

1ST AND 2ND ORDER										3RD AND 4TH ORDER									
PROCESSES AND PROPERTIES INDEX																			
<i>ca</i> The dynamics of crystal growth and change in form with gypsum and calcium carbonate as examples. M. I. Chepuravskii and B. B. Evlina. <i>Trans. Sci. Inst. Metallurgy and Metallurgy</i> (U. S. S. R.) No. 137, 127-44 (1937). --A review and a discussion of the bearing of the dynamics of crystal growth on the crystn. of apatite with its various lamellae. J. S. Joffe																			
ASB-1A METALLURGICAL LITERATURE CLASSIFICATION																			
1ST ORDER										2ND ORDER									
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1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCESSES AND PROPERTIES INDEX																			
Current problems in inorganic chemical analysis. M. L. Chupelavskii. Zuretskaya Lab. 7, 521-5 (1958). A Chay. Blanc																			
ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION										EXT. CITE. INDEX									
SUBMITTER'S INDEX										SUBMITTER'S INDEX									
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Physical chemical investigations of fertilizers. M. L. Chelpevskii, *Nauka Inst. Fertilizers and Trapped Elements of the Soil*. P. Samoilova 1919-20, 63 7 (1939); *Akim. Referat. Zhur.* 1940, No. 6, 80-91. Studies of equilibria, kinetics of decomposition, crystal, and transformation of phases during the decomposition by acids of natural phosphate in systems: $\text{CaO-P}_2\text{O}_5\text{-SO}_3\text{-H}_2\text{O}$, $\text{CaO-P}_2\text{O}_5\text{-Na}_2\text{O-H}_2\text{O}$, $\text{CaO-P}_2\text{O}_5\text{-H}_2\text{O}$, $\text{Fe}_2\text{O}_3\text{-P}_2\text{O}_5\text{-SO}_3\text{-H}_2\text{O}$, $\text{Al}_2\text{O}_3\text{-P}_2\text{O}_5\text{-H}_2\text{O}$, etc., showed that superphosphate contains CaSO_4 and that by changing the concn. of H_2PO_4 and the temp. during extrn. it is possible to obtain the anhydride, the hemihydrate or *gypsum*. The permissible contents of Fe_2O_3 in natural phosphate for fertilizer manuf. by acid treatment were detd. Methods for producing $\text{Na}_2\text{CO}_3\text{-(NH}_4\text{)-SO}_4$ and mirabilite were developed by studying the systems: $\text{Na}_2\text{SO}_4\text{-NH}_4\text{HCO}_3\text{-H}_2\text{O}$, $\text{Na}_2\text{SO}_4\text{-NH}_4\text{-H}_2\text{O}$, $\text{Na}_2\text{SO}_4\text{-(NH}_4\text{)}_2\text{SO}_4\text{-NH}_4\text{-H}_2\text{O}$, and $\text{Na}_2\text{SO}_4\text{-NH}_4\text{-CO}_2\text{-H}_2\text{O}$. The formation and stabilization of emulsions were studied, and marginal angles of wetting, heats of wetting, fluidities and adhesive properties of insecticide powders were measured. The systems $\text{Na}_2\text{O-As}_2\text{O}_5\text{-H}_2\text{O}$ and $\text{Na}_2\text{O-CaO-As}_2\text{O}_5\text{-H}_2\text{O}$ were studied to det. methods for producing As preps. Theoretical studies were carried out to det. methods for regenerating caustic alkalis and for producing FeSO_4 and Na_2SiF_6 . The presence of Al salts in the soln. decreases the effect of salting out Na_2SiF_6 . Thermal treatment of phosphates is discussed.

W. R. Henn

ASB-51A METALLURGICAL LITERATURE CLASSIFICATION

6-00000000

Abstract

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444

1157: 030473

21111 500 000 13

1ST AND 2ND ORDERS																										3RD AND 4TH ORDERS																									
PROCESSES AND PROPERTIES INDEX																																																			
<i>CA</i> Analytical chemistry in U. S. S. R. M. I. Chuprovskiy. Trudy Vsesoyuz. Konfrentsi. Anal. Khim., Abad. S. S. R. 1, 6-40(1939); Khim. Referat. Zhur. 1939, No. 6, 64.—The paper reviews the most important investigations in Russia with respect to chem. industry. W. R. Henn																																																			
RESEARCH AND DEVELOPMENT RESEARCH AND DEVELOPMENT																																																			

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCESSES AND PROPERTIES INDEX																			
ca										2									
Hidden periods of crystallization and the equation for the velocity of formation of crystal nuclei. M. I. Chetaniyev. <i>J. Phys. Chem. (U. S. S. R.)</i> 19, 361-71 (1959).—Engel. data are given on the hidden periods of crystals of several salts as a function of supersat. and temp. The hidden period is a criterion of the rate of formation of nuclei and their growth. The equation for the rate of growth of crystal nuclei is analogous to that for the condensation of a liquid from a vapor with the addn. of a viscosity factor. From these data the approx. sp. surface energies of crystals on the boundary solid body-liquid are calcd. to be $4(1/3) \times 10^{-12}$; $\text{Ca}(\text{H}_2\text{PO}_4)_2 \cdot \text{H}_2\text{O}$, 2.4; $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$, 7.2; CaCO_3, 60; CaSO_4, 65; $\text{FePO}_4 \cdot 2\text{H}_2\text{O}$, 98; BaSO_4, 104; and CaF_2, 132, in good agreement with calcs. from hardness data according to Hertz, Auerbach and Frenkel. F. H. Rathmann																			
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION																			
FROM STEINBRONN										FROM SCHLITZ									
GROUP 1A										GROUP 1B									

CA	Titration by maximum turbidity with a photoelectric colorimeter. I. M. L. Chepelevskii. <i>Zarodshaya Lab.</i> 11, 498-503 (1945).—Addn. of each portion of the pptg. agent to the solu. of the sample results in the formation of a ppt. and an increase in the turbidity of the solu. (absorption of light). Increase in the absorption of light takes place until the region close to the equiv. point is reached. A further addn. of the pptg. agent does not increase the absorption of light, but decreases the degree of the turbidity, owing to the diln. of the suspension by excess of the pptg. agent and to the partial coagulation of the suspension. The method of titration by max. turbidity can be used in cases where volumetric analysis based on oxi. and oxidation-reduction reaction is not possible (in detn. of stable cations and such anions as SO_4^{2-}, PO_4^{3-}, and F^-, forming pptg. in the presence of impurities which interfere with the titration in neutral solu.). Titration of H_3PO_4 with BiOCl_3 according to the reaction $\text{H}_3\text{PO}_4 + \text{BiOCl}_3 = \text{BiPO}_4 + \text{HClO}_4 + \text{H}_2\text{O}$ produced satisfactory results. To prevent hydrolysis, 10 ml. of 6% HClO_4 was added before the titration. The ppt. was sufficiently dispersed and formed rapidly. Addn. of 5 ml. of 2.5% HNO_3 increased the percentage error to -8%, relative. The presence of chlorides interfered with the detn. Fluorides can be titrated with Th salts, using alizarin as indicator. They can be titrated also with $\text{Pb}(\text{AcO})_2$ in the presence of excess NaCl and alc., forming a highly dispersed ppt. of PbFCl. 10 references.
	W. R. Henn

CA

7

TITRATION BY MAXIMUM TURBIDITY BY MEANS OF THE PHOTOELECTRIC COLORIMETER. II. DETERMINATIONS OF SULFATE ION IN PHOSPHATE SOLUTIONS. M. L. Chepuravetskii, S. S. Rubinsova, and B. B. Ryzkina, *Zashchita Lab.* 11, 783-7 (1945); cf. C.A. 40, 2407^a.—Analysis of phosphate solns. for SO_4^{--} by titration by max. turbidity yields more accurate and reliable results than are obtained gravimetrically from rats. acidified with HCl. Add 3 ml. of 3% HCl to a definite vol. of the sample ext. in a photometer cuvette (5 or 10 ml. of the content of SO_4 ; if up to 1%, 3 ml. if 1–3%, 2 or 1 ml. if 2–5%), add water to yield approx. 75 ml. of soln., set the mixer in motion, add $\frac{1}{2}$ of the vol. of 0.25 or 0.1 N BaCl₂ soln. required for the pptn., select the light filter so that the readings of the logarithmic sector are within the first 20 scale divisions, and add BaCl₂ in portions of 0.2–0.3 ml. After the addn. of each portion, let stand for 1.0–1.5 min., compensate the illumination of the 2nd photometer by means of the logarithmic sector until the photocurrents are in equil., and record simultaneously the readings of the logarithmic sector and those of the buret. Construct on a mm. paper that segment of the titration curve which is proximate to the end of pptn. and det. graphically the no. of ml. of BaCl₂ soln. required for the pptn. The time required for the analysis is considerably less than that required for gravimetric detn.

W. R. Henn

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

CHEPELEVETSKIY, M.L.

③ 4

Solubility isochrones-isotherms of apatite in the system $H_2SO_4-H_3PO_4-H_2O$ as example of topochemical kinetics, and the production of superphosphate. E. B. Bratskiy and M. L. Chepelevetskiy, *Izvest. Sektora Fiz.-Khim. Anal., Akad. Nauk S.S.S.R.* 20, 383-8 (1950).—When apatite is digested with H_2SO_4 in the production of superphosphate, too high a concn. of the acid or too high a temp. stops the reaction. This is attributed to deposition of $CaSO_4$ formed in initial stages of the process on grains of apatite and thereby protecting it from the action of H_2SO_4 . The reaction was studied at 50, 70, and 90°, and 4 and 7 hrs. duration. Plotting H_2SO_4 concn. vs. P_2O_5 in soln. showed that for each of these temps. there are concns. of H_2SO_4 at which the reaction is arrested. As the temp. rises this zone of complete passivation of the apatite is shifted toward lower concns. Addn. of H_3PO_4 lowered the soly. of apatite still further. For greatest efficiency in superphosphate production it is essential to feed the ingredients, H_2SO_4 and apatite, at controlled rates. Cf. Ch., *et al.*, *C.A.* 38, 5039.
M. Hoseh

CHEPELEVETSKIY, M.L.

Nuclear Science Abstracts
July 15, 1954
Chemistry

①
✓ EXPERIMENT ON THE CONSTRUCTION OF A PERIODIC
SYSTEM OF ATOMIC NUCLEI. M. L. Chepelevetskiy.
Zhur. Ekspitl. i Teoret. Fiz. 23, 327-38(1953) Sept. (In
Russian)

For a foundation of a series of investigations of nuclear
properties a periodic system of isotopes of atomic nuclei is
given. The dimensions of the periods of the nuclei were de-
termined by the number of neutrons and protons in the
periods, and in the same way the connection between the
forms of the periodicity was determined. The hypothesis
states that the construction of a period is accomplished by
means of finishing the construction with nucleons of surface
layers of nuclei with subsequent finishing of its deep layers.
(tr-auth)

RMZ

9-21-54

AMELIN, A.G.; BALNYEV, A.V. [deceased]; BRUTSKUS, Ye.B.; KEL'MAN, P.N.;
OSHEROVICH, R.Ye.; STEPANOV, M.N.; CHEPELEVITSKIY, M.L.; CHIRNO-
BAYEVA, M.M.; MIKHAL'CHUK, B.V., redaktor; LEBOT'YEVA, K.D., re-
daktor; SHPAK, Ye.G., tekhnicheskii redaktor.

[Methods of analysing and controlling the production of sulfuric
acid and superphosphates] Metody analiza i kontrolya proizvedstva
sernoi kisloty i superfosfata. Ser. A.G.Amelin i dr. Pod red.
B.V.Mikhal'chuka. Moskva, Gos.nauchno-tekhn. izd-vo khim. lit-ry,
1955. 159 p. (MLRA 9:5)

1. Moscow. Nauchnyy institut po udebreniyam i insektitsidam.
(Sulphurec acid) (Phosphates)

CHEPELEVETSKIY, M. L.

USSR/Chemical Technology - Chemical Products and Their Application. Fertilizers,
I-6

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 62121

Author: Chepelevetskiy, M. L., Bol'ts, Ts. S., Vasilenko, N. A., Rubnova,
S. S.

Institution: None

Title: Solubility of Crystallhydrate $\text{CaHPO}_4 \cdot 2\text{H}_2\text{O}$ and Rate of Its Con-
version to Anhydrous Salt in Connection with Some Problems of the
Production of Phosphorus Fertilizers

Original
Periodical: Sb. Issledovaniya po prikl. khimii, Moscow-Leningrad, Izd-vo AN
SSSR, 1955, 175-183

Abstract: Study of solubility of metastable at 40° $\text{CaHPO}_4 \cdot 2\text{H}_2\text{O}$ (I) in phos-
phoric acid solutions and also of the rate of phase conversion (PC)
of I to anhydrous salt (II). For the investigations were prepared
at $2-3^\circ$ solutions of monocalcium phosphate which decompose on rise
in temperature with separation of I as solid phase. The derived

Ca, Card 1/2

CHEPELEVETSKIY, M.L.

VASILENKO, N.A.: CHEPELEVETSKIY, M.L.

Solubility of lanthanum phosphate in aqueous solutions of phosphoric acid at 80°. Zhur. neorg. khim. 2 10:2486-2489 0 '57. (MIRA 11:3)

1. Nauchno-issledovatel'skiy institut po udobreniyam i insektofungitsidam, Laboratoriya fiziko-khimicheskogo analiza.

(Solubility) (Lanthanum phosphate) (Phosphoric acid)

CHEPELEVETSKIY, Mark Leybovich, prof.; BRUTSKUS, Yelena Borisovna;
SOKOLOVSKIY, A.A., red.; LUR'YE, M.S., tekhn.red.

[Superphosphate; physicochemical production principles]
Superfosfat; fiziko-khimicheskie osnovy proizvodstva. Moskva,
Gos.nauchno-tekhn.izd-vo khim. lit-ry, 1958. 272 p. (MIRA 12:2)
(Phosphates)

KHAMSKIY, Ye.V.; CHEPELEVETSKIY, M.L.

Effect of the formation conditions of calcium sulfate semihydrate
on kinetics of its conversion to dihydrate in phosphoric acid
solutions. Zhur. prikl. khim. 31 no.7:976-980 J1 '58.

(Calcium sulfate) (Phosphoric acid)

(MIRA 11:9)

AUTHORS: Chepelevetskiy, M. L., Gimmel'farb, B. M., 20-119-1-36/52
Kuperman, M. Ye., Krasil'nikova, Z. V.

TITLE: An Electron-Microscope Investigation of the Structure of
Phosphorites From the Kara-Tau Basin (Elektronno-mikroskopi-
cheskoye issledovaniye struktury fosforitov basseyna Kara-
-Tau)

PERIODICAL: Doklady Akademii Nauk SSSR, 1958, Vol. 119, Nr 1, pp. 133-135
(USSR)

ABSTRACT: The phosphorites of this basin (deposits Ak-Say, Kok-su and
Chulak-Tau) contain larger quantities of dolomite (mostly
10-18%), whereby the consumption of sulfuric acid per ton of
assimilable P_2O_5 in superphosphate increases. Thereby the
quality of this fertilizer is impaired as well with regard
to the assimilable P_2O_5 as to its physical properties: it
becomes hygroscopic and smeary. At present 2 methods of the
enrichment of these phosphorites exist: flotation and the
chemical method. By flotation it was possible to attain a
concentrate with a highly reduced magnesium content (Ak-Say),
whereas the phosphorites of the Chulak-Tau deposit still
yield concentrates with an MgO-content of 1,5% and higher.

Card 1/3

An Electron-Microscope Investigation of the Structure of
Phosphorites From the Kara-Tau Basin

20-119-1-36/52

These difficulties may be explained by the grain size of the phosphate substance of these phosphorites. The respective ores were inspite of a similar geological age and belonging to the same series of phosphorites intensively changed by a contact-metamorphism (nearness of a granite-intrusive), especially their phosphates were recrystallized. The structural peculiarities of the Chulak-Tau phosphorites were investigated under an electron-microscope. The structure of the phosphorites of the two remaining deposits were studied for comparison under an ordinary microscope. The characteristics of the Kara-Tau phosphorites are given in table 1. Polished sections of phosphorite samples were produced, impressions were made by the polystyrene-quartz and the collodium-quartz method and then etched, and again impressions made. The investigation showed that the size of the phosphate grains in all 5 samples from Chulak-Tau lies between 0,1 and 4,0 (figure 2). As the production of concentrates is due to the grain size in Chulak-Tau rendered difficult, the flotation shall be combined with a refinement by diluted acids, especially H_2SO_4 . There are 2 figures, 1 table.

Card 2/3

An Electron-Microscope Investigation of the Structure of
Phosphorites From the Kara-Tau Basin

20-119-1-36/52

ASSOCIATION: Nauchnyy institut po udobreniyam i insektofungisidam
(Scientific Institute for Fertilizers and Insecticides).
Gosudarstvennyy institut gornokhimicheskogo syr'ya
(State Institute for Mining-Chemical Raw Materials)

PRESENTED: June 11, 1957, by S. I. Vol'fkovich, Member, Academy of
Sciences, USSR

SUBMITTED: June 5, 1957

Card 3/3

BRUTSKUS, Ye.B.; CHEPELEVITSKIY, M.L.

Water - salt systems including orthophosphates. Itogi
nauki: Khim.nauki 4:109-121 '59. (MIRA 13:4)
(Systems (Chemistry)) (Phosphates)

5(2)

SOV/80-32-5-2/52

AUTHORS:

Khamskiy, Ye.V., Chepelevetskiy, M.L.

TITLE:

On the Crystallization of Potassium Sulfate From Solutions of Extraction Phosphoric Acid

PERIODICAL:

Zhurnal prikladnoy khimii, 1959, Vol 32, Nr 5, pp 948-952 (USSR)

ABSTRACT:

The study of the crystallization of potassium sulfate from solutions of phosphoric acid is closely connected with the production of H_3PO_4 by the method of sulfuric acid extraction. There are the dihydrate, polyhydrate and anhydrite methods of production [Ref 1]. The principal condition for the dihydrate method is the absence of the polyhydrate of potassium sulfate in the bottom phase at the end of the production process. The technological conditions for producing phosphoric acid of high concentration by the dihydrate method are investigated here. It has been shown (Table 1) that in distinction from solutions of chemically pure H_3PO_4 the dihydrate of potassium sulfate is first precipitated in the bottom phase. This change is explained by the presence of admixtures. If the ratio CaO/SO_3 is increased, the crystallization of calcium sulfate in the form of dihydrate takes place at more concentrated solutions of phosphoric acid

Card 1/2

SOV/80-32-5-2/52

On the Crystallization of Potassium Sulfate From Solutions of Extraction Phosphoric Acid.

and at higher temperatures. The increase of this ratio in the liquid phase raises the stability of the dihydrate in the solutions of phosphoric acid. The experiments showed that the admixtures facilitate the crystallization of calcium sulfate as dihydrate and that the increase of the ratio CaO/SO_3 in the liquid phase improves the stability of the dihydrate. At 70°C and a molar ratio of CaO/SO_3 phosphoric acid containing 34-38% P_2O_5 can be produced. There are: 3 tables and 5 Soviet references.

ASSOCIATION: Nauchnyy institut po udobreniyam i insektofungisidam im. prof. Ya.V. Samoylova (Scientific Institute of Fertilizers and Insectofungicides imeni Professor Ya.V. Samoylov)

SUBMITTED: December 21, 1957

Card 2/2

YANYISHEVA, V.S.; YUZHNAIA, Ye.V.; CHEPELEVETSKIY, M.L.

Electron microscope study of calcium sulfate films formed at the surface of calcite crystals during the decomposition of the latter by sulfuric acid. Dokl. AN SSSR 141 no.5:1161-1162 D '61. (MIRA 14:12)

1. Nauchnyy institut po udobreniyam i insektofungisidam im. Ya.V. Samoylova. Predstavleno akademikom S.I. Vol'fkovichem. (Calcium sulfate) (Electron microscopy)

CHEPELEVETSKIY, M.L.; KHARITONOVICH K.F.

Solubility product of lead molybdate. Zhur. anal. khim.
18 no.3:357-359 Mr'63. (MIRA 17:5)

1. Moskovskiy institut tonkoy khimicheskoy tekhnologii
imeni Lomonosova.

LAMP, V.N.; CHEPELEVETSKIY, M.L.

Solubility in the system $K_2SO_4 - CaSO_4 - H_3PO_4 - H_2O$ (isotherm
80°C). Zhur. prikl. khim. 36 no.12:2650-2656 D '63.
(MIRA 17:2)

CHEPELEVETSKIY, M.L.; MAKAREVICH, V.M.

Determination of the sulfate ion in extractive phosphoric acid
by photodurbidimetric titration. Zav. lab. 30 no.8:935-937 '64.
(MIRA 18:3)

1. Nauchno-issledovatel'skiy institut po udobreniyam i insekto-
fungisidam imeni Samoylova.

KHARITONOVICH, K.F.; CHEPELEVETSKIY, M.L.

Study of calcium precipitation as molybdate by phototurbidimetric titration. Zhur. anal. khim. 20 no.6:743-745 '65.

(MIRA 18:7)

1. Moskovskiy institut tonkoy khimicheskoy tekhnologii imeni Lomonosova.

CHEPELEVETSKIY, M.L.; KHARITONOVICH, K.F.

Titimetric attachment to a photocolormeter. Zav. lab. 31 no.2:253-
254 '65. (MIRA 18:7)

1. Moskovskiy institut tonkoy khimicheskoy tekhnologii im. M.V.Lomonosova.

LAMP, V.N.; CHEPELEVSKIY, M.L.

Solubility of salts in a quaternary system
 $K_2SO_4 - CaSO_4 - H_3PO_4 - H_2O$. Isotherm 60° . Trudy NIUIF no.208:
104-108 '65.

Solubility of salts in the ternary systems
 $K_2SO_4 - KH_2PO_4 - H_2O$, $KHSO_4 - H_2PO_4 - H_2O$, and
 $KHSO_4 - KH_2PO_4 - H_2O$, constituents of the quaternary system
 $K_2O - SO_3 - P_2O_5 - H_2O$. Isotherms 80° . Ibid.:108-115

Conversion of the binary salts of syngenite
($K_2SO_4 \cdot CaSO_4 \cdot H_2O$) and pentasalt ($K_2SO_4 \cdot 9CaSO_4 \cdot H_2O$) to
soluble sulfates. Ibid.:115-122 (MIRA 18:11)

NISHCHENKO, Ya.S.; CHEPELEVSKIY, N.I.

Solubility of ammonium sulfate in the presence of sulfuric
and phosphoric acids and also of ammonium phosphates. Trudy
NIUIF no. 208:16-30 '65.

Solubility in the system

$(\text{NH}_4)_2 \text{HPO}_4 - (\text{NH}_4)_2 \text{SO}_4 - \text{H}_2\text{SO}_4 - \text{H}_3\text{PO}_4 - \text{H}_2\text{O}$. Ibid. 190-42
(MIRA 18:11)

FILATOVA, I.N.; CHEPELEVETSKIY, M.L.

Formation in aqueous solutions of a binuclear ferrophosphate complex and its properties. Dokl. AN SSSR 166 no.1:140-143 Ja '66.
(MIRA 19:1)

1. Vsesoyuznyy institut khimicheskikh reaktivov i sposoby chistykh khimicheskikh veshchestv. Submitted June 14, 1965.

L 36482-66 EWT(m)/EWP(t)/ETI IJP(c) JD

ACC NR: AP6027073

SOURCE CODE: UR/0076/66/040/002/0307/0316

AUTHOR: Chepelevetskiy, M. L.

ORG: none

TITLE: Periodic system of atomic nuclei

SOURCE: Zhurnal fizicheskoy khimii, v. 40, no. 2, 1966, 307-316

TOPIC TAGS: nucleus, isotope, electron shell, neutron, proton, heavy nucleus, stoichiometry, transuranium element

ABSTRACT: A previously published periodic system of atomic nuclei¹⁹ is examined in greater detail. The development of a periodic system of nodal isotopes from the experimental data is shown both on the basis of analogies and in contrast to the structure of the electron shells in the Mendeleev system. The numbers of neutrons and protons filling the 1'-subgroups and the 1"-subgroups respectively can be found from the difference between the sizes of the nucleus periods for neutrons and protons correspondingly. These regularities emanate from the principle of shell layers completion, analogous for the electron and nucleon periodic systems. For average and heavy nuclei the filling numbers of neutrons in each given 1'-subgroup are equal to the sum of the filling numbers of protons in the given 1"-subgroup and in the previous one, i.e., 1"-1. This indicates the simple stoichiometry of the proton-neutron ratios of 1:1 between each two neighboring 1', 1"-subgroups -- a characteristic without analogy in electron shells. The periodic system of nuclei with an odd number of neutrons or protons is shown using nuclear isomerism as an example. The system can be continued beyond the limits of the discovered transuranium elements. The isotopic and mass numbers of nodal isotopes, predicted back in 1947, of earlier unsynthesized elements were subsequently confirmed upon their discovery. Orig. art. has: 1 figure, 3 formulas and 6 tables.

SUB CODE: 20, 07 / SUBM DATE: 21Dec64 / ORIG REF: 016 / OTH REF: 026 [JPRS: 36,455]

Card 1/1, 11/11

UDC: 541.20

USTICH, A.F., starshiy elektromekhanik; KNYAZEV, M.N., starshiy
inzhener; CHEPELEVICH, M.I.

From the editors' mail. Avtom., telem. i svyaz' 5 no.3:44 Mr
'61. (MIRA 14:9)

1. Mineralovodskaya distantziya signalizatsii i svyazi Severo-Kavkazskoy dorogi (for Ustich).
2. Nadezhdinskaya distantziya signalizatsii i svyazi Sverdlovskoy dorogi (for Knyazev).
3. Nachal'nik laboratorii signalizatsii i svyazi Litovskoy dorogi, (for Chepelevich).

(Railroads--Electric equipment)

(Railroads--Communication systems)

(Railroads--Signaling--Interlocking systems)

CHEPELEVSKIY, I., (Engr-Lt Col)

Coauthor with Engr-Lt Col A. INFELITSYN* of article, " Tent for Repairing Equipment," construction of a tent to be used in the field when repairing equipment. (Tankist, Moscow, No 4, Apr 54)

SO: SUM NO. 239, 13 Oct 1954

CHEPELEVSKIY, I.

STOYANOV, V., inzhener-podpolkovnik; CHEPELEVSKIY, I., inzhener-podpolkovnik.

Repairing electric equipment in the field. Tankist no. 5:38-43 My
'56.

(MIRA 11:3)

(Tanks (Military science)--Electric equipment)

CHEPELEVSKIY, I.

SHIR, V., Kandidat tekhnicheskikh nauk; CHEPELEVSKIY, I., inzhener.

Single design for automatic thermoregulators used in electrolytic
baths. Avt.transp. 35 no.7:25 J1 '57. (MLRA 10:8)
(Thermostat) (Electroplating)

~~CHEPELEVSKIY, I. F.~~

ZHIKH, V.A., kand. tekhn. nauk; CHEPELEVSKIY, I.F., inzh.

Automatic control of temperatures in a galvanic bath. Vest. mash. 38
no. 4:63-64 Ap '58. (MIRA 11:3)

(Thermostat)

CHEPELEVSKIY, V.; SARKHOSH'YAN, G.

Special exhibition "New techniques in the repairing of units
of motor vehicles." Avt.transp. 40 no.9:58 S '62.

(MIRA 15:9)

(Motor vehicles--Maintenance and repair)

CHEPELEVSKIY, V., inzh.; TUMANOV, I., inzh.

Unified specifications for the receiving for overhauling, and the
delivery after overhauling of motor vehicles, their parts and units.
Avt.transp. 41 no.10:28-29 O '63. (MIRA 16:10)

~~CHEPELEVSKIY, Vladimir Natanovich; TUMANOV, Ivan Aleks-vevich;~~
~~SARKHOSH'YAN, Gorgen Nikitovich; RUMYANTSEV, Aleksey~~
~~Nikolayevich; KLEVENSKIY, Aleksandr Iosifovich;~~
~~BELOTSERKOVSKAYA, S.I., red.; SHUPLYAKOV, S.I., red.~~

[New developments in the technology and equipment used
in motor-vehicle repair] Novoe v tekhnologii i oborudo-
vanii dlia remonta avtomobilei. Moskva, Transport, 1964.
127 p. (MIRA 18:1)

L 11780-66(11) EWT(1)/EWT(m)/EWP(e)/EWP(t)/EWP(b) IJP(c) JD/WH/GG/WH

ACC NR: AP6003253

SOURCE CODE: UR/0020/65/165/006/1336/1339

AUTHOR: ^{44,55}Bogomolova, L. D.; ^{44,55}Lazukin, V. N.; ^{44,55}Chepeleyva, I. V.; ^{44,55}Bal'skaya, L. A.ORG: ^{44,55}Moscow State University im. M.V. Lomonosov (Moskovskiy gosudarstvennyy universitet)TITLE: ^{21, 44, 55}Electron paramagnetic resonance of manganese ions in the As-Se-Ge glass system53
52
B

SOURCE: AN SSSR. Doklady, v. 165, no. 6, 1965, 1336-1339

TOPIC TAGS: EPR spectrum, manganese ion, nonsilica glass, arsenic selenium germanium glass, glass structural property

ABSTRACT: EPR spectra of Mn^{2+} in the glasses of the As-Se-Ge system containing 12.5—40 at% Ge have been studied in the 293—77K range to determine the type of chemical bonding of Mn and Ge atoms in relation to heat treatment and crystallization of the glasses. Glass samples were synthesized from high-purity materials by melting and slow cooling in evacuated quartz ampuls. All samples contained 1 at% Mn. The resonance absorption lines with g-factors of 2 and about 4 were observed in the EPR spectra of all samples. The lines with g-factor of 2, which broadened greatly with a decrease in temperature, were attributed to antiferromagnetic, small-size inclusions of $MnSe$ crystals. The "residual" line with a g-factor of 2 in the EPR spectra at 77K, especially

Card 1/2

UDC: 541.67-161.6:538.113

L 11780-55

ACC NR: AP6003253

strong in the sample with 40 at% Ge, was correlated with Mn in the glass skeleton. The EPR lines with g-factor of 4, which become more intense with an increase in Ge concentration, were associated with an increase in concentration of $[\text{GeSe}_4/2]$ and $[\text{GeGe}_4/4]$ tetrahedral nodes in the glass structure. The presence of Mn may contribute to the increase in the tetrahedral nodes content by a mechanism analogous to that theoretically established for Fe^{3+} in silicate glasses. The EPR line with g-factor of about 4.3 was observed earlier by Soviet and Western scientists in the Fe^{3+} containing silicate glasses. Mn in the glass lattice may be bound to As by a semipolar bond and to Se by a covalent bond. The EPR line with g-factor of 10 was observed in only one glass sample at 77K and was attributed to heat treatment. Orig. art has: 2 figures and 2 tables. [JK]

SUB CODE: 07/ SUBM DATE: 12May65/ ORIG REF: 004/ CTH REF: 001/ ATD PRESS: 4/78

HW
Cord 2/2

CHEPELIK, V., arkhitektor (g.Kiyev).

Practices of maintaining dormitories. Zhil.-kom. khoz. 10 no.10;
6-7 '60. (MIRA 13:10)
(Dormitories--Maintenance and repair)

1. CHEPEL'KIN, I.
2. USSR (600)
4. Brooms and Brushes
7. Spiral wire brush for cleaning the pipes of a vacuum apparatus. Moloch. prom. 14, no. 3, 1953.

9. Monthly List of Russian Accessions, Library of Congress, May 1953. Unclassified.

CHEPELKIN, N.A., agronom.

Wide horizons. Zemledelie 27 no.8:61-64 Ag '65.

1. Kolkhoz "Rassvet", Kirovskogo rayona, Mogilevskoy oblasti. (MIRA 18:11)

7

CHEPEL'KIN, Yu. N.

ca

Rapid determination of sulfite in bisulfite, sulfite and thiosulfate ligures. M. A. Portnov and Yu. N. Chepel'kin. *Zashchita Lab. S.* 561-5 (1939).—The tauridine method was found to be suitable for detg. SO₂ ions in bisulfite, sulfite and thiosulfate. The content of SO₂, SO₃ and SO₄ ions in the mother liquors obtained in the manuf. of thiosulfate can be detd. in about 1 hr. by the combined methods of Kinteraker-Wolach (cf. C. A. 21, 1041) and Dabbs (cf. C. A. 20, 3164).

B. Z. Kamich

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

REGION 57103319

SECTION 449 ONE COL

CLASSIFICATION

1041 30440

57111 ONE ONE ALL

1ST AND 2ND SERIES										1ST AND 2ND SERIES									
PROCESSING AND PROPERTIES INDEX																			
CHEPEL'KIN, YU. N. 1 CA Protective devices for glass electrodes, Yu. N. Chepel'kin, <i>Sovetskaya Lot.</i> 15, 373-4(1940).—The depts. are readily done in vessels in which the electrodes are permanently fixed in perforated stoppers on the top, while the bottom of the vessel is provided with a stopcock through which the soln. is easily removed after depts. A side tube is used for washing and for addn. of new solns. A portable frame in which the electrodes are fixed and surrounded by a protective glass screen can be used for immersion depts. in beakers, etc. G. M. Kuznetsov																			
ASB-51A METALLURGICAL LITERATURE CLASSIFICATION 8-27-44-1000																			
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CHEPEL'KIN, YU. N.

CA

Industrial-type calomel electrodes. N. K. Nikolskii and
~~Yu. N. Chepel'kin~~, *Zhurnal Fiz. Khim.* 13, 613 (1949).--
These electrodes have been specially designed for
detg. pH of flotation slurries. A characteristic feature is a
capillary constriction in the electrode tube to prevent loss of
contact between the Pt wire and Hg. The half-cell con-
sists of the calomel electrode and a jacket filled with satd.
KCl soln. The electrode proper consists of a 6-mm. (out-
side-diam.) glass tube with a partition in which the Pt wire
is held. One end of the Pt wire is joined to a Cu lead wire
and the other end is immersed in Hg in the capillary constric-
tion. Modifications to withstand vibrations during tests
are shown. B. Z. Kamich

3
4 Determination of hydrogen-ion concentration. N. K.
Nikol'skiy and Yu. N. Cheselkin. U.S.S.R. 105,257,
Apr. 25, 1957. To make the meas. independent of the kind
of electrode used, the meter is calibrated according to the
electrode potential and to the gradient potential. To this
end, 2 resistances are included in the setup: one in the
measuring circuit and the other in the comparison circuit.
Both resistances are of the variable type. - M. Haseh
4E4
177 conf

CHEPELKIN, Yu. N.

N. K. Nikol'skiy, I. P. Kell', Yu. O. Tennison and Yu. N. Chepelkin (Mekhanobr)

"The determination of the residual sulphur-ion concentration in the pulp with the aid of a silver-sulphide electrode"

report presented at the 4th Scientific and Technical Session of the Mekhanobr
Inst, Leningrad, 15-18 July 1958

KONDRAT'YEV, Nikolay Yevgen'yevich, starshiy nauchnyy sotrudnik;
CHKBOTAREV, A.I., otv.red.; ~~CHKPELKINA~~, L.A., red.;
YASNOGORODSKAYA, M.M., red.; ~~SERGEEV~~, A.W., tekhn.red.

[Design of reservoir coast changes; practical manual] Raschet
beregovykh pereformirovani na vodokhranilishchakh; prakticheskoe
posobie. Leningrad, Gidrometeor.isd-vo, 1960. 62 p.

(MIRA 14:1)

1. Gosudarstvennyy gidrologicheskiy institut (for Kondrat'yev).
(Coast changes) (Reservoirs)

BARANOV, Ivan Vasil'yevich; CHEPELKINA, L.A., red.; SERGEYEV, A.N.,
tekhn. red.

[Limnological types of lakes of the U.S.S.R.] Limnologicheskie
tipy ozer SSSR. Leningrad, Gidrometecizdat, 1961. 275 p.
(MIRA 15:5)

(Lakes)

(Limnology)

BYKOV, Nikolay Ivanovich; BORSUK, V.N., otv. red.; CHEPELKINA, L.A.,
red.; ALEKSEYEV, A.G., tekhn. red.

[Agronomic characteristics of soil moisture conditions in
the middle Volga Valley] Agrogidrologicheskie svoistva pochv
Srednego Povolzh'ia; spravochnik. Leningrad, Gidrometeoiz-
dat, 1962. 225 p. (MIRA 15:11)
(Volga Valley--Soil moisture)

VERIGO, Stefaniya Antonovna; RAZUMOVA, Lyubov' Aleksandrovna; KULIK, M.S.,
oty. red.; CHEPELKINA, L.A., red.; VOLKOV, N.V., tekhn. red.;
SUVOROVA, L.D., tekhn. red.

[Soil moisture and its role in agricultural production] Pochven-
naya vlaga i ee znachenie v sel'skokhoziaistvennom proizvodstve.
Leningrad, Gidrometeoizdat, 1963. 288 p. (MIRA 16:6)
(Soil moisture) (Agriculture)

SHUL'TS, Viktor L'vovich; KOZIK, Ye.M., otv. red.; CHEPELKINA,
L.A., red.; NIKOLAYEVA, G.S., tekhn. red.

[Rivers of Central Asia] Reki Srednei Azii. Leningrad,
Gidrometeoizdat. Pt.1. 1963. 301 p. (MIRA 16:10)
(Soviet Central Asia--Rivers)

IVANOV, K.Ye., doktor geogr. nauk, prof.; ROMANOV, V.V., kand. tekhn. nauk; SIDORKINA, L.M., kand.geogr. nauk; SHIFMAN, N.M., inzh.; BAVINA, L.G., inzh.; GALINOVSKAYA, I.A., inzh.; KOZHINA, Z.M., red.; CHEPELKINA, L.A., red.; SHATILINA, M.K., red.; BRAYNINA, M.I., tekhn. red.

[Hydrological calculation in the drainage of bogs and swampy soils] Gidrologicheskie raschety pri osushenii bolot i zabolochen-nykh zemel'. Pod red. K.E.Ivanova. Leningrad, Gidrometeoizdat, 1963. 447 p. ____ [Supplement no.9. Maps] Prilozhenie no.9. Karty. (MIRA 16:12)

1. Leningrad. Gidrologicheskiy institut.
(Drainage)

SKLYAROV, V.M., otv. red.; GRIBANOV, N.N., red.; MUROMTSEV, A.M., red.; POGOSYAN, Kh.P., red.; PROTOPOPOV, V.S., red.; RUDNEV, G.V., red.; SOKOLOV, A.A., red.; SOLOV'YEV, V.A., red.; USMANOV, R.F., red.; ZHDANOVA, L.P., red.; RUSAKOVA, G.Ya., red.; CHEPELKINA, L.A., red.; KOLESOVA, Z.M., tekhn.red.

[Man and the elements; hydrometeorologic desk calendar for 1964] Chelovek i stikhiia; nastol'nyi gidrometeorologicheskii kalendar' 1964. Leningrad, Gidrometeorologicheskoe izd-vo, 1963. 154 p. (MIRA 17:2)

SALAZANOV, Vladimir Vasil'yevich; PODVISHENSKAYA, N.Ya., kand.
tekhn. nauk, otv. red.; CHEPELKINA, L.A., red.

[Spring runoff of the rivers in the upper Dnieper Basin;
conditions governing their formation and the forecasting
methods] Vesennii stok rek basseina Verkhnego Dnepra; us-
lovia formirovaniia i metody prognozov. Leningrad, Gid-
rometeoizdat, 1964. 141 p. (MIRA 17:6)

SHMIDTS, Sergey Viktorovich; CHEPELKINA, L.A., red.

[Safety techniques in hydrological work] Tekhnika bez-
opasnosti pri gidrologicheskikh rabotakh. Leningrad,
Gidrometeoizdat, 1964. 170 p. (MIRA 17:9)

GOTLIB, Yakov L'vovich; ZAYMIN, Yevgeniy Yevgen'yevich; RAZZORENOV,
Fedor Fedorovich; TSEYTLIN, Boris Semenovich; CHEPELKINA,
L.A., red.

[Thermal properties of ice on the Angara River] Ledotermika
Angary. [By] IA.L.Gotlib i dr. Leningrad, Gidrometeoizdat,
1964. 196 p. (MIRA 17:6)

KHMALADZE, Grigoriy Nikolayevich; YEGIAZAROV, I.V., akademik,
retsenzent; LOPATIN, G.V., doktor geogr. nauk,
retsenzent; LISITSYNA, K.N., nauchn. sotr., retsenzent;
BOGOLYUBOVA, I.V., nauchn. sotr., retsenzent;
KHERKHEULIDZE, I.I., red.; CHEPELKINA, L.A., red.

[Suspended sediments of the rivers of the Armenian S.S.R.]
Vzveshennyye nasosy rek Armianskoi SSR. Leningrad, Gidro-
metecizdat, 1964. 245 p. (MIRA 17:9)

1. Laboratoriya nanosov Gosudarstvennogo gidrologicheskogo instituta (for Lisitsyna, Bogolyubova).

DEN SOV, Yeviy Mikhaylovich SHUL'TS, V.L., red.; CHEPELKINA,
L.A., red.

[System for calculating the discharge hydrograph of
mountain rivers] Skhema rascheta gidrografa stoka gor-
nykh rek. Leningrad, Gidrometeoizdat, 1965. 102 p.
(MIRA 18:7)

SHUL'TS, Viktor L'vovich; KOZIK, Ye.M., otv. red.; CHEPELKINA, L.A.,
red.

[Rivers of Central Asia] Reki Srednei Azii. Leningrad,
Gidrometeoizdat. Pts. 1 - 2. 1965. 691 p.
(MIRA 18:5)

GRUSHEVSKIY, M.S., red.; KUCHMENT, L.S., red.; CHEPELKINA, L.A.,
red.

[Electronic computers in hydrology; a collection of
translations] Elektronnye vychislitel'nye mashiny v gidro-
logii; sbornik perevodov. Leningrad, Gidrometeoizdat,
1965. 233 p. (MIRA 18:10)

BYKOV, Vasil'y Dmitriyevich; VASIL'YEV, Andrey Vasil'yevich;
CHEBOTAREV, A.I., otv. red.; CHEPELKINA, L.A., red.

[Hydrometry] Gidrometriya. Izd. 2., perer. i dop. Leningrad, Gidrometeoizdat, 1965. 498 p. (MIRA 19:1)

LIOZINA, Ye.M.; CHEPELOVA, M.A.; TARTAKOVSKAYA, B.E.

Volume of circulating blood in some diseases of the hemopoietic
organs; isotope method. Vest.rent. 1 rad. 31 no.5:21-26 S-0 '56.
(MLRA 10:1)

1. Iz otdela klinicheskoy gematologii i laboratorii izotopov (sav. -
prof. D.N.Yanovskiy) Ukrainskogo instituta klinicheskoy meditsiny
imeni akad. N.D.Strazhesko (dir. - prof. A.L.Mikhnev)
(BLOOD VOLUME, determ.
isotope method)

L 24667-65 EWT(1)/T/EEG(b)-2 IJP(o)/SSD(c)/AFWL/SSD/RAEM(a)/FSD(ga)/ESD(t)

ACCESSION NR: AP4046665

S/0185/64/009/009/1026/1026

16
15
10

AUTHOR: Zakharko, Ya. M.; Chepelyev, V. V.

TITLE: The asymmetry of the impulse amplitude (pulse-height) differential distribution curves of x-ray detectors of NaI(Tl) crystals. ↗

SOURCE: Ukrayins'kyh fizychny'y zhurnal, v. 9, no. 9, 1964, 1026

TOPIC TAGS: NaI(Tl) crystal, pulse height curve, x ray detector, scintillation detector, luminescence quenching center, photomultiplier, single channel scintillation spectrometer

ABSTRACT: NaI(Tl) crystals produced for operation as x-ray detectors in the 4-20 kev energy region were characterized by distinct asymmetry from the low amplitude side regardless of the light reflecting material or method of preparing the unit surface. Curve 1 of the enclosed figure shows a typical differential distribution of the impulse amplitude of a single channel scintillation spectrometer with an FEU-35 photomultiplier for CuK_α radiation. The parameters established after

Cord 1/3.

L 24667-65

ACCESSION NR: AP4046665

6-8 months at room temperature are shown in curve 2: the curve became symmetrical and its amplitude peak was increased. It was indicated that the observed effects are associated with nonuniformity of the scintillation due to uneven distribution of the activating additive or some unidentified luminescence quenching centers during the NaI(Tl) crystal growing process. Asymmetry from the high amplitude side described by J. T. Wright (J. Sci. Instrum., 31, 462, 1954) and by W. Hink (Zs. Phys., 169, 289, 1962), was not observed in these detectors except when the scintillation intensity was artificially reduced, e. g., by introduction of neutral filters between the container outlet window and the photomultiplier. Orig. art. has: 1 figure

ASSOCIATION: L'vis'ky*y derzhuniversy*tet im. I. Franka (Lvov State Institute)

SUBMITTED: 27Mar64

ENCL: 01

SUB CODE: NP, GC

NO REF SOV: 001

OTHER: 003

Card 2/3

L 24667-65

ACCESSION NR: AP4046665

ENCLOSURE:01

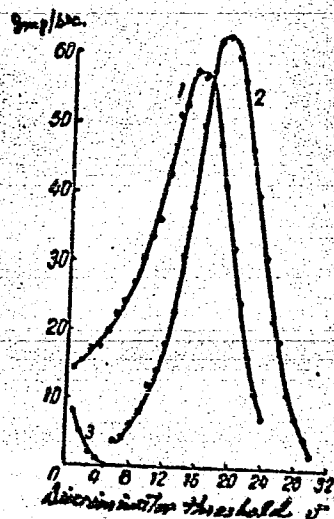


fig. 1

Effect of natural aging of crystal on the differential distribution of the impulse amplitude of the x-ray detector for CuK - radiation.

1. detector prepared from fresh monocrystal
2. the same detector after 6 months storage
3. distribution of FEU-35 noise

Card 1/1

L 28879-66 EWT(m)

ACC NR: AF6016047

SOURCE CODE: UR/0185/66/011/005/0538/0539

AUTHOR: Zakharko, Ya. M.; Chepelyev, V. V.

ORG: L'vov State University im. I. Franko (L'vivs'kyi derzhuniversitytet)

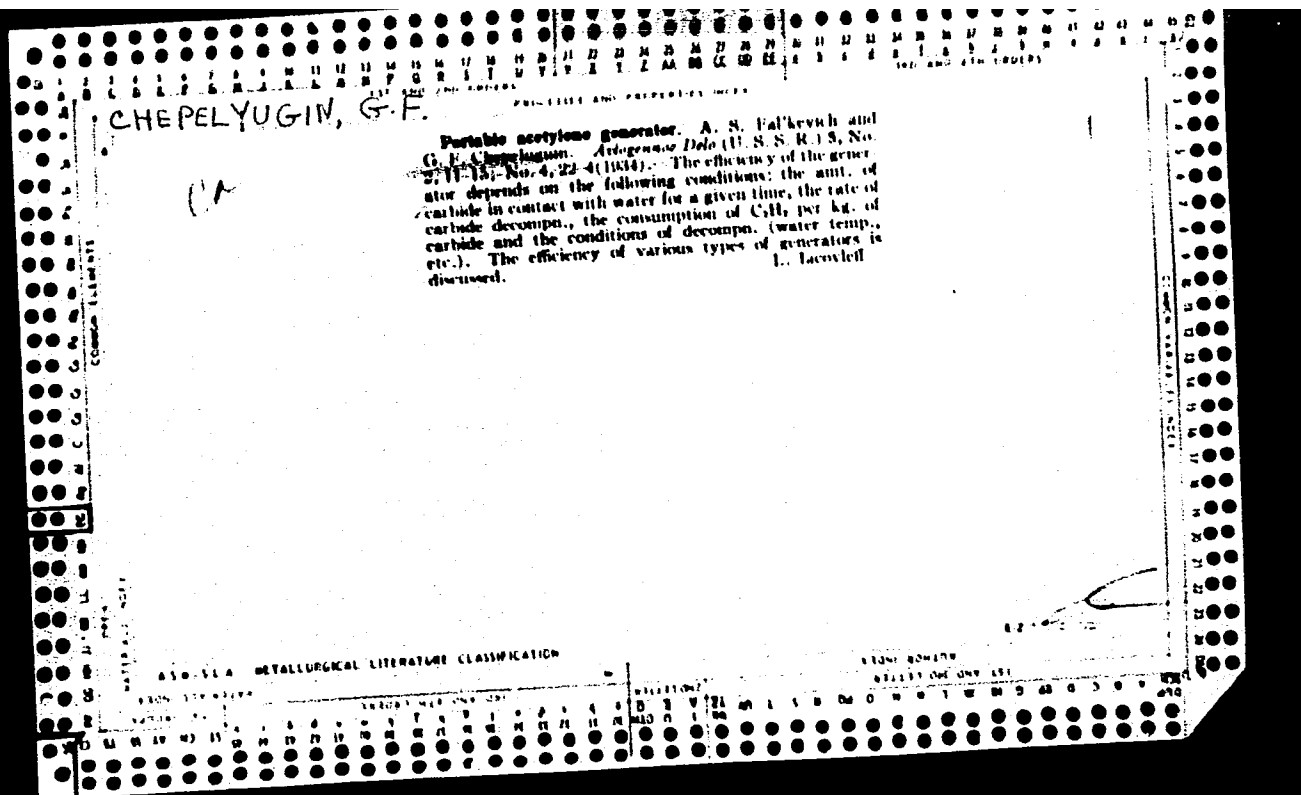
TITLE: Influence of x-irradiation on resolution of a NaI(Tl) scintillator

SOURCE: Ukrayins'kyi fizychnyy zhurnal, v. 11, no. 5, 1966, 538-539

TOPIC TAGS: x ray effect, scintillator, activated crystal, Gamma spectrometer

ABSTRACT: This is a continuation of earlier work (Izv. AN SSSR ser. fiz. v. 29, 78, 1965) where it was shown that exposure of NaI(Tl) crystals to hard x-irradiation affects adversely the resolution of γ spectrometers. The present article reports the experimentally obtained intrinsic resolution of irradiated crystals and the causes of this effect. Round plates of 30 mm diameter and 5 mm thickness cut from crystals grown by the Kiropoulos method were exposed to doses of 300-350 r/min. The resolution was measured with a scintillation spectrometer using an FEU-43 photomultiplier several hours after the irradiation, when the phosphorescence of the crystal completely stopped. Several weeks after a 60-minute irradiation the resolution was still 40-45% below the initial value, and the amplitude of the photopeak was ~15% below the initial value. The small decrease in the average photopeak amplitude, compared with the appreciable deterioration of the resolution, is the consequence of inhomogeneities of the optical yield, due to the different absorption of the x-rays in the scintillator material. This was confirmed by a separate test of the variation of the light yield with the thickness of the scintillating crystal. Orig. art. has: 1 figure and 1 table. [02]

Cord 1/1 SUB CODE: 20/ SUBM DATE: 30Jun65/ ORIG REF: 005/ ATD PRESS: 5005



CA
CHEPELYUGIN, G.F.

A comparative study of welding torches. A. S. Falke-
vich, G. F. Chepeyugin and V. P. Tret'yakov. *Argon-
now* (U.S.S.R.), No. 7, 7-11 (1935). The torches
under investigation proved to be of similar quality and were
characterized by the following features: the relation of
acetylene to O in the gas mixt. at 840°-1020°F. was
1.1-1.3; the velocity of the gas current leaving the torch
was 110-160 m./sec.; the pressure in the acetylene pipe
was 450-650 mm. L. Jacobell

ASB-51A METALLURGICAL LITERATURE CLASSIFICATION

21

CHEPELYUGIN, G. F.

The possibility of using pseudobutylene as a fuel in the cutting of metals. A. S. Fatkeev and G. F. Chepe-lyugin. *Avances. Dolo* 8, No. 10, 1730 (1957). *Chem. Zentr.* 1958, II, 2344 G. Pseudobutylene is suitable for use as a fuel for cutting steel pieces up to 300 mm. in thick-ness. It is comparable to C_2H_2 and benzine as regards cutting velocity and gas consumption. The combustion process is represented by the equation: $C_4H_8 + 2 O_2 = 4 CO + 4 H_2$ and $4 CO + 4 H_2 + 4 O_2 = 4 CO_2 + 4 H_2O$. The only adjustment required is an enlargement of diam. of the injector of the cutting tool. The operation is quick and the flame can be easily regulated, especially if the cyl-inder contg. the fuel is fitted with a reducing valve for regulating the pressure and is warmed with hot water (40-50°) in the winter. M. G. Moore

ASB-514 METALLURGICAL LITERATURE CLASSIFICATION

CHEPELYUGINA, M. F.

"The Role of Afferentation in Construction of Motive Actions," Thesis for degree of
Cand. Biological Sci. Sub 22 Jun 50, Acad Med Sci USSR

Summary 71, 4 Sept 52. Dissertations Presented for Degrees in Sci. and Engineering in
Moscow in 1950. From Vechernaya Moskva. Jan-Dec 1950.

SERDYUKOVA, M.; CHEPELYUK, B.

Projection camera for a trichinelloscope. Mias. ind. SSSR
32 no.3:47 '61. (MIRA 14:7)

1. Vinnitskaya myasokontrol'naya stantsiya No.1.
(Trichinella spiralis)
(Meat industry—Equipment and supplies)

CHEPILYUK, M.A.; BAKKER, E.T.

Q fever in Pavlodar Province; preliminary report. Zdrav.
Kazakh. 17 no.6:46 '57. (MIRA 12:6)

1. Iz Pavlodarskoy oblastnoy sanitarno-epidemiologicheskoy
stantsii.

(PAVLODAR PROVINCE--Q FEVER)

EXCERPTA MEDICA Sec 17 Vol 5/7 Public Health July 59

1894. MATERIAL FOR THE STUDY OF Q FEVER IN THE PROVINCE OF PAVLODARSK (Russian text) - Chepelyuk M. A. and Bekker E. T. Pavlodarsk Prov. Sanit.-Epidemiol. Station, Pavlodarsk, USSR - ZDRAVOOKHR. KAZ. 1958, 18/6 (20-23)

The diagnosis of Q fever was first confirmed in the region by positive CFT with patients' sera and the rickettsial antigen of *C. burnetii*. The incidence of Q fever is seasonal (March-June), probably owing to seasonal frequencies of exposure to ticks, among which *Dermacentor marginatus* was found to be naturally infected with Q fever rickettsiae. Q fever was highly prevalent among meat-packers, and of low incidence in dairy personnel.

Anigstein - Galveston, Tex. (L, 17)

CHEPELYUK, M.A.; BEKKER, E.T.

Bacteriological characteristics of brucellosis in Pavlodar Province. Zdrav. Kazakh. 22 no.9:58-61 '62.

(MIRA 17:2)

1. Iz Pavlodarskoy oblastnoy sanitarno-epidemiologicheskoy stantsii.

Subject : USSR/Aeronautics - tactics AID P - 4591
Card 1/1 Pub. 135 - 3/23
Author : Chepelyuk, S. G., Hero of Sov. Union, Candid. in Mil.
Sci., Guards Maj.
Title : Destruction of ground targets from hedge-hopping flight
Periodical : Vest. vozd. flota, 3, 19-25, Mr 1956
Abstract : The advantages of hedgehopping attacks and the methods
of carrying out of such attacks are described in detail.
Two sketches. The article is of informative value.
Institution : None
Submitted : No date

BOGDANOVICH, B.N.; CHEPENETS, E.N.

Ligation of the main bronchus in pulmonary tuberculosis.
Probl. tub. 40 no.6:103-104 '62 (MIRA 16:12)

1. Iz tuberkuleznogo sanatoriya "Sosnovka" Ministerstva zdравo-okhraneniya BSSR (glavnyy vrach T.S.Shatkovskaya).

CHERENETS, E.N.

Case of bronchogenic cyst of the mediastinum and tuberculoma
of the lung. Prob. tub. no.1:85-86 '65. (SERA 18:12)

1. Torakal'noye otdeleniye legochnogo tuberkuleza 4-go protivotu-
berkuleznogo dispansera (glavnyy vrach G.I. Luchev), Chelyabinsk.

DEVIATNIN, V.A.; MEL'NIKOVA, Ye.Ya.; CHEPENKO, A.I.

Vitamin A, B₁ and B₂ content of cheese. Trudy VNIIV 6:240-242
'59. (MIRA 13:7)

1. Vsesoyuznyy nauchno-issledovatel'skiy vitaminnyy institut.
Khimiko-analiticheskaya laboratoriya.
(CHEESE) (VITAMINS)

AKCHURINA, R.M.; CHEPENKO, N.K.

Advanced method for painting and drying metal containers in
lacquer and paint factories. Lakokras. mat. i ikh prim. no.6:
64-66 '61. (MIRA 15:3)
(Painting, Industrial--Equipment and supplies)

MAYZEL', Boris Isaakovich; OKUN', Boris TSalerovich CHEPENKO,
Nata Konstantinovna; EFROS, M.M., red.

[Use of the combustion products of natural gas in convec-
tion drying chambers for drying protective paint coatings]
Konveksionnye sushil'nye kamery s ispol'zovaniem produktov
sgoraniia prirodnogo gaza dlia sushki lakokrasochnykh po-
krytii. Leningrad, 1965. 25 p. (MIRA 18:7)

OKHATSIMSKAYA, M., RASTRUSIN, Y., RAKITYANSKY, I., CHEPETNOV, R.

"Laws of excitation of short-period oscillations in middle latitudes."

report presented at the Intl. Association of Geomagnetism and Aeronomy, Symposium on Rapid Geomagnetic Variations, Utrecht, Netherlands, 1-4 Sep 59.

CHEPIAKOV, V.D.

Case of degeneration of myositis ossificans into a malignant tumors.

Khirurgia 35 no. 11:122-123 N '59.

(MIRA 14:1)

(MUSCLES—DISEASES) (LEGS—CANCER)

18 3100

25425

S/137/61/000/006/019/092
A005/A101

AUTHORS: Tsyb, P.P., Getskin, L.S., Vartanyan, A.M., Fel'dman, V.G., Anosova, T.V., Akylbekov, A.A., Levina, A.A., Chepick, M.N.

TITLE: Extracting indium from lead plant dusts

PERIODICAL: Referativnyy zhurnal. Metallurgiya, no. 6, 1961, 19, abstract 60166
("Sb. nauchn. tr. Vses. n.-i. gornometallurg. in-t tsvetn. met",
1960, no. 6, 377 - 388)

TEXT: The authors describe a technological system of In extraction from
dusts of lead production, using the method of dust sulfatizing at the beginning
of the process. Extraction of In into 1-st class metal from the content in the
initial dust (In 0.003 - 0.006%) is about 60%.

G. Svodtseva

[Abstracter's note: Complete translation]

Card 1/1

L 46243-66 EWT(m)/EWP(e)/EWP(t)/ETI IJP(c) JD/HW

ACC NR: AP6023916

SOURCE CODE: UR/0363/66/002/007/1218/1224

AUTHOR: Kuz'ma, Yu. B.; Chepiga, M. V.; Flakhina, A. M.

45B

ORG: L'vov State University im. Iv. Franko (L'vovskiy gosudarstvennyy universitet)

TITLE: Phase equilibria in the systems Cr-Co-B, Mn-Fe-B, and Mn-Co-B

27 27 27 27 27

SOURCE: AN SSSR. Izv. Neorg materialy, v. 2, no. 7, 1966, 1218-1224

TOPIC TAGS: phase equilibrium, metal phase system, chromium compound, boron compound, iron compound, manganese compound, cobalt compound

ABSTRACT: The study constitutes a part of systematic investigations being carried out in the Inorganic Chemistry Department of L'vov University (Kafedra neorganicheskoy khimii L'vovskogo universiteta), concerned with the phase diagrams of ternary systems of two transition metals with boron and the crystal structures of the ternary compounds formed. The systems Cr-Co-B, Mn-Fe-B, and Mn-Co-B were studied by x-ray structural analysis and in part by microstructural analysis, and the isothermal sections of these systems at 800°C were plotted. The compound $\text{Cr}_2\text{Co}_{21}\text{B}_6$ (τ phase), having a $\text{W}_2\text{Cr}_{21}\text{C}_6$ -type structure ($a = 10.471 \text{ \AA}$), exists in the Cr-Co-B system. The boride Co_2B dissolves up to 30 at. % Cr. The presence of continuous solid solutions $(\text{Mn}, \text{Fe})_2\text{B}$ and $(\text{Mn}, \text{Fe})\text{B}$ was confirmed in the Mn-Fe-B system. A ternary compound (τ phase) with a $\text{W}_2\text{Cr}_{21}\text{C}_6$ -type structure ($a = 10.518-10.641 \text{ \AA}$) is formed in the Mn-Co-B system; the

Card 1/2

UDC: 541.123.3

L 46243-66

ACC NR: AP6023916

0

region of homogeneity of the τ phase is located between 10 and 40 at. % Mn. The existence of $(\text{Mn}, \text{Co})_2\text{B}$ and $(\text{Mn}, \text{Co})\text{B}$ solid solutions was confirmed. Orig. art. has: 5 figures and 2 tables.

SUB CODE: 11/ SUBM DATE: 14Oct65/ ORIG REF: 010 / OTH REF: 004

hs

Card 2/2

CHEPIGA, Ya.																									
1ST AND 2ND ORDERS													PROCESSING AND PROPERTIES INDEX												
ca 27 Single pressing of castor beans in the 2000' screw press. Ya. Chepiga. <i>Masloborn-Zhurnal</i> Prom. 10, No. 5-6, 10-21 (1940). Oil can be expressed from pretreated castor beans by a single pressing in a screw press, leaving only 7-7.5% oil in the cake if the screw speed is kept down to 9 r. p. m. The pretreatment consists in drying the beans to 3.5-5% moisture and steaming at 101-104°. The ex- pressed oil, when filtered, fully meets quality specifica- tions. Julian F. Smith																									
ASB-5LA METALLURGICAL LITERATURE CLASSIFICATION 1ST AND 2ND ORDERS PROCESSING AND PROPERTIES INDEX																									

USSR/ Engineering ~ Cast-iron shafts

Card : 1/1

Authors : Ceredenko, B. N., Cand. Tech. Sc.; Chepigin, G. V., Can. Tech. Sc.

Title : Resistance to wear and dependability in operation of crankshafts of high-strength cast iron.

Periodical : Vest. Mash., 34, Ed. 6, 65 - 68, June 1954

Abstract : Numerical values are given for the resistance of high-strength cast iron to various kinds of strains. An analysis is made of the results obtained by experimenting both with steel and high-strength cast iron for tractor engine crankshafts, and such experiments lead to the conclusion that the high-strength cast iron may be used as a substitute for steel in crankshafts and other parts. Graphs; tables; drawing; illustration.

Institution : ...

Submitted : ...

CHEFIGIN, G.V., kand.tekhn.nauk; GUL', N.S., inzh.; CHIZHOV, A.P., inzh.
KHESIN, A.Ya.

Results of the operational tests of a full-flow RMTs device on the
SMD diesel engine. Trakt. 1 sel'khoz mash. 32 no.6:12-14 Je '62.
(MIRA 15:6)

1. Dnepropetrovskiy sel'skokhozyaystvennyy institut (for Chepigin,
Gul', Chizhov). 2. Gosudarstvennoye spetsial'noye konstruktorskoye
byuro po dvigatelayam (for Khesin).
(Tractors—Oil filters)

CHEPIGIN, G. V., inzh.; NEKHAY, S. M., inzh.; GUL', N. S., inzh.;
CHIZHOV, A. P., inzh.

Replacing the double-cleaning oil filter with a full-flow
centrifuge. Mashinostroenie no.5:95 S-O '62.
(MIRA 16:1)

(Tractors—Engines—Oil filters)

CHEPIGIN, G.V., kand. tekhn. nauk; GUL', N.S., inzh.; CHIZHOV, A.P., inzh.

Use of cast-iron crankshafts in motor-vehicle and tractor engines.
Mashinostroenie no.5:112-113 S-0 '63. (MIRA 16:12)

1. Dnepropetrovskiy sel'skokhozyaystvennyy institut.

FINKEL'SHTEYN, David Naumovich; RYABCHIKOV, D.I., otv. red.;
CHEPIGO, K.V., red.

[Pure substance] Chistoe veshchestvo. Moskva, Nauka,
1965. 167 p. (MIRA 19:1)

1. Chlen-korrespondent AN SSSR (for Ryabchikov).

L 35912-66 EWT(m)/EWP(j) RM

ACC NR: AP6014893

SOURCE CODE: UR/0076/65/039/012/2951/2957

AUTHOR: Nurmukhametov, R. N.; Chepigo, O. S.; Shvayka, O. P.ORG: Moscow Physico-chemical Scientific Research Institute im. L. Ya. Kerpov (Moskovskiy nauchno-issledovatel'skiy fiziko-khimicheskiy institut)TITLE: The structural luminescence and absorption spectra of solutions of aryloxydiazoles and some aryl ethylenes at 77°K

SOURCE: Zhurnal fizicheskoy khimii, v. 39, no. 12, 1965, 2951-2957

TOPIC TAGS: absorption spectrum, luminescence spectrum, ethylene

ABSTRACT: The spectra were taken by the Shpol'skiy method in n-hydrocarbons and methyl cyclohexane at 77°K on a Hilger spectrometer. The average concentration of the solutions was approximately 10^{-4} moles/liter. The spectra obtained are exhibited in a number of figures. It was established that a majority of the compounds studied exhibited only fluorescence, while phosphorescence was absent. The structure of the spectra has a periodic form. The article interprets other vibrational frequencies by analogy with known interpretations of the spectra of aryl ethylenes. In compounds with the general structure

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L 35912-66

ACC NR: AP6014893

R_1 --X-- R_2 (where R_1 and R_2 are aryl nuclei and the X group is either ethylene or oxydiazole) there is observed an identical nature of the π -bond, which presupposes an identical mechanism for the formation of the vibrational structure of the bands of these compounds. Orig. art. has: 2 figures and 2 tables.

SUB CODE: 07, 20/ SUBM DATE: 01Aug64/ ORIG REF: 011/ OTH REF: 003

card 2/2 *llb*

CHEPIGO, S. V.

Chepigo, S. V. "For the development and the complete utilization of planned capacities in hydrolysis plants", *Gidroliz. prom-st'* SSSR, 1948, No. 6, p. 1-4.

So: U-3261, 10 April 53, (*Letopis 'Zhurnal 'nykh Statey*, No. 12, 1949).

CHEPIG-P, S.V.

✓ Production of charcoal from hydrolytic lignin. S. V. (2)
 Chernov, I. N., Zaitseva, and A. A. Danilova. *Gidroliz. Lignosulfon. Prem.* 8, No. 3, 12-14 (1955).—Charcoal (I) of high discoloration power (II) can be obtained from various sources (cottonseed and sunflower seed hulls, corn husks, straw, sawdust, lignite) by treatment with concd. H_2SO_4 . The quality of I depends on the amt. and concn. of acid, time and temp. of the dehydration reaction, the fineness and moisture content of the raw material, and the degree of purification. The optimum II is reached at 250% of H_2SO_4 , based on the dry lignin. The yield depends on the temp. of the reaction mass, the time of reaction, and the concn. of acid. An acid with the sp. gr. of less than 1.75 gives a product of inferior quality. The optimum temp. for the formation of I corresponds to the max. temp. evolved during dehydration (160-180°). Keeping the reaction temp. at 170-180° for more than 10 min. lowers the yield of I, and prolonged heating (30 min.) diminishes II. The dimension of the particle of the starting material does not affect II significantly. The temp. of the wash water and the amt. of washing do not lower II. The equipment for a continuous process consists of a reaction drum with a shaft mixer, a filter tank, and storage for the raw material and products. T. Jurek

CHEPIGO, S. V.

USSR/Chemical Technology - Chemical Products and Their Application. Wood Chemistry Products. Cellulose and Its Manufacture. Paper, I-23

Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 63364

Author: Chepigo, S. V.

Institution: None

Title: Production of Activated Charcoal - Collactivite from Vegetable Materials

Original

Periodical: Tr. in-ta lesokhoz. problem AN Latv. SSR, 1955, 8, 69-79

Abstract: Collactivite (K) is obtained on treating vegetable materials with H_2SO_4 . In comparison with activated charcoals (AC) it contains less carbon (63% in lieu of 88-98%) and more oxygen and hydrogen. Adsorption characteristics of K are higher than those of AC. In addition to adsorption, due to the presence in K of active functional groups ($COOH$, OH , SO_3 , CO) there take place chemisorption and ion-exchange reactions. It is recommended to store K in moist condition since drying lowers the adsorptive properties of K.

Card 1/1

CHEPIGO, S.V.

Reduction in losses in the manufacture of xylose. N. A. Vasyunina, S. V. Chepigo, and S. V. Pavlov. *Gidroliz. i Lesokhin. Prom.* 9, No. 4, 15-17:1950. -- In order to increase the amt. of xylose, obtained by hydrolysis of cottonseed hulls, each step of the process was carefully studied and recommendations made as follows: (a) more effective mixing of the hydrolyzed mass during the addn. of accurately measured milk of lime; (b) improved filter presses; (c) continuous evapn. of the xylose sirup; (d) a more suitable ion-exchange purifying bed, etc. Elisabeth Barabash

All Union Sci Res. Inst. Hydrolysis & Sulphur.
Industry.