

440029/27

(A)

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21

AUTHOR: Dobryakova, G. V.; Panchenko, I. Ya.; Povalyayov, A. P.

ORG: nono

TITLE: Ratio of strontium to calcium upon passage from rations into dog skeletons

SOURCE: Radiobiologiya, v. 6, no. 4, 1966, 625-627

TOPIC TAGS: dog, isotope, calcium, strontium, biologic metabolism, food ration, bone mineral metabolism had been raised to a slightly positive calcium balance. The dogs were fed bone meal from animals who had received strontium-90 for a long time. Doses for groups 1 and 2 were  $4.4 \times 10^{-9}$  and  $4.4 \times 10^{-8}$  curie and 1 and 10 g calcium respectively. The animals were sacrificed after 30, 90 and 180 days and samples of tissues, urine and feces were studied radiometrically and chemically. In group 2, the absolute strontium-90 accumulation in bone was only 2 fold that of group 1 although its content in the ration was 10 fold. Calcium and strontium absorption in group 2 was 1.5-2 fold that of group 1. No statistically valid difference was found for the calcium content of bones from the animals of groups 1 and 2. The same applied to soft tissues and blood. Results show that the uptake of calcium and strontium from the gastrointestinal tract and their accumulation in the

Card 1/2

ANISIMOV, A.A.; FUZINA, Ye.K.; DOBRYAKOVA, L.A.; LIKHOVIDOVA, Ye.V.

Diurnal periodicity of the translocation of assimilates. Dokl.  
AN SSSR 146 no.6:1441-1444 0 '62. (MIRA 15:10)

1. Gor'kovskiy gosudarstvennyy universitet im. N.I. Lobachevskogo.  
Predstavлено академиком А.Л. Курсановым.  
(Plants—Assimilation)

ANISIMOV, A.A.; DUBOVSKAYA, T.S.; DOERYAKOVA, L.A.

Effect of nitrogen and phosphorus nutrition of wheat on the incorporation of C<sup>14</sup> into assimilates and their translocation.  
Fiziol. rast. 11 no.5:793-799 S-O '64. (MIRA 17:10)

1. Gorky State University.

DOBRYAKOVA, L.I.

VELIKOVSKAYA, Ye.M.; VELIKOVSKIY, D.S.; PEGANOV, A.A.; DOBRYAKOVA, L.I.;  
KUROCHKINA, Z.V.; LISOVSKIY, I.I.

Synthetic drying oils. Patent U.S.S.R. 77,050, Dec. 31, 1949.  
(CA 47 no.19:10244 '53)

DOBRYAKOVA, L. I., Cand Tech Sci -- (diss) "Study of the Properties of Limestone Treated with Silicofluorides and Certain Silico-<sup>to</sup> Organic Compounds ~~for~~ Increase ~~of~~ Its Durability." Mos, 1957.  
21 pp (Acad of Construction and Architecture USSR, Sci Res Inst of New Construction Materials, Finishing and Outfitting of Buildings), 150 copies (KL, 49-57, 113)

KRESTOV, M.A.; DOBRYAKOVA, L.I.; KOSHKIN, V.G.; YEVDOKIMOV, A.A.;  
IVANOVA, V.V.; KHMELEVSKIY, V.A.; KOSTOCHKINA, T.V.; PFLAUMER,  
O.E., kand.tekhn.nauk, nauchnyy red.; SEVORTSOVA, I.P., red.  
izd-va; TEMKINA, Ye.L., tekhn.red.

[Finishing large panels and blocks using colored concretes]  
Otdelka krupnykh panelei i blokov s primeneniem tavetnykh beto-  
nov. Moskva, Gos.izd-vo lit-ry po stroit., arkhit. i stroit.  
materialam, 1959. 87 p. (MIRA 13:3)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut novykh stroi-  
tel'nykh materialov. 2. Institut novykh stroitel'nykh materialov  
(for Krestov, Dobryakova, Kosshkin, Yevdokimov, Ivanova, Khmelevskiy).  
3. Institut betona i zhelezobetona (for Kostochkina).  
(Building blocks)

KRESTOV, M.A., kand. arkh.; MAKOTINSKIY, M.P., kand. arkh.; TSILLI,  
L.B., kand. arkh.; Prinimali uchastiye: BOGUSLAVSKIY, A.I.,  
inzh.; DOBRYAKOVA, L.I., kand. tekhn. nauk; LIVSHITS, A.M.,  
inzh.; MUNTS, V.O., kand. arkh.; L'VOV, G.N., inzh., retzen-  
zent; POPOV, A.N., retsenzent; GURVICH, E.A., red.izd-va;  
TEMKINA, Ye.L., tekhn. red.

[Catalog of finishing materials and elements] Katalog otde-  
lochnykh materialov i izdelii. Moskva, Gosstrooiizdat.  
Pt.6.[Concrete and mortars] Betony i rastvory. 1962. 46 p.  
(MIRA 16:8)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut novykh  
stroitel'nykh materialov. 2. Deystvitel'nyy chlen Akademii  
stroitel'stva i arkhitektury SSSR (for Popov).  
(Finishes and finishing)

DOBRYAKOVA, Lyudmila Ivanovna, kand. tekhn. nauk; YEVDOKIMOV,  
Aleksey Aleksandrovich, inzh.; LOPOVOK, Lev Isayevich,  
kand. arkhitektury; MILOVZOROV, Aleksey Konstantinovich,  
arkh.; ORLOV, Aleksandr Mikhaylovich, kand. tekhn. nauk;  
KHMELEVSKIY, Vladimir Aleksandrovich, arkh.; GLEZAROVA,  
I.L., red.; BOROVNEV, N.K., tekhn. red.

[Industrial finishing of buildings] Industrial'naia ot-  
delka zdaniii. Moskva, Gosstroizdat, 1963. 106 p.  
(MIRA 16:11)

(Buildings—Finishing)

DOBRYAKOVA, N. S.

Chemical Abstracts.  
Vol. 48 No. 5  
Mar. 10, 1954  
Biological Chemistry

Action of antimony trioxide on the organism. I. D. Gadaskin, N. S. Dobryakova, I. F. Kreps, E. I. Lyublina, and Z. N. Favortseva, Research Inst. Inst. and Professional Hyg., Leningrad. *Gigiena i Sanit.* 1953, No. 10, 23-7. Expts. with rabbits and observations on human cases show that  $Sb_2O_3$  is a toxic substance whose concn. in the atm. cannot exceed thousandths of mg. per/l. or less. Toxic effects are evident after prolonged inhalation of air contg. hundredths of mg./l. Skin deformations are among the symptoms of intoxication. G. M. Kosolapoff.

DOBRYAKOVA, N.S.

LAZAREV, N.V.; ALEKSANDROV, I.S.; LYUBLINA, Ye.I.; AKKERBERG, I.I.; ZAKA-  
BUNINA, M.S.; GADASKINA, I.D.; DOBRYAKOVA, N.S.; KREPS, I.F.; KARASIK,  
V.M.; LEVINA, E.N.; DANISHEVSKIY, S.L.; YEGOROV, H.M.; RYLOVA, M.L.,  
starshiy nauchnyy sotrudnik; KARPOV, B.D.; ANDREYEV, V.V.; LYKHINA,  
Ye.T.; ZAMESHAYEVA, G.I.; ANISIMOV, A.N.; FRIDLYAND, I.G.; DANETSKAYA,  
O.L.; BOGOVSKIY, P.A.; TIUNOV, L.A.; MIKHEL'SON, M.Ya.; ABRAMOVA, Eh.I.,  
GEIGOR'YEVA, L.M.; KLINSKAYA, K.S.

Third Leningrad conference on the problems of industrial toxicology.

Farm. i toks. 16 no.2:59-62 Mr-Ap '53.

(MLRA 6:6)

(Poisons)

DOBRYAKOVA, N.S.

Effective illumination of working areas of relay adjusters.  
Dig. 1 san. 24 no.3:37-41 Mr '59. (MIRA 12:5)

1. Iz sanitarno-epidemiologicheskoy stantsii Oktyabr'skoy  
zheleznay dorogi.

(ILLUMINATION,  
of indust. relay-control areas (Rus))

(INDUSTRY AND OCCUPATIONS,  
illumination of relay-control areas (Rus))

*DOBRYAKOVA, N.Ye.*

STARIKOVA, Ye.V.; DOBRYAKOVA, N.Ye.; KOROBKO, V.A.; AL'TMAN, A.A.;  
ROMANOVA, N.V., vedushchiy redaktor; POLOSINA, A.S., tekhnicheskiy  
redaktor

[Methods of testing petroleum products] Metody ispytaniia nefte-  
produktov. Moskva, Gos. nauchno-tekhn. izd-vo neftianoi i gorno-  
toplivnoi lit-ry, 1953. 389 p. [Microfilm] (MLRA 7:9)  
(Petroleum products--Testing)

S/065/61/000/002/008/008  
E194/E284

AUTHOR: Dobryakova, N. Ye.

TITLE: A Conference on the Design of Lubricating Grease Plants of Increased Output Using Improved Methods of Manufacture

PERIODICAL: Khimiya i tekhnologiya topliv i masel, 1961, No. 2,  
pp. 71-72

TEXT: A conference of the above title was held from September 26 to October 1, 1960 in Lvov, it was organised by the GNTK of the Council of Ministers of the RSFSR and the GNTK of the Council of Ministers of the UkrSSR. The conference was attended by representatives of GOSPLAN USSR and UkrSSR, the GNTK of the USSR, the Lvov Council of National Economy, GLAVNEFTESNAB, the Experimental-Council of Ministers of the RSFSR, TsP NTO UkrSSR, the Scientific Industrial Production Trust 'Neftemaslozavody', Scientific Research and Design Institutes, the Academy of Sciences of the UkrSSR, Grease Plants and others. More than 100 persons attended the conference and 12 reports were read of which the main ones were as follows: Recent investigations and trends in development and manufacture of greases in the USSR and abroad. The design  
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S/065/61/000/002/008/008  
E194/E284

A Conference on the Design of Lubricating Grease Plants of Increased Output Using Improved Methods of Manufacture

of high output grease plants and the reconstruction of existing plants. Scientific research and experimental work on new methods of grease manufacture. The design of equipment for new methods of grease manufacture. Automation and mechanization of laborious processes in grease manufacture. Methods of handling and packing greases produced in large and small quantities. Experience of a number of works in grease manufacture including the Berdyansk and Moscow Neftegaz Plants. It was noted that a great deal of work had been done on the development and supply of greases but reconstruction of grease plant was being hindered by the absence of the necessary equipment and instruments. Raw materials for grease manufacture was not yet standardized which impairs the quality and gives rise to production troubles. Not enough research is being done on new methods of manufacture, apparatus and equipment. In the АННИИ НП(All-Union Scientific Research Institute of the Petroleum Industry) the development of continuous methods of manufacture of lithium greases with the use of homogenisers is only

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S/065/61/000/002/008/008  
E194/E284

A Conference on the Design of Lubricating Grease Plants of Increased Output Using Improved Methods of Manufacture

in the laboratory stage and so no data is available for the design of full-scale plants. Mechanization and automation of packing and handling is inadequate, particularly in the case of packages up to 1 kg. The following measures were recommended. In the current 7-year plan to get out designs for grease plants ranging from 5 to 50 000 tons per year output using batch and continuous processes. Recommend GOSPLAN RSFSR to obtain modern imported equipment both for grease manufacture and laboratory testing. New methods of manufacturing lithium based greases should be developed and introduced including the application of ultrasonics. In Ukrniiprojekte in 1960-1961 Scientific Research Laboratory with a staff of 20-25 should be instituted to develop methods of manufacturing large tonnages of grease. The All-Urion Research Institute of the Petroleum Industry should, in 1961, intensify research work on new greases and in the first place waterless calcium greases, and sodium greases containing sodium nitrite. Antioxidant, anti-wear and anti-corrosion additives should be developed for greases. Raw materials for greases should be standardized.

Card 3/3

MALOLETKOV, Ye.K., inzh.; KRASAVIN, I.A., inzh.; DOBRYAKOVA, Ye.M.,  
tekhnik

[Method of estimating the operational qualities of single-  
bucket construction excavators while designing them] Meto-  
dika otsenki ekspluatatsionnykh kachestv odnokovshovykh  
stroitel'nykh ekskavatorov pri proektirovani. Moskva, Gos-  
stroizdat, 1964. 36 p. (MIRA 17:7)

1. Moscow. Nauchno-issledovatel'skiy institut organizatsii,  
mekhanizatsii i tekhnicheskoy pomoshchi stroitel'stva.

M.  
DOBRYANSKAYA, Ye.; SOTNIK, A.

Driving through an incline by using deep blast holes. Mast.ugl.2 no.11:17-18  
N '53. (MLRA 6:11)

1. Nauchnyy sotrudnik DenUGI (for both). (Coal mines and mining) (Blasting)

Contractor  
Coworker  
СОТРУДНИК

DOBRYANSKAYA, Ye.M., inzhener.

Experience with multiple shift work in combine-equipped mines. Mekh.trud.  
rab. 7 no.8:28-32 Ag '53.

(MLRA 6:8)  
(Coal-mining machinery)

USSR/Mining

Card 1/1

Authors : Dobryanskaya, E. M.

Title : Leading Shaft Sinking Cadres of Donets Coal Fields

Periodical : Mekh. Trud. Rab. Ed 3, 13 - 16, Apr - May 1954

Abstract : Studies conducted by Donets Coal scientific-investigational institute on problems involving the productivity and efficiency of individual working cadres employed in the coal shaft sinking operations. The compiled graphs and tables indicate the productivity of each individual cadre, time consumed during the specific operation, type of operation, and the methods and machinery used during shaft sinking. Tables; graphs.

Institution : ....

Submitted : ....

RUBINSKIY, Yu.M., kand.ekon.nauk; DOBRYANSKAYA, Ye.M., kand.tekhn.nauk

Tasks in revising technical work norms in coal mines. Ugol'  
Ukr. 3 no.4:39-41 Ap '59. (MIRA 12:?)

1. Dnepropetrovskiy gornyy institut (for Rubinskiy). 2. Donetskiy  
ugol'nyy institut (for Dobryanskaya).  
(Coal mines and mining)

DOBRYANSKAYA, Ye.M., kand.tekhn.nauk; BIRENEBERG, B.M.; SERDYUK, A.I.

Causes of actual labor productivity being less than that predicted  
in mines of the Donetskugol' Combine. Sbor. DonUGI no.28:50-79  
'62.

(MIRA 16:8)  
(Donets Basin--Coal mines and mining—Labor productivity)

DOBRYANSKAYA, Ye.M., kand.tekhn.nauk; BIRENBERG, B.M., gornyy inzh.;  
SERDIUK, A.I., gornyy inzh.

Effect of individual factors on the labor productivity and  
coal production costs in coal mines; "collection of articles.  
Reviewed by E.M. Dobrianskaia, B.M. Birenberg, A.I. Serdiuk.  
Ugol' 37 no.9:62-63 S '62. (MIRA 15:9)

(Coal mines and mining--Costs)  
(Coal mines and mining—Labor productivity)

DOBRYANSKAYA, Ye. M., kand. tekhn. nauk

Reasons for the decline of actual labor productivity from the  
level specified by the plans for mine reorganization. Sber.  
DonUGI no.32:39-56 '63.  
(NRA 17:10)

BRAUN, M.P., prof.; KOSTYRKO, O.S.; DOBRYANSKAYA, Ye.P.; KONDRAZHEV, A.I.

Efficient heat treatment process for hot rolling mill rolls.  
Izv.vys.ucheb.zav.; chern.met. 2 no.8:105-112 Ag '59.  
(MIRA 13:4)

1. Ukrainskaya Akademiya sel'skokhozyaystvennykh nauk.  
(Rolls(Iron mills)) (Steel--Heat treatment)

18.7100

77596  
SOV/129-60-2-9/13

AUTHORS: Braun, M. P. (Professor, Doctor of Technical Sciences),  
Kostyrko, O. S., Dobryanskaya, Ye. P., Kondrashev, A.  
I. (Engineers)

TITLE: Rational Heat Treatment Rates for Hot Rolling Rolls

PERIODICAL: Metallovedeniye i termicheskaya obrabotka metallov,  
1960, Nr 2, pp 48-52 (USSR)

ABSTRACT: At Novo-Kramatorskiy Plant (Novo-Kramatorskiy zavod) in Kramatorsk protracted heat treatment of hot rolling rolls failed to remove flakes. In order to study the effect of cooling rates on flake formation after forging 55Kh-steel specimens, the authors tested four different heat treatment methods (see Fig. 2).

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Rational Heat Treatment Rates for Hot  
Rolling Rolls

77596  
SOV/129-60-2-9/13

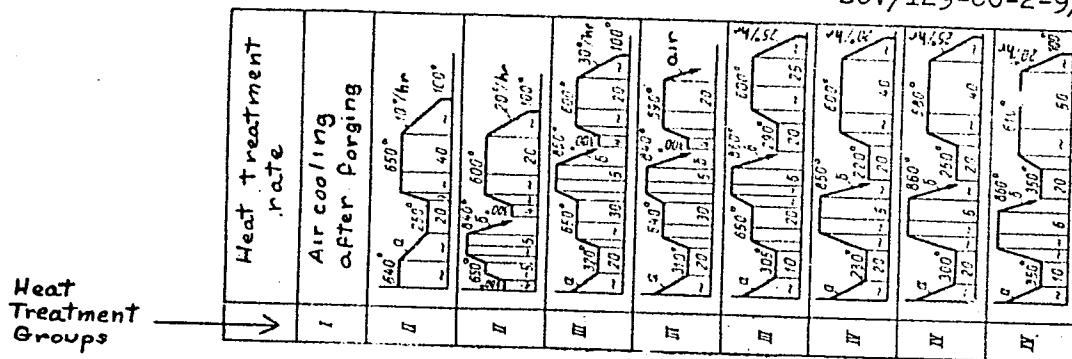


Fig. 2. Experimental rates. Cooling (a) with furnace and (b) in air.

Specimens of different weight were taken from ingots used for the production of rolls. Specimens as well

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## Rational Heat Treatment Rates for Hot Rolling Rolls

77596  
SOV/129-60-2-9/13

as rolls were forged with the same degree of reduction. Tensile tests allowed the observations of hardness changes along the cross section of specimens. Flakes were detected by means of magnetic defectoscope. Table 1 shows data relating to weight and chemical composition of specimens.

Key to Table 1: (A) Heat treatment group; (B) ingot weight in tons; (C) specimen weight in tons; (D) contents of elements in %.

(A)	(B)	(C)	(D)		
			C	Mn	Cr
I	42	6.2	0.56	0.54	1.20
II	42	6.4	0.57	0.37	1.12
II	42	6.4	0.57	0.37	1.12
III	32	7.5	0.52	0.54	1.17
III	32	7.3	0.52	0.54	1.17
III	36	36	0.56	0.54	1.20
IV	42	6.4	0.57	0.37	1.12
IV	42	7.3	0.56	0.54	1.20
IV	32	8.2	0.56	0.55	1.33

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Note: Si--0.26 to 0.32%; S--0.020 to 0.33%; P--0.016 to 0.025%.

Rational Heat Treatment Rates for Hot  
Rolling Rolls

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For a complete analysis of test results, the authors calculated the amount of H escaping from a forging with 1,000 mm diam at various temperatures of iso-thermal holding. The period during which H escaped was calculated according to a formula by N. M. Chuyko (see Ref 1 Stal', 1951, Nr 3). The authors estimated that 100 g 55Kh-steel contains 8 cm<sup>3</sup> H and maximum 4 cm<sup>3</sup> H after heat treatment. Calculations showed that H is liberated slowly from large forgings during austempering. Most flakes were identified in air-cooled forgings and a minimum number or none in specimens heat-treated according to method IV with the following characteristics:

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Rational Heat Treatment Rates for Hot  
Rolling Rolls77596  
SOV/129-60-2-9/13

Key to Table 2. (a) Heat treatment group; (b) specimens taken from; (c) tensile strength, kg/mm<sup>2</sup>; (d) yield point, kg/mm<sup>2</sup>; (e) elongation; (f) reduction of area; (g) impact strength, kgm/cm<sup>2</sup>; (h) mean, kg/mm<sup>2</sup>; (i) number of flakes; (j) surface; (k) 1/3 radius; (l) 2/3 radius; (m) center part.

(a)	(b)	(c)	(d)	(e) %	(f) %	(g)	(h)	(i)
IV	(j)	84,4	41,1	18	26,7	2,6		
	(k)	82,1	42,3	16,2	23,4	1,9	228-241	75
	(l)	83,4	39,7	13,4	24,8	2,1		
	(m)	81,8	39,1	14,2	21	1,9		
Card 5/6	(j)	90,1	42,9	13,5	21,4	2,1		
	(k)	84,1	36,8	12,8	21,6	2,3	228-252	3
	(l)	75,6	39,4	12,3	19,3	2,8		
	(m)	74,8	36,5	10,3	19,8	3,1		

Rational Heat Treatment Rates for Hot  
Rolling Rolls

77596  
SOV/129-60-2-9/13

It was found that isothermal holding immediately after forging failed to prevent flake formation. By heating steel after the initial overcooling, flakes have no time to develop. Subsequent overcooling promotes considerable H liberation. The minor amounts of H which remain in the steel do not enhance flake formation. The authors recommend the application of the above heat treatment rates which combine annealing and normalization and reduce the time of heat treatment of large-size forgings by 40%. Hundreds of rolls have already been heat-treated by the above method, and considerable saving was achieved at the plant. There are 4 figures; 2 tables; and 4 Soviet references.

ASSOCIATION: Novo-Kramatorskiy Machine Building Plant (Novo-Kramatorskiy mashinostroitel'nyy zavod)

Card 6/6

DOBRYANSKAYA, YEP

## PHASE I BOOK EXPLOITATION

SOV/5511

Nauchno-tekhnicheskoye obshcheshchstvo makhinostroitel'nyy proizvodstvennosti.  
Kievskoye oblastnoye pravleniye.

Metallovedenye i termicheskaya obrabotka (Plastica, Metallo) i nest  
Treatment of Metals) Moscow, Naukiz, 1961. 350 p. Errata slip  
Inserted. 5,000 copies printed.

Sponsoring Agency: Gendinistrstrenyy nauchno-tehnicheskly komitet  
Soveta Ministriv UkrSSR. Nauchno-tehnicheskoye obshcheshchstvo  
makhinostroitel'nyy proizvodstvennosti. Kiyavskoye oblastnoye  
pravleniye.

Editorial Board: M. P. Braun, Doctor of Technical Sciences, I. Ya.  
Dobryak, Doctor of Technical Sciences, D. A. Drayser, Doctor of  
Technical Sciences, I. S. Kazenichhev, Engineer, Ye. A. Markov-  
skiy, Candidate of Technical Sciences, V. G. Permyakov, Doctor  
of Technical Sciences, and A. V. Chernopol, Candidate of Technical  
Sciences; Ed.: M. S. Sorkin; Tech. Ed.: M. S.  
Sordyuk, Engineer; Chief Ed., Naukiz (Southern Dept.): V. K.  
Sordyuk, Engineer.

Card 1/10

PURPOSE: This collection of articles is intended for scientific  
workers and technical personnel of research institutes, plants,  
and schools of higher technical education.

COVERAGE: The collection contains papers presented at a convention  
held in Kiev on problems of physical metallurgy and methods of  
the heat treatment of metals applied in the machine industry.  
Plane transformations in metals and alloys are discussed, and  
results of investigations conducted to ascertain the effect of  
heat treatment on the quality of metal are analyzed. The possi-  
bility of obtaining metals with given mechanical properties  
is discussed, as are problems of steel brittleness. The collec-  
tion includes papers dealing with kinetics of transformation,  
heat treatment, and properties of cast iron. No personalities  
are mentioned. Articles are accompanied by references, mostly  
Soviet.

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## Physical Metallurgy (cont.)

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## Physical Metallurgy (Cont.)

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DOBRYANSKIY, A. F.

Dobryanskiy, A. F. "Clearing the Russian technical language of foreign terminology," Vestnik vyssh. shkoly, 1948, No. 12, p. 19-22

SO: U-3264, 10 April 1953, (Letopis 'Zhurnal 'nykh Statey, No.3, 1949)

DOHRYANSKIY, A.F., prof., red.; TISHCHENKO, V.V., dots., red.;  
GAVRILOV, B.G., dots., red.; PIASTRO, V.D., red.; ZHUKOVA,  
Ye.G., tekhn. red.

[Proper storage of machinery] Kak pravil'no khranit' mashiny.  
Moskva, Mosk. rabochii, 1962. 35 p. (MIRA 15:10)  
(Agricultural machinery—Storage)

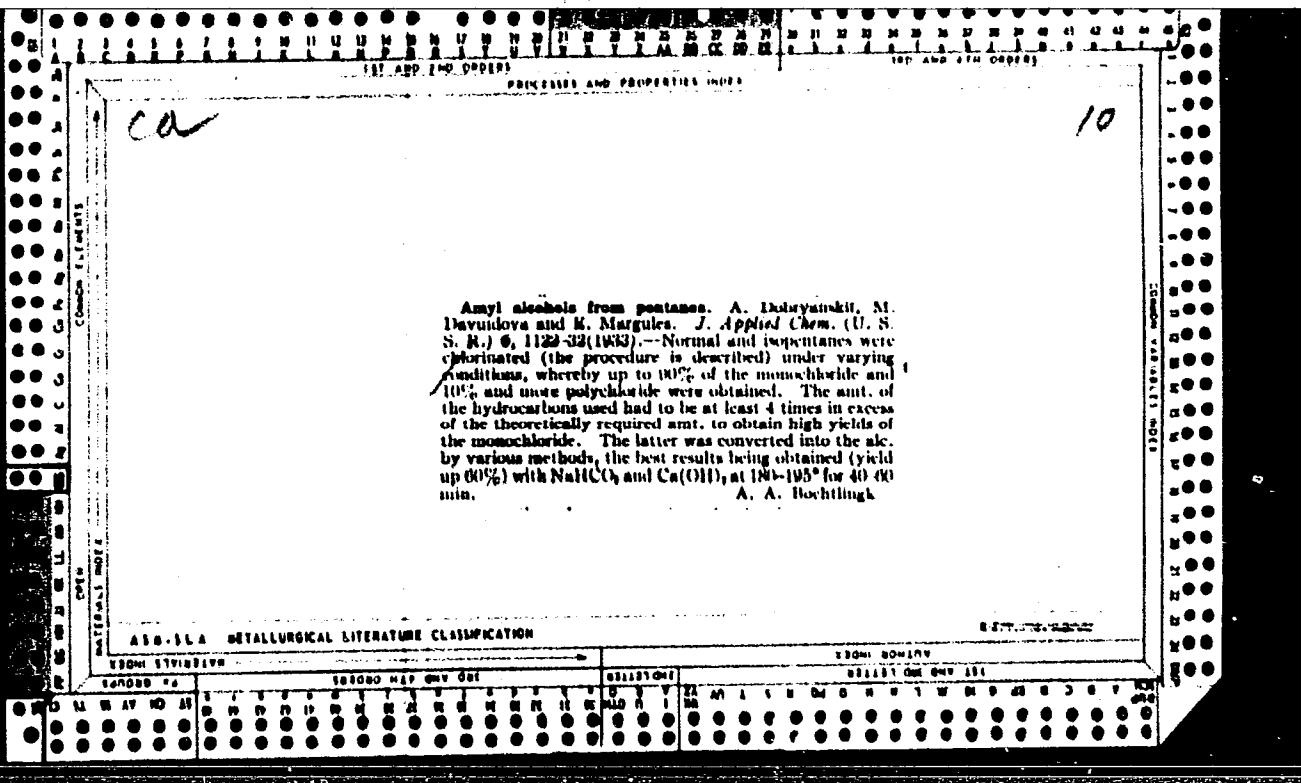
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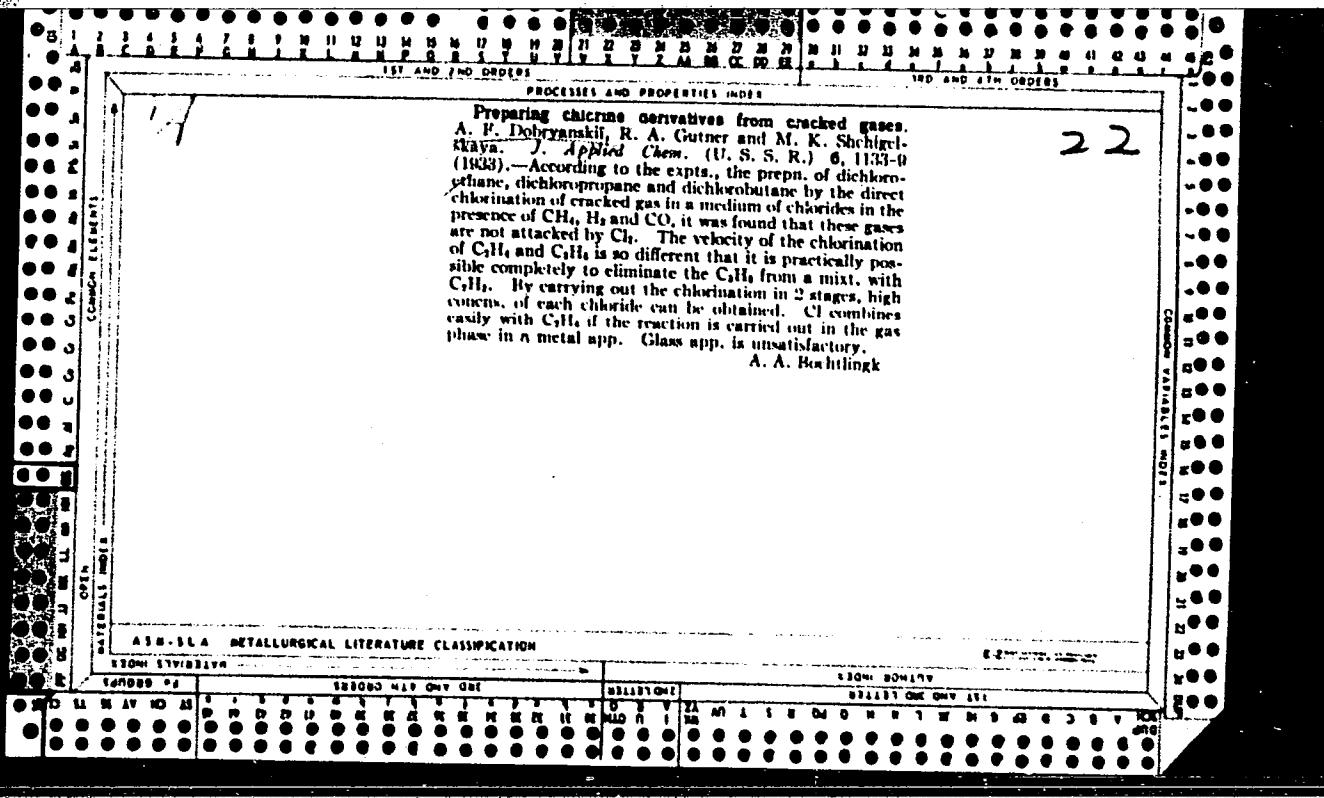
**Cracking with aluminum chloride.** A. F. Dobrynynikov and N. I. Zelenin. *Khim. Tverdogo Zapovedn. 4*, 606-18 (1933).—A kerosene distillate freed from low-boiling fractions was heated with 10%  $\text{AlCl}_3$ , yielding 46.6% of a gasoline of 74.6 sp. gr. The yield can be raised by prolonging the time. The following stocks were also used, although the yields of gasoline were lower: Bimta gas oil, Syvash Ostrov gas oil, Surakhani fuel oil, cracked kerosene, cracked polymers and "green" oil. The gasoline contained about 2%  $\text{C}_4\text{H}_8$ , 8.4%  $\text{PbMe}$  and 5.7%  $\text{C}_4\text{H}_9\text{Me}$ . The gas contained about 94% butane, 5.7% H and 0.3% heavy hydrocarbons. A. A. Boehlting.

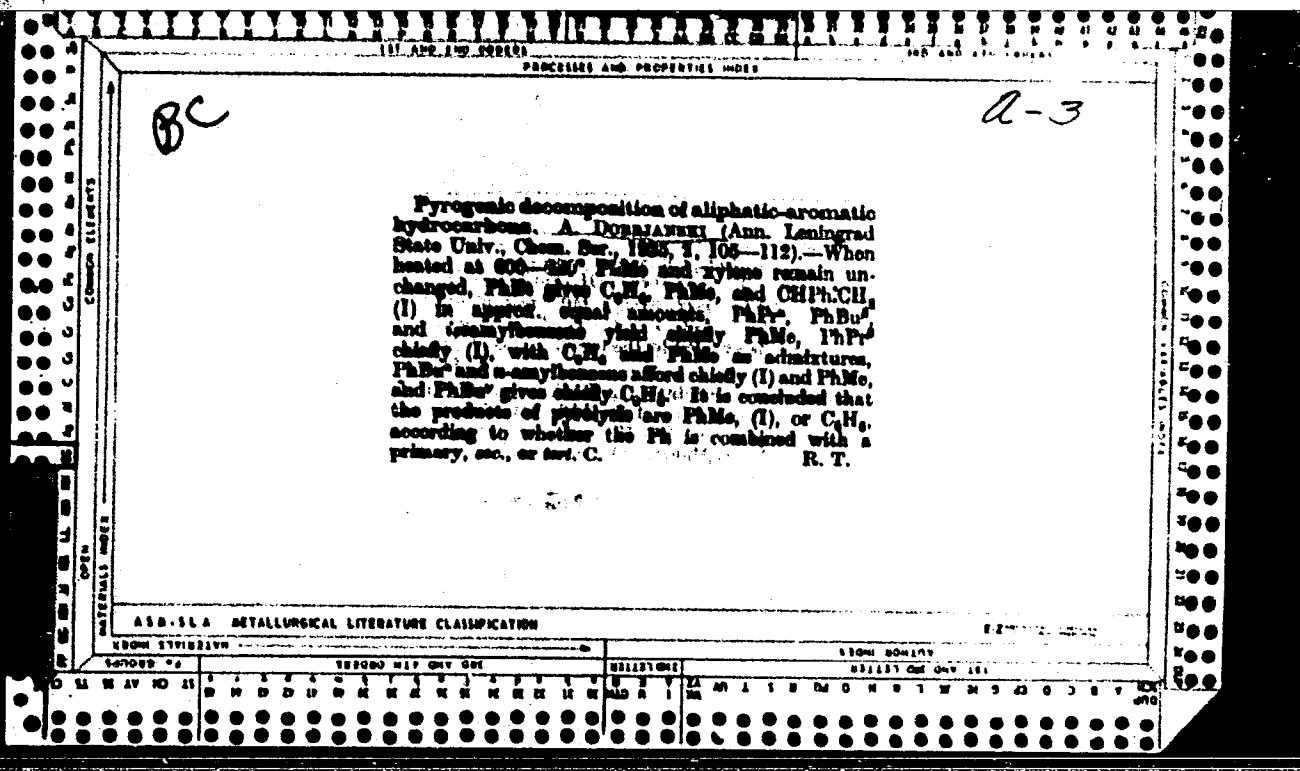
43B-51A METALLURGICAL LITERATURE CLASSIFICATION

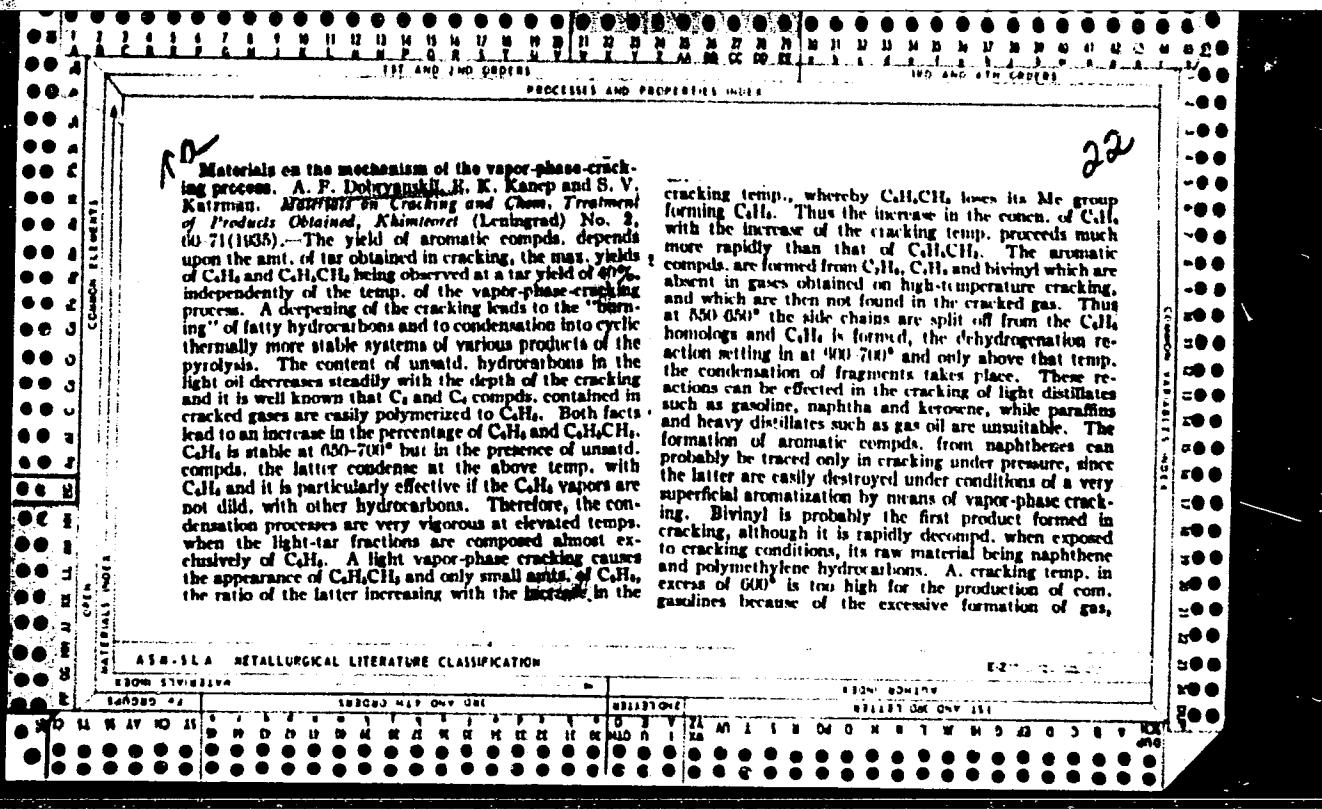
**APPROVED FOR RELEASE: 06/13/2000**

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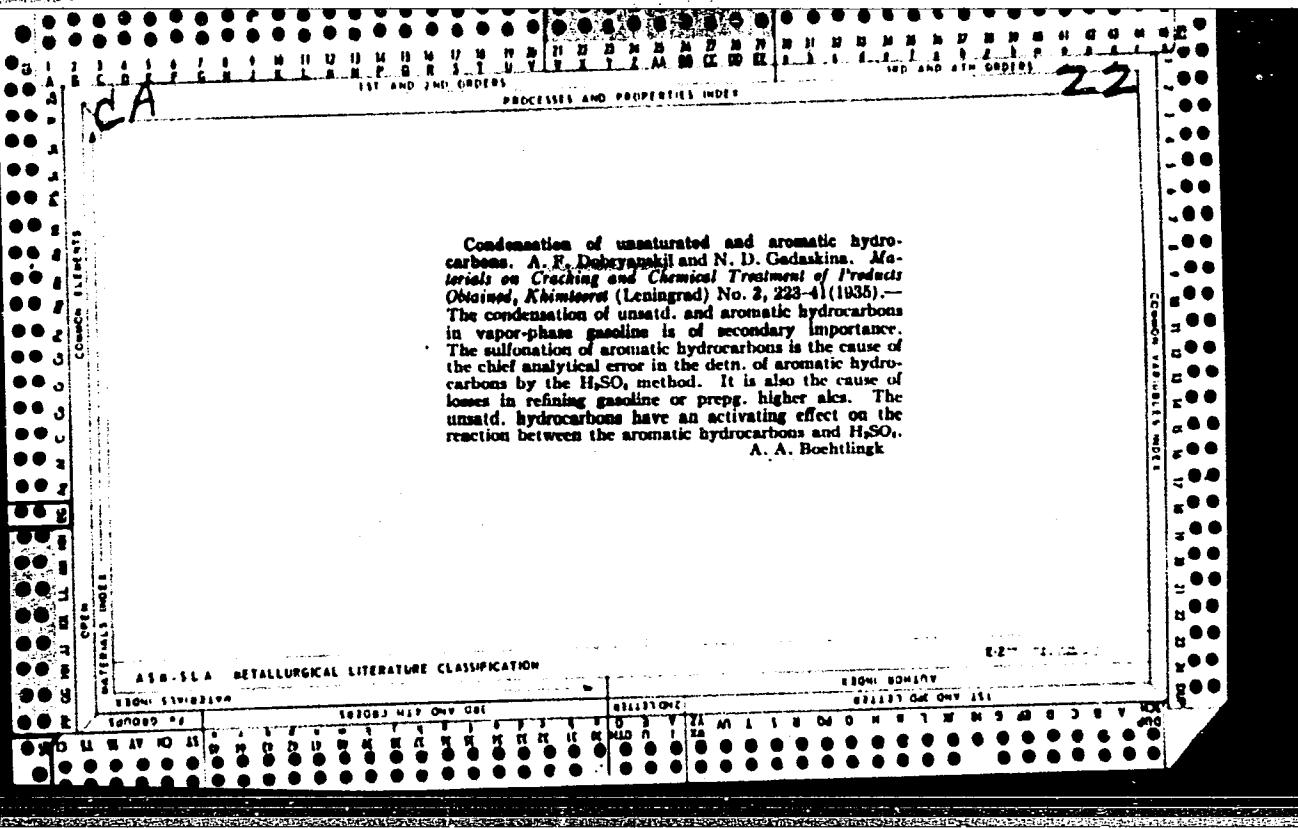






which should not exceed a total of 15-20%. These conclusions were made on the basis of a series of expts. (by various authors) to be reported in the succeeding papers. The app. used in the pyrolysis had one iron or porcelain tube (100 cm. long) with an inner diam. of 34 mm., 2 upright Liebig condensers arranged in series (each 100 cm. long) and 2 flushing towers of the same length. The condensate was collected in the usual manner and the reflux was not recycled but analyzed immediately after the withdrawal from the system. The gases were collected and analyzed.

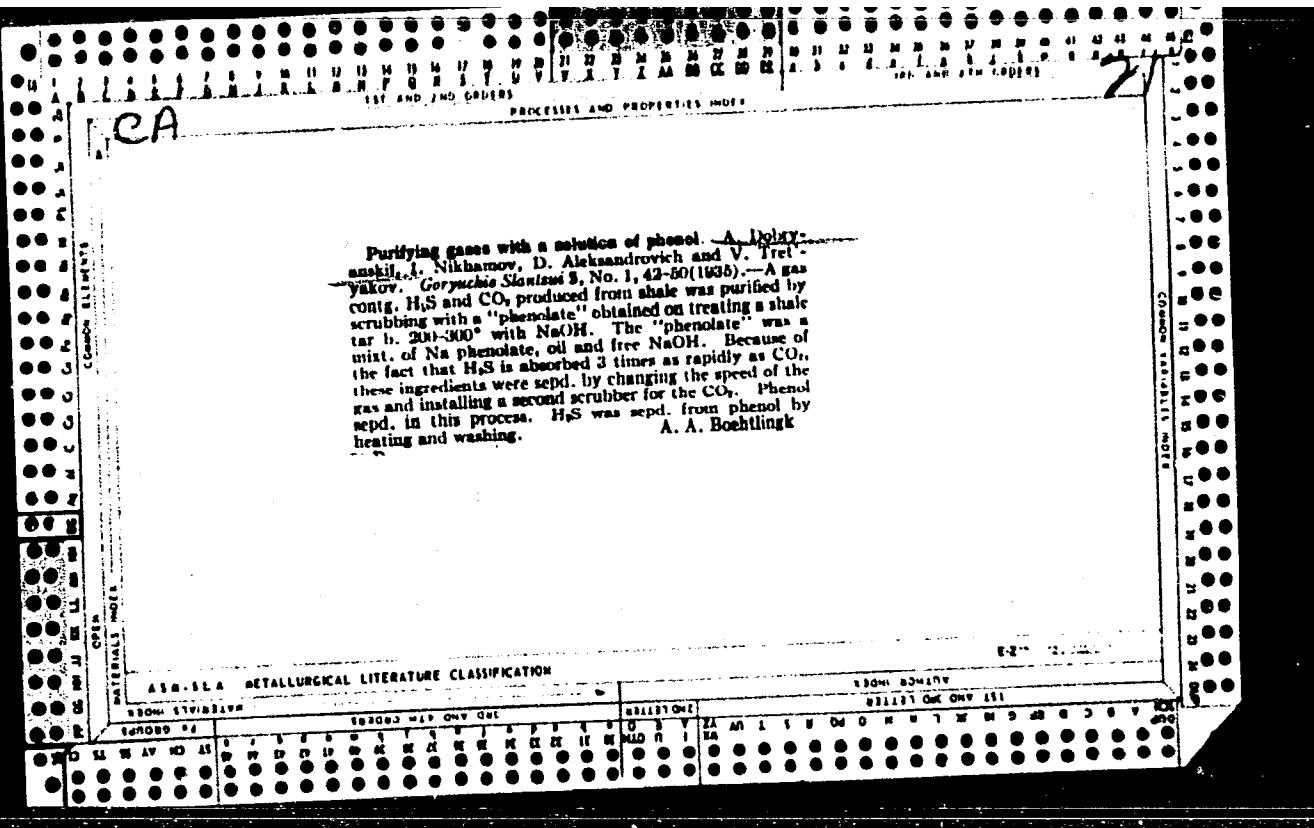
A. A. Roehling

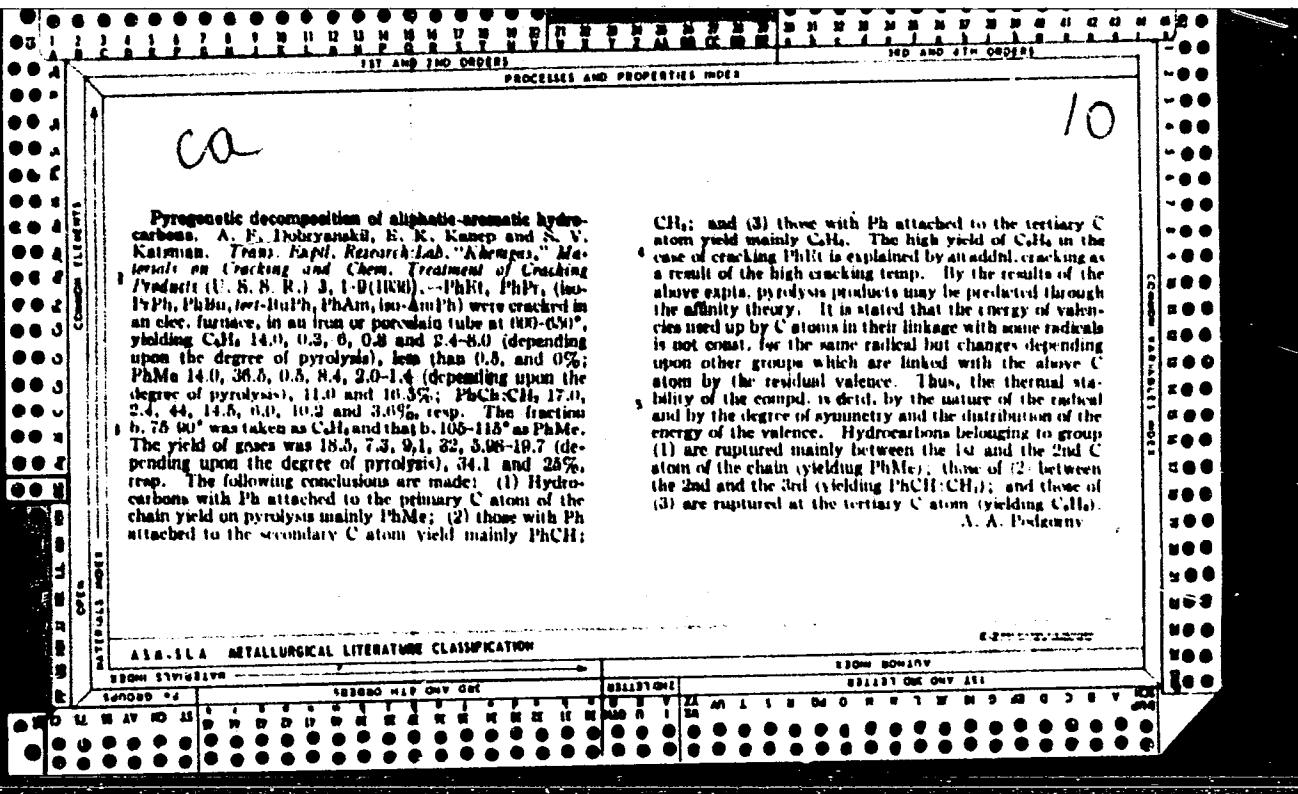


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<p style="text-align: center;">22</p> <p>Determination of aromatic hydrocarbons in saturated gasolines. A. P. Dobryanskiii and N. T. Tikhomirov- Dubrovskii. <i>Azovskikhnefte Neftyanoe Khimicheskoe</i> 1935, No. 5, 84-9. The amt. of <math>C_6H_6</math>, <math>C_8H_8CH_3</math>, and <math>C_8H_10</math> (<math>CH_3</math>)<sub>2</sub> added to a gasoline free from aromatic compds. was checked by the picric acid method. Accordingly, the gasoline was broken up into fractions b, 85-95° ("ben- zene" fraction), 90-125° ("toluene" fraction) and 125- 155° ("xylene" fraction). The individual fractions, after the addition of various amts. of the above aromatic compds. (from 1 to 100%), were sntd. with picric acid, and 10 cc. of the fraction to be analyzed was pipetted into the sntd. picric acid, followed by agitation for 5 min. The gasoline mixt. was transferred (the sediment was left in the flask) into a second flask with 35 cc. of distd. <math>H_2O</math>; another 10 cc. was used for rinsing the funnel. Three drops of phenom- philalein was then added and the mixt. agitated for the transfer of the acid into the aq. soln., the titration being carried out with 0.01 N NaOH to the appearance of a permanent (4-5 min.) raspberry-red color in the lower <math>(H_2O)</math> layer. The results are accurate to ±0.02%.</p> <p style="text-align: right;">A. A. Boettlingk 6</p>																																																																																																																																																			
<p style="text-align: center;">ASB-51A METALLURGICAL LITERATURE CLASSIFICATION</p> <table border="1" style="width: 100%; border-collapse: collapse;"> <tr> <th colspan="12">SECOND DIVISION</th> <th colspan="12">THIRD DIVISION</th> </tr> <tr> <th colspan="6">SUBDIVISION</th> <th colspan="6">SUBDIVISION</th> <th colspan="6">SUBDIVISION</th> <th colspan="6">SUBDIVISION</th> </tr> <tr> <th colspan="2">SUBDIVISION</th> </tr> <tr> <td colspan="2">I</td> <td colspan="2">II</td> <td colspan="2">III</td> <td colspan="2">IV</td> <td colspan="2">V</td> <td colspan="2">VI</td> <td colspan="2">VII</td> <td colspan="2">VIII</td> <td colspan="2">IX</td> <td colspan="2">X</td> <td colspan="2">XI</td> <td colspan="2">XII</td> <td colspan="2">XIII</td> </tr> <tr> <td>1</td><td>2</td><td>3</td><td>4</td><td>5</td><td>6</td><td>7</td><td>8</td><td>9</td><td>10</td><td>11</td><td>12</td><td>13</td><td>14</td><td>15</td><td>16</td><td>17</td><td>18</td><td>19</td><td>20</td><td>21</td><td>22</td><td>23</td><td>24</td></tr> </table>																								SECOND DIVISION												THIRD DIVISION												SUBDIVISION		I		II		III		IV		V		VI		VII		VIII		IX		X		XI		XII		XIII		1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18	19	20	21	22	23	24																																																
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DOBRIANSKII, A. F.

RT-1632 (Production of dichloroethane from chlorine and ethylene (in semi-commercial plant) ) Poluchenie dikhloroetana iz khlora i etilena (na poluzavodskoi ustanovke).  
TRUDY GOSUDARSTVENNOGO IN TITUTA PRIKLADNOI KHMII 24: 32-47, 1935





Determination of the aromatic hydrocarbon content by means of specific gravity. A. F. Dobryanski and Yu. N.

Petrova. *Trans. Exptl. Research Lab. Khemgas, Materials on Cracking and Chemical Treatment of Cracking Products U.S.S.R.*, 3, 400-8 (1938); cf. *C. A.* 30, 3060<sup>a</sup>. The content ( $y$ ) of PhH, MePh and Me<sub>2</sub>C<sub>6</sub>H<sub>5</sub> in mixts. can be detd. by the formula:  $y = [(d_1 - d_1)/(d_2 - d_1)] \times 400 + K$ , where  $d_1$ ,  $d_2$  and  $d_3$  are ds. of the gasoline residue, of the sample under investigation, and of the detd. aromatic hydrocarbon, resp.;  $K$  is the correction factor for the difference between the calcd. av. d. of the mixt. of the gasoline with different amts. of aromatic hydrocarbons and that experimentally found. The method is discussed and data are tabulated. A. A. Podgorny

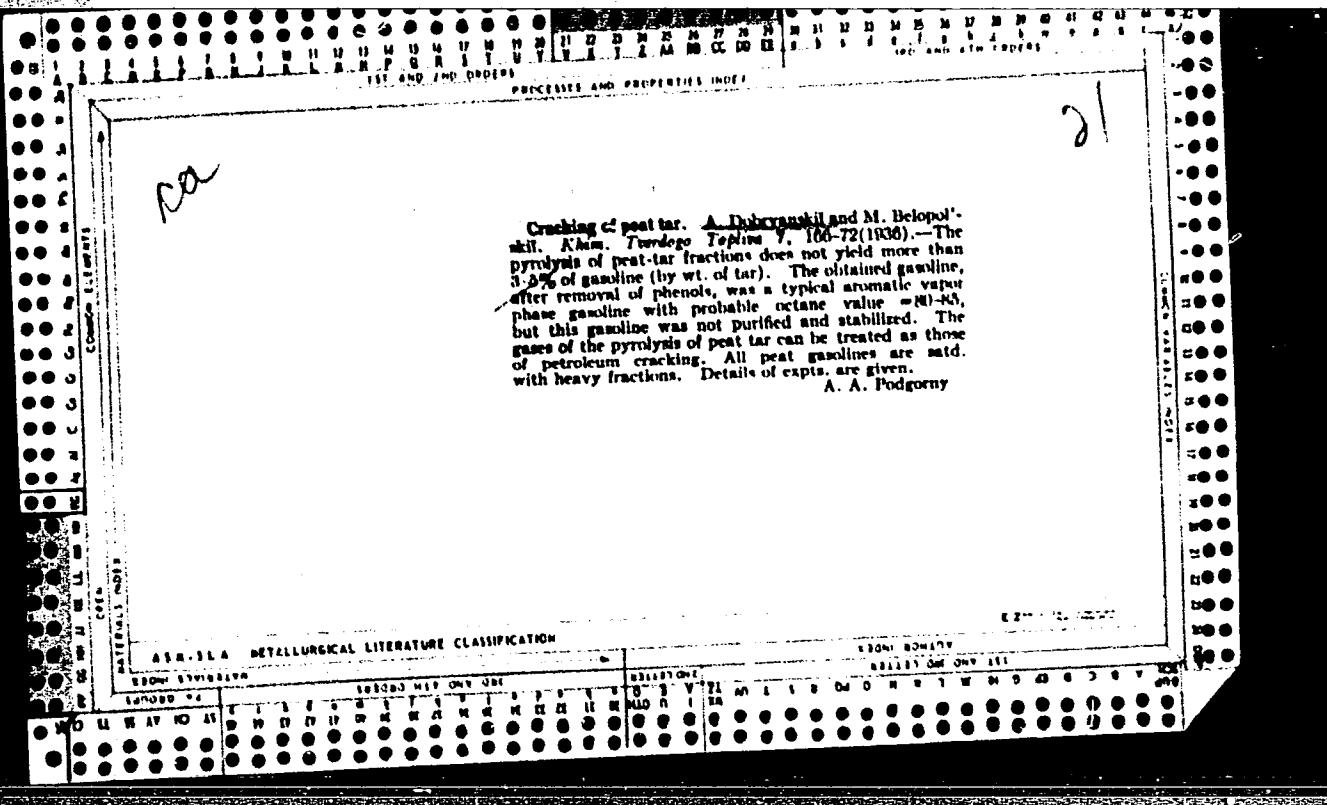
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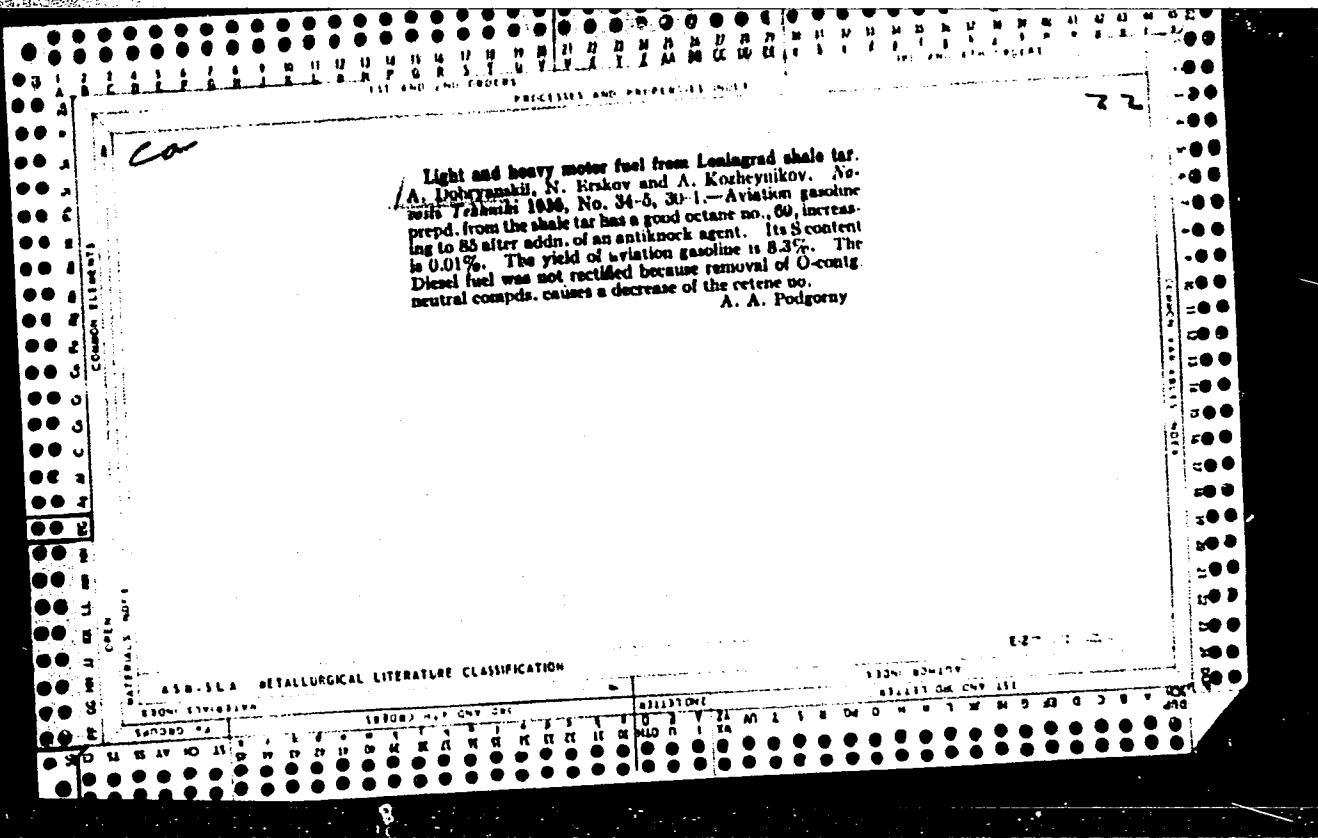
**AM-SEA METALLURGICAL LITERATURE CLASSIFICATION**

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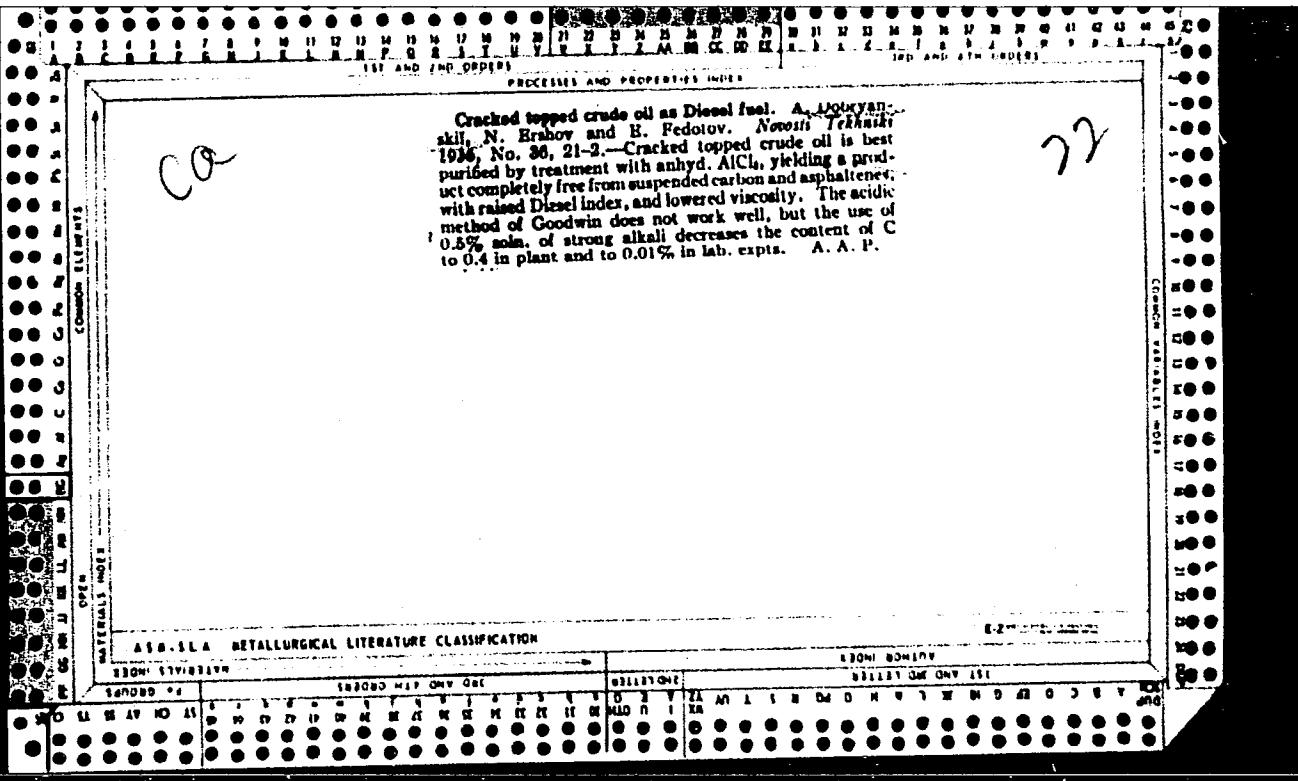


**Stabilization of gasoline.** A. Dobryankil, N. Ershov and A. Kochevnikov. *Naučni Tiskarstvo 1936*, No. 30, 20-1. Peat- and shale-tar phenols, freed from acids and neutral oils, are strong inhibitors, giving an induction period of 6 hrs. (0.01% of phenols in gasoline). The addn. of 0.01% of phenols to an aviation gasoline decreases the amt. of potential gum to the standard level. The most effective inhibitors are the peat-tar phenols. All inhibitors are sol. in gasoline, insol. in water, do not

change the color of gasoline and have all the requirements for "universal inhibitors." A. A. Pidgorsky

## ASA-SEA METALLURGICAL LITERATURE CLASSIFICATION

APPROVED FOR RELEASE: 06/13/2000 CIA-RDP86-00513R000410710006-2"



Chlorination of propene oxide. A. V. Polyanyi,  
M. I. Davydova and Z. T. Papkina. J. Gen. Chem.

(U. S. S. R.) 7, 201-7 (1937).— $\text{MeCH}_2\text{CH}_3\text{O}$  treated with 2 mols of dry  $\text{Cl}$  at  $0^\circ$  afforded a very complex reaction mixt., contg. chiefly  $\text{MeCOCH}_2\text{Cl}$ , b. 118-21° (semicarbazone, m. 141-8°) and  $\text{MeCH}(\text{OH})\text{CH}_2\text{Cl}$ , b. 127°. The work is being continued. Chas. Blane

CROWN ELEMENTS  
1993 INDIA

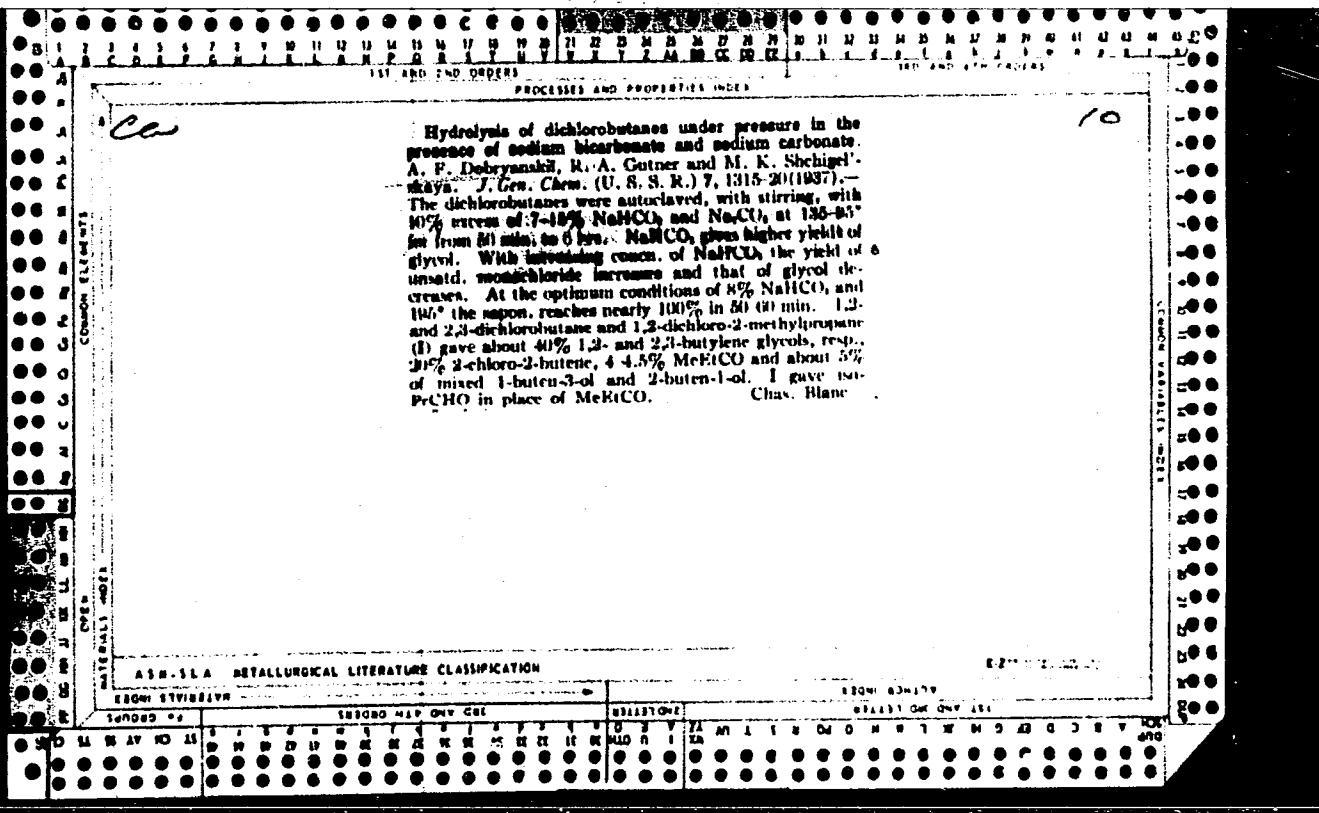
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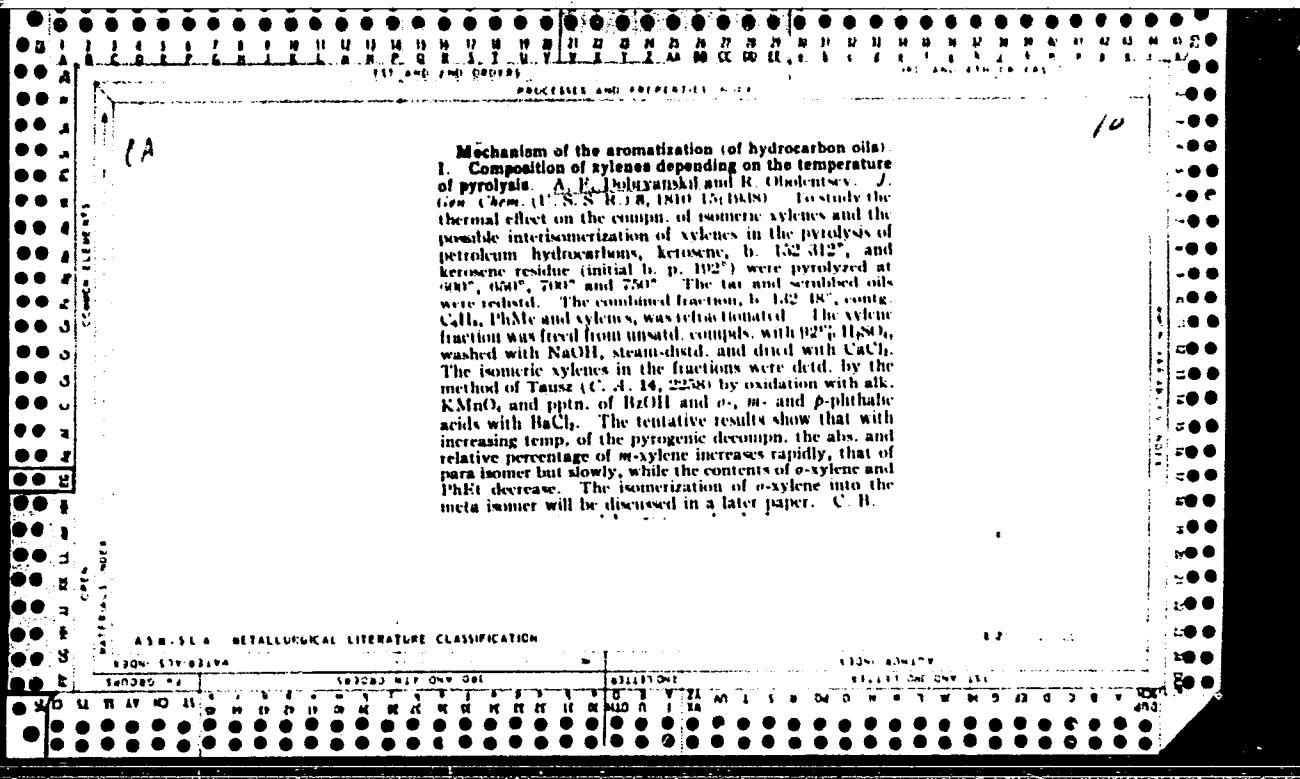
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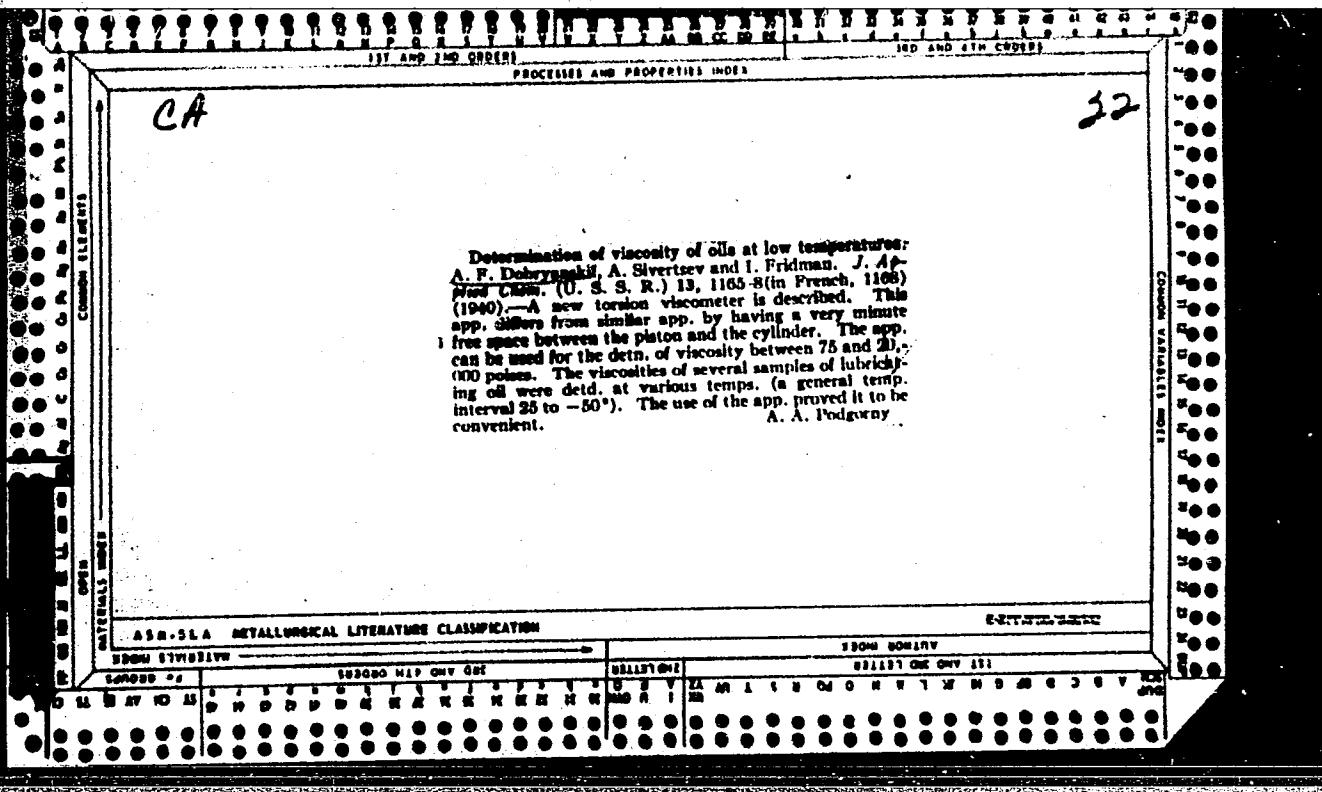


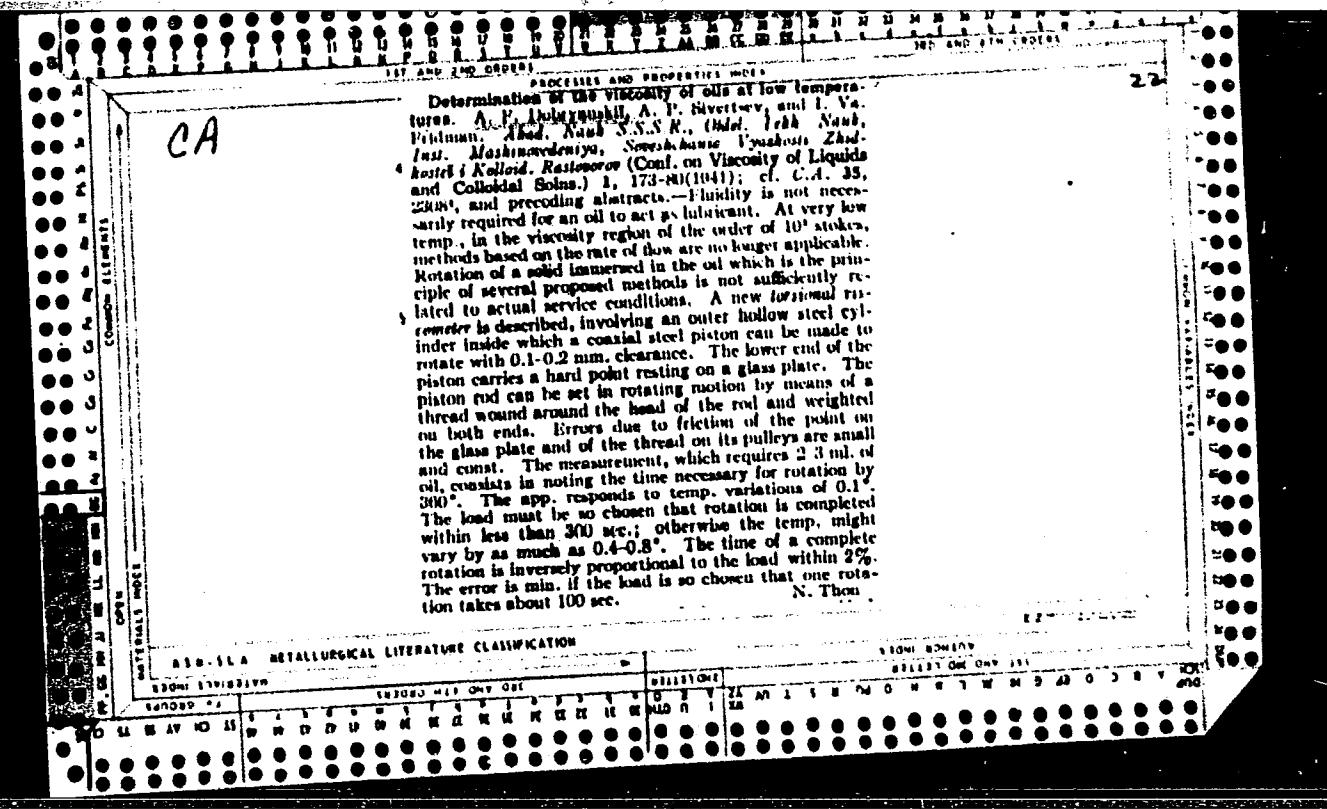


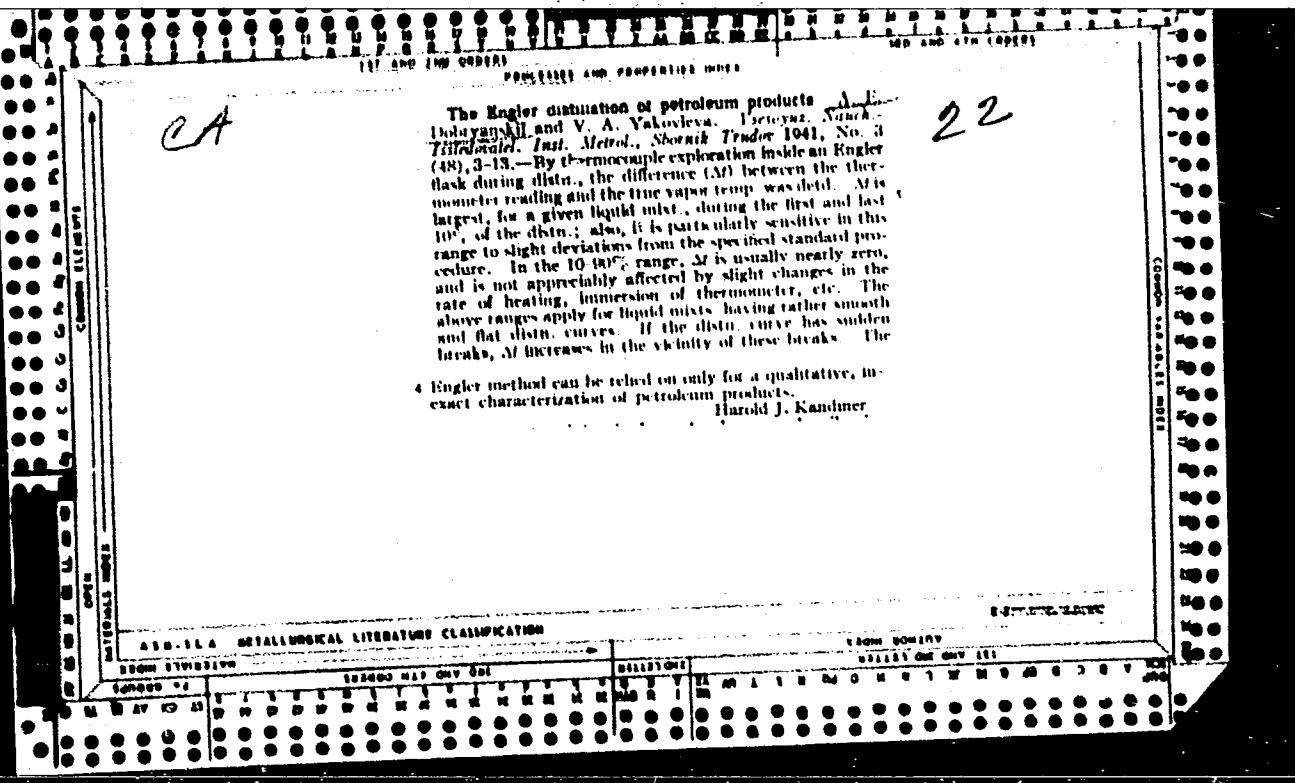
*Br. abr*

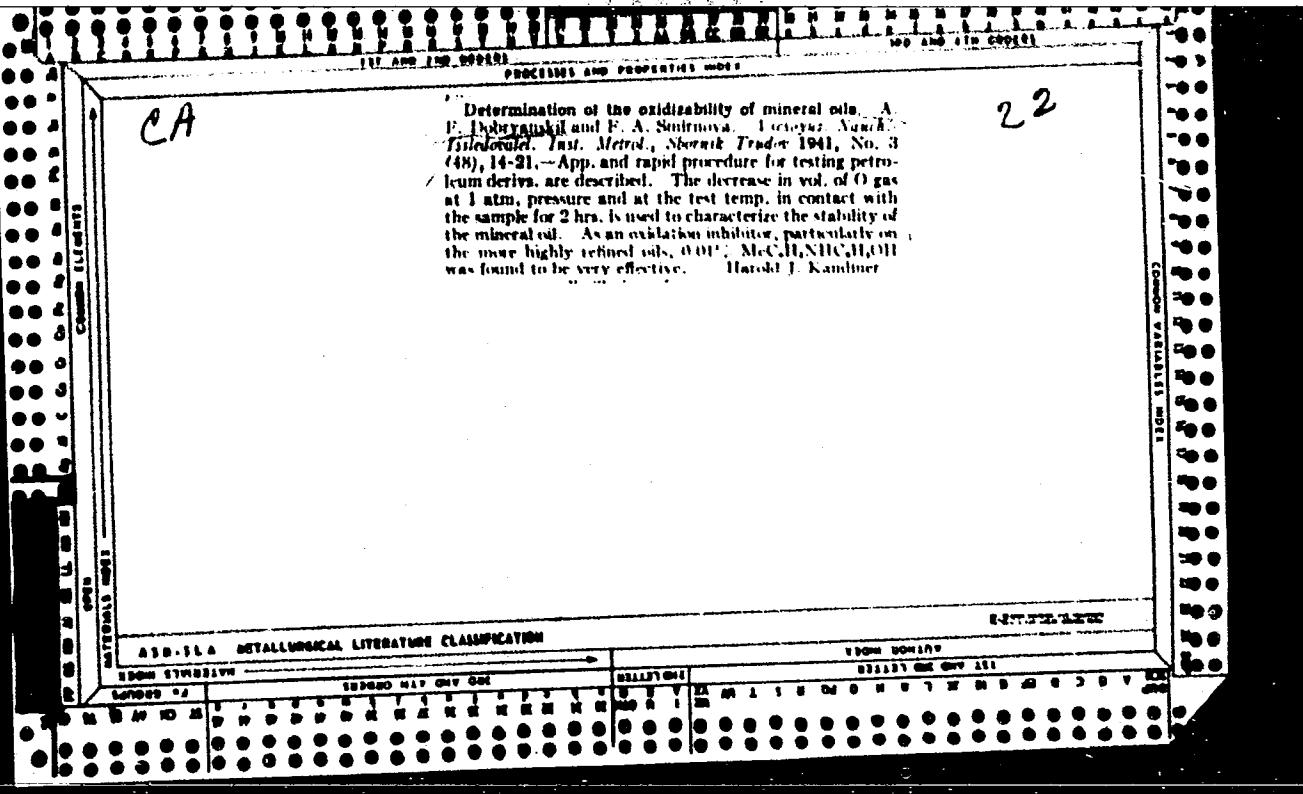
*A.S. II. Geochemistry.*

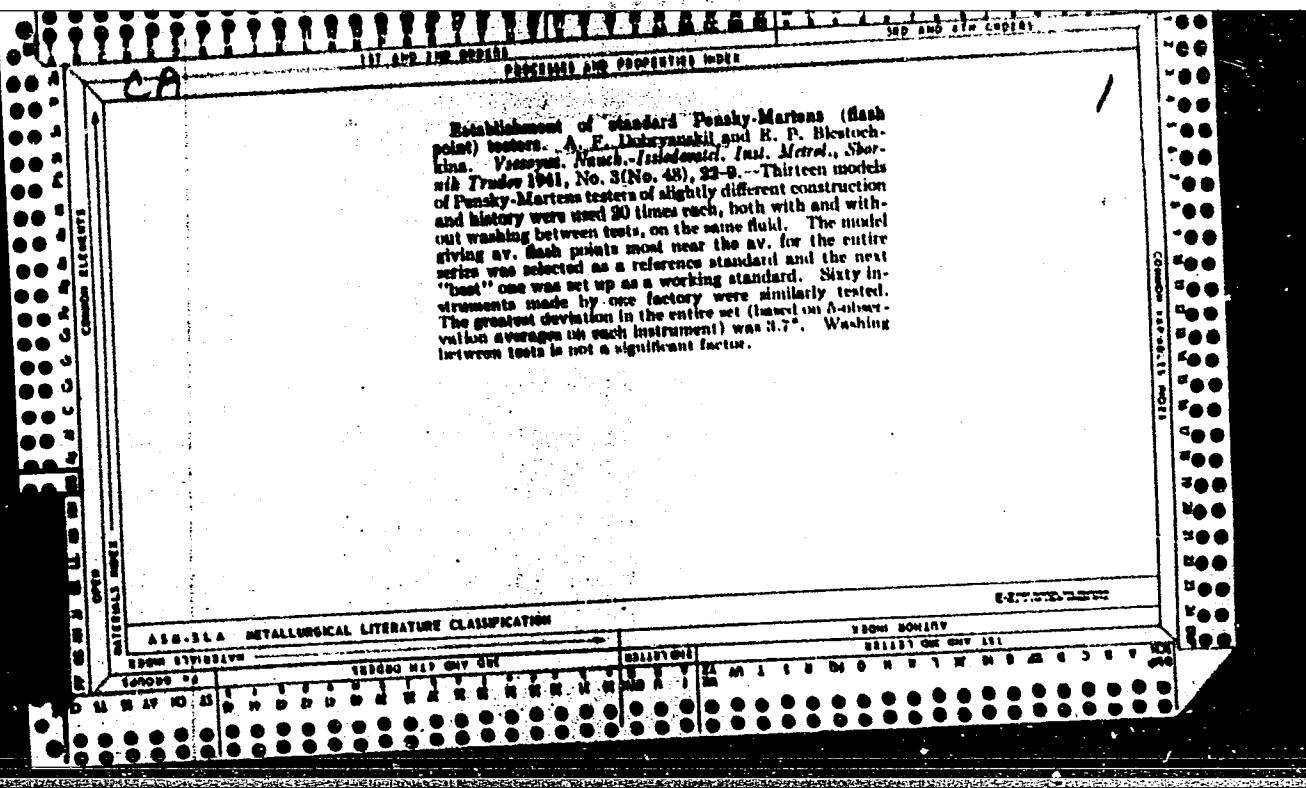
**Genetic classification of combustibles.** A. N. Dobrojanski (U. S. N. S. C. Am., Mass., 1940, 28, 443—468).—If the analyses of the organic part of wood and fossil fuels are represented in a triangular diagram with C, H, and (O + S + N) in the summits, it is seen that they form two distinct branches which touch each other at the beginning (at small C contents). To one branch belong wood, peat, lignite, coal, and anthracite; the other includes sapropel, oil shale, sapropelite, asphalt, petroleum, and asphaltites. These facts serve to solve the problem of the origin of petroleum and asphaltites. Petroleum is a derivative of sapropel, and asphaltites are formed by oxidation and evaporation of asphalts and petroleum. J. J. B.











DOBRYANSKIY, A.F.; BLOKH, L.S.; BLESTOCHKINA, Ye.P. [deceased]

Relationship between kinematic viscosity and viscosity according  
to Engler. Trudy VNIIM no.5:22-32 '47. (MIRA 12:1)  
(Lubrication and lubricants) (Viscosity)

**DOBRYANSKIY, A.F.; KUCHINSKIY, V.N.**

Determining the kinematic viscosity of mixtures of oils. Trudy  
VNIIM no.5:46-51 '47. (MIRA 12:1)  
(Lubrication and lubricants) (Viscosity)

DOBRIANSKY, A.

Dobriansky, A., and Sivertzev, A.-"Action of Aluminium Chloride upon the Esters"  
(p. 912)

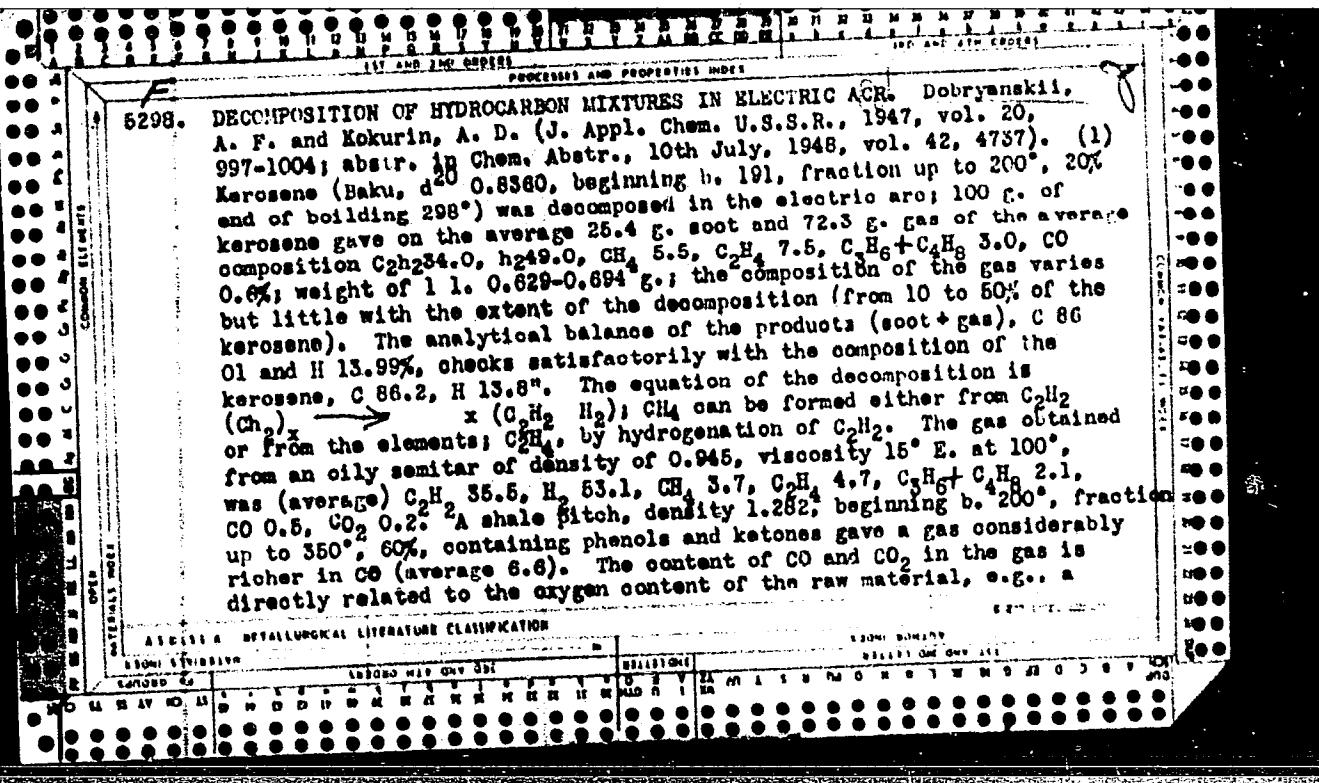
SO: Journal of General Chemistry, (Zhurnal Obshchey Khimii), 1947, Vol. 17, No. 5

**Aluminum naphthenate.** A. F. Dobryanski and R. V. Malinovskaya. *J. Applied Chem. (U.S.S.R.)*, 30, 837 (1947) (in Russian).—The method of increasing lubricating oil viscosity by addn. of Al naphthenate (Leonit "cv. Ross. Pat. 56,128) which contains free  $\text{Al}(\text{OH})_3$ , is based on erroneous data. The f.p. of various grades of oil with addn. of 0.5-2% Al naphthenates was lowered but slightly in comparison with untreated specimens. Crude naphthenic acids were distd. and the fraction b. 140-255° was used to prep. Me esters, b. 161-70°,  $d_4^{20} 0.9482$ ,  $n_D^{20} 1.4602$ . Sapon. of the Me esters was followed by distn. of the regenerated acids, of which a fraction b. 149-8°,  $d_4^{20} 0.9770$ , mol. wt. 222, was used to prep. the Al salts according to  $3\text{RCOOH} + 3\text{KOH} + \text{KAl}(\text{SO}_4)_2 \rightarrow \text{Al}(\text{OEt})_3 + 2\text{K}_2\text{SO}_4$ . The product had the appearance of

Silica gel; it was a semitransparent amorphous solid, containing 7.40% Al, which, however, was shown to be heterogeneous by repeated treatment with 80% RCOH (free RCOH could be leached out). The alk.-washed salt corresponded closely to  $\text{RCO}_2\text{Al}(\text{OH})_5$ . Use of excess alkali ( $\text{Na}_2\text{CO}_3$  or  $\text{NaHCO}_3$ ) leads to copious, free  $\text{Al}(\text{OH})_3$ , which can be sepd., by treatment of the ppt. with  $\text{AmOAc}$ , the soin, config. the salt,  $\text{RCO}_2\text{Al}(\text{OH})_5$ . Although addn. of the salt config.  $\text{Al}(\text{OH})_3$  tends to increase lubricating oil stability, the phenomenon is not stable and the viscosity drops again sharply on standing or agitation. Addn. of the purified salt gives more stable formulations, but the viscosity increase is much smaller; thus spinelle oil (0.0045 Stoke at 20°) gave the following values (in Stokes): 1.0483 with 0.6%<sup>o</sup>, 1.1348 with 1.0%<sup>o</sup>, 1.0724 with 1.5%<sup>o</sup>, and 2.1860 with 2.0%<sup>o</sup> additive, while the last soin., stirred 8 hrs. at 3000 r.p.m., dropped to 2.0648 Stokes.

S. M. Knopoff

1ST AND 2ND ORDERS												3RD AND 4TH ORDERS												
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COMPOUNDS ELEMENTS												COMPOUNDS ELEMENTS												
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Aluminum naphthenate. A. F. Dobryanski and N. V. Malinovskaya. J. Applied Chem. (U.S.S.R.) 30, 837-840 (1947) (in Russian). The method of increasing lubricating oil viscosity by addn. of Al naphthenates (Leont'ev, Russ. Pat. 56,128) which contain free Al(OH) <sub>3</sub> is based on erroneous data. The l.p. of various grades of oil with addn. of 0.5-2% Al naphthenates was lowered but slightly in comparison with untreated specimens. Crude naphthenic acids were distd. and the fraction b.p. 140-205° was used to prep. Me esters, b.p. 161-70°, d <sup>25</sup> 0.9489, n <sup>D</sup> 1.4602. Sapon. of the Me esters was followed by distn. of the regenerated acids, of which a fraction b.p. 142-5°, d <sup>25</sup> 0.9770, mol. wt. 222, was used to prep. the Al salts according to 3RCOOH + 3KOH + KAl(RO <sub>2</sub> ) <sub>2</sub> → Al(O <sub>2</sub> CR) <sub>3</sub> + 2K <sub>2</sub> CO <sub>3</sub> . The product had the appearance of white gell., it was a semitransparent amorphous solid, contg. 7.48% Al, which, however, was shown to be heterogeneous by repeated treatment with 80% KOH (free RCOOH could be leached out). The alc.-washed salt corresponded closely to RCO <sub>2</sub> Al(OH) <sub>3</sub> . Use of excess alkali (Na <sub>2</sub> CO <sub>3</sub> or NaHCO <sub>3</sub> ) leads to copptn. of free Al(OH) <sub>3</sub> , which can be sepd. by treatment of the ppt. with AmOAc, the soln. contg. the salt, RCO <sub>2</sub> Al(OH) <sub>3</sub> . Although addn. of the salt contg. Al(OH) <sub>3</sub> tends to increase lubricating oil stability, the phenomenon is not stable and the viscosity drops again sharply on standing or agitation. Addn. of the purified salt gives more stable formulations, but the viscosity increase is much smaller; thus spindle oil (0.9845 stoke at 20°) gave the following values (in stokes): 1.0483 with 0.8%, 1.1348 with 1.0%, 1.0724 with 1.5%, 2.1800 with 2.0% additive, while the last soln., stirred 8 hrs. at 3000 r.p.m., dropped to 2.0848 stokes.												G. M. Koosaloff												
ADM-31A METALLURGICAL LITERATURE CLASSIFICATION												E-27-1727-2-2022												
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SYNTHETIC												SYNTHETIC												
VOLUME 94												VOLUME 94												
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1	2	3	4	5	6	7	8	9	10	11	12	1	2	3	4	5	6	7	8	9	10	11	12	



Diesel fraction, a shale pitch, and a peat pitch with 0.1.2, 5.2, and 12.6 %, respectively, gave CO 3.2, 6.6, and 8.6, CO<sub>2</sub> 0.4, 0.6, and 1.1% respectively. In all experiments, up to 50% of the material remained undecomposed; the density of the residue is the higher, the deeper the decomposition; e.g., kerosene, decomposition 10, 15, 20, and 30%, density of residue 0.8386, 0.8383, 0.8382, and 0.8387; the temperature of beginning boiling of the residue is 4-5° higher than that of the initial kerosene, but the fractional composition remains unchanged; thus, the change of density is due solely to removal of the lightest fractions, without any cracking of the main mass taking place. (2) EtOH (96%) gave a gas C<sub>2</sub>H<sub>2</sub> 14.4, H 51.3. CH<sub>4</sub> 6.2, C<sub>3</sub>H<sub>8</sub> 4.7, C<sub>3</sub>H<sub>6</sub> C<sub>4</sub>H<sub>8</sub> 0, CO 22.7, CO<sub>2</sub> 0.7%; no traces of M<sub>2</sub> CHO were found; the undecomposed alcohol is unchanged. It is noteworthy that dehydration to C<sub>2</sub>H<sub>4</sub> is very low. No soot is formed. The reaction consists in the main in 2 EtOH → 3CO + 6H<sub>2</sub> + C<sub>2</sub>H<sub>2</sub> + CH<sub>4</sub>. (3) C<sub>2</sub>H<sub>2</sub> is a primary product and its high yield is determined by its fast removal from the high temperature zone where it might suffer decomposition. Its formation is possibly the result of the recombination of free radicals.

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000410710006-2

DOBRYANSKIY, A. F.

"The Geochemistry of petroleum", (Geokhimiya nefti), Tostopizadt, 1948.

APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000410710006-2"

DOBRYANSKIY, A. F.

PA 43/49T18

USSR/Chemistry - Petroleum  
Chemistry - Mixtures Oct 48

"Volume-Ratio of a Mixture of Petroleum Fractions," A. F. Dobryanskiy, N. V. Strigaleva, N. I. Georgieidi, Leningrad Tech Inst, All-Union Petroleum Sci Res Geol Res Inst, Leningrad, 6 pp

"Zhur Prilied Khim" Vol. XXX, No 10

Investigates expansion of a mixture when various quantities of petroleum fractions are mixed together, and when gasoline fractions are mixed with aromatic hydrocarbons. Expansion is proportional to increase of specific weight or index of refraction. Maximum expansion occurs for a 50% mixture. Shows that

43/49T18

USSR/Chemistry - Petroleum (Contd) Oct 48

Molin-Gurvich principle applies for all physical properties of hydrocarbon mixtures not conforming to the law of additivity. Shows that determination of aromatic hydrocarbons in a petroleum fraction mixture can be sufficiently accurate by application of investigated method. Submitted 22 Mar 48.

43/49T18

"APPROVED FOR RELEASE: 06/13/2000

CIA-RDP86-00513R000410710006-2

DOBRIANSKIY, ALEKSANDR FLAVIANOVICH.

...COMBUSTIBLE SHALES OF THE USSR...ABSTRACTED BY SIRION MLOSKY.  
WASHINGTON, 1949

125, 61 L. MAPS, DIAGRS., TABLES. (US BUREAU OF MINES. INTRA-BUREAU  
REPORT)

BIBLIOGRAPHIES: P. 123-125; 60-61.  
TYPED COPY.

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DOBRYANSKIY, A. F.

32364 DOBRYANSKIY, A. F. i BOGOMOLOV, A. I. i SHKLYAR, I. V. Kataliticheskoye Vliyaniye  
Pered Na izmeneniye sostava Nefti. Zhurnal Prikl. Khimii, 1949, No. 10, s. 1124-32  
Bitliogr: 8 NAZV

SO: Letopis' Zhurnal'nykh Statey, Vol. 44

DOBRYANSKIV, A.F., professor; GAVRILOV, B.G., dotsent.

Catalytic conversions of petroleum hydrocarbons. Nauch. biul. Len.  
un. no.23:13-19 '49.  
(MLRA 10:4)

1. Kafedra tekhnicheskoy khimii.  
(Petroleum) (Hydrocarbons)

87-2 Series 10006

Bul. Ado.

Petroleum-like liquid from the Tkriveli coal deposits. A. F. Dobrynin and A. I. Bogomolov *J. Appl. Chem., U.S.S.R.*, 1949, 22, 630-643).—Oil exuding from Tkriveli (Georgian SSR) cannel coal resembles shale oil, and has  $\delta^{\text{P}}$  0.9676, S 0.98, N 0.14, aromatic 8.0, naphthenic 9.2, and paraffinic hydrocarbons 47.4, and paraffin wax content 14.2%; 88% of the oil distills from 134° to 350°.

R. Truscoz.

B.1 also

R.D. & C. Leake

Catalytic influence of titan on the change of composition of petroleum. A. N. Dzhaparov, A. I. Bogunov, and L. V. Shlyar (U. S. S. R., 1970, No. 1124-1132).—Polycyclic hydrocarbons are decyclized by heating at 200° with an Al silicate (beatoite) catalyst. Compounds of lower bp, and simpler molecular species are formed, with redistribution of H, the proportion of naphthalenes being increased. The metamorphism of petroleum in strata is believed to take place by a similar mechanism.

I. H. Butow.

DOBRYANSKIY, A.F.

PHASE I BOOK EXPLOITATION

SOV/4941

Mezhvuzovskoye soveshchaniye po khimii nefti, Moscow, 1956.

Sbornik trudov Mezhvuzovskogo soveshchaniya po khimii nefti  
(Collection of Transactions of the Inter-University Con-  
ference on Petroleum Chemistry) [Moscow] Izd-vo Mosk.  
univ., 1960. 313 p. Errata slip inserted. 1,600 copies  
printed.

Organizing Committee of the Conference: Chairman: B. A.  
Kazanskiy, Academician; Vice-Chairman: S. I. Khromov,  
Docent; G. M. Panchenkova, Professor; A. F. Plate, Pro-  
fessor; Secretary: Ye. S. Balenkova, Scientific Worker.  
Editorial Board: Resp. Ed.: A. F. Plate; I. V. Gostun-  
skaya, I. N. Tits-Skvortsova, L. A. Erivanskaya.

PURPOSE: This collection of articles is intended for the  
teaching staff of universities and schools of higher ed-  
ucation training specialists for the petroleum and petrol-  
eum-refining industries.

-Card #17-

Collection of Transactions (Cont.)

SOV/4941

COVERAGE: The collection includes articles dealing with the present state of the petroleum industry, the scientific research problems in petroleum chemistry, the chemistry of petroleum, the composition of petroleum and petroleum products, the scientific principles of refining petroleum into motor fuels and lubricants, and the manufacture of synthetic products from hydrocarbon gases and petroleum. One article discusses the effect of chemical composition and additives on fuel combustion in jet engines. The material was presented at the Inter-University Conference on Petroleum Chemistry, held at the Moscow State University imeni M. V. Lomonosov November 26-28, 1956. No personalities are mentioned. References accompany most of the articles.

TABLE OF CONTENTS: None given

The authors and the titles of articles are as follows:

Introduction by B. A. Kazanskiy, Academician

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## Collection of Transactions (Cont.)

SOV/4941

Fedorov, V. S. [Deputy Minister of the USSR Petroleum Industry (Currently Chairman of the State Committee of the USSR Council of Ministers for Chemistry)]. Present State of the Petroleum Industry and Scientific Research Problems in the Field of Petroleum Chemistry

5

Mamedaliyev, Yu. G., Academy of Sciences, Azerbaydzhanskaya SSR. Organic Synthesis Based on Hydrocarbons of Petroleum

25

Dobryanskiy, A. F., Leningradskiy gosudarstvennyy universitet im. A. A. Zhdanova (Leningrad State University imeni A. A. Zhdanov). Conversions of Hydrocarbons at Low Temperatures as the Cause of the Diverse Types of Petroleum

61

Bogomolov, A. I., K. I. Panina, and L. I. Khotyntseva, Vsesoyuznyy nauchno-issledovatel'skiy geologorazvedochnyy institut (All-Union Scientific Research Institute for Geological Exploration). Catalytic Conversions of Acids in Contact With Aluminum Silicates (Aspect of the Problem

~~Card 3/7~~

DOBRYANSKIY, A. F.

24004      DOBRYANSKIY, A. F.. Opyt analiza nekotorykh polozheniy Gipotezy  
prevrashcheniya neftey na materiale vtorogo Baku. Trudy Vsesojuz.  
Neft. Nauk.-Issled. Geol.-razved. IN-TA, Novaya seriya, VIP,  
28, 1949, S. 5-27.

SO: Letopis, No. 32, 1949.

(A)

Transformations of petroleum in nature as the cause of variation of its properties. A. F. Dobryanski. Vestnik Leningrad Univ. 1950, No. 1, 31-8. "Variations" of the properties of petroleum are explainable by the natural changes which occur during submergence in the ground. Since chem. changes can occur in petroleum in the feasible range of 150-300°, such reactions as H disproportionation, radical shifts in cyclic structures, and decompr. of larger mols. under the influence of catalytically active clays also become feasible. The formation of methane-type petroleum, naphthalene types, and those contg. aromatic compds. are readily explainable on the above basis. The oldest geological formations are low in petroleum and the Tertiary deposits appear to be the richest. The industrial refining of petroleum is seen as a parallel of similar natural processes of transformations along chemically similar lines.  
G. M. Kostanoff

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15

Thermal condensation of acetylene in the presence of porous fillers. A. V. Dobryanski and A. E. Drubkin (Lensovet Technol. Inst., Leningrad). *Zhur. Obshchei Khim.* (J. Gen. Chem.) 20, 2255-60 (1950).—Passage of C<sub>2</sub>H<sub>2</sub> through a heated porcelain tube produces condensation of C<sub>2</sub>H<sub>2</sub> already at 400° and reaches a max. (65.8%) at 700°, above which tar formation predominates. At 400° no aromatic hydrocarbons form, but at 500° or above, progressively greater amts. of C<sub>6</sub>H<sub>6</sub> appear, while at 600° small amts. of PhMe and xylenes appear. The optimum contact time at 700° is 12 sec. If the tube is filled with porcelain rings (45 vol.-%) no significant differences arise. A filler of silica gel (76 vol.-%) causes a decline of the catalyst; thus at 700° only 16.9% condensate forms, and neither contact time nor temp. variation causes a rise of yield. An activated charcoal filler gives a similar result and at 700° all C<sub>2</sub>H<sub>2</sub> is decompl. (0.0% at 600°), giving 50.66% gaseous products and only 0.03% condensate, along with much C deposition (about 30%). Apparently the porous fillers cause rapid condensation, yielding high-boiling products that gradually fill the pores of the filler, after which the process resembles that obtained with an open tube. The C deposit is not catalytic. The pore-closing takes place after 10-20 hrs. of operation.

G. M. Kosolapoff

1951

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The thermal condensation of acetylene in the presence of  
porous fillers. A. V. Dohryanski and A. S. Drabkin.  
*J. Gen. Chem. U.S.S.R.* 20, 2345-51 (1950) (Engl. transla-  
tion).—See C.A. 45, 61004d. B.I.M.

DOBRYANSKIY, A.F.; GARVILOV, B.G.

Thermal conversion of alkyl benzenes. Uch.sap.Len.un. no.155:261-  
269 '52.  
1.Kafedra tekhnicheskoy khimii.  
(Benzene)

Dobryanskiy, A. I.

Catalytic cleavage of *syn*-diphenylethane under the influence of aluminum chloride. A. I. Dobryanskiy and Yu. L. Kosolapova (Leningrad Technol. Inst., Leningrad). Sov. Fiz. Khim. Oshchel. Khim., Akad. Nauk S.S.R., 1, 311-14 (1953).—Heating ( $\text{C}_1\text{I}_4\text{Pn}$ ) (I), with  $\text{AlCl}_3$  to 230-50° or 340-50° results in 2 simultaneous processes: cleavage and condensation. I undergoes solely the unsym. cleavage, regardless of conditions, yielding  $\text{C}_6\text{H}_5$  and condensation products of linear structure. The yields depend on conditions. If the low boiling material is not removed continuously, the yield of distillable material (mostly  $\text{C}_6\text{H}_5$ ) declines. The yield of distillable material rises with increase of catalyst, even, from 1% to 10%. Oxidation of the condensation products gave  $\text{BzOH}$  and  $p\text{-C}_6\text{H}_4(\text{CO}_2\text{H})$ .

G. M. Kosolapov

Dobryanskiy, A.F.

Condensation of ethylene chloride with aromatic hydrocarbons in the presence of aluminum chloride. A. N. Dobryanskiy and Yu. I. Kornilova (Leningrad Technical Institute, Leningrad). Sbornik Statei Obshchel Khim., Akad. Nauk S.S.R. 1, 815-19 (1953).—The reaction of  $(\text{CH}_2\text{Cl})_2$  with  $\text{C}_6\text{H}_6$  in the presence of  $\text{AlCl}_3$  yields  $(\text{PhCH}_2)_2$  and related or condensation products.  $(\text{PhCH}_2)_2$  (90 g.) and 234 g.  $\text{C}_6\text{H}_6$  with 10 g.  $\text{AlCl}_3$  gave, after the usual acid treatment, 55%  $(\text{PhCH}_2)_2$ , b.p. 162-8°, and 40 g. product, b.p. 155-300°. Some higher boiling residue was left. Repeated distn. gave  $\alpha\text{-C}_6\text{H}_4(\text{CH}_2\text{CH}_2\text{Ph})_2$ , b.p. 212°, m. 47-8.0°, and  $(\rho\text{-PhCH}_2\text{CH}_2\text{C}_6\text{H}_4\text{CH}_2)_2$ , b.p. 270-8°, m. 78-85° (possibly a mixt. of isomers). Oxidation gave terephthalic acid and  $\text{Bi}_2\text{O}_3$ . The same oxidation products were obtained from the tarry distn. residue. Similarly  $\text{MePh}$  gave up to 65% ditolylethene, b.p. 142°, d<sub>4</sub> 0.9031, which oxidized with  $\text{KMnO}_4$  to terephthalic and biphenyl acids, with traces of toluic acids, thus indicating the formation of *m*- and *p*-isomers in the condensation. The higher boiling products yielded 1,3-di[2-(*p*-methylphenyl)ethyl]-5-methylbenzene, b.p. 229-8°, m. 63-6. G. M. Kosolapoff

DOBRYANSKIY, A.F.

The action of aluminum chloride on esters of dibasic acids. A. F. Dobryanskiy and Yu. I. Kornilova. *Sbornik Statei Obshchel Akademii Nauk S.S.R. 1, 320-1(1953).*  $\text{AlCl}_3$  (5 g.) and 10 g.  $\text{CH}_3(\text{CO}_2\text{Et})_2$  heated on a steam bath 7-10 min. until the reaction commenced and the mixture allowed to stand until gas ( $\text{EtCl}$ ) evolution ceased gave a spongy yellow mass, which, extd. with  $\text{Et}_2\text{O}$ , yielded a residue of *Al malonate*,  $\text{C}_4\text{H}_4\text{O}_4\text{Al}_2$ , which was extremely hygroscopic. Similarly 4.7 g.  $\text{AlCl}_3$  and 23 g.  $\sigma\text{-C}_6\text{H}_4(\text{CO}_2\text{Bu})_2$  yielded after the above-described treatment an unstated amount of *Al phthalate*,  $\text{C}_8\text{H}_8\text{O}_4\text{Al}_2$ , a very hygroscopic solid, along with  $\text{BuCl}$ .

*M. J. G.*

D.OBRYANSKLY, A.F.

Oxidation of aromatic hydrocarbons by esters of dicarboxylic acids. A. F. Dobryanski and Yu. I. Kurnilova, *Soviet Osnches Khim., Akad. Nauk S.S.R.*, 1, 329-4 (1953).—Heating 36 g.  $\text{AlCl}_3$ , 40 g.  $\text{C}_6\text{H}_6$ , and 30 g.  $(\text{CO}_2\text{Et})_2$  until the reaction started (total duration about 0.5 hr.) followed by aq. treatment of the mixt. after cessation of gas evolution, gave 10 g.  $\text{EtPh}$ . Similarly 20 g.  $\text{MePh}$  gave 10 g. mixed  $m$ - and  $p$ - $\text{MeC}_6\text{H}_4\text{Et}$ , with predominance of the former (as shown by oxidation to the dicarboxylic acids). Com. xylene similarly gave 5-ethyl- $m$ -xylene, b. 184-7°, which on oxidation gave pure trimelic acid, m. 348-51°. A similar reaction with  $\text{CH}_3(\text{CO}_2\text{Et})_2$  and  $\text{C}_6\text{H}_6$  gave very little  $\text{EtPh}$ .

G. M. Kosolopoff

MA Jan

DOBRYANSKIY, A. F.

Catalytic decomposition of symmetric ditolylethane under the influence of aluminum chloride. A. F. Dobryanskiy and Yu. I. Kornilova (Lenovet Technol. Inst., Leningrad); Sbornik Sistem Osnicheskoi Khim. Akad. Nauk S.S.R. 1; 325-0 (1953).—Heating mixed 1,2-di(*m*-tolyl)ethane and 1,2-di(*p*-tolyl)ethane with AlCl<sub>3</sub> to 230–50° led to cleavage of the former in 2 directions. Predominant reaction was cleavage of MePh<sub>3</sub>; a lesser reaction was the cleavage of xylene (*m*- and *p*-isomers as identified after oxidation to the acids). The extent of the reaction rises with time, and with duration, as well as with increase of the proportion of AlCl<sub>3</sub> used. Attempts to oxidize the high-boiling products failed to yield any conclusive results. G. M. Kiselevoff

DOBRYANSKIY, A.P.

Thermocatalytic conversions of hydrocarbons. Part 1. Conversions of phenylcyclohexane. Zntr. ob. khim. 23 no. 7:1116-1119 Jl '53. (MLRA 6:7)

1. Kafedra tekhnicheskoy khimii Leningradskogo Gosudarstvennogo universiteta.  
(Cyclohexane) (Catalysis)

DOBRYANSKIY, A.F.; GAVRILOVA, E.K.

Thermocatalytic conversions of hydrocarbons. Part 2. Conversions of  
tertiary-butyl benzene. Zhur. ob. khim. 23 no.7:1118-1119 Jl '53.  
(MLRA 6:7)

I. Kafedra tekhnicheskoy khimii Leningradskogo Gosudarstvennogo universi-  
teta.  
(Butyl benzene) (Catalysis)

DOBRYANSKY, A.F.

△ Thermocatalytic transformations of hydrocarbons. III.  
Transformations of diisobutylmethane. A. F. Dobryanskiy  
and P. N. Kolomitshev (Lensovet Technol. Inst., Lenin-  
grad). *Zhur. Obrabotki Khim.* 23, 1360-70 (1953); cf. *C.A.*  
47, 12214c.— $\text{Ph}_2\text{CH}_2$  was heated, either with distn. of the  
volatiles or in closed autoclave at 200°, with activated gum-  
beet catalyst. The products were the same in both cases;  
the reaction yielded  $\text{C}_6\text{H}_5$  and mainly  $p\text{-C}_6\text{H}_4(\text{CH}_2\text{Ph})_2$  (I),  
 $b_1 243$ - $50^\circ$ , m.  $88^\circ$ .  $\text{C}_6\text{H}_5$  and I begin to form even at 120°.  
No  $\text{MeC}_6\text{H}_4$  is formed. In a closed app., equil. is established;  
at 120° in 10 hrs. the mixt. contains 2.2%  $\text{C}_6\text{H}_5$ ; at 150°  
0%, at 170° 21%. When I is heated with  $\text{C}_6\text{H}_5$  and the  
catalyst in a closed autoclave at 170° a similar equil. mixt.  
forms with indications that the catalyst can be readily re-  
cycled without loss of activity. G. M. Kosolapoff

(1)

DOBRYANSKIY, A.F.; PONOMAREV, L.A.; DYBKIN, L.D.

Thermocatalytic conversions of hydrocarbons. Part 4. Conversions of diphenyl-  
ethane and ditolylethane. Zhur. ob. khim. 23 no.10:1632-1635 O '53.  
(MLRA 6:11)

1. Leningradskiy Gosudarstvennyy universitet.

(Ethane)

DOBRYANSKIY, A.F.

The work of D.I. Mendeleev in the chemistry of petroleum.

A. F. Dobryanskiy. *Vestnik Leningrad. Univ.*, 9, No. 11,

Ch. Ser. Mat., Fiz., i Khim. No. 3, 107-75 (1951).—A brief review and discussion of Mendeleev's work and hypotheses concerning the origin of petroleum. — I. A. K.

DOBRYANSKIY, A. F.

AID - P-103

Subject : USSR/Chemistry

Card : 1/1

Authors : Dobryanskiy, A. F., and Zvyagintsev, O. Ye.

Title : Bibliography

Periodical : Zhur. Prikl. Khim. 27, no. 4, 466-468, 1954

Abstract : A book Chemical Refining of Petroleum by P. Goldstein (translated into Russian by N. S. Dabagov) is reviewed. Some articles from Transactions of the Scientific Research Institute of Glass, no. 32, 1953. Transactions of the Moscow Institute of Fine Chemical Technology im. M. V. Lomonosov, no. 3, 1952, and Transactions of the Khar'kov Polytechnic Institute im. V. I. Lenin. Vol. 1, Chemical-technological Series, no. 1, 1952 are listed.

Institution : None

Submitted : No date

DOBRYANSKIY, A.F.

Catalytic conversion of paraffin and ceresin over gamma-  
A. F. Dobryanskiy and G. Ya. Vorob'eva. *J. Appl. Chem.*  
U.S.S.R. 27, 689-93 (1954) (Engl. translation).—See C.A.  
48, 12304f.

B. M. R.

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URKOVINSKIY, A.

U.S.R.

1953. ANALYTIC TRANSLATION OF THE SIGHTING OF  
LICHENSKI, A.F. AND VOROBIEVA, G.YA. IN THE U.S.S.R. DURING  
THE 1953 U.S. AIR FORCE SURVEY OF THE CHUKOTKA PENINSULA.  
TRANSLATED BY: R. J. BROWN, JR., 625-607, 1000, 1000, 1000,  
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Andreyev, Pavel Fedorovich; Bogomolov, Aleksey Ivanovich; Dobryanskiy,  
Aleksandr Flavianovich; and Kartsev, Aleksey Aleksandrovich

Prevrashcheniya nefti v prirode (Conversion of Petroleum in Nature)  
Leningrad, Gostoptekhizdat, 1958. 416 p. 3,100 copies printed.

Ed.: Dobryanskiy, A.F.; Executive Ed.: Chizhov, A.A.; Tech. Ed.:  
Yashchurzhinskaya, A.B.

PURPOSE: This book is intended for specialists in geochemistry and  
petroleum geology.

COVERAGE: The book gives a systematic approach to problems related to  
the transformations of present-day petroleum deposits as systems  
of active substances. A.F. Kartsev wrote Chapters I, II and V  
(pt.1); P.F. Andreyev - Chapters III, IV and V (pt.2), A.I Bogo-  
molov - Chapters VI and VII; A.F. Dobryanskiy - Chapters VIII  
and IX. References are given at the end of each Chapter.

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## Conversion of Petroleum in Nature

1178

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