLE:	Hydrogen in Steel.
and the effect of the individual alloying additions (W, V, 11 and of) on SH at a pressure of 100 atm and a temperature of 500° C are investigated. An installation is described for saturating metals with hydrogen permitting the fast cooling of samples from high temperatures to room temperature without reducing the gas pressure. The results obtained are presented in graphs and tables, from which follows that small additions of W (1.61%) and V (0.45%) increase SH 5.5 times and twice, respectively, an admixture of 0.48% Ti practically does not affect SH, and the increase in the Cr content to 13% increases SH only insignificantly. In alloyed steels of the austenite class SH is by 4 times higher than in carbon steel of		PERIODICAL:	Tr. Leningr. tekhnol. in-ta im. Lensoveta, 1958, Nr 46, pp 125 - 132
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ARCHAKOV, Yu. I. Cand Teoh Sci -- (diss) "Study of the strength of steels used in machines and apparatus of the petroleum and ohemical industries under high temperatures and hydrogen pressures." Len, 1959. 15 pp (Min of Higher Education USSR. Len Order of Labor Red Banner Technological Inst im Lensovet), 150 copies (KL, 41-59, 104)

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"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000101920014-3

18.7100 77508 SOV/80-33-1-17/49 AUTHORS: Archakov, Yu. I., Grebeshkova, I. D., Teodorovich, V. P. TITLE: The Effect of Heat Treatment on Decarburization and Cracking of Steels While Under Hydrogen at 500-600° C and 800 kg/cm² Pressure Zhurnal prikladnoy khimii, 1960, Vol 33, Nr 1, pp 89-94 PERIODICAL: (USSR) ABSTRACT: This study was started under the supervision of Professor V. V. Ipat'yev. Steel of type 40X (0.41% C; 0.85% Cr) and melt 7475 (0.18% C; 3.05% Cr) were hardened and subsequently tempered at various temperatures which allowed the obtaining of various Cr-content in the carbide phase and solid solution. The tests were made in an autoclave under hydrogen pressure of 800 atm. The carbon content, structure, and composition of the carbide phase before and after the tests were determined. The composition of the carbide phase was determined by means of the electrolytic dissolution (N. M. Popova, Carbide Analysis of Steel--Karbidnyy analiz stali --, Card 1/3

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The Effect of Heat Treatment on Decarburization and Cracking of Steels While Under Hydrogen at $500-600^{\circ}$ C and 800 kg/cm^2 Pressure

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Oborongiz, 1957) and by means of chemical, and in some instances, X-ray analysis. It was established that the amount of chromium in the carbide phase increased with the tempering temperature, and the rate of decarburization decreased. Practically full decarburization was achieved with tempering temperature equal to 550-700° C. Metallographic investigation showed that the size of microscopic cracks appearing along the grain boundaries increased with the tempering temperature and with the chromium content in the carbide phase. It was found that steel with 0.18% C alloyed with up to 3% Cr resisted the corrosive action of hydrogen better than steel 40X. The investigated steels showed low resistance, however, at 600° and 800 atm hydrogen pressure; evidently, the amount of the alloying elements was insufficient to bind all the carbon into carbides. Addition of 0.5-1.0% molybdenum to steel with 0.16% C and 3% Cr did not increase its resistance against the action of hydrogen

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The Effect of Heat Treatment on Decarburization and Cracking of Steels While Under Hydrogen at $500-600^{\circ}$ C and 800 kg/cm^2 Pressure

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under the above-mentioned conditions. It was also found that a new phase, a so-called "carbide precipitate," was formed in all decarburized steels on electrolytic dissolution. In chromium steels, this precipitate contained considerable amounts of chromium, and in chromiummolybdenum steels, considerable amounts of Cr and Mo. This new phase requires further studies. There are 2 tables; 1 figure; and 3 references, 1 U.K., 1 German, 1 Soviet. The U.K. reference is: N. Inglis, W. Andreus, J. Ir. St. Int., 128, 383 (1933).

ASSOCIATION: All-Union Scientific Research Institute for Petrochemical Processes (Vsesoyuznyy nauchno-issledovatel'skiy institut neftokhimicheskikh protsessov)

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SUBMITTED: January 20, 1959

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				S/ AC	/080/60/033/01 003/A001	.1/007/014	
	AUTHOR:	Archakov, Yu	<u> </u>	· · ·	1		
	TITLE:	An Investiga Pipes	tion of the Rate	of Hydrogen Cor	rosion in Car	bon Steel	
	PERIODICAL:	Zhurnal prikl	ladnoy khimii, 19	60, Vol. 33, No	. 11, pp. 254	7-2552	
: • •	TEXT: of his work tion of carb (Ref. 8). C the pipes we was investig first series and a temper creep limit decarbonizat. pressures fry	The work was in (Refs. 1 ar on steel was i arbon steel of re cut and the ated under a m of experiment ature of 525 C of steel form ion depth and om 50 to 300 k	guided by profest and 2). The effect investigated here f the grades 20 and butt sections we microscope with a ts was made at hyd C. The data obtain a straight line (the thickness of tg/cm ² , a temperation the decarbonization	sor <u>V. V. Ipat</u> t of external f . The experime and 30 was used. ere ground. The magnification drogen pressure: ined with stress (Fig. 1). The the pipe wall	<u>ve</u> v and is a actors on the ntal method i After the e e decarbonizat of 87 diameters s from 50 to 8 ses not exceed dependence betwas investigat	continuation decarboniza- s given in xperiments tion depth rs. The 200 kg/cm ² ling the tween the ted at	
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		s/080/60/033/011/0 A003/A001	007/014	
An Investigat	ion of the Rate of Hydrcgen Cor	rosion in Carbon Steel Pipes		
There are 7 f	igures, 3 tables and 9 referen	ces: 6 Soviet, 2 English, 1	American.	
ASSOCIATION:	Vsesoyuznyy nauchno-issledovat protsessov (<u>All-Union Scientif</u> Processes)	el'skiy institut neftekhimic 10 Research Institute of Pet	heskikh rochemical	
SUBMITTED:	January 16, 1960			
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s/080/60/033/011/008/014 A003/A001

AUTHOR: Archakov, Yu. I.

TITLE: The Derivation of a Formula for Calculating the Decarbonization Depth

PERIODICAL: Zhurnal prikladnoy khimii, 1960, Vol. 33, No. 11, pp. 2553-2557

TEXT: A general kinetic equation is derived for the dependence of the decarbonization depth in the <u>corrosion process</u>, which takes place in pipes made of steel 20 at increased temperatures and high pressures of commercial hydrogen, on the temperature, the pressure, the duration of action, the wall thickness and the inner pipe radius. The work is a continuation of Ref. 1. The basic formula used here: $lgK = -A^{1:} \frac{1}{1} \cdot 10^{1} + B$ is identical with Formula (5) in Ref. 1. The value A' in this formula is a constant, the value B depends on the change of the pressure the inner radius of the curvature of the pipes and the thickness of the wall. The dependence between the free term B and the pressure P at temperatures of 350 - 630°C was found. From a graph of the experimental data the following formula was derived: $B = n \cdot lgP - C_1$, where n = 1.12, $C_1 = 2.04$. Using other empirical data, the following final formula is obtained:

$$\Delta 1 = A \cdot e^{-\overline{RT}} \cdot \tilde{\iota}^{\alpha} \cdot p^{n} \cdot 1^{m} \cdot R_{2}^{\mu}$$
(10)

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s/080/60/033/011/008/014 A003/A001

The Derivation of a Formula for Calculating the Decarbonization Depth

where Δl is the depth of the decarbonized layer (in mm), T is the absolute temperature (in ^oK), \tilde{l} is time (in hours), P is the pressure of the gas (in kg/ cm²), 1 is the thickness of the pipe wall (in mm), R₂ is the inner pipe radius (in mm), E is the apparent activation energy of the decarbonization process for steel 20 which is 7,200 cal/g-atom.

$$A = A_{0} \cdot (T_{0}^{-\infty} \cdot P_{0}^{-n} \cdot 1_{0}^{-m+1} \cdot R_{2}^{-m}),$$

where \mathcal{L}_{0} , $P_{0,1}$ lo and R_{20} are scales for measuring the values. For $\tilde{\mathcal{L}} = 1$ hour, $P_{0} = 1$ kg/cm², lo = 1 mm and $R_{20} = 1$ mm, $A = A_{0} = 0.0076$ (for steel 20), $\ll = 0.5$, n=1.12, m=0.71, $\mu=0.58$. The discrepancy between the experimental and calculated values does not exceed 20-25%. The empirical equation (10) is valid for hydrogen pressures of 100-800 kg/cm² and temperatures of 350-600°C for the pipe samples investigated. It reduces the number of tests necessary for selecting steel used in the manufacture of apparatus and pipelines subjected to the action of hydrogen. There are 4 figures, 2 tables and 2 Soviet references.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut neftekhimicheskikh protsessov (All-Union Scientific Research Institute of Petrochemical

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Processes) January 16, 1960

CIA-RDP86-00513R000101920014-3

18.3000 18.1151 // 3110 33835 8/137/62/000/001/171/237 A006/A101

AUTHORS: Archakov, Yu. I., Grebeshkova, I. D., Teodorovich, V. P.

TITLE:

Hydrogen corrosion of ferrochrome alloys at 400 - 800 kg/cm hydrogen pressure and 600° C temperature

PERIODICAL: Referativnyy zhurnal, Metallurgiya, no. 1, 1962, 76, abstract 11541 (V sb. "Stal", Moscow, Metallurgizdat, 1961, 424-435)

TEXT: The authors analyze failure of carbon and alloyed steels under the effect of H_2 of high-parameters. The mechanism of the process is explained. The investigation was made at up to 4,000 hours holding time and 400 - 800 kg/cm² pressure of regularly renewed H_2 . The authors investigated 15 alloys containing C 0.05 - 0.42% and Cr 1.25 - 19.8%; the alloys had been previously normalized (at 900 - 1,000°C) and tempered (at 600 - 720°C) with different times of holding. Me₇C₃ type carbide which was previously considered to be stable, broke down under the described conditions. The authors studied the relationship between the stability of the alloys and their location on the Fe-Cr-C phase diagram. Stability is determined by not only a certain C and Cr ratio (about 1 : 30) but also by their absolute content. Under the experimental conditions corrosion

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"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000101920014-3 33855 Hydrogen corrosion of ferrochrome ... 3/137/62/000/001/171/237 A006/A101 was not observed in an alloy with 0.4% C and about % Cr, having a Me₂₃C₆ carbide constituent. There are 13 references. Ye. Bukhman [Abstracter's note: Complete translation]

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۰. s/184/61/000/002/004/008 18.8300 1138, 1454 A110/A033 Archakov, Yu. I., Candidate of Technical Sciences; Teodorovich, V. P. AUTHORS: Candidate of Chemical Sciences TITLE: On the problem of the beginning of hydrogen corrosion of steel PERIODICAL: Khimicheskoye Mashinostroyeniye, no. 2, 1961, 35 - 38 TEXT: The purpose of this paper was to investigate the conditions marking the beginning of hydrogen corrosion of carbon and 30XMA(30KhMA) steels at temperatures of 200 - 300°C. The issue is of great importance as, if it is known when decarbonization is due to set in, it is possible to determine the service life of equipment and pipes operating under various conditions. The dependence of the incubation period on the hydrogen pressure and temperature of 35 steel is expres- $\overline{v}_0 = \frac{e^{\frac{2}{T}}}{3/2}$ (1)Card 1/9

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no signs of decarbonization after 8,800 hours, which confirms the accuracy of calculations of the length of the incubation period according to Formula (1), i.e. 10,000 hours for fine-grained steel. According to the same formula, the incubation period of carbon steel is 50,000 hours at 220°C and 800,000 hours at 170°C. P. S. Perminov (Ref. 4: "Corrosion and its Prevention", vol. 2, no. 1, 1936) states that the tendency to fractures grows proportionally to the increasing carbon content of steel, owing to differing specific volumes of cementite and ferrite. Ref. 4 also contains data on the begin of hydrogen corrosion of carbon steel at various temperatures and pressures (Figure 2). From reports contained in References 3 and 4 follows, that at pressures up to 500 atm the lower temperature limit of hydrogen effect on carbon steel lies around 200°C. These results have been confirmed by tests carried out by the Gosudarstvennyy Institut Vysokikh Davleniy, GIVD (State Institute of High Pressures) in 1940. Grade "25" steel pipes were fully decarbonized after 1.5 years of operation in a nitrogen-hydrogen---ammonia medium at 300 atm and 450 - 525°C. Similar corrosion appeared in grade "20' steel subjects to the same medium at 200 atm. The unreliability of short--term tests is demonstrated by means of results obtained by the GIVD with grade "20" steel, and by a non-specified ammonia plant with carbon steel. According to the

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author's calculations, the incubation period of grade "20" steel should be 10,000 hours at 150 - 250°C and 275 - 300 atm. Data obtained by F. Nauman (Ref. 5, "Steel and Iron", vol. 57, no. 32, 1937) is considered unreliable because of the briefness of pertinent tests (100 hours). Production experiments confirm the accuracy of results given in Ref. 4, and established temperature limits and hydrogen pressure in respect of various steels listed by G. A. Nelson (Ref. 14: "Trans the ASME", February, 1951). According to O. van Rossum (Ref. 11:"Chem. Ing. Eng. ug Techn." vol. 25, no 8/9, 1953). Inner stresses and preliminary cold working decrease the hydrogen resistance of steel, consequently carbon steel should not be used under pressures exceeding 300 atm and temperatures above 220°C. This statement is borne out by the findings given in References 4 and 14 and appears in respect of smaller machine parts made of fine-grained steel. The longer the soaking of samples in hydrogen, the lower the temperatures and pressures at which fractures take place. The use of carbon steel in the shape of coarse-grained large billets is particularly dangerous. Temperatures up to 150°C are considered safe for carbon steel equipment used in hydrogenation plants. For short-term operations at higher temperatures, the strength of machinery and pipe walls should be selected taking into account hydrogen corrosion. The effective power of the

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activated decarbonization process in respect of chromium and molybdenum alloyed steels increases from 7,200 cal/g-atom ("20"steel) to 15,000 cal/g-atom (30KhMA) steel). Figure 4 shows the dependency of decarbonization depth Δ 1 on pressure p for various pipe walls: Yu. I. Archakov, (Ref. 15: "Journal of Applied Chemistry", vol. 33, no. 11, 1960) developed an equation linking the depth of decarbonization with pressure, temperature, wall strength and inside radius of a "20" steel pipe

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$$\Delta l = Ae \quad \mathcal{L}^{a} p^{n} \delta^{m} R_{2}^{\mu}$$
(3)

 Δl - depth of decarbonized layer; T - absolute temperature in ^OK; T- duration in hours; p - hydrogen pressure in kg/cm²; δ - strength of walls and machinery casings in mm; R₂ - inside radius of pipe in mm; e - base of natural logarithms; E = 7,200 cal/g-atom effective power of activated decarbonization process; R≈1.987 cal/g-atm ^OC, gas constant; A = 0.0076; a = 0.5; n = 1.12; m = 0.71; μ = 0.58. Equation (3) is applicable only after completion of the incubation period, 1.e., during the decarbonization process. The empirical equation applies

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to 4, 7, 10 and 12 mm pipes with inside diameters of 4, 8, 10, 16 and 18 mm. According to V.V. Ipatyev et al (Ref. 1: Zhurnal prikladnoy khimii, vol. 31, no. 12, 1958) the kinetic aspect of decarbonization of 30XMA (30KhMA) steel is expressed by

 $lgK = -\frac{15,000}{2,3RT} + 0.0034 p + 1.13$ (4)

 $K = \frac{\Delta 1}{1\sqrt{\tau}}$, showing the ratio of the decarbonization rate of 30 XNA(30KhMA)

steel pipes to the pressure and temperature. Equation (4) makes it possible to determine the rate and depth of decarbonization of $24 \times 7 \text{ mm } 30 \times \text{MA}$ (30KhMA) pipes and the approximative computation of their service life, depending on operating conditions. All equations are applicable to hydrogen affected steel pipes. Under production conditions the steam-gas mixture contains 50 - 80 % (volumetric) hydrogen; the rest are hydrocarbons which delay hydrogen corrosion. The resistance to hydrogen corrosion of carbon steels exposed to high temperatures and pressures can be improved by addition of strong carbide-forming elements, of which the increase of chromium content proved most effective. Tests proved that a slow de-

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carbonization process takes place already at $200 - 250^{\circ}$ C, consequently the safe operation range for carbon steels in hydrogen media at 300 kg/cm^2 pressure is limited to temperatures not exceeding 150° C. $30 \times M \wedge (30 \times M \wedge)$ chromium-molybdenum steel can be used in hydrogenation plants and subjected to 300 kg/cm^2 hydrogen pressures at 300° C. There are 6 figures and 16 references: 10 Soviet-bloc and 6 non-Soviet-bloc. The reference to the English-language publication reads as follows: G. A. Nelson "Trans the ASME, February 1951; 0il and Gas Journal, vol. 57, no. 22, 1959.

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26861 s/080/61/034/004/005/012 1413, 1418, 2808, 4016 11 A057/A129 b., Teodorovich, V. P. 18,8100 Archakov, Yu. I., Grebeshkova, I. Determination of the rate of hydrogen diffusion through 1X18H9T AUTHORS: (1Kh18N9T) steel at high temperatures and pressures Zhurnal prikladnoy. khimii, v. 34, no. 4, 1961, 821 - 825 TITLE: A method was developed for determining the hydrogen permeability of 1Kh18N9T (AISI 321) steel under high-temperature (up to 1,000°C) and high-pres-PERIODICAL: sure (up to 300 kg/cm²) conditions. It was found that the pressure dependence of the hydrogen diffusion rate is parabolic and the temperature dependence is exponenthe hydrogen diffusion rate is parabolic and the temperature dependence is ex tial, having a temperature coefficient of 21,650 cal·8⁻¹ atom⁻¹. The method allows diffusion rates through metals at high temperatures and pressures to be determined with sufficient accuracy. It is essential that welding joints can be avoided and the amount of diffused gas can be measured. The present investigation was necessary since literature data refer only to tests at high temperature and low pressure or vice versa. No data are available for high-temperature and high-pressure conditions. In the method presented tubular samples (Figure 1) are used without welding joints and temperature and pressure were kept constant during Card 1/5

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the test. The testing assembly contained the diffusion unit, placed into an electric oven, a high-pressure system and the device for measuring the amount of diffused hydrogen. In all experiments technical grade hydrogen gas was used. The sample (Figure 1) was installed into the diffusion unit by welding the hydrogen gas inlet-tube at point E to the sample and inserting the sample into a vertical cylindrical tube which was placed in the vertical electrical oven. Hydrogen gas thus entered into the sample and diffusion occurred through the walls between C-D (see Figure 1). From the cylindrical tube the diffused hydrogen passed through an outlet-tube to the gas-measuring device. The high-pressure assembly used in the present experiments was described in a previous paper (Ref. 10: Tr. LTI. im. Lensoveta, Goskhimizdat, XXXVIII, 204-215, 1957; Ref. 11: ZhPKh, 32, 12, 2667, 1959). The temperature was regulated during the experiments by means of a $3A\Pi-17$ (EDP-17) electronic potentiometer and registered by a 311-09 (EPP-09) automatic electronic potentiometer. The amount of diffused hydrogen was measured in a device containing for smaller gas amounts a 2 cm³ microburet calibrated in 0.01 cm³, while for greater gas quantities a 100 cm³ buret graduated in 0.2 cm³ was used. Measurementswere carried out in certain time interval and the amount of diffused gas was estimated per time unit and for normal conditions. The observation made by P.L. Chang et. al. (Ref. 3: J. Iron and Steel Inst., 3, 170, 205, 1952) that

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stationary gas diffusion is quicker attained by cooling the sample was proved in preliminary experiments in the present work. Thus diffusion characteristics were estimated starting at high temperatures. The present tests were carried out on 6 tubular samples made of 1Kh18N9T steel. The following temperatures and pressures were investigated: 500°C and 300 kg/cm², 600 - 800°C up to 200 kg/cm², at 900°C up to 100 kg/cm², 950°C up to 75 kg/cm², and at 1,000°C up to 50 kg/cm². The maximum difference in the results between the six investigated samples was 5 %. The results shown in Figure 3 and in a table indicate a parabolic diffusion rate. In Figure 4 the temperature dependence of the diffusion rate is demonstrated. Since all isobars in Figure 4 are parallel to each other, the dependence can be expressed by $V = k \cdot e^{-21650/RT}$. Thus the apparent activation energy, estimated from the inclination angle, is 21,650 cal·g⁻¹.atom⁻¹. H. Bennek and G. Klotzbach (Ref. 1: Stahl und Eisen, 61, 25, 597, 1941) found for the diffusion of hydrogen at high temperature and 1 atm pressure through 18 % Cr, 8.9 % N1 steel an activation energy of 18,850 cal·g⁻¹.atom⁻¹. There are 4 figures, 1 table and 11 references: 4 Soviet-bloc and 7 non-Soviet-bloc.

ASSOCIATION:

Vsesoyuznyy nauchno-issledovatel'skiy institut neftekhimicheskikh protsessov (All-Union Scientific Research Institute of Petrochemical Processes)

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S/737/61/000/000/003/010 AUTHORS: Archakov, Yu.I., Grebeshkova, I.D., Teodorovich, V.P. Hydrogen corrosion of ferrochromium alloys at hydrogen pressure of TITLE: $400-800 \text{ kg/cm}^2$ and a temperature of 600° C. Stal', sbornik statey. Ed. by A. M. Yampol'skiy. Moscow. 1961, SOURCE: 424-435 Experimental findings show that the effect of H on ferrochromium (FC) TEXT: alloys on the stability of the carbide component therein is not controlled by the ratio of the C and Cr contents alone. FC alloys with a C content <0.4% are H-corrosion resistent at T=600°C and p_{H} =800 kg/cm², provided the Cr content is > 9%. Decarbonization of the alloy occurs at the said temperature and pressure, even when only trigonal Cr carbide (Cr, Fe)7C3 is present. A brief state-of-theart report on the H-corrosion problem of C steel and means for its minimization are given. The objective of this study is an investigation of the H corrosion of various FC alloys under more severe conditions than those employed in antecedent investigations. The tests defined in the title were conducted for a period of 1,000-4,000 hours. The specimens were fully exposed (on all sides, not only along an interior cavity) to the H pressure. The method used for the investigation of the Card 1/3

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Hydrogen corrosion of ferrochromium alloys...

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effect of the alloy composition on the H-corrosion process was that of Yu.I.Archakov (Leningradskiy tekstil'nyy institut im. Lensoveta. Trudy, no.38, Goskhimizdat, 1957), except for a thermostatically controlled electric furnace in which a 36-mm OD, 18-mm ID, tube of 1X18H9T (1Kh18N9T) steel was set to serve as a H-pressurized reaction tank. Fresh H was circulated from one end of the reaction tube once a day to replace any H that might have diffused through the tube and to eliminate any traces of methane that might have formed. The degree of H corrosion was judged before and after soaking by microscopic examination, mechanical testing, and chemical analysis for C. The pre-test heat treatment of various FCalloy specimens is tabulated. The carbide phase was analyzed chemically and by X-ray before the test. The precipitate was separated electrolytically and chemically, and the Cr and Fe contents were determined in the precipitate and in the separate carbide components. A tabulation of the data obtained by N.M. Popova's method (Karbidnyy analiz stali - Carbide analysis of steel. Oborongiz, 1957) and by N.A. Saverina's method (TsNIITMash, book 36, Mashgiz, 1950) is adduced for 9 alloys, and the two methods are briefly defined. X-ray analysis revealed the presence of the Me₇C₃ with hexagonal lattice, the $Me_{23}C_6$ phase, and some weak lines of an unknown phase. The changes in mechanical properties after H soaking are tabulated for the two series of tests performed at increasing H pressures and longer soaking times. FC alloys with an initial C:Cr ratio of 1:25 and 1:29 were

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Hydrogen corrosion of ferrochromium alloys... S/737/61/000/000/003/010 totally decarbonized by the H. Alloys with a C:Cr ratio as high as 1:35 were still partly decarbonized. Yet, an alloy with a C:Cr ratio of only 1:24 exhibited only an insignificant surficial decarbonization under the most severe testing conditions. It was concluded that the C:Cr-ratio criterion alone (cf. Inglis, N., Andrews, W., Journal of the Iron & Steel Institute, v. 128, 1933, 383-408, and 2 Soviet references) is inadequate, but that a definite Gr threshold value of 9% is a dependable criterion for the H-corrosion stability of FC alloys containing up to 0.4% C and a $(Cr, Fe)_7C_3$ carbide component. There are 3 figures, 5 tables, and 13 references (9 Russianlanguage Soviet, 2 English-language, 2 German). ASSOCIATION: Vsesoyuznyy n.-i. institut neftekhimicheskikh protsessov (All-Union scientific research institute for petrochemical processes). Card 3/3

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		39071 5/080/62/035/006/009/013 D204/D307
	11 1220	D2047 D701
	11.3110	
	AUTHOR:	Archakov, Yu. I.
		The effect of alloying elements on the stability of
	TITLE:	steels to hydrogen
		500010 to 10
	PERIODICAL:	Zhurnal prikladnoy khimii, v. 35, no. 6, 1962,
		1280-1285
	·	0.15 - 0.23 C, $0.14 - 0.23$ C, $0.14 - 0.23$ C, $0.14 - 0.22$
	TEXT: The	stability of steels containing 0.15 - 0.23 C, 0.14 - .38 - 0.50 Mn, 0.18 - 0.25 Ni, 0 - 0.11 Cu, 0.022 - .39 - 0.50 Mn, 0.18 - 0.25 Ni, 0 - 0.11 Cu, 0.022 -
	0.38 S1, 0.000	0.001 - 0.022% P, alloyed with Cr, ", v and - 800 kg/cm,
	otudied at	stability of Boold - 0.25 Ni, 0 - 0.11 Cu, 0.022 .38 - 0.50 Mn, 0.18 - 0.25 Ni, 0 - 0.11 Cu, 0.022 a 0.001 - 0.022% P, alloyed with Cr, W, V and Ti was 2 a 0.001 - 0.022% P, alloyed with Cr, W, V and Ti was 2 600° C under pressures of H ₂ equal to 400 - 800 kg/cm, (100) Kg/cm) (10
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	over 1000	n measurement of mechanical properties and carbides
	examination	n, measurement of mechanical properties and carbides after the experiments. Phase analyses for carbides that after the experiments at 800 kg/cm ² showed that
	steels con	alter the choice hour tests at 800 kg/cm ² should not carried out. 4000-hour tests at 800 kg/cm ² should not test the carried out. 4000-hour tests to H_2 if the Cr contents taining ~0.2% C were unstable to H_2 if the Cr contents 4.5 or 5.4%, but were stable if Cr >8.4%. The important
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~	L 19303-63 EWP(q)/EWT(m)/BDS ASD/AFFTC JD ACCESSION NR: AR3006906 S/0137/63/000/007/10153/1053	5
	SOURCE: RZh. Metalurgiya, Abs. 71363	
	TITLE: Hydrogen corrosion of alloyed steel CITED SOURCE: Vestn. tekhn. i ekon. inform. N-i. in-t tekhnekon. issled Gos.	
1	kom-ta Sov. Min. SSSR po khimii, no. 12, 1962, 26-31 TOPIC TAGS: corrosion, hydrogen corrosion, alloyed steel, ferro-chromium steel, hydrogen stability, carbide, interstitial carbide	
	TRANSLATION: The influence of Cr, W. V, and Ti on the stability to hydrogen corrosion of ferro-chromium steel was studied, with Cr (1.45-18.8%) and C (0.11-0.72%) contents in various ratios. Heat-treated samples were tested under	
-	the influence of H_2 for 1000-4000 hours at pressures of 400-800 kg/cm ² and 6003. The deciding factors responsible for the hydrogen stability of the steel are the nature of the carbide phase of the steel, the presence of other elements dissolved in the carbide component, and the absence of cementite in the steel. The greatest	
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	hydrogen s type (TiC Bibliograp	and VC), a and VC), a phy of 13 t	s posse s well itles.	essed as wi N. I	by ste ith chr ukashi	ols w comium na.	th carb	arbides ides of	of the " the type	inter of C	stitia r2306.	<u>1"</u> >7	
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ACCESSION NR: AT4013970 S/2659/63/010/000/0305/0313		
AUTHOR: Archakov, Yu, I.; Grebeshkova, I. D.		
TITLE: Influence of alloying elements on the long-term hydrogen stability of steel		
SOURCE: AN SSSR. Institut metallurgii. Issledovaniya po zharoprochny*m splavam, v. 10, 1963, 305-313		
TOPIC TAGS: hydrogen stability, alloying element, hydrogen, chromium, tungsten, vanadium, niobium, titanium, steel, steel stability		
ABSTRACT: Steel strength drops significantly under the physical and chemical action of hydrogen penetration, leading to the dissociation of the carbide phase and to decarbonization. The present investigation considers the influence of separate alloying elements (cr, W, V, Nb, Ti), and also of the combined action of Cr and Mo, Cr and W, Cr and Nb, Cr and V Cr and Ti cr the		
steel at a temperature of 600C, hydrogen pressure of 800 atm and test duration of 1,000-4,000 hours. It was shown that under conditions of high temperature inter- action with hydrogen, steel with 0.15%C is not decarbonized when it contains 8.4%		
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AUCESSION NR: A14	013970	f	
carbonized (conten Under the same con with up to 0.48%V, the carbon content	the (Cr, Fe) 23C6 type. Second, under a hydrature of 600C for 1,000 hours the steels are t: 0.2%C, with addition of 0.4%W, 0.75%W, 1.1 ditions steel with approximately 0.2%C and 3% 1.5% Mo and 1.5%W. Third, under the influer is not lowered in steels containing approxim Ti and 1.97% or in steels containing approxim	completely de- 57%W, or 0.48%V). % Cr is decarbonized nce of hydrogen,	
2% Nb, and 2%T; Fo stability of steel when the entire cap	Ti and 1.97% or in steels containing approxim purth, the main factor ensuring the durabilit is the type of carbide phase. Steel is not bon content is chemically united in TiC and r, Fe) 23°C6. Orig. art. has: 1 chemical for	, 3% Cr, 1.5%V, ty of the hydrogen decarbonized	
2% Nb, and 2%T; Fo stability of steel when the entire can chromium carbide ((and 3 tables.	burth, the main factor ensuring the durabilit is the type of carbide phase. Steel is not	, 3% Cr, 1.5%V, ty of the hydrogen decarbonized VC carbides or rmula, 4 figures,	
2% Nb, and 2%T; Fo stability of steel when the entire can chromium carbide ((and 3 tables.	burth, the main factor ensuring the durabilit is the type of carbide phase. Steel is not bon content is chemically united in TiC and r, Fe) 23C6. Orig. art. has: 1 chemical for	, 3% Cr, 1.5%V, ty of the hydrogen decarbonized VC carbides or rmula, 4 figures,	
2% Nb, and 2%T; Fo stability of steel when the entire can chromium carbide ((and 3 tables. ASSOCIATION: Insti	burth, the main factor ensuring the durabilit is the type of carbide phase. Steel is not bon content is chemically united in TiC and r, Fe) 23C6. Orig. art. has: 1 chemical for tut metallurgii AN SSSR (Institute of Metall	, 3% Cr, 1.5%V, ty of the hydrogen decarbonized VC carbides or rmula, 4 figures, lurgy AN SSSR)	

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ACCESSION NR: AR4041619 S/0137/64/000/005/1071/1071 SOURCE: Ref. zh. Metallurgiya, Abs. 51410 AUTHOR: Archakov, Yu. I. TITLE: Hydrogen corrosion of steel CITED SOURCE: Sb. Vliyaniye vodoroda na sluzhebn. svoystva stali. Irkutsk, 1963, 6-21 TOPIC TAGS: steel, steel corrosion, corrosion, hydrogen corrosion TRANSLATION: Questions of beginning of hydrogen corrosion of steel, influence of chickness of wall, stresses and alloying elements on hydrogen stability of steel are con-sidered. Results are presented which allow us tentatively to calculate magnitude of incubation period for steel 20 and 30khMA at 200 - 600° and with pressures of id decarbonizing of steel. With stresses, causing plastic flow of metal, the Card 1/2
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TRANSLATION: Questions of beginning of hydrogen corrosion of steel, influence of thickness of wall, stresses and alloying elements on hydrogen stability of steel are con-sidered. Results are presented which allow us tentatively to calculate magnitude of incubation period for steel 20 and 30KhMA at $200 - 600^{\circ}$ and with pressures of $12 \text{ of } 100 - 500 \text{ kg/cm}^2$. Increase of thickness of wall of pipes increases speed of decarbonizing of steel. With stresses, causing plastic flow of metal, the
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$\frac{L 15176-55}{ASC(f)-2/ASD(m)-3/AFMDC/AFTC(f)} III = 11ACCESSION NR: AT4045964$		F -4/Pa-4 3 3D/ 00/0019/0043	
AUTHOR:Archakov, Yu. 1.; Grebeshko	wa, l. D.		
TITLE: Hydrogen resistant steels of	low scarcity	6	
SOURCE: Novytye materialy v mashin manufacturing). Moscow, izd-vo Mash	ostroyenii (New materiais in Inostroyeniye, 1964, 19-43	machinery	
TOPIC TAGS: steel, alloy steel, hyd <u>decarbonization</u> , stainless steel, ca	rogen resistance, embrittlen rbon diffusion	nent, <u>corrosion,</u>	
ABSTRACI: Processes taking place un at elevated temperatures, so that bo have to be considered in evaluating i	th heat resistance and hydro	oger resistance	
environment. The two main problems a hydrogen embrittlement, the former be the influence of alloying elements on	are hydrogen corrosion (deca eing the most harmful. in t n the hydrogen resistance of	he present paper, steel was investi-	
gated on specimens subjected to hydro various test durations. Mechanical hydrogen showed the corresponding ch of tests were carried out: "(1) 'var steels (hydrogen pressure 400800 k	ogen at various pressures at tests performed before and a anges in physical properties lation of the chromium conte	600C, and for ofter heating in Four series	

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its decald standards and second services 111 e tre service A Hart & Barris & Barris L 15176-65 ACCESSION NR: AT4045984 of the tungsten, columbium, vanadium, and titanium content (800 kg/cm²; 1000 hours); (3) variation of the molybdenum, tungsten, columbium, vanadium, and titanium content in steel with 0.2% C, 3% Cr (400--750 kg/cm²; 1000--2000 hours); (4) variation of the sillcon, malybdenum, tungsten, vanadium, and titanium content in steel with 0.2% C, '6% Cr' (800 kg/cm²; '4000 hours). From the obtained test results it is concluded that the hydrogen resistance of steel is determined mainly ' by the type of <u>carbide phase</u> the character of the interatomic ties between carbon and the alloying element, and secondarily by the velocity of carbon diffusion in chromium-containing ferrite. Tests showed that additions of Cr, Mo, W, V, Cb, and TI increase the stability of the carbide phase; at the same time, the velocity of carbide phase coagulation decreases. The degree of dispersion of the carbide component is apparently not important. To stabilize steel with 0.2% C, 3-6% Cr against hydrogen corrosion (at 600 C), the addition of stronger carbide-forming elements such as V, Cb, and TI, or the addition of more than 9% Cr, is required. However, at the present time, the authors are limiting their recommendations for the conditions 600C and 800 kg/cm². Under such conditions, not less than 8.4% Cr should be added to steels with 0.2% C (or less) and not less than 10% Cr should be added to steels with 0.2-0.4% carbon. There are not yet sufficient data available to establish a quantitative correlation between the temperature and pressure, on the one hand, and the maximum hydrogen resistance on the other. Orig. art. has: 12 figures and 12 tables. Card 1

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"APPROVED FOR RELEASE: 06/05/2000 L 15176-65 ACCESSION NR: AT4045964 \odot 4 Ì 27 ASSOCIATION: none 00 SUB CODE: MM SUBMITTED: 16May64 ENCL: OTHER: 004 NO REF SOV: 009 Cord 3/3

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$\frac{L 7054-66}{ACC NR} = \frac{EWT(m)/EWP(w)/EWA(d)/T/EWP(t)/EWP(z)/EWP(b)}{IJP(c) MJW/JD/RM/WH}$		
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(Engineer) (Candidate of technical sciences); Grebeshkova, I. D.		
ORG: none		
TITLE: Investigation of the tendency towards carburization of steels in methane at		
SOURCE: Khimicheskoye i neftyanoye mashinostrovenime		
TOPIC TAGS: carburization, alloy steel math		
TOPIC TAGS: carburization, alloy steel, methane, steel, iron/ 12KhlMF steel, NMI	v	
ABSTRACT: The tendency of a number of chromium and <u>carbon steels</u> towards carburiza-		
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ted contained and , metallurgizdat, 7067) m		
The elemental and 10 wounded from containing a and 10, 20, and 30.10		
presented graphically (see Fig. 1). It was found that the experimental results are	-	
decreases the ductility but increases the strength of the steel. Carburization of 10-		
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UDC: 65.046.56:547.211		
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ARCHAKOV, Yu.I.

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Protection of metals from the effect of hydrogen. Zashch. met. 1 no.5:587-590 S-0 '65. (MIRA 18:9)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut neftekhimicheskikh protsessov.

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IJP(c) JD/WB/EM	EWA(d)/EPR/T/EWP(t)/EWP(b)/EWA(c) Pr-4 UR/0032/65/031/007/0873/0875
ACCESSION NR: AP5017096	620.194 35 32
AUTHOR: Archakov, Yu. I.; Golubov	skiy, B. S.; Il'ichev, V. I.
TITLE: Nethod of investigating th	e effect of stress on the gas corresion of steal
SOURCE: Zavodskaya laboratoriya,	v. 31, no. 7, 1965, 873-875
TOPIC TAGS: stressed state, volume	astric stress, plane stress, stress test, corro- lecarburized steel, steel corrosion, gas corro-
sion, corrogive medium	and the literature any information on the ef-
ABSTRACT: The authors could not if fect of stress on the rate of the	find in the literature any information on the ef- ghemical, gas corrosion of metals (in particular, nd of steel by hydrogen) in the presence of high fill this gap the authors tried to develop tech-
ABSTRACT: The authors could not a fect of stress on the rate of the on the rate of the <u>decarburization</u> temperatures and pressures). To niques of investigating the effect at temperatures of up to 500°C an negating the performed in a spe	n of steel by hydrogen in the presence of high rill this gap the authors tried to develop tech- t of stresses on the decarburization rate of steel d hydrogen pressure of up to 500 kg/cm ² . The ex- cially designed setup (Fig. 1). Clamped between
ABSTRACT: The authors could not a fect of stress on the rate of the on the rate of the <u>decarburization</u> temperatures and pressures). To niques of investigating the effect at temperatures of up to 500°C an negating the performed in a spe	n of steel by hydrogen in the presence of high fill this gap the authors tried to develop tech- t of stresses on the decarburization rate of steel t of stresses on the decarburization rate of steel

CIA-RDP86-00513R000101920014-3

L 57735-65 AP5017096 ACCESSION MR: the region of contact between the gas and specimen is 30 mm. Washers 5, inserted in the nipple grooves, serve to improve the clamping tightness. One nipple is linked to a high-pressure hydrogen system and the other, to a high-pressure nitrogen system. Pressure in both parts of the apparatus is measured with the aid of standard manometer 6. Chromel-alumel thermocouples 7 and portable potentiometers 8 serve to measure the test temperature and are regulated by means of electronic potentiometer 9 connected to the circuit of heating furnace 10. The setup is provided with cocks 11 for regulating gas pressure. Commercial hydrogen at a fixed pressure of 200 kg/cm² is used as the corresive-aggressive medium, while nitrogen at the same and higher pressures is used to produce stresses in the investigated plate. The whole setup is placed in an electric furnace and heated to 500°C. Each experiment lasted 24 hr. After each experiment microsections of the specimens were examined and found to display a distinct decarburized zone In the region of exposure to hydrogen. The material on the specimen surface in contact with the aggressive medium (H2) experiences two forms of stressed state: volumetric (due to the pressure of hydrogen and the counterbalancing pressure of nitrogen) and plane (due to bending of the plate by the excess pressure of nitrogen); the bending causes tangential and radial stresses which can be computed from the corresponding formulas. The principal stresses in the material of the specimens equal the algebraic sum of the rated conditional stresses produced by both forms of Cord 2/5

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	periment the be determined that as the s layer increas or as a func- stresses acc	stresses chang d from them on conditional str ses also. Fig. tion of stresse	e owing to the the basis of d esses increase 2 presents a 5 σ_{TT} . As can coceed of corro	decarburiz ifferent ct the measur curve of th be seen fr sion, thus	because in the c sation of steel) s rength theories. The depth of the d the depth b of deca om this curve, in accelerating the	tresses ca Tests sha ecarburiza rburized crease in	an owed ed lay the	
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1610-66 EMT(m)/EMP(w)/EMA(d)/T/EMP(t) ACCESSION NR: AP5021666	UR/0080/65/038/008/1754/1760 621.0-419	
AUTHOR: Archakov, Yu. I.	45 B	
TITLE: Hydrogen resistance of bimetal	18 14, 44, 55 B	
SOURCE: Zhurnal prikladnoy khimii, v	. 38, no. 8, 1965, 1754-1760	
TOPIC TAGS: bimetal, steel, s <u>tainles</u> bility measurement, gas diffusion/ <u>EI4</u>	s steel, hydrogen embrittlement, permea-	
widely used in the chemical and petrole tion of hydrogen into these steels start		
is a theoretical study of the hydrogen f tical deductions are compared to the e	bermeability of bimetals in which mathema- ermeability of bimetals in which mathema- xperimental data of other investigators. It and pressures, a layer of stainless steel e welding boundaries of the individual com-	
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ponents of the bimetal and con- metal. An equation is derived boundary of the two metal laye hydrogen permeability constan and their thicknesses. Orig.	l which permits calcula rs as a function of cha its of the individual con	ation of the pressu unge in external pr mponents of the big	re at the essure, the metals.
ASSOCIATION: None		•	
SUBMITTED: 08Aug64	ENCL: 00	SUB CODE: M	M
NR REF SOV: 009	OTHER: 007		

ARCHAKOV, Yu.I., kand.tekhn.nauk; GREBESHKOVA, I.D., inzh.

Studying the tendency of steel toward carburizing in methane et increased temperatures and pressures. Khim. i neft. machinostr. (MIPA 18:10) no.9:26-28 S '65.

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where v is the rate of penetration of hydrogen through a unit of surface of a two layer metal, with steady state flow and a concentration gradient of $\partial c/\partial x$. Based on literature data, the article gives extensive tables showing the rate of hydrogen penetration into a large number of steels and alloys. It is concluded that a cladding following manner: the stainless layer of the two layer steel, having a smaller diffusion capacity, reduces the hydrogen pressure (its concentration) at the boundary between the metals, and promotes the removal of the production of reaction, that is, and 6 tables.	r
SUB CODE: 11, 20/ SUEM DATE: none/ ORIG REF: 014/ OTH REF: 007	
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FLORINSKIY, F.V., prof.; VOLOSHINA, L.P., dots.; LYAKHOVITSKIY, S.I., kand. tekhn.nauk; SHIROCHENKO, Ye.V., dots. [deceased]; ARCHAKOVA, L.A., inzh.; GVAY, T.B., inzh.; MURZINA, Z.I., inzh.

> Results of research on screen vibrating in the horizontal horizontal plane. Izv.vys.ucheb.zav.; gor.zhur. no.2:167-170 '60. (MIRA 14:5)

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ARCHAKOVA, L.A., inzh.; KKYUKOV, B.I., kand. tekhn. nauk; FYEHAL'SKIY, Yu.A., kand. tekhn. nauk
Mynamics of vibratory machines with planetary-type binarmonic vibrators. Izv.vys.ucheb.zav.;gor.zhur. 7 no.9:140-143 '64. (MIRA 18:1)
1. Dnepropetrovskiy ordena Trudovogo Krasnogo Znameni gornyy institut ireni Artema. Rekomendovana kafedroy teoreticheskoy i stroitel'noy mekhaniki.

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CIA-RDP86-00513R000101920014-3

ARCHAKOYA, L.I. Further analysis of the myenteric reaction. Dokl. AN BSSR 7 no.3; 208-211 Mr '63. (MIRA 16:6) 1. Institut fiziologii AN BSSR. Predstavleno akademikom AN BSSR 1.A.Bulyginym. (Intestines)

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CIA-RDP86-00513R000101920014-3

GORYUSHINA, V.G.; ARCHAKOVA, T.A. Rapid volumetric determination of beryllium in alloys. Zav.lab. 22 (MIRA 9:8) no.5:532-535 '56. (Beryllium-Analysis) (Titration)

CIA-RDP86-00513R000101920014-3

SOV/32-25-7-5/50 5(2) Goryushina, V. G., Archakova, T. A. AUTHORS: يستوكان والمراجب New Volumetric Method for the Determination of Uranium (Novyy TITLE: ob"yemnyy metod opredeleniya urana) PERIODICAL: Zavodskaya laboratoriya, 1959, Vol 25, Nr 7, pp 789 - 790 (USSR) The new method of determining uranium was elaborated in much ABSTRACT: the same way as the determination of beryllium (Refs 10, 11). The method is based upon a precipitation of uranium in the presence of trilon B with sodium arsenate and a subsequent iodometric titration of the · ion bound with uranium. It was experimentally determined AsC that the crystalline precipitate which is precipitated to acetic uranium solution by addition of sodium arsenate (pHN 3) possesses constant composition (UO::AsO = 1:1). The results of the iodc-metric arsenic titration and conversion into uranium (Table 1) confirmed the applicability of this method to the determination of uranium. The method described has a high selectivity since the elements such as the bivalent metals Fe, Al, Bi, and V, Th, Card 1/2No do not disturb the determination, and Ti by addition of Per-

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New Volumetric Method for the Determination of Uranium SOV/32-25-7-5/50 hydrol and Zr by addition of fluoboric acid (Table 2) can be bound. By addition of a larger quantity of precipitant the influence of the phosphates can also be eliminated. A process of analysis is described. There are 2 tables and 1 Soviet reference. ASSOCIATION: Gosudarstvennyy nauchno-isoledovatel'skiy institut redkikh i malykh metallov (State Scientific Research Institute for Rare and Minor Metals)

Card 2/2

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S/032/61/027/007/001/012 B110/B203

AUTHORS: Goryushina, V. G., Romanova, Ye. V., and Archakova, T. A. TITLE: Colorimetric method for determining zirconium in alloys PERIODICAL: Zavodskaya laboratoriya, v. 27, no. 7, 1961, 795-797

TEXT: The methods, much improved recently, for determining zirconium, e. g., with pyrocatechol violet, xylenol blue, and Arsenazo III, are subject to the effect of various elements contained in the alloys (Table 1). Tin can be used as a reducing agent, particularly in strongly acid media in which hydroxylamine and ascorbic acid are poorly efficient. In the Arsenazo III method, the presence of larger amounts of elements usually admixed to Zr is permissible. In strongly acid medium (2 N HCl), the effect of all bivalent, and many other, elements is eliminated. At a Zr content $\gtrsim 0.1\%$, Table 2). The results obtained agree with control tests performed with pyrocatechol violet. The Ti content may be ≤ 10 mg. At a Zr content of 0.2%, the method is applicable to vanadium and ferrous alloys. In the Card 1/6/

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Colorimetric method for determining ...

S/032/61/027/007/001/012 B110/B203

latter, the iron is previously reduced by hydroxylamine. In Al and Mg alloys, there is no lower limit of the Zr content. With introduction of the same Cu amount of the analytic solution into the zero solutions, a Zr content of up to 0.005% can be determined without Cu separation since the optical density of solutions with 5-25/Zr is preserved in the presence of 50, 100, and 200 mg of Cu. Dissolution in H₂SO₄ is required for Zr alloys

on Cu basis with elevated Cr content. Since a content of only $\leq 100 \text{ mg}$ of SO_4^{2-} is permissible for the Arsenazo III method, the Zr must be (1) precipitated with NH₃ (at low Zr content in the presence of 5 mg of Al or Fe as collector), or (2) determined colorimetrically by means of xylenol orange. Authors' tests showed that $\leq 10 \text{ mg}$ of Cu did not disturb the determination of $10-60 \swarrow$ of Zr in 50 ml of 0.5 N H₂SO₄ with an addition of

2 ml of 0.1% dye solution. The determination may be conducted without Cu separation with an accuracy of 0.01%. For Zr determination in refractory alloys be means of xylenol orange, it is recommended to separate Fe, Ni, and other elements on the Hg cathode with subsequent precipitation of the

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Colorimetric method for determining ...

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hydroxides by means of NaOH. Even in the presence of 5 mg Nb (50 ml of 2 N HCl, 1 ml of 0.1% dye solution), 10-12 Zr can be determined by means of Arsenazo III. With lower Zr content and a high excess of the reagent, the latter may react with Nb; the result of the Zr analysis may already be distorted at a niobium content of 100%. 0.1-0.2 g of Al, Ti, etc., alloy is dissolved in HCl or, (Cu alloy), HNO3. In the presence of Ti, a mixture with H_2O_2 is prepared and boiled to discoloring. Then, it is acidified to 2 N HCl. Part of the solution with 5-25f' of Zr is diluted to 10 ml by means of 2 N HCl, and heated to boiling. In the case of Fe content, hydroxylamine is added until the yellow color disappears. 3 ml of 1% Arsenazo III solution is admixed, and filled up with 2 N HCl to 50 ml; then, the optical density is compared to that of the zero solution (equal gelatin and Arsenazo III amounts in 50 ml of 2 N HCl). To prepare the reagent solution, 10 mg of Arsenazo III in 50-60 ml of H 0 is diluted with 15 ml of HCl (1:5), and filled up with $H_{2}O$ to 100 ml. In the presence of Cu in the analytic solution, Cu salt solution in 2 N HCl is added to the zero solution in a quantity corresponding to the Cu amount Card 3/6

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CIA-RDP86-00513R000101920014-3

s/032/61/027/007/001/012 Colorimetric method for determining B110/B203 in the aliquot analytic solution. Measurements were made with an ΘK -M (FEK-M) photoelectric colorimeter and red light filter in a cuvette, 2 cm long. The first two authors (Ref. 3: Zavodskaya laboratoriya, XXVI, 415 (1960)) plotted a calibration curve for 5-25% Zr in 50 ml of 2 N HCl. There are 2 tables and 7 references: 3 Soviet-bloc and 4 non-Soviet-bloc. The reference to the English-language publication reads as follows: Ref. 6: G. Milner, J. Edwards. Anal. Chim. Acta, 13, 230 (1955). ASSOCIATION: Gosudarstvennyy nauchno-issledovatel'skiy i proyektnyy institut redkometallicheskoy promyshlennosti (State Design and Planning Scientific Research Institute of the Rare Metals Industry) Table 1. Effect of various elements on colorimetric zirconium determinations (measurement by means of $\oint \mathcal{F}K-M(FEK-M)$ colorimeter). Legend: (1) Reagent, (2) conditions of determination, (3) optimum concentration of Zr in 50 ml, \mathcal{M} , (4) permissible amount of the element, mg, (5) pyrocatechol violet, (6) xylenol orange, (7) Arsenazo III, (8) acetate buffer $pH \approx 5.2 + Trilon B$.

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Card 4/6

STATISTICS.

CIA-RDP86-00513R000101920014-3

GORYUSHINA, V.G.; ARCHAKOVA, T.A.

Colorimetric determination of calcium with acid chrome blue K. Zav.lab. 28 no.7:796-798 62. (MIRA 15:6)

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s/180/60/000/004/017/027

18:1210

AUTHORS :

Archakova, Z.N., Romanova, O.A. and Fridlyander, I.N. (Moscow)

TITLE: Investigation of the Properties of Alloys of the Alloy Lingdon System at Room and Elevated Temperatures

82622

E193/E483

PERIODICAL: Izvestiya Akademii nauk SSSR, Otdeleniye tekhnicheskikh nauk, Metallurgiya i toplivo, 1960, No.4, pp.106-110

The alloys studied in the course of the investigation TEXT : described in the present paper contained 0 to 3% Li and 4.0 to 6.5% Cu, the content of other alloying additions being constant and amounting to 0.1% Cd, 0.6% Mn and no more than 0.3% each of The mechanical properties of the alloys were Fe and Si. determined after 4 types of thermal treatment: (1) solution treatment, i.e. quenching from 525 to 535°C; (2) annealing, i.e. cooling from 430 to 150°C in 7 days; (3) ageing at room temperature for 7 days; (4) ageing at temperatures between 150 and 200°C for 12 h at 200°C and 16 h at other temperatures. The mechanical tests were carried out both at room and elevated (200 to 250°C) temperatures on specimens machined from extruded rod and appropriately heat-treated. It was found that Card 1/3

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Investigation of the Properties of Alloys of the Al-Cu-Li-Cd-Mn System at Room and Elevated Temperatures

simultaneous introduction of 0.9 to 1.4% Li and 0.1% Cd greatly increased the strength of the Al-Cu-Mn alloys in the age-hardened condition and, on the basis of the results obtained, the composition of a new, high strength rod alloy, suitable for high temperature The nominal composition of the new alloy VAD23 is: 5.4% Cu, 1.25% Li, 0.6% Mn, 0.15% Cd, remainder Al: its U.T.S. and 0.2% proof stress at 20°C are 60 and 54 kg/mm² respectively. Regarding its room-temperature strength, the new alloy resembles the high strength, Al-Zn-Mg-Cu alloys (type V95); its mechanical properties at high temperatures (150 to 250°C) are better than those of any known Al-base alloy of this type and, what is particularly important, the alloy retains its strength after long periods at these temperatures. Thus, the U.T.S. and elongation 6 of the VAD23 alloy, held at 200°C for 0.5 h . were 46.9 kg/mm^2 and 6.3% respectively, the corresponding figures for the V95 alloy being 40 kg/mm² and 12.6%. After 100 h at the temperature, U.T.S. and δ of the VAD23 alloy were still Card 2/3

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Investigation of the Properties of Alloys of the Al-Cu-Li-Cd-Mn System at Room and Elevated Temperatures

42.4 kg/mm² and 6.3%, whereas U.T.S. of the V95 alloy fell to 19.3 kg/mm², its 5 increasing to 19.8%. The corrosion resistance of the new alloy is not impaired by its lidium content; its specific gravity (2.73 g/cm³) is slightly lower and its molecules of elasticity slightly higher than those of the standard aluminium alloys D16 and V95. There are 5 figures, 2 tables and 10 references: 2 Soviet, 5 English and 3 German.

SUBMITTED: March 17, 1960

Card 3/3

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"APPROVED FOR RELEASE: 06/05/2000 CIA-RDP86-00513R000101920014-3 FRIDLYANDER, I.N. (Moskva); ROMANOVA, O.A. (Moskra); ARCHAKOVA, Z.N. (Moskva); Prinimali uchastiye: REZNIK, P.G.; LEBEDEVA, N.S. Mechanical properties of heat-resistant aluminum alloys with

lithium and cadmium, Izv.AN SSSR. Otd.tekh.nauk. Met.i topl. no.4:82-89 Jl-Ag 62. (MIRA 15:8) (Aluminum alloys--Testing) (Heat-resistant alloys--Testing)

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FRIDLYANDER I.N.: ANDREYEV. A.D.; PAVLOVA, I.K.; ROMANOVA, O.A.; ARCHAKOYA, Z.N.; Prinimali uchastiye: FOMIN, K.N.; POTAPOVA, V.I.; KALININA, Ye.N.

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Selecting a technology and studying the effect of technological factors on the structure and properties of the VAD23 alloy. Alium. splavy no.3:182-193 '64. (MIRA 17:6)

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ACCESSION NR: AT4037659 S/2981/64/000/	003/0182/0193	
AUTHOR: Fridlyander, I. N.; Andreyev, A. D.; Pavlo Archakova, Z. N.	ova, I. K.; Romanova, O. A.;	
TITLE: Selection of a fabrication process and a structure and propert:	tudy of the effects of les of alloy VAD23	
OURCE: Alyuminiyevy*ye splavy*, no. 3, 1964. Def Malleable alloys), 182-193	formiruyemy*ye splavy*	x ·
OPIC TAGS: aluminum alloy, alloy VAD23, alloy str roperty, alloy hardening, alloy aging, alloy cast lloy hot rolling, alloy cold rolling, alloy forgin ropy, high strength aluminum alloy, heat resistant	ing, alloy hot pressing, ng, alloy semiproduct aniso-	•
BSTRACT: Ingots (diameter 300 mm, length 1000 mm) ip-cast (flux refined, kept 60 min. at 745-780C, p dded in mold, liquid flux 46% LiC1 plus 54% KC1, r ip rate 15-18 mm/min), then homogenized for 24 hrs ere then hot pressed into PR306-7 sections (deform 0 min. at 525 \pm 5C, aged 12 hrs. at 170C), panels rd1/32	boured, 1.4% Li and 0.15% Cd mixed, settled at 750-770C, a. at 510 ± 10 C. The ingots mation 94%, 420-440C: hardened	X -

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ACCESSION NR: AT4037660	S/2981/64/000/003/0194/0200
AUTHOR: Fridlyander, I. N.; Romanova, Dronova, N. P.; Petrova, A. A.; By*chko	O. A.; Archakova, Z. N.; Gur'yev, I. I.; va, Z. S.
TITLE: Preparation and testing of intermore resistant aluminum alloy VAD23	ediate shapes from high-strength heat
SOURCE: Alyuminiyevy*ye splavy*, no. 3 alloys), 194-200	, 1964. Deformiruyemy*ye splavy* (Malleable
TOPIC TAGS: aluminum alloy, alloy VAD2 aluminum alloy, alloy mechanical property pressed strip, hot rolled sheet, cold rolled	23, heat resistant aluminum alloy, high strength , hot pressed rod, hot pressed section, hot I sheet, forged piece, double pressing
nto rods (intermediate diameter 127 mm o strips with 25x210 mm cross section and pre	eter 260 mm) of alloy VAD23 (5.1-5.7% Cu, 1.2- 0.15-0.25% Ti) were hot pressed (430-450C) r final diameter 20 mm), sections PR306-7, ssedpanels. The pieces were water quenched Sheets 1.0, 1.5 and 2.0 mm thick were hot
ard 1/2	

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rolled from strips to 6.0-5.5 mm, then cold rolled to desired thickness with intermediate annealing and finally heat treated (water quenched from $523\pm5C$, aged 16 hours at $170\pm5C$). Forgings (90 or 120x200x400 mm) were forged on a vertical press (deformation 65%, preheating 3 hours to 420-440C) from rods (diameter 180 mm) and heat treated as for sheets. Pressed shapes exhibited high tensile strength ($66-70 \text{ kg/mm}^2$) at a relative elongation of 3-4%. It was noted that double pressing (i.e., into intermediate diameter rods, then final shape) reduced the tensile strength and increased the plasticity. Mechanical properties of sheets and forgings were lower than those of the pressed shapes. "K. N. Fomin, N. S. Lebedeva, P. G. Reznik, N. Averkina, L. S. Zheltovskaya, Yu. A. Vorob'yev and N. N. Tyurin also took part in the work." Orig. art. has: 7 tables. ASSOCIATION: none

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SUB CODE: MM	NO REF SOV: 000	OTHER: 000	
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STATISTIC ISCILL

AUTHOR: Fridlyander, I. N. (Docto	r of technical sciences); <u>Romanova, O. A.:</u>	
Archakova, Z. N.	47	
ORG: none	41	
ono. none	BTI	
TITLE: Properties of VAD23 alloy		
SOURCE: Alvuminivevvve splavy, no.	4, 1966, Zharoprochnyye i vysokoprochnyye splavy	
(Heat-resistant and high-strength	alloys), 5-14	
	containing alloy, lithium containing alloy,	
TOPIC TAGS: aluminum alloy, copper	se containing alloy, titanium containing alloy,	
alloy composition, metal property/	VAD23 aluminum alloy	
7	Lithium, cadmium, manganese, titanium, iron, and aluminum alloy have been studied under laboratory	
ABSTRACT: The effects of <u>copper</u> , silicon on the properties of VAD23	3 aluminum alloy have been studied under laboratory	
The state The second state	ACAY CASE IN A WATET-CODIED MOID, WELE CALLUGUE AND	
round bars 10 mm in diameter, which	waried from 0 to 3.0% at copper contents of 4.0%,	
	imitim (1), 100/manganese (V.Va), and creations (******	
To the other ceries of	incots at a constant lithium (1,5%) and copper	
(5.2%) content, the manganese content $(5.2%)$ to $0.3%$	ntent was varied from 0 to 2.0%, cadmium from %, and iron and silicon from 0 to 0.9%. It was	
0 to 5.0%, titanium itom 0 to 0.5.	Ang 1944	
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found that: lithium intensifies the effects of aging; copper at contents of 4%-5%. increases strength; manganese at contents up to 1.0% improves strength and ductility; up to 0.2% cadmium increases strength of aged alloys and intensifies the effects of artificial aging; and titanium at contents of up to 0.3% has no effect on tensile strength but improves rupture strength. Iron and silicon were found to be harmful impurities. On the basis of these results the optimum composition of VAD23 alloy was established as follows: 4.9-5.8% copper: 1.0-1.4% lithium, 0.1-0.25% cadmium: 🚡 0.4-0.8% manganese; a maximum of 0.3% each of iron and silicon: and a maximum of 0.15% titanium. Artificial aging at 150-160C for 10-12 hr produces the best combination of mechanical properties: tensile strength, 51-54 kg/mm²; yield strength, 36-44 kg/mm² with an elongation 10-15%. Cold rolling prior to heat treatment, with reductions from 4% to 10%, promotes intensive grain growth and lowers strength and ductility. At the present, round and flat ingots are produced by continuous casting and processed by rolling and extrusion. (Orig. art. has: 10 figures. [TD] SUB CODE: 11/ SUBM DATE: none/ ORIG REF: 007/ OTH REF: 006/ ATD PRESS:50.56 Card 2/2 ha

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L 46984.66 EWT(m)/EWP(t)/FTI IJP(c) JH/JD
ACC NR: AT6024912 (A, N) SOURCE CODE: UR/2981/66/000/004/0037/0048
AUTHOR: Mikhaylov, K. N.; Kovrizhnykh, V. G.; Archakova, Z. N.; Baranchikov, V. M.; Sandler, V. S.; Shvets, V. A.
ORG: none 40 B+1
TITLE: Preparation of pressed semifinished products from VAD23 alloy
SOURCE: Alyuminiyevyye splavy, no. 4, 1966. Zharoprochnyye i vysokoprochnyye splavy (Heat resistant and high-strength alloys), 37-48
TOPIC TAGS: aluminum alloy, metal pressing, solid mechanical property / VAD23 alumi-
ABSTRACT: In order to determine the possible applications of VAD23 alloy, the influ- ence of various technological factors on its mechanical properties and structure was investigated. The optimum mechanical properties were found to be produced by pressing directly from an ingot which had first undergone homogenization. The optimum pressing temperature of sections with a flange thickness of 5 mm, 470-490°C, i. e., the temper- ature to which the blanks are heated, insures high strength characteristics and a com- paratively good plasticity over the entire length of the section. The elongation per unit length of the sections is practically independent of the pressing temperature of the alloy and of the degree of primary recrystallization./X A change in the pressing rate in the range of 0.5-5.0 m/min at pressing temperatures of 250-430°C does not af-
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iy. in order quenching of	ticity of VAD23 to slow down th thin sections pr coefficient of	ne recrystalli ressed at 470-	zation of t 490℃.it i	he structure d s necessary to	uring heating fo	220
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L $46985-66 = WT(m)/EWP(t)/ETI IJP(c) JD/JT$ ACC NR. AT6024913 (A.N.) SOURCE CODE: UR/2981/66/000/004/0049/004	55
BT	
ORG: none	
TITLE: Study of pressed sections and panels of <u>D20</u> alloy <i>17</i> SOURCE: Alyuminiyevyye splavy, no. 4, 1966. Zharoprochnyye i vysokoprochnyye splav (Heat resistant and high-strength alloys), 49-56	vy
TOPIC TAUS: metal pressing, aluminum alloy, copper alloy, manganese containing allo	י ד
ABSTRACT: The effect of technological factors (elongation during pressing, pressing temperature, homogenization of initial ingot, heat treatment conditions) on the stru- ture and mechanical properties of pressed sections of D20 alloy (of the <u>Al</u> _Ou_Mn sys- tem) 2 and 5 mm thick was studied. In order to obtain the optimum mechanical proper ties, the heat treatment of the sections should consist of quenching after heating a $535^{\circ} \pm 5^{\circ}$ C and artificial aging at 160-170 °C for 16 hr. This schedule does not cau any tendency in the alloy to corrode under stress. Different elongations (from 14 t 43.4), pressing temperatures (320, 370, 420, and 480 °C), and homogenization of the i tial ingot do not appreciably affect the structure of the initial ingot or the mecha- cal properties of sections with wall thicknesses of 2 and 5 mm. Stretching of the s tions after quenching raises the yield point substantially, but the tensile strength	
Card 1/2	