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GAILATNO, T. M., MADVADAV, J. S.

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"Initiation of emulsion copolymerization with encyllydroperoxide," a paper presented at the 9th Congress on the Chemistry and Physics of Nigh Polymors, 20 Jan-2 Feb 57, MoscoW, Karpev, Inst.

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B-3,024,395

GRIZEHKAV, V.M. [Hryzenkov, V.M.]. agronom-ekonomiat Effectiveness of growing vegetables in greenhouse heated by electric lamps. Makh. sil'. hosp. 9 no.10:10-11 0 '58. (WIRA 11:10) (Vegetable gardening) (Greenhouse management)

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23332 3. C. B. + 1. C. C. M. G. S. + 1663 +CO. 21. 24.6600 (1057,1482) Griznko, V.M., Sikera, D.I., Shkeda-Ulivanov, J.A., Atearetkov, A.D., Parlag, A.M., Shramenko, B.I., Fisur, A.N. AUTHORS: An arrempt to determine oncess sectors, of The new time in lead by using a very thick target and a month-watge to whether the TITLES Referativnyy zhurnal Fizika, no. 6, 1961, 96, approach 68392 ("Dekl. 1 sonbanch, Uzhgoredak, un-t. Ser (tiz. matem, v. ", 1960, no.3, 1...) PERIODICAL The authors discuss preliminary results of taleilations of the cross section of reaction (T, n) in PD from the dita, obtained by them earlier, on the yield of photoneutrons from a very thick lead target using a monoenergetic electron beam (RZhFiz, 1961, 18471). The authors are of the opinion that the accuracy of reproducioility of $\mathcal{C}(\mathcal{Y}, n)$ in the region > 15 MeV is by no means worse than in the region of lower energies. They point out that the method of "difference of photons", which was apulted formerly for calculations of the president of the the acturaly by 20 30% popror in the region of energies count the grant real schance, this can live the smoothing out of a consister sectrolary maximum. The Card 1/2 ereterne va



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TITLE:	Precision monitor of electron beams		
PERIODICAL:	Referativnyy zhurnal. Fizika, no. 7, 1961, 43, abstract 7B84 ("Dokl. i soobehch. Uzhgorodsk. un-t. Ser. fizmatem.n", 1960, no. 3, 5-7)		
fier with 100 and the block	The authors describe a monitor of electron beams which represents Faraday cylinder connected with an integrating circuit (d-c ampli- % negative feedback). A detailed description of the monitor design c-diagram of the integrator are presented. The test of the monitor Linear accelerator has shown that the precision of monitoring the m amounts to 0.7%, and the errors arising due to ionization currents i 0.05%.	ЪВ	
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[Abstracter'	s note: Complete translation]		1
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TITLE: Determination of the Yield of Photoneutrons From Lead Under the Action of Electrons Having Energies From 10.4 to 20.5 Mev (Method of Thick Absorber)	· .	8356 2	
ADTHORS:Abramenkov, A. D., Shramenko, B. L. Pisun A. N.TITLE:Determination of the Yield of Photoneutrong Prom Lead Under the Action of Electrons Having Energies From 10.5 to 20.5 Mev (Method of Thick Absorber)PERIODICAL:Zhurnal eks, erimental nog i teoreticheskey fizike 1960. Vol. 38, No. 5, pp. 1370-1373TEXT:In an earlier publication (Ref. 1), some of the authors have cal- culated the photoneutron yield for some elements with the self of the Studied the yield of photoneutrons from a lead block that is prottically studied the yield of photoneutrons from a lead block that is prottically 	6.224		
the Action of Electrons Having Emergies Final for the 20.5 Mev (Method of Thick Absorber) PERIODICAL: Zhurnal eksierimentalindy i teoreticheokey fizike 1960. Vol. 38; No. 54 pp. 1370-1373 TEXT: In an earlier publication (Ref. 1), some of the authors have cal- culated the photoneutron yield for some elements with the delp of the Belen'kiy-Tamm equilibrium spectrum. Now the authors have experimentally studied the yield of photoneutrons from a lead block that to practicular of infinite thickness and absorbs the monochromatic electron beam for pletely, and compared the results with those of the theiry. The prosent paper describes this work. The experimental method is essentially that suggested by V. I. Guldanskly and V. A. Shkuda Ullyanov. The experiments, arrangement is schematically shown in Fig. 1; the beam catcher on the used simultaneously as a monitor of the electron beam and as the source of	6.2 240 AUTHORS:	Abramenkov, A. D., Shramenko, B. L., Pisun as house	
Vel. 38, No. 5, pp. 1370-1375 TEXT: In an earlier publication (Ref. 1), some of the authors have cal- culated the photoneutron yield for some elements with the self of the Belen'kiy-Tamm equilibrium spectrum. Now the authors have experimentally studied the yield of photoneutrons from a lead block that to practically of infinite thickness and absorbs the monochromatic electron beam com pletely, and compared the results with those of the theory. The proper paper describes this work. The experimental method is essentially that suggested by V. I. Goldanskly and V. A. Shkuda Ulyanov. The experimente- arrangement is schematically shown in Fig. 1; the beam catcher and be used simultaneously as a monitor of the electron beam and as the source of	TITLE;	the Action of Electrons Having Energies ring for the 20.5 Mev (Method of Thick Absorber)	
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tion threshold The experiment of Ref. 9. An tion yielded t the cross sect A. S. Litvinen <u>Comonay</u> , and A and <u>I. A. Cris</u> and V. I. Gol!	tor was calibrated for absolute of for oxygen and carbon according al results agree better with those estimate of the integral photoneu he value 2.6 b. Mev. For this est ion reaches its maximum value for ko, A. I. Charkin, V. A. Skubko, . M. Parlag for their assistance hayev for their interest and disc danskiy for their advice. There a Soviet and 6 US.	to an activation method. se of Ref. 10 than with those atron production cross sec- timate, it was assumed that r 13.8 Mev. The authors thank V. L. Auslender, V. I. in the work; A. K. Valter cussions; and L. Ye. Lazarev	-
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ASSOCIATION:	Fiziko-tekhnicheskiy institut (Institute of Physics and Techn Sciences Ukrainskaya SSR). Uzh universitet (Uzhgorod State Un	nology of the Academy of gorodskiy gosudarstvennyy	

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s/058/63/000/001/015/120 A062/A101 24.6730 Sinel'nikov, K. D., Grishayev, I. A., Grizhko, V. M., Fisun, A. N., AUTHORES Zykov, A. I., Kitayevskiy, L. Kh. A 30 MeV energy linear travelling-wave electron accelerator TITI :: PERCEINSE Referativnyy zhurnal, Fizika, no. 1, 1963, 39 - 40, abstract 1A374 (In collection: "Elektron. uskoriteli." Tomsk, Tomskiy un-t, 1961, 3 - 9) The authors describe a 30 MeV linear electron accelerator designed TWEE at the Physico-technical Institute of the Academy of Sciences of the Ukrainian MUNE. The accelerator consists of two sections connected with each other - the injector section and the main section (with a constant wave phase speed); the length of the main section is 2.8 m, the value ka = 2.48 (k - wave vector, a -- waveguide radius). The two sections are energized by one klystron power amplifier, excited by a magnetron generator. The power dissipated in the main section and in the output load is ~10 Mw (in the load 3.3 Mw); the field intensity is then 100 kV/em, The accelerating system is composed of separate resonators; the Card 1/2

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A 30 FeV energy linear travelling-wave...

electrical contact between them is realized by mechanical ties in the places where the cystem is connected to the input and output matching transformers. The resonations of the main section are disposed tightly in a copper tube which is also a vacuum housing. The precision of manufacture of the accelerating system (diameter of the resonators and disphragm apertures) is ± 0.01 mm. The source of electrons is an electron gun operating under the tension of 79 kV (the corresponding electron velocity is 0.5c); the current is 1 amp. in a pulse. The pumping out of the vacuum volume of the accelerator is effected by 5 diffusion pumps; the operating pressure in the klystron amplifier is 2.10^{-7} mm Hg, in the remaining space 3.1 ± 10^{-7} mm Hg. Measurements have shown that the maximum intensity and emergy are attained in the accelerator at the frequency 2796 Mc/s. The mean current of the percent (at the output) under the optimum focusing is 3 - 4 mm, the spectrum withh $= \frac{25}{2}$.

A. Fateyev

[Abstructor's note: Complete translation]

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	FSD(gs)/AEDC(a)/ESD(t)/AFWL ACCESSION NR: AP4046356 8/0C57/64/034/010/1903/1905		•
	AUTHOR; Grizhko, V.M.; Vishnyakov, V.A.; Grishayev, I.A.; Yeremenko, Ye.V.; Kuznet-		
1	sov, G.F.; Ostrovskiy, Ye.K.; Khvorostenko, Y.I.		•.
	TITLE: A 40 MeV linear electron accelerator 19		•
	SOURCE: Zhurnal tekhnicheskoy fiziki, v.34, no.10, 1964, 1903-1905		
	TOPIC TAGS: linear accelerator, electron accelerator		
· · · · · · · · · · · · · · · · · · ·	ABSTRACT: The authors briefly describe a linear accelerator which, operating at 2797.2 Mc/sec, produces 1.5 microsec, 80 mA pulses of 40 MeV electrons at repetition rates of up to 50/sec. The electrons are produced in a two-electrode gun with a tam talum cathode and are accelerated to 5 MeV in an 83 cm long injector containing an experimentally adjusted longitudinal magnetic field for focusing. The principal accelerator is a 450 cm long constant phase velocity iris waveguide. Each of the two sections is fed through a 72 x 34 mm ² vacuum waveguide by a 20 megawatt klystron amplifier, each excited by the same magnetron oscillator. The working vacuum of bet ter than 5 x 10 ⁻⁶ mm Hg is maintained by a battery of titanium pumps. The beam energy can be smoothly varied from 5 to 40 MeV by varying the power supplied to the		
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GRIZIK, A. A., BYKCVSKAYA, Yu. I., and MARUNINA, N. I.

"Use and methodology of radioactive indicators."

report presented at The Use of Radioactive Isotopes in Analytical Chemistry, Conference in Moscow, 2-4 Dec 1957 Vestnik Ak Nauk SSSE, 1958, No. 2. (author Rodin, S. S.)

	507/81-59-10-34412
Translation	from: Referativnyy zhurnal. Khimiya, 1959, Nr 10, p 85 (USSR)
AUTHORS:	Plyushchev, V.Ye., Kuznetsova, T.P., Grizik, A.A.
TITLE:	The Study of the Ion-Exchange Capacity of the <u>Cationites</u> ' SBG, MSF, KU-1, <u>KN</u> and <u>RF</u> in Solutions of Chlorides of Alkali Metals
PERIODICAL:	Tr. Mosk. in-ta tonkoy khim. tekhnol., 1958, Nr 7, pp 73-80
AESTRACT:	The absorption of alkali metals by H-forms of the resins SBS, MSF, Ku-1 and RF at various pH of the initial solution (in a non-buffer system) has been studied under static conditions. It is assumed that for industrial con- ditions these data characterize the ionite better than the dependence of the absorption on the pH of the equilibrium solution.
	M. Arkhangel'skiy
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<u>. 33 5. (1480), 33 6. (178 - 118 667</u> 667	แรงและและและและและและและและและและและและและแ

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SOV/117 59-2-4052 Translation from: Referativnyy zhurnal. Metallurgiva, 1959, Nr.2 p.247 (USSR) AUTHORS: Grizik, A. A. Marunina, N. L. Methodological Problems in the Use of Radioactive Tracers for the TITLE: Control of Processes of the Recovery of Rare Metals (Metodicheskiye voprosy primeneniya radioaktivnykh indikatorov dlya kontro'ya tekhnologii polucheniya redkikh metailor) Tr. Komis, po analit, khimil, AN SSSR, 1958, Vol 9 (12), pp 433-PERIODICAL: 340 The authors examine some methodological problems that atise in ABSTRACT: the use of radioactive tracers (RT) for the control of processes of the recovery of rare metals and examine in detail the advantages and shortcomings connected with the application of RT. Recommendations are made for the selection of methods for the mass analysis of specimens for Ta and Nb with a view of decreasing the consumption of these materials. The design of a device for measuring the activity at different points of the ingot examined and the results of experiments conducted in the investigation of distribution of Sb along a Ge single crystal which was obtained by drawing from a melt are described. An extensive Card 1/2

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lethodological Problems in the Use of Radioact	tive Tracers for the Control (cont.)
pplication of the autoradiographic method is no f the homogeneity of alloys based on rare meta escribed. It is pointed out that the method of i or obtaining correct control results. The trac ame chemical state in the feed substance. The uction of RT is selected for each particular ca ntroduction of RT into niobium pentoxide, whic	introduction of RT is ery important er and the admixture should be in the erefore a specific method of intro-
f metallic Nb.	Z. F.
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AUTHORS: <u>Grizik, A.A., Piyushchev, V.Ye.</u> TITLE: On the Synthesis of <u>Lithium Metazirconate</u> and Metahafnate FERIODICAL: Zhurnal vsesoyuznogo khimicheskogo obshchestva im.D.I. Mendeleyeva, 1960, Vol. 5, No. 3, PP. 349 - 350 TEXT: There are literature data (Refs 2-8) on the existence of three lithium zirconates, but the existence of only one, viz., Li20'ZrO2 was proved. This zir- zirconates, but the existence of only one, viz., Li20'ZrO2 was proved. This zir- used here to obtain pure zirconates, and their properties were studied. ZrO2 (with a content of 0.03% HfO2) and HfO2 (with a content of < 1% ZrO2) were employed as initial materials. Chemically pure Li2CO3 was used instead of Li2O. It de- as initial materials. Chemically pure Li2CO3 was used instead of Li2O. It de- 1,200°C. Above 900°C alundum crucibles were used, because platinum vessels are 1,200°C. Above 900°C alundum crucibles were used, because platinum vessels are destroyed. The analysis of the reaction products showed that the composition of the products obtained at a Li2O:MeO2 ratio from 1:1 to 1:4 corresponds to the for- mula Li2MeO3. Pure meta-compounds are obtained in the interaction of MeO2 and Li2CO3 (taken in the ratio Li2O:MeO2 = 1:1) at 1,100± 20°C. The analysis by the Card 1/2			83254 5/063/60/005/003/002/003 A003/A001
TITLE: On the Synthesis of <u>Lithium Metazirconate</u> and Metahafnate FERIODICAL: Zhurnal vsesoyuznogo khimicheskogo obshchestva im.D.I. Mendeleyeva, 1960, Vol. 5, No. 3, pp. 349 - 350 TEXT: There are literature data (Refs 2-8) on the existence of three lithium zirconates, but the existence of only one, viz., Li20'ZrO ₂ was proved. This zir- conate is used in the silicate industry (Ref 9). High-temperature synthesis was used here to obtain pure zirconates, and their properties were studied. ZrO ₂ (with a content of 0.0% HfO ₂) and HfO ₂ (with a content of $< 1\%$ ZrO ₂) were employed as initial materials. Chemically pure Li ₂ CO ₃ was used instead of Li ₂ O. It de- somposes at high temperatures to Li ₂ O. The experiments were carried out at 800- 1,200°C. Above 900°C alundum crucibles were used, because platinum vessels are destroyed. The analysis of the reaction products showed that the composition of the products obtained at a Li ₂ O:MeO ₂ ratio from 1:1 to 1:4 corresponds to the for- mula Li ₂ MeO ₃ . Pure meta-compounds are obtained in the interaction of MeO ₂ and Li ₂ CO ₃ (taken in the ratio Li ₂ O:MeO ₂ = 1:1) at 1,100± 20°C. The analysis by the	5.2100		
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FERIODICAL: Zhurnal vsesoyuznogo khimicheskogo obshchestva im.D.I. Mendeleyeva, 1960, Vol. 5, No. 3, pp. 349 - 350 TEXT: There are literature data (Refs 2-8) on the existence of three lithium zirconates, but the existence of only one, viz., Li20'ZrO2 was proved. This zir- conate is used in the silicate industry (Ref 9). High-temperature synthesis was used here to obtain pure zirconates, and their properties were studied. ZrO2 (with a content of 0.03% HfO2) and HfO2 (with a content of $< 1\%$ ZrO2) were employed as initial materials. Chemically pure Li2CO3 was used instead of Li20. It de- composes at high temperatures to Li20. The experiments were carried out at 800- 000°C. Above 900°C alundum crucibles were used, because platinum vessels are 1,200°C. Above 900°C alundum crucibles were used, because platinum vessels are destroyed. The analysis of the reaction products showed that the composition of the products obtained at a Li20:MeO2 ratio from 1:1 to 1:4 corresponds to the for- mula Li2MeO3. Pure meta-compounds are obtained in the interaction of MeO2 and Li2CO3 (taken in the ratio Li20:MeO2 = 1:1) at 1,100± 20°C. The analysis by the		On the Synthesis of Lithium Metazirconate	and Metahafnate
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Card 1/2	conate is use used here to (with a conte as initial ma composes at 1 1,200°C. Ab destroyed.	There are literature data (Refs 2-8) on the but the existence of only one, viz., Li20'Z bed in the silicate industry (Ref 9). High- boobtain pure zirconates, and their properti- tent of 0.03% HfO ₂) and HfO ₂ (with a content materials. Chemically pure Li ₂ CO ₃ was used high temperatures to Li ₂ O. The experiments bove 900°C alundum crucibles were used, beca The analysis of the reaction products showe s obtained at a Li ₂ O:MeO ₂ ratio from 1:1 to	temperature synthesis was es were studied. ZrO_2 of $\langle 1\% ZrO_2 \rangle$ were employed instead of Li20. It de- were carried out at 800- nuse platinum vessels are ed that the composition of 1:4 corresponds to the for-
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weight method 80.82. Li ₂ ZrO 87.52. Li ₂ HfO white color wi were determine 6.45 <u>3+</u> 0.003. is hydrolyzed in diluted (1. X-rays cause a references:	showed the following results (weight %)found: Li20 19.54; ZrO2 by deloulated Li20 19.52; ZrO2 80.48. Found LiO2 12.50; HfO2 calculated Li20 12.43; HfO2 87.57. Crystalline powders of the a high refractive index are obtained. The following densities by the picnometric method (200C): Li2ZrO3 4.12340.003; Li2HfO3 Both compounds melt above 1,500°C and are not hygroscopic. Li2MeO3 by water. Metazirconate and metahafnate are decomposed by boiling (1) HCl and H2SO4, concentrated H2SO4 decomposes them in the cold. Soviet, 3 English, 3 German, 2 American and 1 Canadian.
ASSOCIATION:	Meskevskiy institut tenkey khimicheskoy tekhnologii im.M.V.Lemeneseva (Moscow Institute of Fine Chemical Technology imeni M.V. Lomenesev)
SUBMITTED:	November 1 ⁴ , 1959
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15.2230	B121/B101
UTHORS:	Grizik, A. A., Plyushchev, V. Ye
TITLE:	Lithium meta-zirconate and lithium meti-hafrate
PERIODICAL:	Zhurnal neorganicheskoy khimii, v. 6, no. 10, 1961, 2249-2253
	hod for preparing lithium moto (Ref. 12: A. A. Grizik and previously been described (Ref. 12: A. A. Grizik and hchev: Zh. Vsesoyuzn, khim, obshch-va im, Mendeleyeva 5, 349 me properties of these compounds are discussed in the present me properties of these compounds are discussed in the present

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not hygroscop water than li the pH with an solubility of hydrochloric a acids decompo- hafnate are s tables and 17 recent refere J. Schenck. N Bull., 27, 49 (1954).	ic. Lithium meta-zirconate thium meta-hatnale. The tes n MR-5 (LP-5) potentiometer lithium meta-zirconate and m acids was studied. Concents se lithium meta-hafnate. L table in alkaline solutions references: 6 Soviet and nees to English-language pul ucleonics, 10, N8, 54 (1952 2 (1948); C. E. Curtis et a	neta nainate in sulluric and rated hydrochloric and sulfu- ithium meta-zirconate and me (0.1 - 10 N KOH). There are 11 non-Soviet. The three mo	ric taa J gt
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21.2000 AUTHORS: <u>Crizik. A. A.</u>, Plyushchev, V. Ye. TITLE: Synthesis and properties of sodium zirconate and hafnate PERIODICAL: Zhurnal neorganicheskoy khimii, v. 7, no. 5, 1962, 1054-1061 TEXT: To study the formation of compounds in the system Na₂O - ZrO₂ the reaction of Na₂CO₃ with ZrO₂ between 800 and 1400°C was investigated, the Na₂CO₃/ZrO₂ ratio being varied between 1:4 and 4:1. The mixtures of the initial substances powdered to 200 mesh were pressed into tablets and heated. Only the compound Na₂ZrO₃ was found to form, which was detected by Debye pattern after 1-hr keeping at 800°C. The Na₂O sublimation observed confirmed a direct reaction in the system Na₂O - ZrO₂. The use of Na₂CO₃ yielded no pure metazirconates. At too low a temperature, the product contained Na₂O, at too high a temperature, the metazirconate Card 1/3

Synthesis and properties of sodium	S/078/62/007/005/008/014 B101/B110
decomposed. Thus, experiments with NaNO ₃ optimum conditions were found for the oron Na ₂ ZrO ₃ : heating of Na ₂ CO ₃ +ZrO ₂ , ratio 1. addition of 10% NaNO ₃ , and short heating Na ₂ O can be washed out with ethylene glyc was prepared analogously. Both compounds refractive indices are: 1.76 for Na ₂ ZrO ₃ , $d_{Na_2}^{2O} = 4.060\pm0.003$; $d_{Na_2HFO_3}^{2O} = 5.763\pm$	were made, and the following duction of coarse-crystalline 5 : 1, at 1200 - 1300°C, at 1000°C. The small content of ol or absolute alcohol. Na ₂ HfO ₃ show birefringence. The mean 1.78 for Na ₂ HfO ₃ ; 0.005. The interplanar spaces bebye patterns. Both compounds
of the two compounds were determined by hydrolyze quickly and completely in water 20° C, with formation of polyzirconates wh compounds Na ₂ Ke ^{IV} O ₃ decompose slowly owin CO ₂ , the metazirconate being more stable Card 2/3	at 100 C, but only signal air, the nich are being studied. In air, the ng to reaction with water vapor and

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 heating to 1300°C, decomposition of the metazirconate was about 12%, that of the metahafnate 37.3%. There are 7 figures and 2 tables.
 AssocIATION: Moskovskiy institut tonkoy khimicheskoy tekhnologii im. M. V. Lomonosova (Moscow Institute of Fine Chemical Technology imeni M. V. Lomonosov)

 SUBMITTED:
 June 1, 1961

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TEXT: The formation and properties of $K_2 0.2rO_2$ (I) were studied using the starting materials and procedure described in previous papers (Zh. neorgan. khimii, 7, 1962, 2095, 2086 and 1054). Separation of I from the trizirconate which forms equally in the $K_2 0 - 2rO_2$ system was achieved by exploiting their different behaviour in hydrolysis. At room temperature I is hardly hydrolized at all, being almost completely so at 100°C. Pure I was obtained for the first time from compounds with a molecular $K_2 0:2rO_2$ ratio of 1.5 : 1 and a large excess of free $K_2 0$ at 1000 - 1100°C by	•	5/078/62/007/012/006/022 B144/B180	
TITLE: Synthesis and some properties of potassium dizirconate PERIODICAL: Zhurnal neorganicheskoy khimii, v. 7, no. 12, 1962, 2702-2708 TEXT: The formation and properties of $K_2^{0.2rO_2}$ (I) were studied using the starting materials and procedure described in previous papers (Zh. neorgan. khimii, 7, 1962, 2095, 2086 and 1054). Separation of I from the khimii, 7, 1962, 2095, 2086 and 1054). Separation of I from the trizirconate which forms equally in the $K_2^{0} - ZrO_2$ system was achieved by trizirconate which forms equally in the K ₂ 0 - ZrO ₂ system was achieved by is hardly hydrolized at all, being almost completely so at 100°C. Pure I is hardly hydrolized at all, being almost completely so at 100°C. Pure is obtained for the first time from compounds with a molecular $K_2^{0:2rO_2}$ ratio of 1.5 : 1 and a large excess of free K ₂ 0 at 1000 - 1100°C by	រយាមលាន :	Grizik, A. A., Plyushchev, V. Ye., Pleskova, I. A.	
PERIODICAL: Zhurnal neorganicheskoy khimii, v. 7, no. 12, 1962, 2702-2700 TEXT: The formation and properties of $K_2^{0.2}TO_2$ (I) were studied using the starting materials and procedure described in previous papers (Zh. neorgan. khimii, 7, 1962, 2095, 2086 and 1054). Separation of I from the trizirconate which forms equally in the $K_2^{0} - ZrO_2$ system was achieved by exploiting their different behaviour in hydrolysis. At room temperature I is hardly hydrolized at all, being almost completely so at 100°C. Pure I was obtained for the first time from compounds with a molecular $K_2^{0:2}TO_2$ ratio of 1.5 : 1 and a large excess of free K_2^{0} at 1000 - 1100°C by sintering them for 1 hr at 1000°C, removing the free K_2^{0} with methanol and acetone, and drying in air at 50 - 70°C. $K_2^{0.2}TO_2$ forms white oblong	_	Cunthesis and some properties of potassium dizirconate	
starting materials and procedule cost 1054). Separation of I from the khimii, 7, 1962, 2095, 2086 and 1054). Separation of I from the trizirconate which forms equally in the $K_2O - ZrO_2$ system was achieved by exploiting their different behaviour in hydrolysis. At room temperature is hardly hydrolized at all, being almost completely so at 100°C. Pure I is hardly hydrolized at all, being almost completely so at 100°C. Pure I was obtained for the first time from compounds with a molecular $K_2O:ZrO_2$ I was obtained for the first time from the free K ₂ O at 1000 - 1100°C by ratio of 1.5 : 1 and a large excess of free K ₂ O at 1000 - 1100°C by	PERIODICAL:	Zhurnal neorganicheskoy khimii, v. 7, no. 12, 1962, 2702-2703	
Card 1/2	starting ma khimii, 7, trizirconat exploiting I is hardly I was obtai ratio of 1 sintering (acetone, a	terials and procedule 1054). Separation of I from the 1962, 2095, 2086 and 1054). Separation of I from the e which forms equally in the $K_2O - ZrO_2$ system was achieved by their different behaviour in hydrolysis. At room temperature hydrolized at all, being almost completely so at 100°C. Pure hydrolized at all, being almost completely so at 100°C. Pure ned for the first time from compounds with a molecular $K_2O:ZrO_2$.5 : 1 and a large excess of free K_2O at 1000 - 1100°C by .5 : 1 and a large excess of free K_2O at 1000 - 1100°C by	

S/078/62/007/012/006/022 Synthesis and some properties of B144/B180 prisms insoluble in water and the usual solvents. It dissolves complete when heated in dilute mineral acids. When heated to 900° C, it decompose into the trizirconate and free K ₂ O with a sharp drop in the free K ₂ O content during the first 30 min. X-ray analysis showed that I is thermostable only in the presence of free K ₂ O. Density at 20° C: 3.376 [±] O.003; refractive index: >1.78. The x-ray diffraction data are given. Disagreement with published data is attributed to the use of	ely ses
prisms insoluble in water and the usual solvents. It dissolves complet when heated in dilute mineral acids. When heated to 900° C, it decompose into the trizirconate and free K ₂ O with a sharp drop in the free K ₂ O content during the first 30 min. X-ray analysis showed that I is thermostable only in the presence of free K ₂ O. Density at 20° C:	el y ses
when heated in dilute mineral acids. When heated to 900° C, it decomposes into the trizirconate and free K ₂ O with a sharp drop in the free K ₂ O content during the first 30 min. X-ray analysis showed that I is thermostable only in the presence of free K ₂ O. Density at 20° C:	ses
when heated in dilute mineral acids. When heated to 900°C, it decompose into the trizirconate and free K_20 with a sharp drop in the free K_20 content during the first 30 min. X-ray analysis showed that I is thermostable only in the presence of free K_20 . Density at $20^{\circ}C$:	368
content during the first 30 min. X-ray analysis showed that 1 is thermostable only in the presence of free K ₂ O. Density at 20 ^o C:	
impure trisirconate in those works. There are 5 figures and 2 tables.	
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CIA-RDP86-00513R00051701



ID/WW/JG UR/0078/65/010/006/1312/1319 ACCESSION NR: AP5015014 UR/0078/65/010/006/1312/1319 AUTHOR: Grizik, A.A.; Plyushchev, V. Ye.; Kamenskaya, A. N. Image: Comparison of the second seco						
ID/WH/JG UR/0078/65/010/006/1312/1319 ACCESSION NR: AP5015014 UR/0078/65/010/006/1312/1319 AUTHOR: Grizik, A.A.; Plyushchev, V. Ye.; Kamenskaya, A. N. Image: Source: Shurnal neorganicheskoy khimii, v. 10, no. 6, 1965, 1312-1319 TOPIC TAGS: rubidium dizirconate, zirconium dioxide, x-ray phase analysis ABSTRACT: A systematic search for individual compounds in the system Rb2O-ZrO2 was undertaken in order to determine the interaction between the components and find out whether the separation of individual phases becurs. The reactions of ZrO2 with RbNO3 whether the separation of individual phases becurs. The reactions of Laropounds formed, X-ray phase analysis showed the formation of three different phases characterized by individual crystal lattices; one of them was identified as rubidium dizirconate Rb20. 2ZrO2, and its composition was confirmed by chemical analysis. The effect of the reaction temperature, duration of sintering, initial molar ratio of the components, and additional sintering on the extent of the reaction between ZrO2 and RbNO3 and on the composition of the products formed was investigated. Data were obtained on the hydrolyza-	L 59238-65 EPA(s)-2/EWT(m)/EPF(n)-2/T/EWP	(t)/EMP(b)/EWA(c)	Pt-7/Pu-4	IJP(c)	- - 5	
TITLE: Rubidium dizirconate SOURCE: Zhurnal neorganicheskoy khimii, v. 10, no. 6, 1965, 1312-1319 TOPIC TAGS: rubidium dizirconate, zirconium dioxide, x-ray phase analysis ABSTRACT: A systematic search for individual compounds in the system Rb ₂ O-ZrO ₂ was undertaken in order to determine the interaction between the components and find out whether the separation of individual phases becurs. The reactions of ZrO ₂ with RbNO ₃ (at 700-1200C) and Rb ₂ CO ₃ were carried out by sintering and fusion. In the case of Rb ₂ CO ₃ , the reaction was too weak to permit any conclusiono regarding the compounds formed. X-ray phase analysis showed the formation of three different phases characterized by individual crystal lattices; one of them was identified as rubidium dizirconate Rb ₂ O · 2ZrO ₂ , and its composition was confirmed by chemical analysis. The effect of the Rb ₂ O · 2ZrO ₂ , and its composition was confirmed by chemical analysis. The effect of the reaction temperature, duration of sintering, initial molar ratio of the components, and additional sintering on the extent of the reaction between ZrO ₂ and RbNO ₃ and on the addition of the products formed was investigated. Data were obtained on the hydrolyza-	JD/W/JO ACCESSION NR: AP5015014	UR/0078/65/01	10/006/1312			
SOURCE: Zhurnal neorganicheskoy khimii, v. 10, no. 6, 1965, 1312-1319 TOPIC TAGS: rubidium dizirconate; zirconium dioxide, x-ray phase analysis ABSTRACT: A systematic search for individual compounds in the system Rb ₂ O-ZrO ₂ was undertaken in order to determine the interaction between the components and find out whether the separation of individual phases beccurs. The reactions of ZrO ₂ with RbNO ₃ (at 700-1200C) and Rb ₂ CO ₃ were carried out by sintering and fusion. In the case of Rb ₂ CO ₃ , the reaction was too weak to permit any conclusions regarding the compounds formed. X-ray phase analysis showed the formation of three different phases characterized by individual crystal lattices; one of them was identified as rubidium dizirconate Rb ₂ O · 2ZrO ₂ , and its composition was confirmed by chemical analysis. The effect of the Rb ₂ O · 2ZrO ₂ , and its composition of sintering, initial molar ratio of the components, and additional sintering on the extent of the reaction between ZrO ₂ and RbNO ₃ and on the composition of the products formed was investigated. Data were obtained on the hydrolyza-	AUTHOR: Grizik, A.A.; Plyushchev, V. Ye.;	Kamenskaya, A. N.		\mathcal{B}	· · · · ·	
TOPIC TAGS: <u>rubidium dizirconate</u> , <u>zirconium dioxide</u> , x-ray phase analysis ABSTRACT: A systematic search for individual compounds in the system Rb ₂ O-ZrO ₂ was undertaken in order to determine the interaction between the components and find out whether the separation of individual <u>phases</u> bccurs. The reactions of ZrO ₂ with RbNO ₃ (at 700-1200C) and Rb ₂ CO ₃ were carried out by sintering and fusion. In the case of Rb ₂ CO ₃ , the reaction was too weak to permit any conclusiono regarding the compounds formed. X-ray phase analysis showed the formation of three different phases characterized by individual crystal lattices; one of them was identified as rubidium dizirconate Rb ₂ O \cdot 2ZrO ₂ , and its composition was confirmed by chemical analysis. The effect of the Rb ₂ O \cdot 2ZrO ₂ , and its composition was confirmed by chemical analysis. The effect of the cation temperature, duration of sintering, initial molar ratio of the components, and additional sintering on the extent of the reaction between ZrO ₂ and RbNO ₃ and on the additional sintering on the products formed was investigated. Data were obtained on the hydrolyza-	TITLE: Rubidium dizirconate	·	· · · · · · · · · · · · · · · · · · ·		at star	
ABSTRACT: A systematic search for individual compounds in the system Rb_2O-2rO_2 was undertaken in order to determine the interaction between the components and find out whether the separation of individual phases occurs. The reactions of ZrO_2 with RbNO ₃ (at 700-1200C) and Rb_2CO_3 were carried out by sintering and fusion. In the case of Rb_2CO_3 , the reaction was too weak to permit any conclusiono regarding the compounds formed. X-ray phase analysis showed the formation of three different phases characterized by individual crystal lattices; one of them was identified as rubidium dizirconate $Rb_2O \cdot 2ZrO_2$, and its composition was confirmed by chemical analysis. The effect of the $Rb_2O \cdot 2ZrO_2$, and its composition was confirmed by chemical analysis. The effect of the additional sintering on the extent of the reaction between ZrO_2 and RbNO ₃ and on the additional sintering on the extent of the reaction between ZrO_2 and RbNO ₃ and on the products formed was investigated. Data were obtained on the hydrolyza-	SOURCE: Zhurnal neorganicheskoy khimii, v.	10, no. 6, 1965, 13 dioxide, x-ray pha	12-1319 se analysis			
Card 1/2	ABSTRACT: A systematic search for individual was undertaken in order to determine the inter- whether the separation of individual phases box (at 700-1200C) and Rb ₂ CO ₃ were carried out by Rb ₂ CO ₃ , the reaction was too weak to permit a formed. X-ray phase analysis showed the form by individual crystal lattices; one of them was Rb ₂ O \cdot 2ZrO ₂ , and its composition was confir- reaction temperature, duration of sintering, in	al compounds in the a action between the co- surs. The reactions y sintering and fusion my conclusiono rega- nation of three differ identified as rubidiu- med by chemical ana nitial molar ratio of	system Rb2O omponents and of ZrO2 with a. In the cau- rding the con- cent phases of m dizirconal lysis. The the compone PbNO2 and	h RbNO3 se of npounds characterize effect of the nts, and on the		
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	bility, thermal stability, and the reactions of rubidium dizirconate with a series of reagents (methanol, methanol + water, other homologous alcohols). Methanol was found to be the (methanol, methanol + water, other homologous alcohols).	-	
	(methanol, methanol + water, other homologous alcohols). Methanol web tructure of rubidium best solvent for Rb2O . 2ZrO2. The physiccohemical properties and structure of rubidium dizirconate were determined, and the corresponding x-ray data are tabulated. Orig. art. has: 5 figures and 4 tables.		
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IOFFE, I.I.; DOBROVOL'SKIY, S.V.; LEVIN, Ya.S.; GRIZIK R.M.; KAMBULOVA, V.A.; KRONICH, I.G.; SOKOLOVA, Ye.V.

Similarity of reactions catalyzed by liquid and solid acids. Probl. kin. i kat. 10:294-297 '60. (MIRA 14:5)

1. Nauchno-issledovatel'skiy institut organicheskikh poluproduktov i krasiteley. (Acids) (Naphthylamine) (Naphthol)

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Su: Monthly List of East European Accessions, Vol. 3, no. 3, Library of Congress, March 1954. Uncl.

Gulao, A.
<u>urizo, A.;</u> Jovanovic, M.; Tecilazic-Stevanovic, M. "The influence of the electrolyte on the viscosity and plasticity of Arandjelovac clays." p. 403. (<u>Friroda</u>. Vol. 18, no. 6/7, 1953. Zagreb.)
S0: <u>monthly list of East surgrean Accessions</u>, Vol. 3, no. 5, Liorary of Congress. March 1954. Uncl.

CONTRACTOR OF A DESCRIPTION OF A DESCRIPTION

GRIZO, A.; NIKOLIC, Z.; DELIC, D.

Importance of determining free lime in coment, a survey of methods of determination. p. 248. Vol. 11, No. 2, 1956. THENIKA. Beograd, Yugoslavia.

SOURCE: East European Accessions List, (EEAL) Library of Congress, Vol. 5, No. 8, August, 1956.

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CIA-RDP86-00513R00051701

GRIZO, Aleksandar, inz. (Skopje, Elektrohemijski kombinat "Biljana"); DELIC, Dejan, dr. inz., prof.

> Adsorption capacity of some indigenous coals for various phenols. Tehnika Jug 17 no.7:Suppl.: Hemindustrija 16 no.7:1361-1366 J1 '62.

1. Tehnicki direktor Elektrotehnickog kombinat "Biljana", Skopje (for Grizo). 2. Tehnoloski fakultet. Univerziteta u Beogradu (for Delic).

GRIZO, Aleksandar, inz.

Purification of waste water polluted by nitro and amino compounds. Tehnika Jug 18 no.5:Suppl.:Hemindustrija 17 no. 5:917-923 My '63.

1. Tehnicki direktor Elektrohemijskog kombinata "Biljana", Skopje.

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ITLE:	Investigation of the Dye H-Resordinol as a Reagent for the Photometric Determination of Boric Acid (Izucheniye azekrasitelya H-rezortsina kak reaktiva dlya fotometricheskogo opredeleniya bornoy kisloty)		
PERIODICAL:	Zhurnal analiticheskoy khimii, 1950, Vol. 13, Nr 4, pp. 434– 438 (USSR)		
ABSTRACT :	For the photometric determination of boric acid hydroxy- anthraquinones containing a hydroxyl group in a peri-position to the quinene group are used. The intensely colored solutions of these compounds in concentrated sulfuric acid change their color at an addition of boric acid. Some other organic com- pounds, however, containing a hydroxyl group in the vicinity of a carbonyl group (Refs 1, 2) also react with boric acid in con- centrated sulfuric acid Other methods of determining boric acid make use of its reaction with curcumin, a derivative of dibenzoyl methane, or with derivatives of salicylic acid (Ref 3). As a contrast to all these color reactions, that were carried out in concentrated sulfuric acid or after elimination of the		
Card $1/4$	water by concentrating, reactions were found which can be car-		

Investigation termination o	SOV/75-13-4-10/29 of the Dye H-Resorcinol as a Reagent for the Photcmetric De- f Boric Acid	
	ried out in slightly acetous solutions (Ref 4). Compounds con taining 2 hydroxyl groups in the peri- and ortho-position next to an ago or agomethine group serve as reagents. The authors of the present paper investigated the reaction of the ago dye from diagotized H-acid and resorcinol ("H-resorcinol") with borig acid as well as the reaction with the agomethine compound, which develops from H-acid and salicylic aldehyde. In the first case the color of the solution changes from yellow to pink, in the second case the acetous solutions become light yellow. The au- thors investigated the composition of the compound from H-re- sorcinol and borig acid. The photometric measurements were car- ried out on a universal photometer of the type f M using light filters M-53 (λ_{max} =530m μ). The highest absorption occurs at a molar ratio of borig acid and the reagent of 1:1. Therefore in the new compound one molecule of borig acid falls to one mole- cule of H-resorginol. The determination of the dissociation constant of the complex in solution showed that neither the degree of dissociation a nor the dissociation constant K_D at	
Card 2/4	$p_{\rm H}$ -values of 2,2 up to 3,0 is dependent on the $p_{\rm H}$ -value. The	
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SOV/75-13-4-10/29 Investigation of the Dye H-Resorcinol as a Reagent for the Photometric Letermination of Boric Acid mean value of $K_{\rm m}$ in this range is 6,33.10⁻⁵, for a a value of 0,658 was found. A high excess of the reagent is detrimental as the solution of the reagent highly absorbs in that range of wave-lengths, in which also the light absorption of the complex is measured (530m μ). The best conditions for photometric determination of boron according to this method turned out to be a quantity of 4-5 ml of a 10^{-3} molar solution of H-resorcinol and 5 ml 1 n acetic acid for $1, 1\mu = 66\mu$ boron in a total volume of 50 ml. The intensity of the color increases with time and only after 6 hours reaches a practically constant value. Therefore the solution to be investigated has to be left to stand for 6 hours before measuring. It is not necessary to heat the solution. As the dependence of light absorption of the complex on the concentration of boron is not rectilinear, it is necessary to establish a calibration curve for the determination. There are 4 figures, 4 tables, and 8 references, 7 of which are Soviet. Card 3/4

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	SOV/75-13-4-10/29
Investigation termination of	of the Dye H-Resorcinol as a Reagent for the Photometric Le- 5 Boric Acid
ASSOCIATION:	Odesskiy farmatsevticheskiy institut (Odessa Fharmaceutical Institute)
SUBMITTED:	November 10, 1956
	1. ResorcinolChemical reactions 2 Reagents-Performance
	1. ResorcinolChemical reactions 2 Redgened Forcal reactions 3. Boric acidDetermination 4. Boric acidChemical reactions 5. Photometry
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GRIZO, V.A.; KUBLANOVSKIY, S.I. Conditions for the formation of rubidium iodomercurates. Trudy OGMI no.27;21-28 '61. (HTMA 16:6) (Rubidium salts) (Iodomercurates)



SAVITSKAYA, M.M. [Savyts'ka, M.M.]; KHOLODOVA, Yu.D.; POSTORONKO, A.I.; GRIZODUB, A.P. [Hryzodub, A.P.]

New congulating agents for the acceleration of brine purification in the production of soda. Khim. prom. [Ukr.] nc.3:32-35 J1-S ⁴63. (MIRA 17:8)

1. Ukrainskiy nauchno-issledovatel'skiy institut fiziologii rasteniy (for Savitskaya, Kholodova). 2. Slavyanskim sodovyy kombinat (for Postoronka, Grizodub).

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PODOSYNKIN, P.A.; FOSTORONKO, A.I.; GRIZODUB, A.P. [Hryzodub, A.P.]; KAL'NA, Z.P.; LYAPINA, A.G. [Liapina, A.H.]

Purification of waste waters from the washing of the electric filters of lime kilns. Khim. prom. [Ukr.] no.3:82-84 J1-S '63. (MIRA 17:8)

1. Slavyanskly sodovyy kombinat.

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RETSEPTOR, Ta. (g.Moskva); SHAKIROV, O.; HOAK, A.; SEREBRYANIKOV, G., ekonomist; KHAIT, M.; FILIPPENKO, A.; SULMYMANOV, A. (Dagestanskaya ASSR); GRIGOR'YEV, A.; DZHURINSKIY, N. (g.Kishinev); MALTUKHA, L. (g.Klin); POLISHCHUX, I. (g.Pervoural'sk, Sverdlovskoy obl.); ORIZODUB, Yu. (G.Frunze); CHIGAREV, A. Letters to the editors. Sots. trud 6 no. 1:136-141 Ja '61. (MIRA 14:1) 1. Glavnyy inzh.shakhty No. 31 tresta Kirovugol', g.Karaganda (for Shakirov). 2. Kachal'nik planovogo otdela shakhty No. 31 tresta Kirovugol', g. Karaganda (for Noak). 3. Glavnyy bukhgalter stroitel'nogo upravleniya "Tyashmashstroy", g.Kramatorsk, Stalinskoy obl. (for Khait). 4. Kachal'nik otdela truda 1 zarabotnoy platy vol'skogo zavoda "Metallist" (for Filippenko). 5. Nachal'nik ofdela truda i zarabotnoy platy leningradskogo zavoda "Kinap" (for Origor'yev). 6. Pavinskiy l'nozavod Kostromskoy oblasti (for 'higorev). (Mage payment systems) (Industrial management)

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TR IVI ., GRIZODUB, Yu. N. 1 Jun 50 USSR/Physics - Hydraulics "Determination of the Coefficient of Oscillatory Damping of a Fluid Column, Due to Internal Friction in the Walls of a Pipe," Yu. N. Grizodub "Dok Ak Nauk SSSR" Vol LXXII, No 4, pp 649-650 Derives formula of damping that involves elastic oscillation of pipe due to motion of fluid in pipe, both thick- and thin-walled. Submitted 25 Mar 50 by Acad A. I. Nekrasov. 1651105

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GRIZODUE, Yu. N. USSR/Hydrology - Pumps May 51 USSR/Hydrology - Pumps "Computation of Harmonic-Disturbance Propagation in "Computation of Harmonic-Disturbance Propagation in "Computation of Harmonic-Disturbance Propagation in Disturbance Propagation in Pumping System With Pipes Con- minute Filling a Pumping System With Pipes Con-	
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GRIZODUBOV, N.I.; MIL'KOVA, Z.A.

Determining the fineness in lime grinding. Sakh. prom. 36 no.7:40-43 J1 162. (MIRA 17:1)

1. Romanskaya gruppovaya laboratoriya.



GRITSENKO, Ye.M.; GRIZODUECV. N.I.; MIL'KOVA, Z.A.; TYAZHELOVA, G.F.; STASEYEV, G.I.
Problem deserving attention. Sakh. prom. 37 no.10:28-33 0 '63. (MIRA 16:12)
1. Ramonskaya gruppovaya laboratoriya (for Gritsenko, Grizodubov).
2. Voronezhskiy tekhnologicheskiy institut (for Mil'kova).
3. Ramonskiy sakharnyy zavod (for Tyazhelova, Staseyev).



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CRIZCAOZ.KIY, A. M.

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COUNTRY CATEGORY	Chemical Tools	
AB3. JOUR.	Their Applications. Chemical Products and : AZKhim., 10, 23 1939, 10, 83467	
ATTICA	: Grjarnev, A.	
TILL	: New Methods of Coal Tretrestment for Coling	
ORIG. PUB.	: Pelive, 1958, 38, No 7, 238-243	
ABCTRACT	: Effect of fine grinding and of other coal (C) pretreatment methods on the quality of coke (K) has been investigated. A rew principle for the pretreatment is indicated which con- sists in fine grinding:- the necessity of de- coarseness with the quantity of fine perticles feld to a minimum. The method of selective crushing (SC) is characterized in detail re- solting in the 3-0 pm size and its	
ARD:	Solid Fossil Fuels.	
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2.4. 107.1 2.4. 107.1	:
ABD. JOUR.	: RZKA18., No. 23 1959, No. 83467
AUNIOR IEST. TITLI	
0AIS. 208.	
A33TRACT Con'd Cand:	effectiveness. Presented are flow dispress for SC of rew charge (R) and its components. Depending on the composition of R and the re- quired K autity, processing schemes include SC or fine crinding and SC with consequent treatment and compacting of R. In certain in- stances it is permissible to add gas C into R is: 15-25% quantity and as 12-4 mm size perticles. Described Plas is the preliminary prebent of R up to 100-2000, that leads to the increase of density and of structural strength of K and increases the cohing rate. 2/2Ya. Sctunovskiy.
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2 (* 1972) Graddiaeth	: Czechoslovakia :	H-27
And. Jour.	: RERNIM., Ho. 21 1959, No.	26448
AVTROR IJ39. PICAA	: Grjaznov, V. P. : Not Riven : The Identification of Aldehydes, Es Higher Alcohols in the Products fro fication of Ethyl alcohol	ters, and m the Nosti
otin, pup,	fication of Ethyl Alcohol : Kvasny Prumysl, 5, No 3, 58-61 (195	1
A 331 AAOF -	The author has shown by piper chrome the crude alcohol obtained from grain and from molasses contains the acets methyl, ethyl, and isoamyl alcohols methyl, ethyl, propyl, isobutvl, and hols. Chemical and chromatographic qualitative and quantitative content esters, and alcohols in the alcoholo plates of the purifying and rectifyin 2-column continuous still are given a	stography that in-potato mash ate esters of as well as a isoamyl alco- indexes of the of aldohydes, on the various
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GRKOVIC, LJ.; NIKSIC, M.

C #254/10/10/10/17/19 12/24/26

Strawberries as a raw material for processing. p. 1398. (Tehnika, Vol. 11, no. 9, 1956. Beograd, Yugoslavia)

SO: Monthly List of East European Accessions. (EEAL) LC, Vol. 6, No. 7, July 1957. Uncl.

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Country : CZECHOSLOVAKIA Category: Plant Diseases. Diseases in Cultivated Plants.

Abs Jeur: NZhBiol., No 18, 1558, No 82695

Author : Grkovic, Staneje Inst : -Title : Apple Rust - a New Disease in Dolenjsko Hegion (Lewer Slovenia, Yugoslavia)

Orig iub: Sadjar., vinar., vrtuer., 1957, 44, No 12, 343-347

Abstract: A description is given of apple rust, caused by Gymnosporangium tremchoides and first noted in Lover Slovenia. Methods to control the disease are shown, and among them, in addition to spraying, one reconmends extermination of the juniper in the surrounding

Card : 1/2

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Country : CZECHOSLOVAKIA Category: Plant Diseases. D'seases in Cultivated Plants.

Abs Jour: hZhBiol., No 18, 1058, No 82695

APPROVED FOR RELEASE: Thursday, July 27, 2000 CIA-RDP86-00513R00051 fruit gardens, which is the intermediate plant-hest. -- G.A. D'ynkeva

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GRLIC, Lj.

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Vitamin values of edible wild plants common in Yugo-slavia. I. Ascorbic acid and carotene content of edible wild greens. Lj-Grilć. Acta Pharm. Jugoslav. 2, 112-25 (1032).—The species itudied showed considerable differ-ence in ascorbic acid content but within a plant family these variations were limited. The highest ascorbic acid content was found in leaves of princose (Prinula), in wild rocket (Diplotaxis), and in other Cruciferae. Legunitotae and Chenopodiaceae are also fich in vitamin C, while Com-positae. Boraginaceae, and Plantaginaceae are relatively poor. The plants had the highest content of vitamin C during the month of June. The carotene content was highest in leaves of deadnetile (Lomium), allaila (Medicago). nettle (Unica), mallow (Maise), and plantain (Plantago major). Although considerable differences found in vitamin C and A contents of wild plants, they appear to be better sources of these vitamins than the cultivated plants. Greferences.

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Report. Acta pharm. jugosl. 4 no.3:115-118 1954.
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(VITAMIN C, determ.
in edible wild greens)
(CAROTENE, determ.
in edible wild greens)
(PLENTS
edible wild greens, vitamin C & carotene content)
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Grlic, Ly. Milkweed (Ascleplas syriaca) as a vegetable. Lj. Grlić, and S. Čmelik (Central Inst. Hyg., ⁷agreb, Yugosluvia). Farm. Glasnik 10, 379-394(1954).— Milkweed spronts con-tain H₂O 90.70, ash 0.78, crude fiber 1.37, protein 0.85, ether ext. 0.51, and N-free ext. 5.79%. During the grow-ing period (from the end of April till the beginning of July) the vitamin C content fluctuates from 149 to 359 ing. ⁶₂ (av. 218 mg. ⁶₂). The carotene content varies from 0.93 to 2.86 (av. 1) mg. ⁶₂. It is believed that if any twic gluco-side is present, it will be removed by boiling with water, and pouring off. ----222
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Teaching and research on distory of medicine in Yugoslavia, Lijecn. vjesn. 84 no.1:5-21 '62.

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Evaluation of the effective temperature increase of neutrons by means of thermodynamic irreversible processes. Jaderna energie 9 no.74234 Jl 463.

1. Ustav jaderneho vyzkumu, Ceskoslovenska akademie ved, Rez u Prahy.



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GREEK, Mirke Brazzen OCHEK, I.D. [translat of; Matter Parks prof.; otv. ret. [Gerontology, the star of old and songevity.

Translated from the Groatian with sup tementary notorial translated from the English] Germanic lin - u corless storesti i delpoletil. Poskva, Nauka, 1951. - 5.2 p. Constants at the set

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SYNAK, Yu. [Synak, J.] (Bratislava, Chekhoslovatskaya Sotsialisticheskaya Respublika); <u>GRNCHAR, Ya.</u> [Hrncar, J.] (Bratislava, Chekhoslovat-skaya Sotsialisticheskaya Respublika); KORAL'CHUK, I.I. [translator] New herbicide. Zashch. rast. ot vred. i bol. 8 no.4:52-53 Ap '63. (MIRA 16:10) (Triazine) (Herbicides)

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Material Limit in Furniture Factories. LEKA PROMISHLENOST (Light Industry) 4:6:April 55

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DFLGARIA / Chunical Tucknology. Charles Development H-6 and Their Application. Saruby and Depitation.

Abs Jour: Rof Zhur-Khimiya, No 23, 1958, 78104.

Abstract: in the heart, and 2) Swelling of Lungs, componsatory, solutions interstitish, ediastinal or tissue emphysical, dilatation of h. rt, changes in the Avecardium which disappear in resportion with the disappearance of changes in lungs. The import nee of the roomtgenologic study of the thorax for the disappears of poisening with MGs is emphasized. -- From the authers' such ry.

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н-28 Gzechoslovakia COUNTRY CATEGORY ž 19900 ABS. JOUR. : RZKhime, No. 5 1960, No. : Grner, F. AUTHOR s llot given INST. Impurities in Milk and Dairy Products TITLE 3 CRIG. PUB. : Ceskoslov Hyg Mliecnych Produktocn, 4, No 7, 582-391 (1955) A review article with a bibliography listing to ABGTRACT 1 titles. A. Progorevich 375 CARD: 1/1

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