

GVOZDEV, A.A., prof., doktor tekhn.nauk; KRYLOV, S.M., kand.tekhn.nauk;
PETROVA, V.V., red.izd-va; SHERSTNEVA, N.V., tekhn.red.

[Instructions for calculating statically indeterminate reinforced concrete construction elements taking into account the redistribution of stresses] Instruktsiia po raschetu staticheski neopredelimykh zhelezobetonnykh konstrukttsii s uchedom pereraspredeleniia usilii. Moskva, Gos.izd-vo lit-ry po stroit., arkhitekt. i stroit. materialam, 1960. 109 p. (MIRA 13:6)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut betona i zhelezobeta, Perovo. 2. Deyatvitel'nyy chlen Akademii stroitel'stva i arkhitektury SSSR (for Gvozdev).
(Reinforced concrete) (Strains and stresses)

82196

S/097/60/000/03/01/003

15.3200

AUTHORS: Gvozdev, A.A., Professor, Doctor of Technical Sciences, Mikhaylov, K.V., Nikula, I., Candidates of Technical Sciences

TITLE: Glass-Reinforced Plastics for Reinforcing of Concrete Structures

PERIODICAL: Beton i Zhelezo-Beton, 1960, No. 3, pp. 105 - 111

TEXT: The article deals with the latest development in the field of reinforced concrete, consisting in the employment of non-metal reinforcing material on the basis of fiber-glass. The idea was first expressed by architect A.K. Barov in 1941; since that time extensive research work has been done in the USSR and abroad. Scientists have come to the conclusion, that fiber-glass alone is unsuitable as reinforcement; it can be used, however, in the shape of bars with plastic binders; for better adhesion fiber-glass is treated with silicohydride combinations (silanes). From 1958 to this date special investigations were conducted by the Institute of Concrete and Reinforced Concrete of the ASiA of the USSR. For reinforcing material strips of glass veneer sheets of the Leningradskiy zavod sloistykh plastikov (Leningrad Laminated Plastics Plant) were used. This veneer contained about 67% of fiber-glass by weight, and 40% by volume. The tensile strength of the fiber-glass of non-alkaline composition 14-16 μ in

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diameter is given as being 8,000-9,000 kg/cm² (in the air); the same fiber-glass, but calcium-chloride-dried, increases in tensile strength to 13,500-15,000 kg/cm². The arithmetic mean value of tensile strength of glass veneer strips is equal to 11,750 kg/cm², in accordance with tests conducted on 150 samples of strips 50-70 cm long. The article gives a formula which shows to what extent the tensile strength depends upon the duration of stress being applied. At a given time the tensile strength has decreased by 35%. Tests have shown that after 10 hours of stress application the coefficient of decrease of tensile strength is 0.7. Tensile strength of fiber-glass is known to depend upon the medium in which it is situated. In this connection tests were conducted in different mediums, which are characteristic of the process of concrete production. In using Ca(OH)₂ which is similar to the action of concrete on reinforcement, it appeared that alkali and water have little effect on the tensile strength of fiber-glass veneer, reducing it by 7-8%. The reaction to high temperature was unfavorable, proving that steaming or autoclave treatment are prohibitive for fiber-glass reinforcement. Tests have shown that in using fiber-glass veneer, it is necessary to allow for reduction in tensile strength up to 40% of the initial nominal strength. Tests were also performed to determine the development of deformation of fiber-glass veneer under brief and long-lasting load. It can be seen that the ratio

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between the plastic part of the deformation and the final one remains an almost constant value until the breaking point. Tests which consisted in pulling out fiber-glass veneer strips 10-15 cm long from the concrete revealed that the average value of tensile strength of adhesion does not exceed 10 kg/cm^2 . The pulling effort on the strips did not extend to a greater depth than 5-7 cm. Due to the low module of elasticity of glass reinforcing plastics, the summary losses of tension due to creeping and shrinkage of concrete did not exceed 5-6% as compared to 15-20% in case of high-grade wire. Table 2 shows the results of tensile strength and crack resistance tests performed on 15 concrete bars 70-200 cm long, reinforced with glass plastics. The results of the tests agree with the calculated theoretical values, which proves that in designing glass-plastics reinforced concrete it is possible to use the calculating device of Instruction CH 10-57 (SN 10-57). It is economically justifiable to employ glass-plastics reinforcement, when for technical considerations it is not possible to use steel reinforcement. A great deal of research work is yet to be done, until glass-plastics reinforcement can compete with steel reinforcement. The Institut betona i zhelezobetona (Institute of Concrete and Reinforced Concrete) of ASIA USSR

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Glass-Reinforced Plastics for Reinforcing of Concrete Structures

YuzhNII, KhIIKS, Institut stroitel'stva i arkhitektury AN BSSR (Institute of Construction and Architecture) and others are engaged in further development work. There are 4 photographs, 8 graphs, 4 tables and 1 Soviet reference.

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GVOZDEV, A.A., doktor tekhn.nauk, prof.

Current problems in developing precast prestressed reinforced
concrete construction elements. Bet.i zhel.-bet. no.7:295-297
Jl '60. (MIRA 13:7)

1. Deystvitel'nyy chlen Akademii stroitel'stva i arkhitektury.
(Precast concrete)

GVOZDEV, A.A., doktor tekhn.nauk, prof.; DMITRIYEV, S.A., kand.tekhn.nauk

Designing cross sections according to the tendency to form cracks.
Bet.i zhel.-bet. no.7:331-332 JI '60. (MIRA 13:7)
(Strains and stresses)

GVOZDEV, A.A., prof., doktor tekhn.nauk; KRYLOV, S.M., kand.tekhn.nauk;
KLIMOVA, G.D., red.izd-va; KASIMOV, D.Ya., tekhn.red.

[Instructions for calculating statistically indeterminate reinforced-concrete construction elements taking the redistribution of stresses into account] Instruksia po raschetu staticheski neopredelimykh zhelezobetonnykh konstruktaii s uchetom pereraspredelenia uailii. Izd.2. Moskva, Gos.izd-vo lit-ry po stroit., arkhitekt. i stroit.materialam, 1961. 109 p.

(MIRA 15:2)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut betona i zhelezobetona, Perovo. 2. Tsentral'naya laboratoriya teorii zhelezobetona i armatury Nauchno-issledovatel'skogo instituta betona i zhelezobetona Akademii stroitel'stva i arkhitektury SSSR (for Gvozdev, Krylov).

(Strains and stresses)

(Reinforced concrete)

TEMKIN, L.Ye., inzh., nauchn. red.; OVSYANKIN, V.I., red.; STRELETSKIY, N.S.,
prof., red.; GVOZDEV, A.A., prof., red.; IVANOV, Yu.M., red.; SE-
MENTSOV, S.A., kand. tekhn. nauk, red.; GALKIN, Ya.G., red.; KRASIL'-
NIKOV, P.A., red.; MURASHEV, V.I., red. [deceased]; NIKITIN, N.V.,
red.; TAL', K.E., kand. tekhn. nauk, red.; VILKOV, G.N., red. izd-va;
GARNUKHIN, Ye.K., tekhn. red.

[Papers from the International Conference on Designing Building
Elements] Materialy Mezhdunarodnogo soveshchaniya po raschetu stroitel'-
nykh konstruksii. Moscow, 1958. Moskva, Gos. izd-vo lit-ry po stroit.,
arkhit. i stroit. materialam, 1961. 258 p. (MIRA 14:7)

1. Mezhdunaroye soveshchaniye po raschetu stroitel'nykh konstruksiy.
Moscow, 1958. 2. Deystvitel'nyy chlen Akademii stroitel'stva i arkhitek-
tury SSSR (for Streletskiy, Gvozdev). 3. Chlen-korrespondent Akademii
stroitel'stva i arkhitektury SSSR (for Sementsov, Tal')
(Building)

GVOZDEV, A.A., doktor tekhn. nauk, prof.; CHINENKOV, Yu.V., kand. tekhn. nauk; IFTINKA, G.A., red. izd-va; KLIMOVA, G.D., red. izd-va; MOCHALINA, Z.S., tekhn. red.

[Handbook on the design of three-dimensional reinforced concrete thin-walled roofs and spans] Instruktsiia po proektirovaniu zhelezobetonnykh tonkostennykh prostranstvennykh pokrytii i perekrytii. Moskva, Gos. izd-vo lit-ry po stroit., arkhitekt. i stroit. materialam, 1961. 334 p.

(MIRA 15:2)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut betona i zhelezobetona, Perovo. 2. Nauchno-issledovatel'skiy institut betona i zhelezobetona Akademii stroitel'stva i arkhitektury (for Gvozdev, Chinenkov). 3. Deyatvital'nyy chlen Akademii stroitel'stva i arkhitektury SSSR (for Gvozdev).

(Reinforced concrete construction)
(Roofing, Concrete)

GVOZDEV, A.A., prof.

Stability of thin shells. Bet. i zhel.-bet. no.10:441-445 0
'61. (MIRA 14:12)

1. Deystvitel'nyy chlen Akademii stroitel'stva i arkhitektury
SSSR.

(Roofs, Shell)

MULININ, N.M., kand.tekhn.nauk; DMITRIYEV, S.A., kand.tekhn.nauk;
KRASOVSKAYA, G.M., inzh.; GVOZDEV, A.A., doktor tekhn.nauk, prof.;
KLIMOVA, G.D., red.izd-va; RUDAKOVA, N.I., tekhn. red

[Temporary instructions on the use of thermally strengthened ribbed
cable in prestressed concrete elements] Vremennye ukazaniia po pri-
meneniiu termicheski uprochnennoi katanki periodicheskogo profil'ia
v predvaritel'no napriazhennykh zhelezobetonnykh konstruktsiakh.
Moskva, Gosstroizdat, 1962. 11 p. (MIRA 15:6)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut betona i
zhelezobetona, Perovo. 2. Deystvitel'nyy chlen Akaderii stroitel'stva
i arkhitektury SSSR (for Gvozdev).

(Concrete reinforcement)

GVOZDEV, A. A.

"A Survey of the Results of Scientific Research of Pre-stressed Reinforced Concrete."
report presented at the Intl. Federation of Prestressing, 4th Intl. Congress, Rome and Naples, 27 May - 2 June 1962.

GVOZDEV, A.A., doktor tekhn.nauk, prof.; DMITRIYEV, S.A., kand.tekhn.nauk;
NEMIROVSKIY, Ya.M., kand.tekhn.nauk

Calculation of the displacements (deflections) of reinforced
concrete elements according to the draft of the new standards
(SNiP 2-V.1-62). Bet. i zhel.-bet. 8 no.6:245-250 Je '62.
(MIRA 15:7)

(Precast concrete)
(Flexure)

GVOZDEV, A.A.

Basic objectives of the theory of reinforced concrete for the next twenty years. Izv.ASiA 4 no.4:3-8 '62. (MIRA 16:1)

1. Deystvitel'nyy ohlen Akademii stroitel'stva i arkhitektury SSSR.

(Reinforced concrete construction)
(Building research)

GVOZDEV, A.A., doktor tekhn.nauk, prof., red.; ZUBKOVA, M.S., red.
izd-va; GOL'BERG, T.M., tekhn. red.

[Experimental and theoretical research on reinforced concrete structures] Eksperimental'no-teoreticheskie issledovaniia zhelezobetonnykh konstruksii. Pod red. A.A.Gvozdeva. Moskva, Gosstroizdat, 1963. 253 p. (MIRA 16:7)

1. Akademiya stroitel'stva i arkhitektury SSSR. Institut betona i shelezobetona, Perovo. 2. Daystvitel'nyy chlen Akademii stroitel'stva i arkhitektury SSSR (for Gvozdev).
(Reinforced concrete construction)

GVOZDEV, A.A., prof.

"Physical bases for the theory of the strength of concrete and reinforced concrete" by O.I.A. Berg. Reviewed by A.A.Gvozdev.
Bet. i zhel.-bet. 9 no.2:92 F '63. (MIRA 16:5)

1. Deystvitel'nyy chlen Akademii stroitel'stva i arkhitektury SSSR.
(Concrete--Testing) (Berg, O.I.A.)

TEMKIN, L.Ye., inzh., red.; GVOZDEV, A.A., doktor tekhn. nauk,
prof., red.; SHITOVA, L.N., red.izd-va; RODIONOVA, V.M.,
tokhn. red.

[Construction specifications and regulations] Stroitel'nye
normy i pravila. Moskva, Gosstroizdat. Pt.2. Sec.V. ch.1.
[Specifications for the design of concrete and reinforced-
concrete structures] Betonnye i zhelezobetonnye konstruksii;
normy proektirovaniia (SNiP II-V. 1-62). 1962. 100 p.
(MIR: 16:12)

1. Russia (1923- U.S.S.R.) Gosudarstvennyy komitet po delam
stroitel'stva. 2. Gosudarstvennyy komitet Soveta Ministrov
SSSR po delam stroitel'stva (for Temkin). 3. Nauchno-
issledovatel'skiy institut betona i zhelezobetona Akademii
stroitel'stva i arkhitektury SSSR (for Gvozdev).

(Concrete construction)
(reinforced concrete construction)

GVOZDEV, A.A. (Moscow) :

"Creep of concrete".

report presented at the All-Union Congress on Theoretical and Applied Mechanics, Moscow, 29 Jan - 5 Feb 64.

KLYACHKO, A.L., inzh.; ODINOV, M.I., inzh.; GLUKHOVSKIY, K.A.,
kand. tekhn. nauk, inzh., red.; GVOZDEV, A.A., doktor
tekhn. nauk, prof., red.; GORENSHTEYN, B.V., kand.
tekhn. nauk, red.; KOSTYUKOVSKIY, M.G., kand. tekhn.
nauk, red.; KRYLOV, N.A. doktor tekhn. nauk, red.;
KUREK, N.M., kand. tekhn. nauk, red.; LEVINSKIY, L.G.,
inzh., red.; LOBANOV, N.D., inzh., red.; MOROZOV, A.P.,
inzh., red.; ONIASHVILI, O.D., doktor tekhn. nauk, prof.,
red.; SAKHNOVSKIY, K.V., doktor tekhn. nauk, prof., red.;
FILIN, A.P., doktor tekhn. nauk, prof., red.; YEFIMOV,
A.D., inzh., nauchn. red.

[Three-dimensional structural elements in the U.S.S.R.;
materials of the All-Union Conference on Precast
Reinforced Concrete Three-Dimensional Elements held in
November 13-17, 1962 in Leningrad] Prostranstvennye kon-
struktsii v SSSR; po materialam pervogo Vsesoiuznogo so-
veshchaniia po sbornym zhelezobetonnyim prostranstvennym
konstruktsiiam, sostoiavshegosia 13-17 noiabria 1962 g.
v Leningrade. Leningrad, Stroiizdat, 1964. 461 p.

(MIRA 17:11)

1. Nauchno-tekhnicheskoye obshchestvo stroitel'noy indu-
strii SSSR. Leningradskoye otdeleniye.

GVOZDEV, A.A., doktor tekhn. nauk, prof., red.; GORYACHEVA, T.V.,
red.

[Calculating and building reinforced concrete structural
elements; materials on the justification and explanation
of new regulation norms for designing concrete and re-
inforced concrete elements SNiP II-V. 1-62] Raschet i
konstruirovaniye elementov zhelezobetonnykh konstruktsii;
materialy po obosnovaniyu i raz"iasneniyu novykh polozhe-
niy norm proektirovaniya betonnykh i zhelezobetonnykh
konstruktsiy SNiP II-V. 1-62. Moskva, Stroiizdat, 1964.
214 p. (MIRA 17:11)

TAL', K.E., kand. tekhn. nauk; LESSIG, N.N., kand. tekhn. nauk; Prinsipali uchastiye: GVOZDEV, A.A.; ALEKSANDROVSKIY, S.V.; BORISHANSKIY, M.S.; DMITRIYEV, S.A.; KRILOV, S.M.; MIKHAYLOV, K.V.; MULIN, N.M.; NEMIROVSKIY, Ya.M.; CHISTYAKOV, Ye.A.; VASIL'YEV, B.F.; BOGATKIN, I.L.; ZALESOV, A.S.; NIKITIN, I.K.

New standards SNiP II-V. 1-62 for the design of concrete and reinforced concrete elements. Bet. i zhel.-bet. 9 no.3:97-102
Mr. '63. (MIRA 16:4)

1. Nauchno-issledovatel'skiy institut betona i zhelezobetona Akademii stroitel'stva i arkhitektury SSSR (for all except Vasil'yev, Bogatkin, Zalesov, Nikitin). 2. Gosudarstvennyy institut tipovogo proyektirovaniya i tekhnicheskikh issledovaniy (for Vasil'yev, Bogatkin, Zalesov, Nikitin).

GVOZDEV, A.A., doktor tekhn. nauk, prof., red.

[Crack resistance and deformability of normal and pre-stressed concrete elements] 'Treshchinostoikost' i deformativnost' obychnykh i predvaritel'no napriazhennykh zhelezobetonnykh konstruksii. Moskva, Stroiizdat, 1965.
281 p. (MIRA 18:8)

GVOZDEV, A.A.

Direction of displacements in a longitudinal wave propagating
along the free boundary of an elastic medium. Izv. AN SSSR.
Fiz. zem. no.2:65-68 '65. (MIRA 18:6)

1. Institut fiziki Zemli AN SSSR.

L 00350-66 EWT(m)

ACCESSION NR: AP5018153

UN/0097/65/000/007/0001/0008
666.97.017:539.376

AUTHORS: Gvozdev, A. A. (Doctor of technical sciences, Professor);
Aleksandrovskiy, S. V. (Candidate of technical sciences); Bagriy, E. Ia. (Engineer)

TITLE: Creep of ¹⁵concrete under time-varying stresses

SOURCE: Beton i zhelezobeton, no. 7, 1965, 1-8

TOPIC TAGS: concrete, creep characteristic, creep mechanism, construction material

ABSTRACT: A study is made of creep in concrete under time-varying stresses, for example, stresses due to temperature and humidity fluctuations. Special experiments were set up by the author while working in the Tsentral'naya laboratoriya teorii zhelezobetona NIIZhB, Gosstroya SSSR (Central laboratory of Reinforced Concrete Theory, NIIZhB, Gosstroy SSSR), and the results of these tests are reported. Creep was studied under conditions of centered compression of vibrated concrete of weight content 1:1.9:4.4 with a water-cement ratio of 0.65. The origin and content of the cement and aggregates are given (no special analysis was made of the water). Sample specimens were made in the form of prisms of

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dimensions 7 x 7 x 60 cm. The specimens were sealed in paraffin, petroleum jelly, and polyethylene film after a 72-hour cure. A special lever apparatus was developed for use in loading and unloading the concrete specimens. Brief descriptions of instrumentation and temperature-humidity control devices are given. The modulus of elasticity of the concrete under compression obeys the equation

$$E(\tau) = 3.3(1 - 0.575e^{-0.07\tau}) 10^3 \text{ kg/cm}^2,$$

where $E(\tau)$ is in kg/cm^2 and τ is the compressive stress. Several specimens were exposed to constant loading only in order to provide experimental control. A review is made of certain theoretical creep equations, and experimental measurements are compared with theory. Empirical parameters for certain creep equations are estimated on the basis of experimental observations. The authors conclude that the applicability of the theory of an elastically creeping body depends upon accurate selection of the analytic expressions for the "hereditary" functions of the theory. Orig. art. has: 7 figures and 7 equations.

ASSOCIATION: none

SUBMITTED: 00

ENCL: 00

SUB CODE: MI

NO REF SOV: 010

OTHER: 001

Card 2/2 *ju*

LEVANOV, Nikolay Mikheylovich, prof., doktor tekhn. nauk;
SIVORIGIN, Dmitriy Grigor'yevich, dots., kand. tekhn.
nauk; KUZNETSOV, G.F., prof., doktor tekhn. nauk;
GVOZDEV, A.A., prof., doktor tekhn. nauk

[Reinforced concrete elements] Zhelezobetonnye kon-
struktsii. Moskva, Vysshaya shkola, 1965. 871 p.
(MIRA 18:10)

GVOZDEV, A.F., otv. za vypusk

[Report to the Fifth World Congress of Trade Unions on the activities of the World Federation of Trade Unions] Otchet V Vsemirnomu kongressu profsoiuzov o deiatel'nosti Vse-mirnoi federatsii profsoiuzov, 1957-1961. n.p. Profizdat, 1961. 674 p. (MIRA 15:6)

1. World Federation of Trade Unions.
(Trade unions)

s/148/60/000/002/008/008

18.7520

AUTHORS: Minkevich, A.N., Gvozdev, A.G.

TITLE: Titanizing of Steel in a Molten Salt Bath

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy, Chernaya metallurgiya,
1960, Nr 2, pp 151 - 156

TEXT: 14 Information is given on results of experiments on titanizing of 08 and 50 grade steels in a molten salt bath. Titanizing was successfully performed at 950° and 1,100°C during 0.5 - 2 hours in a bath containing 80 - 90% molten sodium chloride and 10 - 20% fine granulated TiO_x alloy (containing about 10% (at.) O₂). Melting and utilization of the bath was conducted under an argon shield. It is desirable in further experiments to investigate the possible utilization of the bath without a shielding gas. It is also recommended to check the possibility of replacing the powder of a specially molten titanium alloy with oxygen by titanium powder contaminated with oxygen to about the required concentration. Without the indicated experiments the recommendation of an extended use of the aforementioned bath will be premature. The titanized layer formed on 08 steel in the bath consists of a

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Titanizing of Steel in a Molten Salt Bath

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thick layer of columnar grains separated from the core by a boundary line, and of a thin harder external zone. The columnar grains are solid solutions of titanium in α -iron. In a part of this zone, adjacent to the external thin zone, sometimes the separation of excessive titanides, precipitating during slow cooling, was observed. Data of spectral analysis, carried out under the supervision of V.G. Koritskiy, show that the external zone contains about 30% titanium. According to data from X-ray analysis, the external zone of the layer on O8 grade steel is FeTi_2 titanide; on 50 grade steel it is titanium carbide, TiC. The titanized layer formed on the steel in the investigated bath has high corrosion and acid resistance. There are: 4 graphs, 2 sets of microphot, 1 table and 7 references, 2 of which are Soviet and 5 English.

ASSOCIATION: Moskovskiy institut stali (Moscow Steel Institute)

SUBMITTED: October 6, 1958

Card 2/2

AVRAAMOV, Yu.S.; GVOZDEV, A.G.; LIVSHITS, B.G.

Surface energy of single crystals of the Fe - 3% Si alloy.
Izv. vys. ucheb. zav.; Chern. met. 8 no.9:142-145 '65.

(MIRA 18:9)

1. Moskovskiy institut stali i splavov.

AYRAMOV, Yu.S.; ENOLSKY, A.G.; LEVSHIN, B.I.

Diffusive creep of a single crystal of iron-ferrous steel.
Izv. vyz. ucheb. zav.; Chern. met. 8 no.11:1113-1117 '61.
(MIRA 1811)

1. Moskovskiy institut stali i splavov.

L 12015-66 EWT(1)/EWT(m)/EWA(d)/T/EWP(t)/EWP(2)/EWT(b)/EWA(c) TSP(c) JD
ACC NR: AP5028278 SOURCE CODE: UR/0020/65/165/002/0315/0318

AUTHORS: Avraamov, Yu. S.; Gvozdev, A. G.; Livshits, B. G.

ORG: Institute of Steel and Alloys (Institut stali i splavov)

TITLE: Concerning the role of the limiting and surface energy in secondary recrystallization

SOURCE: AN SSSR. Doklady, v. 165, no. 2, 1965, 316-318

TOPIC TAGS: crystal growth, surface property, recrystallization

ABSTRACT: The authors have attempted to determine more accurately the ratio of the grain-boundary and surface contribution to the crystal-growth moving force during different stages of secondary recrystallization, without taking into account the retarding action of inclusions and thermal-etching grooves. The uniform growth of a two-dimensional grain from a stabilized matrix is analyzed to derive the grain-growth moving force. It is shown that the main contribution to the moving force is made by the surface energy when $R \leq 1.5 r$ (R -- radius of the growing grain, r -- radius of the matrix grains), and that when $R > 1.5r$ the contribution of the boundary energy is larger. At the end of the secondary recrystallization, the fractions due to the surface and

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UDC: 532.614

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ACC NR: AP5028278

boundary energies amount to 24 and 76 per cent respectively if $\Delta\sigma$, the difference of the surface energy of the growing grain and of the matrix grains is assumed to be 3 per cent of the surface energy (σ). This compares with values of 88 and 12 per cent estimated by Dunn and Walter (Acta met., v. 8, 497, 1960), and with 37 and 63 per cent obtained by the authors in the case of grain oriented transformer steel (Izv. vyssh. uchebn. zaved, Chernaya metallurgiya, No. 9, 1965). This report was presented by G. V. Kurdyumov. Orig. art. has: 3 figures and 4 formulas.

SUB CODE: 20/ SUBM DATE: 08May65/ NR REF SOV: 001/ OTH REF: 007

Card

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LARIONOV, K.A., doktor ekonom. nauk, prof.; GVOZDEV, A.M., kand. ekonom. nauk, ILYUKHINA, N.A., kand. ekonom. nauk; KOGAY, A.V., kand. ekonom. nauk; NIKOLAYEV, N.I., kand. ekonom. nauk; TSAPKIN, N.V., kand. ekonom. nauk, dots.; VASYUTIN, V.F., prof., red.; KOKOSHKO, A.G., red.; NAUMOV, K.M., tekhn. red.

[Planning the local economy and cultural development of a region] Planirovaniye mestnogo khoziaistva i kul'turnogo stroitel'stva raiona; uchebnoe posobie. Moskva, Izd-vo VPSH i AON pri TsK KPSS, 1961. 382 p. (MIRA 14:11)

1. Kommunisticheskaya partiya Sovetskogo Soyuza. Vysshaya shkola.
2. Kafedra sovetskoy ekonomiki Leningradskoy Vysshey partynoy shkoly (for Larionov, Gvozdev, Ilyukhina, Kogay, Nikolayev, TSapkin). (Russia—Economic policy) (Russia—Culture)

FAYNBERG, A.I.; REZNIK, A.I.; GVOZDEV, A.M.; FILATOV, N.L.;
USHENKO, V.S., red.; SALAZKOV, N.P., tekhn. red.

[Problems on the methodology for planned calculations and
analysis of administrative operations in communal housing and
services] Sbornik zadach po metodike planovykh raschetov i ana-
lizu khoziaistvennoi deiatel'nosti v kommunal'nom khoziaistve.
[By] A.I. Fainberg i dr. Moskva, Izd-vo M-va kommun. khoz.
RSFSR, 1962. 233 p. (MIRA 15:12)
(Housing management--Accounting)
(Municipal services--Accounting)

GVOZDEV, A.P.

Methods of determining the economic efficiency of underground coal gasification. Podzem.gaz.ugl. no.1:72-74 '58. (MIRA 11:4)

1. Glavpodzengaz.
(Coal gasification, Underground--Accounting)

Gvozdev, A.P.
GVOZDEV, A.; SHAPIRO, M.

Natural gas as turbine fuel. Tekh.mol. 26 no.2:19 '58.

(MIRA 11:2)

1.Zamestitel' nachal'nika Glavpodzemgaza (for Gvozdev). 2.Starshiy inzhener Glavpodzemgaza (for Shapiro).
(Gas turbines)

GVOZDEV, Anatoliy Petrovich; CHERNYAVSKIY, Leonid Merkur'yevich;
SMIRNOV, O.S., red.; STRYZHKOVA, N.I., red.; GALAKTIONOVA,
Ye.N., tekhn. red.

[Organizing a centralized operation service] Organizatsia
tsentralizovannoi ekspluatatsionnoi sluzhby. Moskva, Avto-
transizdat, 1962. 79 p. (MIRA 15:9)
(Transportation, Automotive)

S/270/63/000/002/011/020
A001/A101

AUTHORS: Obiralov, A. I., Gvozdev, A. V.

TITLE: The effect of changing temperature of instruments in the differentiated method on the precision of stereoscopic processing of aerial photosurvey data

PERIODICAL: Referativnyy zhurnal, Geodeziya, no. 2, 1963, 24 - 25, abstract 2.52.172 ("Tr. Mosk. in-ta inzh. zemleustroystva", 1962, no. 16, 59 - 66)

TEXT: The error magnitudes δ_p depend on the conditions of photograph illumination (upper, lower), power and disposition of lamps, etc. In order to secure high-precision measurements, it is necessary, during the period of instrument warming, to specify the initial reading by the parallactic screw after definite time intervals. The stabilization of measurement results corresponds to the instant of temperature stabilization of individual instrument parts, after which p magnitudes will be measured on another scale with a constant error eliminated in forming differences of longitudinal parallaxes. Changes in tem-

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The effect of changing temperature of...

S/270/63/000/002/011/020
A001/A101

perature of the instrument do not markedly affect the accuracy of measuring the magnitudes of transverse parallaxes q .

I. Mityachkin

[Abstracter's note: Complete translation]

Card 2/2

BABUSHKIN, A.A.; GVOZDEV, B.A.; GLAZUNOV, P.Ya.

Spectrophotometric equipment for continuous absorption analysis and for recording gas concentrations. Fiz. sbor. no.3:360-363 '57.
(MIRA 11:8)

1. Institut fizicheskoy khimii AN SSSR.
(Spectrophotometer)

GVOZDEV, B. A., SHUBIN, V. N.

"The Effect of Accelerated Electrons on Solutions of $KMnO_4$ " p.73

Trudy Transactions of the First Conference on Radioaction Chemistry, Moscow,
Izd-vo AN SSSR, 1958. 330pp.
Conference -25-30 March 1957, Moscow

GVOZDEV, B. A.

p3

PHASE I BOOK EXPLOITATION

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Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk

Deystviye ioniziruyushchikh izlucheniya na neorganicheskiye i organicheskiye sistemy (Effect of Ionizing Radiation on Inorganic and Organic Systems)
Moscow. Izd-vo AN SSSR, 1958. 416 p. 7,000 copies printed.

Resp. Ed.: Pshezhetskiy, S. Ya.; Ed. of Publishing House: Bugayenko, L.T.;
Tech. Ed.: Prusakova, T. A.

PURPOSE: This publication is for scientists working in the field of radiochemistry.

COVERAGE: This collection of articles represents contributions of Soviet scientists in the field of radiochemistry. The papers are concerned with the effect of ionizing radiation on organic and inorganic substances in solutions and in the solid phase. These papers were completed in the years 1951 - 1956 at the Institute of Physical Chemistry, AS USSR, the Institute of Physics and Chemistry imeni L. Ya. Karpov, the Moscow State University, and other scientific institutions. Most of these works are a continuation of those published in "Sbornik rabot po radiatsionnoy khimii" published in 1955. Ts. I. Zalkind and Yu. M. Malinskiy cooperated in the editing of this symposium.

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Effect of Ionizing Radiation (Cont.)

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TABLE OF CONTENTS:

PART 1. REACTIONS IN AQUEOUS SOLUTIONS
AND RADIATION AND ELECTRO-CHEMICAL PROCESSES

Preface

Duzhenkov, V.I., Dolin, P.I. Effect of X-ray Irradiation on Aqueous Alkali Solutions Saturated With Oxygen

7

The kinetics of accumulation of molecular products formed in the radiolysis of water are studied in this paper. These products are: hydrogen peroxide and hydrogen. The absorption of oxygen in high-purity alkali solutions saturated with oxygen was also taken into consideration. It was determined that the initial yield of hydrogen depends on the concentration of the irradiated KOH solution only for concentrations up to 0.6 - 0.7 N KOH. The same relation was found for H_2O_2 . The material balance of the molecular products showed a strong deviation towards excessive absorption of oxygen. This fact was explained as the formation of higher peroxides, probably HO_2 or the complex $H_2O_2 \cdot HO_2$.

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Effect of Ionizing Radiation (Cont.)

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There are 4 figures and 8 references, of which 6 are Soviet and 2 English.

Gvozdev, B.A., Shubin, V.N. Effect of Accelerated Electrons on Potassium Permanganate Solutions

12

The reduction of aqueous KMnO_4 solutions by accelerated electrons in the concentration range of 10^{-3} to $3 \cdot 10^{-1}$ M is discussed in this paper. The yield of the reaction $\text{Mn(VII)} \longrightarrow \text{Mn(IV)}$ is determined from the relation between the amount of reduced permanganate and the amount of energy absorbed by the solution. The authors explain the effect of the acidity of the solution (in the range $\text{pH} = 0.4$ to 12) on the rate of reduction. The maximum yield was observed for $\text{pH} = 2.05$. It was determined that the yield of reaction does not depend on the dosage rate in the range from $\sim 10^{17}$ to $\sim 10^{20}$ eV/ml·sec for electron energies from 0.1 to 0.7 Mev. There are 8 figures, 3 tables, and 7 references, of which 4 are Soviet, 2 English, and 1 Polish.

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Effect of Ionizing Radiation (Cont.)

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Sidorova, L.P., Zimin, A.V., Proskurin, M.A. Effect of Co^{60} γ -Radiation
on Aqueous Solutions of Salts of Tin

22

The article deals with the effect of irradiation on $SnCl_2$ and $SnCl_4$ solutions, and the role of the salts as acceptors of the products of water radiolysis. Experimental data show that the irradiation of $1 \cdot 10^{-2}$ M solutions of Sn^{2+} in 4N HCl and in 1N NaOH results in a hydrogen yield equal to the yield of bivalent tin oxidation. The yield of hydrogen is 0.53 - 0.60 molecules/100 ev of absorbed energy. The hydrogen yield does not change with the increase of dosage in an acid medium, and decreases in an alkaline medium. The presence of bivalent tin in alkaline solutions increases the hydrogen yield, while it does not show an essential effect in an acid medium. The oxidation of $Sn^{2+} \rightarrow Sn^{4+}$ in acid and alkaline media is apparently conditioned by the interaction of Sn^{2+} with H_2O_2 molecules. There are 3 tables, 1 figure, and 14 references, of which 6 are Soviet, and 8 English.

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Effect of Ionizing Radiation (Cont.)

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Chernova, A.I., Orekhov, V.D., Proskurin, M.A. Oxygen Compounds of Iron Ions and Their Conversion During Radiolysis of Aqueous Solutions

29

This is a study of iron peroxide compounds and their behavior under γ - and ultraviolet irradiation. Variable-valence ions are considered to be carriers of the oxidizing components in radiochemical oxidation. The peroxide iron complex which forms during the irradiation of a Mohr salt solution in 4N sulfuric acid decomposes in the presence of Fe^{3+} to form bivalent ions. This process is intensified with the increase of the Fe^{3+} concentration in the solution. It was determined that the peroxide iron complex can be formed also in Mohr salt solutions in 0.8N acid when an excess of Fe^{3+} is present. Addition of $MnSO_4$, $TiNO_3$, $NaCl$, KBr , KI , $Ce(NO_3)_3$ and $CuSO$ to the irradiated solutions of the Mohr salt in 0.8 and 4N sulfuric acid leads to an increased yield of Fe^{2+} ions. The effect is due to the presence of bromide and iodide ions as well as Ce^{3+} and Cu^{2+} . There are 6 figures and 11 references, of which 4 are Soviet, 4 English, and 3 German.

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Effect of Ionizing Radiation (Cont.)

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Sharpatyy, V.A., Orekhov, V.D., Proskurin, M.A. Sensitization of the Radiolytic Conversion of Sodium Nitrate in Aqueous Alkaline Solutions

37

The subject of this paper is the effect of the temperature of the solution on the yield of radiolytic conversion of nitrate in aqueous alkaline solutions at temperatures from 20° to 90°. The same process was studied with glycerin as acceptor of OH radicals. An increase from 20° to 40° in 1M NaNO₃ / 1M KOH causes a sharp increase of the nitrate yield: from ~ 3.0 to ~ 6.5 equiv./100ev. The increase in yield reaches its limit value at 80° and equals about 8 equiv./100ev. The sensitizing effect of glycerin is apparent only when its concentration is ~ 5.10⁻⁴ M and remains constant for concentrations up to 10⁻¹M. The presence of molecular oxygen (air) inhibits this effect. There are 5 figures and 10 references, of which 6 are Soviet and 4 English.

Sharpatyy, V.A., Orekhov, V.D., Proskurin, M.A. Radiolytic Reduction of Sodium Nitrate in Concentrated Aqueous Solutions

43

This paper considers the radiolytic reduction of nitrate solutions in a wide range of concentrations. Concentrations of NaNO₃ above 1M in the presence of an inert gas (nitrogen) resulted in a yield of 8 to 9 equiv./100ev NO₂⁻. It was found that molecular oxygen inhibits the reduction

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Effect of Ionizing Radiation (Cont.)

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process, which is evident in lower results as compared to the process in an inert atmosphere. There are 2 figures and 6 references of which 3 are Soviet and 3 English.

Chernykh, V.Ya., Pshezhetskiy, S.Ya., Tyurikov, G.S. Kinetics of the Decomposition of Hydrogen Peroxide Under the Action of γ - Radiation

48

The authors studied the kinetics of H_2O_2 decomposition in a wide range of concentrations: 1.78 - 92.2 mol.%, in γ and ultraviolet radiation. The rate of the thermal reaction was also taken into consideration. The rate of the radiation reaction is proportional to the square root of γ -radiation intensity. The activation energy equals 6.5 ± 1.0 kcal/M. The activation energy of the photochemical reaction is 8 - 9 kcal/M. The efficiency of the γ -radiation reaction is 21 (at -4°) to 230 molecules of H_2O_2 (at 50°) per 100ev dosage rate of $1.84 \cdot 10^{18}$ ev/l. sec. It is a chain reaction. The kinetics of the radiation, photo, and thermal reactions can be clarified by assuming electrolytic dissociation of the HO_2 radical, the rate of which depends on the concentration of the solution. There are 20 figures, 4 tables, and

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Effect of Ionizing Radiation (Cont.)

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22 references of which 3 are Soviet, 16 English, and 3 German.

Zalkind, Ts.I., Veselovskiy, V.I. Mechanism of Radiochemical Formation of Stationary Potential Differences in Aqueous Solutions

66

The stationary potential difference of ~ 0.9 is formed in the system Pt/H₂SO₄ saturated with nitrogen/Au and irradiated with Co⁶⁰ γ -radiation. It was shown that the formation of a positive potential at the Au electrode is connected with the radiolytic formation of the OH radical. The oxidation of the electrode during heating facilitates the formation of the positive potential at the Au electrode. The rate of reduction is determined by the rate of the electrode reaction, i.e., the electrochemical discharge stage. There are 6 figures, and 6 references of which 5 are Soviet and 1 English.

Zalkind, Ts.I., Veselovskiy, V.I. Photoelectrochemical and Radiation Electrochemical Processes in Aqueous Solutions of Uranium Salts

74

Uranium salts were irradiated with radon and Co⁶⁰. It was shown that the hexavalent uranium salts show reduction of uranyl ions to pentavalent uranium ions. Due to the ease of oxidation - reduction transitions in the system U(VI) / U(V), the increase of the uranyl ion concentration is followed by a decrease in the amount

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of the formed H_2O_2 , uranium peroxide compounds, and in the oxidation of U(IV) and oxalic acid. Irradiation of the system U(IV) / U(VI) and U(III) / U(IV) results in a shift of equilibrium and the formation of more oxidized forms. There are 14 figures, 7 tables, and 18 references of which 8 are Soviet, 7 English, and 3 German.

Miller, I.B., Veselovskiy, V.I. Radiation Electrochemical Processes in Aqueous Solutions of Uranyl Salts

93

This is a study of the electrochemical nature of the redox components in the radiolysis of uranyl salt solutions. Certain conditions were established for the formation of the "hydrogen" and "oxygen" potentials in this system. A stationary potential of the Pt electrode develops during γ -irradiation due to the emergence of nonequilibrium concentrations of U(V). The stationary potential at the Au electrode in uranyl sulfate solutions is $\sim 1.1v$, while at the Pt electrode it shifts towards negative values. There are 9 figures and 12 references, of which 6 are Soviet and 6 English.

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Rozenfeld, I.L., Oshe, Ye.K. Mechanism of Activation of Electrodes
of Local Cells During Irradiation

103

This paper discusses the effect of radiation on the properties of semiconducting oxide films of the electrodes. It was determined that only irradiation of the cathode intensifies the corrosion of metals in electrolytes. This is due to the sharp increase in the cathode efficiency resulting from "radiation conductivity" in the protective film which is regarded as a semiconductor. The radiation dependence of the corrosion current is expressed by:

$$I_k = I_k^0 + A\sqrt{I},$$

where A is the constant for the given pair of electrodes. Candidate of physical and mathematical sciences V.B. Sandomirskiy participated in this work. There are 10 figures and 6 references of which 1 is Soviet, 3 English, 1 French and 1 Czech.

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Effect of Ionizing Radiation (Cont.)

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Duzhenkov, V.I., Dolin, P.I. Effect of X-rays on Diluted Aqueous Solutions of Organic Substances

114

The effect of organic additives on the yield of molecular products from the radiolysis of water is studied in solutions of varying acidity. It was determined that the initial yield of H_2 and H_2O_2 increases, and their fixed concentrations are lowered. This appears to be due to the fact that the radical products of radiolysis: H , OH , HO_2 react with the organic molecules more vigorously than the molecules H_2 and H_2O_2 . There are 3 figures, 2 tables, and 9 references, of which 3 are Soviet and 6 English.

Chernov, A.I., Orekhov, V.D., Proskurin, M.A. Sensitization and Inhibition of Radiolytic Reduction of the Uranyl Ion in Aqueous Solutions

120

The radiolytic reduction of uranyl ions was sensitized with glucose and glycerin, and inhibited with methylene blue. The reduction of UO_2^{2+} does not exceed 0.5 molecules/100 ev for low concentrations of the uranyl ion ($5 \cdot 10^{-3}$ M) and of glucose or glycerin ($5 \cdot 10^{-3}$ M). Concentration of 0.5 M glycerin leads to reduction of 5 molecules/100 ev. The inhibiting effect of methylene blue is modified by the ratio of concentrations of two substances reacting with H ,

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the product of radiolysis: for $C_{UO_2^{2+}} : C_{dye} = 1:1$ the dosage of inhibition of the dye reaches 10^{16} ev/ml and decreases with the increase of this ratio. The participation of ionized and excited molecules of water in these processes is taken into consideration. Co^{60} γ -radiation was used for irradiation. There are 3 figures and 19 references of which 8 are Soviet and 11 English.

PART 2. REACTIONS OF INORGANIC SUBSTANCES

Buneyev, N.A., Myasnikov, I.A., Pshezhetskiy, Formation of Ozone in Liquid Oxygen Due to γ -Radiation

129

The formation of ozone in LOX due to Co^{60} radiation was studied with respect to duration and intensity of irradiation. It was found that the yield of ozone is 13.6 molecules per 100 ev of the absorbed γ -radiation energy. There are 4 tables, 3 figures, and 3 Soviet references.

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Effect of Ionizing Radiation (Cont.)

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Buneyev, N.A., Pshezhetskiy, S.Ya., Myasnikov, I.A. Formation of Ozone in Gaseous Oxygen Due to Fast Electrons

133

The effect of electrons on oxygen was studied at room temperature and atmospheric pressure, under dynamic and static conditions. The determined stationary concentration of ozone equals 0.1 - 0.2 per cent for 200 Kev electrons and current of 50 - 100 μ a. This is 4 to 8 times less than for LOX under analogous conditions. The efficiency of the reaction is \sim 1.5 molecules per 100 ev. There are 9 figures, 5 tables, and 10 references, of which 3 are Soviet, 4 English, 2 German and 1 Belgian.

Dmitriyev, M.T., Pshezhetskiy, S.Ya. Radiation Oxidation of Nitrogen. Part 1. Kinetics of the Nitrogen Oxidation Reaction due to Electron Impact and the Effect of Ionization Processes

145

This paper deals with kinetics of ionization and oxidation of nitrogen due to the impact of 0 - 400ev electrons at pressures up to 0.1 mm Hg. The reaction rate is defined by an equation of the second order and the activation energy of the reaction equals 7.4 ± 1 kcal/M. The obtained data show that ionization of molecular nitrogen is the primary elementary process leading to oxidation of nitrogen, and that it is followed by direct interaction of the molecular nitrogen ion with a molecule of oxygen and the interaction

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Effect of Ionizing Radiation (Cont.)

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of oxygen with products of the dissociation of the molecular nitrogen ion. There are 18 figures, 12 tables, and 27 references of which 4 are Soviet, 19 English, 3 German, and 1 Belgian.

Dmitriyev, M.T., Pshezhetskiy, S.Ya. Radiation Oxidation of Nitrogen. Part 2. Kinetics of the Oxidation of Nitrogen Due to Fast Electrons

171

The oxidation of nitrogen due to fast electrons is studied in the gaseous phase and at atmospheric pressure. The reaction kinetics are defined by an equation of the second order. The reaction rate is proportionate to the electron flux and to the ratio of gas layer thickness to the maximum electron path in gas. At the temperature of 40° the yield is 1.3 atoms of bound nitrogen for air, and 2 atoms for a 1 : 1 gas mixture per 100 ev of absorbed energy. The differences in the absolute values of reaction rate constants are interpreted as the result of variations in the recombination of ions. There are 10 figures, 2 tables, and 9 references of which 7 are Soviet and 2 English.

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Effect of Ionizing Radiation (Cont.)

790

Bol'shun, Ye.V., Pshezhetskiy, S.Ye., Myasnikov, I.A. Formation of Hydrazine in Liquid Ammonia Due to Fast Electrons

182

The formation of hydrazine due to fast electrons leads to a stationary hydrazine concentration conditioned by the balancing of the rates of direct and inverse reactions. The reaction efficiency is 1 - 1.2 molecules of hydrazine per 100 ev. There are 2 tables, 1 figure, and 6 references of which 3 are Soviet and 3 English.

Baberkin, A.S., Proskurnin, M.A., Orekhov, V.D. The Effect of γ -Radiation on Solid Potassium Nitrate

186

The γ -irradiation of potassium nitrate leads to evolution of gas and nitrite formation. The irradiated salt shows a sharp increase of gas evolution when heated to $\sim 129^\circ$. A decrease in the amount of initially formed nitrite is observed when the temperature rises above 122° (temperature of lattice destruction for KNO_3). The decrease in the nitrite yield is connected with the inverse reaction of nitrate formation which occurs due to the presence of atomic oxygen or free radicals in the irradiated salt.

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Effect of Ionizing Radiation (Cont.)

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There are 4 figures and 12 references of which 2 are Soviet, 9 English, and 1 German.

Baberkin, A.S., Proskurnin, M.A., Orekhov, V.D. Effect of γ -Radiation on Solid Potassium Perchlorate and Chlorate

193

Potassium perchlorate and chlorate powders were subjected to the effect of Co^{60} γ -radiation in doses up to 10 megaroentgens per 1 g of the salt. It was established that the conversion of $KClO_4$ at temperatures up to 100° yielded \sim 1.1 molecules of $KClO_3$ /100 ev and no other products. At temperatures up to 200° the yield of the chlorate drops, the chloride appears and gas evolution occurs. At temperatures above 320° the yield of the chlorate drops to 0.7, and the amount of the other products increases. The irradiated $KClO_3$ yields 0.4 molecules of KCl and 1.2 molecules of $KClO_2$ per 100 ev. At elevated temperatures the yield of the chlorite drops to zero. It is assumed that the thermal decomposition of irradiated salts has a radical character facilitating the conversion of the intermediate products: ClO_4 , ClO_3 , O , and K . Doses not exceeding 10 megaroentgens per 1 g cause coloration of the salts. This is connected with the formation of coloration centers in the crystal lattice which disappear at temperatures above 100°.

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Effect of Ionizing Radiation (Cont.)

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There are 7 figures and 10 references of which 3 are Soviet,
6 English, and 1 German.

PART 3. REACTIONS OF ORGANIC SUBSTANCES

Krongauz, V.A., Bagdasar'yan, Kh.S. Energy Transfer in the Radiolysis of
Benzoyl Peroxide Solutions

205

This paper considers the radiolysis of benzoyl peroxide solutions in benzene, cyclohexane, and ethyl acetate. In the benzene solution of peroxide, the transfer of excitation energy is from the solvent to the solute. This effect was not detected in cyclohexane or ethyl acetate solutions, in which the radiation decomposition of peroxide shows a chain mechanism. The authors determined yields of radicals for the solvents and the peroxide. Small additions of anthracene and phenanthrene to benzene inhibit the decomposition of peroxide. There are 7 figures, 4 tables, and 15 references of which 2 are Soviet, 11 English, and 2 German.

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Effect of Ionizing Radiation (Cont.)

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Mikhaylov, B.M., Tarasova, L.V., Bogdanov, V.S. Radiochemical Conversion of Organic Substances. Part 1. Conversion of Gaseous Aliphatic Hydrocarbons Due to Fast Electrons

218

Methane, ethane, propane, and n-butane were irradiated with a 90 KeV electron beam. Dehydrocondensation is the basic process of radiolysis. Liquid hydrocarbons constitute 50 percent of the conversion products of methane, and 70 - 90 percent of the conversion products of ethane, propane, and n-butane. There are 3 tables, 2 figures, and 7 English references.

Mikhaylov, B.M., Kumova, M.Ye., Bogdanov, V.S. Radiochemical Conversion of Organic Substances. Part 2. Oxidation of Methane with Oxygen Due to Fast Electrons

223

A mixture of methane and oxygen ($\text{CH}_4 : \text{O}_2 = 4 : 1$ and $1 : 1$) was irradiated with fast electrons. CO , CO_2 , H_2 , and H_2O were found in the reaction products. About 50 percent of the methane was converted to the liquid phase. The rate of oxidation increases with methane content (50 - 80%), and with increased pressure (190 - 760 mm Hg). The mechanism of the reaction is regarded as radical.

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Effect of Ionizing Radiation (Cont.)

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There are 9 figures, 4 tables, and 19 references of which 8 are Soviet, 10 English, and 1 German.

Vereshchinskiy, I.V., Bakh, N.A. Oxidation of Organic Compounds with Molecular Oxygen Due to Ionizing Radiation. Part 5. Effect of Radiation From a Nuclear Reactor on Benzene and Cyclohexane in Presence of Molecular Oxygen

234

Liquid benzene and cyclohexane were subjected to radiation from a nuclear reactor (recoil protons and γ -radiation) in the presence and absence of oxygen. Benzene oxidation products included phenol and aldehydes. Cyclohexanes gave hydroperoxides, disubstituted peroxides, cyclohexanone, and acids. Yields are similar to those obtained from low-density ionization. Gaseous products of benzene radiolysis in vacuum contain hydrogen and acetylene. The amount of molecular hydrogen increases in the presence of oxygen. The radiolysis of cyclohexane in vacuum yields hydrogen in amounts equal to number of molecules of cyclohexane which take part in polymerization. The presence of oxygen does not increase the hydrogen yield. There are 11 figures, 1 table, and 31 references of which 8 are Soviet, and 23 English.

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Sarayeva, V.V., Bakh, N.A., Rybin, L.V., Larin, V.A. Oxidation of Organic Compounds with Molecular Oxygen Due to Ionization Radiation. Part 6. Identification of Individual Products Obtained From the Oxidation of N-Heptane and Isooctane

248

This paper discusses the identification of individual carbonyl compounds and acids, and the accumulation of alcohols formed during the radiolytic oxidation of n-heptane and isooctane. It was shown that most of the carbonyls are composed of compounds with the number of carbons close to that of the original hydrocarbon molecule. There are 15 figures, 7 tables, and 15 references of which 7 are Soviet, 7 English and 1 French.

Sarayeva, V.V., Rudenko, B.I. Radiolytic Oxidation of Acetone

263

Acetone was irradiated with Co^{60} γ -radiation in doses of 2×10^{13} and 1×10^{14} ev/cm³. sec and with X-rays of 2×10^{16} ev/cm³ sec. in the presence of oxygen and at temperatures of 18 - 20°. Acids and aldehydes were the products of oxidation. No peroxides were detected. The amount of aldehydes increased with the increase of the dosage rate. The initial aldehyde yield is ~ 80 molecules per 100 ev and it is independent of the dosage rate. There are 2 figures and 6 references, 3 of which are Soviet and 3 English.

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Sarayeva, V.V., Kinetics of Interaction of Organic Peroxides With the Iodide Ion in Acetic Acid and the Determination of Peroxides Obtained From Radiolytic Oxidation of Hydrocarbons

266

This is a study of the rate of interaction of sixteen organic peroxides with potassium iodide in glacial acetic acid. It was determined that the peroxides differ in rate constants but the rate constant for a given type of peroxides varies only slightly with changes in the structure of the hydrocarbon chain of the peroxide. Radiolytic oxidation yields hydroperoxides and peroxides from n-heptane, isooctane and toluene. Cyclohexane and tetralin yield only hydroperoxides. There are 6 figures, 2 tables, and 16 references of which 6 are Soviet, 8 English, and 2 German.

Zimin, A.V., Churmanteyev, S.V. The Effect of γ -Radiation on Benzene-Ammonia Mixture

273

This paper considers the effect of dosage, temperature, and aggregate state on the yield of aniline, hydrogen, and nitrogen. The activity of the Co^{60} radiation source was ~ 80 and ~ 700 g-equiv., and the period of irradiation was 3 to 256 hours. The basic products of radiolysis in the absence of air were aniline, hydrogen, nitrogen, and

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polymerization products. Dosage and temperature increases (to +90°) have no significant effect on the yield of aniline. The benzene - ammonia mixture irradiated in the presence of oxygen shows a sixfold increase in aniline yield. The greater amount of radicals is due to decrease in the recombination of primary products of radiolysis, since atomic hydrogen is bound by molecular oxygen. There are 4 tables, 2 figures, and 8 references of which 3 are Soviet, 3 English, and 2 French.

Zimin, A.V., Verina, A.D. The Effect of γ -Radiation on the Mixture of Benzene With Carbon Tetrachloride

280

The mixture was irradiated with Co^{60} γ -radiation (activity 80, 130, and 1450 g-equiv.). The authors studied the effect of the dosage and molar ratio of components on the yield of HCl and of the non-volatile residue. Variation of the dosage had no effect and the ratio variation of C_6H_6 : CCl_4 from 4 : 1 to 1 : 4 only slightly changed the results. The nonvolatile residue contains 70 percent of condensation products, \sim 20 percent of high boiling fractions, and \sim 10 percent of benzene. The most probable products of the high boiling fractions are monochlorobenzotrichloride isomers with a yield of \sim 0.7 molecules per 100 ev. There are 3 tables and

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3 references of which 1 is Soviet and 2 German.

Vereshchinskiy, I.V., Karpushkin, L.T. Effect of γ -Radiation on the Synthesis of Indophenol

285

The synthesis of indophenol from an irradiated mixture of *a*-naphthol and *n*-phenylene diamine was studied. The radiochemical yield of indophenol is 4.7 molecules per 100 ev, and the yield of the leuco form from the dye solution in vacuum is \sim 1.8 molecules per 100 ev. It was shown that the formation of the dye is a two-stage process. The first stage is the formation of the leuco form. It does not require the presence of molecular oxygen. The second stage is the conversion of the leuco form into the dye. This stage occurs only in the presence of molecular oxygen. Irradiation considerably accelerates the conversion of the leuco form. There are 9 figures and 12 references of which 4 are Soviet, 6 German, 1 English, and 1 French.

PART 4. EFFECT OF RADIATION ON POLYMERS

Slovokhotova, N.A. Infrared Spectroscopic Study of Chemical Changes Occurring in Polytetrafluoroethylene (Teflon) Due to Ionizing Radiation

295

The infrared spectra of Teflon indicated that the conjugate

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Effect of Ionizing Radiation (Cont.)

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C = C double bonds form during irradiation with fast electrons and with Co^{60} γ -radiation. Irradiation in air leads to the interaction of teflon with water vapor with the formation of C = O, OH, and CH groups. Irradiation converts crystalline teflon into amorphous teflon and possibly causes the formation of perfluorocyclobutene rings. There are 7 figures and 30 references, of which 3 are Soviet, 26 English, and 1 French.

Taubman, A.B., Yanova, L.P. Study of the Radiation Stability of High Polymers. Part 1. Effect of Radiation on Diffusion Permeability

307

The effect of radiation on diffusive permeability of high-polymer films is conditioned by the physical state and structure of the polymers. Vitreous polyvinylmethacrylate with low permeability to gases shows a sharp increase in permeability after relatively small radiation doses. This is explained by the radiational instability of the densely packed macromolecules. The coefficients of permeability and diffusion through polyvinyl chloride and polyethylene were computed for hydrogen chloride. There are 3 figures, 2 tables and 9 references of which 3 are Soviet and 6 English.

Card 24/31

Effect of Ionizing Radiation (Cont.)

790

Yanova, L.P., Taubman, A.B. Study of the Radiation Stability of High Polymers. Part 2. Role of Gas Formation in the Destruction of Polymers 314

The role of gas formation was studied in the destruction of polytetrafluorethylene, polymethylmethacrylate and polyethylene due to electron irradiation. The intensity of the process of formation of gaseous destruction products increases sharply in a narrow temperature range of polymer fusion or transition to visco-fluid state. It was shown that the destruction should be regarded as a reversible process: destruction \rightleftharpoons recombination of free radicals. The equilibrium shifts to the left with transition of the polymer into the fluid state. Changes in the mechanical properties of polymers and finally their destruction, are determined by the ratio of decomposition rates to gas formation rates, and, therefore, depends on the permeability to gases of the sample subjected to fast-electron irradiation. Crack formation in the polymer due to irradiation is traced to gas formation and emanation. There are 8 figures, 3 tables, and 9 references of which 6 are Soviet, and 3 English.

Kargin, V.A., Taubman, A.B., Yanova, L.P., Belyayeva, Z.F. Effect of Ionizing Radiation on the Properties of Vinyl Chloride and Vinylidene Chloride Copolymers

325

The effect of irradiation on gas permeability and mechanical

Card 25/31

Effect of Ionizing Radiation (Cont.)

790

properties of the copolymers is connected with changes in the mixed amorphous - crystalline states and microstructure of the copolymers. The presence of crystallizing components increase the microdefects which results in increased permeability to gases. A reverse process can be observed for a specific range of small doses when the crystalline component undergoes fusion and the material becomes amorphous. Therefore, the curves for the permeability coefficient as a function of dosage have a maximum and a minimum. There are 8 figures, 2 tables, and 7 references of which 6 are Soviet, and 1 French.

Nikitina, T.S., Kuz'minskiy, A.S., Karpov, V.L. Radiation Vulcanization of Rubber

333

This is a study of radiation vulcanization of pure rubbers: natural rubber (NK), butadiene-styrene rubber (SKS-30) butadiene rubber (SKB) and butadiene-nitril rubber (SKN-26). The tensile strength of the radiation vulcanizates of NK does not exceed 30kg/cm. Sulfur and certain sulfur compounds (tetramethylthiuram-disulfide) lower the rate of the vulcanization. Compounds containing heavy atoms (zinc oxide, kaolin, chalk) accelerate the vulcanization only when X-rays are used. Gas and lamp blacks are very active fillers.

and 26/31

Effect of Ionizing Radiation (Cont.)

790

The effect of carbon blacks is modified by their degree of oxidation. There are 8 figures, 1 table, and 17 references of which 7 are Soviet, 8 English, 1 French, and 1 German.

Tsetlin, B.L., Sibirskaya, G.K. Effect of Ionizing Radiation on the Thermomechanical Properties of Polyethylene

344

Polyethylene undergoes vulcanization when irradiated with electrons and X-rays. At a temperature corresponding to the melting point of the initial polymer, the irradiated substance converts to a highly elastic state the modulus of which is proportional to the dosage. In the range of 10^{16} - 10^{19} ev/cm sec vulcanization effectiveness follows the principle of equivalent effect of equal doses. The thermal stability of polyethylene increases with increased dosage. The mechanism of radiation vulcanization is related to the recombination of the $R_1 \cdot \dot{C}H \cdot R_2$ radicals which result from primary rupture of the C - H bonds. There are 4 figures, 1 table, and 22 references of which 14 are Soviet, 7 English, and 1 French.

Card 27/31

Effect of Ionizing Radiation (Cont.)

790

Tsetlin, B.L., Yanova, L.P., Sibirskaya, G.K., Korbut, V.M.
Effect of Ionizing Radiation on the Mechanical Properties of
Polyvinyl Chloride and Its Plastics

354

Polyvinyl chloride undergoes vulcanization due to the effect of hard radiation. The changes in mechanical properties result from the formation of a three-dimensional lattice and the occurrence of multiple inner micro-defects (due to gas formation). Polyvinyl chloride base plastics also undergo radiation vulcanization. The rate of the process decreases with the increase of the plasticizer content. There are 5 figures, 1 table, and 16 references of which 10 are Soviet, 3 German, and 3 English.

Tsetlin, B.L., Zaytseva, N.G., Korbut, V.M., Kargin, V.A.
Effect of Ionizing Radiation on Vitreous Polymers

362

This paper reports an experimental study of radiational destruction of vitreous polymers: changes in the thermo-chemical properties, gas formation, dendritic fissures. The modifying factors are: the stabilizing effect of aromatic groups, greater probability of bond rupture in the main chains of macromolecules due to the presence of tertiary

Card 28/31

Effect of Ionizing Radiation (Cont.)

790

carbon atoms in them, decrease in the rate of radiation destruction of polymethylmethacrylate and its analogs with increase in the size of side groups, intensification of the destruction process in the presence of low molecular weight plasticizers. The fissure formation is interpreted as having an adsorption-type mechanism. The process of radiochemical conversion of plexiglass is regarded as irreversible. There are 8 figures, and 27 references of which 11 are Soviet, 14 English, and 2 French.

PART 5. EXPERIMENTAL METHODS

Breger, A.Kh., Belynskiy, V.A., Karpov, V.L., Prokudin, S.D. Equipment for Radiochemical Research. Part 2. Equipment Supplying Doses of up to 300 Roentgen/Sec in 30 ml and up to 100 Roentgen/Sec in 1 Liter, From a Co^{60} γ -Radiation Source With an Activity of 1400 Radium Gram-Equivalent

380

The first part of this paper gives general considerations on the features of units with Co^{60} γ -radiation sources used in radiochemical research. Further, details are given on the new unit K-1400 (improved K-300) which supplies doses of 300 roentgen/sec in 30 ml and 100 roentgen/sec in 1 liter using three standard Co^{60}

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Effect of Ionizing Radiation (Cont.)

790

sources with a total activity of 1440 radium Gram-equiv. A method was developed for safe, "dry" assembling of powerful sources from smaller standard cobalt charges. The K-1400 proved itself efficient safe during one year of operation. There are 6 figures and 22 references of which 9 are Soviet, and 13 English.

Glazunov, P.Ya., Radziyevskiy, G.B. Equipment for the Application of 1 Mev Accelerated Electrons in Radiochemical, Radiobiological, and Other Research Work

395

This paper describes some instrumentation developed and used in the laboratory for working with 1 Mev electrons and X-rays. The accelerator generates continuous and pulsed electron and hard X-ray radiation. The electron flux is measured by means of an ionization chamber (fig. 2). The distribution of electron-flux density is determined by means of densitometers (fig. 7). Directional control of the beam for vertical or horizontal irradiation is achieved by means of a magnetic system (fig. 8) and automatic stabilizing device (fig. 9). Pulse technique with given duration and intervals was achieved with the aid of a pulse regulator (fig. 10).

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Effect of Ionizing Radiation (Cont.)

790

Generation of single pulses of relatively long duration is done by means of a rotating shutter (fig. 12) efficient in the range from 10 millisecc. to 2 sec. Wider versatility of the high-voltage electron accelerator tube was gained by the introduction of a gold target for the generation of hard bremsstrahlung (fig. 13). The X-ray radiation is measured by means of an ionization chamber made of a plastic and lined with aluminum foil. There are 13 figures, and 4 references, 1 Soviet and 3 English.

Zatulovskiy, V.I., Naryadchikov, D.I. X-ray Equipment as Radiation Source for Radiochemical Research

406

The Laboratory of Radiochemistry at the Institute of Physical Chemistry of the USSR (IFKh AN SSSR) developed two types of X-ray apparatus for research purposes. The apparatus and control instruments are described in this paper. The units are: ARKh - 200 (fig. 1) and ARKh - 100 - 20 (fig. 5). There are 7 figures and no references.

AVAILABLE: Library of Congress

Card 51/31

TM/mas
1-30-59

KIRGINTSEV, A.N.; GVOZDEV, B.A.

Cocrystallization of strontium and potassium chromates. Zhur.
neorg. khim. 5 no.10:2374-2376 O '60. (MIRA 13:10)

1. Institut fizicheskoy khimii Akademii nauk SSSR.
(Strontium chromate) (Potassium chromate)
(Crystallisation)

PERELYGIN, V.P.; ALMAZOVA, S.P.; GVOZDEV, B.A.; CHUBURKOV, Yu.T.

[Spontaneous fission with an anomalously short period]
Spontannoe delenie s anomal'no korotkim periodom. Dubna,
Ob"edinennyi in-t iadernykh issl. Vol.2. 1962. 7 p.
(Nuclear fission) (MIRA 15:1)

DONETS, Ye.D.; KARNAUKHOV, V.A.; KUMPF, G.; GVOZDEV, B.A.; CHUBURKOV,
Yu.T.; SARANTSEVA, V.R., tekhn. red.

[Study of the nuclear reaction $\text{Th}_{90}^{232}(\text{Ne}_{10}^{22}, 4n)\text{Fm}_{100}^{250}$] Izuchenie
iadernoi reaktsii $\text{Th}_{90}^{232}(\text{Ne}_{10}^{22}, 4n)\text{Fm}_{100}^{250}$. Dubna, Ob"edinennyi
in-t iadernykh issl., 1962. 10 p. (MIRA 15:4)
(Nuclear reactions)

38856

S/056/62/042/006/008/047
B104/B102

24.6600

(2806)

AUTHORS: Perelygin, V. P., Almazova, S. P., Gvozdev, B. A.,
Chuburkov, Yu. T.

TITLE: Spontaneous fission with anomalously short period. II.

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 42,
no. 6, 1962, 1472 - 1474

TEXT: Fission fragments of the spontaneous fission resulting from the
interaction of 135 Mev Ne²² ions with U²³⁸ in an ionization chamber were
studied using T-1 (T-1) and P-8 (P-8) photographic plates. The U²³⁸
target, 1 mg/cm² thick, was exposed for a few hours to an ion current of
~1 μa from the internal beam of the OIYaI cyclotron. A strong γ back-
ground as well as a background of α particles were detected. 60 tracks of
spontaneous fission fragments were found. The registration efficiency of
the events was 50%. The half-life of the unknown isotope is 17 ± 7 milli-
sec; the production cross section on an interaction of 135 Mev Ne²² with

Card (1/2)

24 (660)

39661
S/056/62/043/001/003/056
B181/3102

AUTHORS: Donets, Ye. D., Karnaukhov, V. A., Kumpf, G., Gvozdev, B. A.,
Chuburkov, Yu. T.

TITLE: The nuclear reaction ${}_{90}^{232}\text{Th}({}_{10}^{22}\text{Ne}, 4n){}_{100}^{250}\text{Fm}$

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 43,
no. 1(7), 1962, 11 - 15

TEXT: Measurements were made of the dependence of the
 ${}_{90}^{232}\text{Th}({}_{10}^{22}\text{Ne}, 4n){}_{100}^{250}\text{Fm}$ reaction cross section on the energy of the
bombarding ions. The ions were extracted from the 300 cm cyclotron of
the OIYaI. A thorium foil, 2 - 2.6 mg/cm² thick, served as a target and
a 3 μ thick silver foil chemically prepared with tencoiltrifluoro acetone,
was used to collect the recoil nuclei. Fermium (yield 50 %) was separated
from the organic phase by anodic precipitation. Fm^{250} was identified
from its 7.43 Mev α-emission. The 7.65 Mev line of Po^{214} was found to
interfere. The reaction cross section has its maximum of $\sim 2.5 \cdot 10^{-31}$ cm²
Card 1/2

The nuclear reaction ...

S/056/62/043/001/003/056
B181/B102

at an ion energy of 107 Mev, and has a half-width of about 11 Mev. The cross section of the reaction ${}_{92}\text{U}^{238}({}_8\text{O}^{16}, 4n){}_{100}\text{Fm}^{250}$, which was investigated earlier (T. Sikkeland, S. G. Thompson, A. Ghiorso, Phys. Rev., 112, 543, 1958; V. P. Pereygin, Ye. D. Donets, G. N. Flerov, ZhETF, 37, 1558, 1959), reached a maximum of 10^{-30} cm^2 , that of the reaction ${}_{94}\text{Pu}^{241}({}_6\text{C}^{13}, 4n){}_{100}\text{Fm}^{250}$ one of $6 \cdot 10^{-30} \text{ cm}^2$. The experiments showed that the maximum cross section decreases much faster with increasing mass of the bombarding particles than is predicted by the theory. This is explained as follows: Either the nucleus is deformed in a collision so that the Coulomb barrier increases, or the system of the two nuclei is excited to perform vibrations so that the probability of fission prior to emission of the first neutron increases. There are 3 figures and 1 table.

ASSOCIATION: Ob"yedinennyy institut yadernykh issledovaniy (Joint Institute of Nuclear Research)

SUBMITTED: January 24, 1962
Card 2/2

ACCESSION NR: AP4009948

S/0186/63/005/006/0712/0715

AUTHOR: Gvozdev, B. A.; Chuburkov, Yu. T.

TITLE: Producing oxide layers of Th, U, Pu and rare-earth elements by electrolytic precipitation from organic solutions

SOURCE: Radiokhimiya, v. 5, no. 6, 1963, 712-715

TOPIC TAGS: heavy ions, metal foil, irradiation targets, thorium oxides, plutonium oxides, uranium oxides, electrolytic precipitation, acetone, methyl ethyl ketone, isobutyl alcohol

ABSTRACT: Layers of lanthanide and similar elements were spread on thin metal foil and used as targets in the study of the nuclear reactions produced by heavy ion irradiation. A new method has been suggested for the production of Th, U, Pu, Am, rare earth and certain other elements on thin aluminum foil (3-6 microns) by electrolytic precipitation from organic media. Salt solutions in acetone, methyl ethyl ketone or isobutyl alcohol are used as electrolytes. The five micron-thick aluminum foil targets with oxides of a number of differ-

Card 1/2

ACCESSION NR: AP4009948

ent elements were able to withstand prolonged irradiation by intensive beams (~5 microampere/cm²) of accelerated heavy ions. A method has been developed for producing oxide layers of lanthanide and certain other elements by electrolytic precipitation from an organic medium. Targets have been developed for a cyclotron of multicharge ions on thin (5 micron) aluminum foil with an area of 5 cm² consisting of thorium, uranium, plutonium and numerous other oxides. "In conclusion, the authors express their sincere gratitude to V. V. Volkov, group leader and senior scientist, for his constant interest in the work." Orig. art. has: 4 figures.

ASSOCIATION: none

SUBMITTED: 09Jun62

DATE ACQ: 07Feb64

ENCL: 00

SUB CODE: EL, NS

NO REF SOV: 003

OTHER: 003

Card 2/2

BREDEL', V.V., V.V.; GVOZDEV, B.A.; FOMICHEV, V.A. [deceased]

Relation of the cross sections of formation of Sc^{44} , Sc^{44m}
isomer pairs in reactions involving heavy ions. Zhur. eksp. i
teor. fiz. 45 no.4:904-909 0 '63. (MIRA 16:11)

1. Ob'yedinennyy institut yadernykh issledovaniy.

ACCESSION NR: AP4042561

S/0056/64/046/006/2017/2022

AUTHORS: Fomichev, V. A. (deceased); Gvozdev, B. A.; Bredel', V. V.

TITLE: Ratio of cross sections for the production of the Sc-44, 44m isomer pair in the reactions of direct interaction with heavy ions

SOURCE: Zh. eksper. i teor. fiz., v. 46, no. 6, 1964, 2017-2022

TOPIC TAGS: scandium, isomerism probability, ion interaction, direct interaction, stripping reaction

ABSTRACT: This is a continuation of an earlier investigation. (ZhETF, v. 45, 104, 1963) devoted to the energy dependence of the production cross sections of Sc^{44,44m} in heavy-ion reactions proceeding via compound-nucleus formation. To determine the energy dependence of the cross sections for the formation of the isomer pair Sc^{44,44m} in reactions with heavy ions, when complete fusion of the nuclei does

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ACCESSION NR: AP4042561

not occur, the authors measured the Sc^{44m}/Sc^{44} isomer ratio in neutron stripping reactions induced by irradiating Sc with O^{16} or Ne^{22} ions. The irradiation was carried out in the internal beam of the U-300 cyclotron of OIYAI, the targets were thin layers of scandium of varying thickness on various substrates, depending on the bombarding gas. The gamma radiation from the specimen was measured in a scintillation spectrometer with a 100-channel analyzer. The results show that in many-nucleon transfers (3pn), the isomer ratio in the $Al^{27} + Ar^{40} \rightarrow Sc^{44,44m}$ reaction becomes nearly equal to 6.5, i.e., larger than in reactions of complete fusion of nuclei. In the case of one-neutron transfer reactions, the isomer ratio is of order of unity and independent of the energy, being somewhat larger for neon ions than for the oxygen ions. "The authors are grateful to G. N. Flerov, S. M. Polikanov, and Ye. D. Donets for a discussion of this work, and to A. N. Filipson and his crew for operating the cyclotron." Orig. art. has: 7 figures and 2 formulas.

Card 2/3

ACCESSION NR: AP4042561

ASSOCIATION: Ob"yedinenny*y institut yaderny*kh issledovaniy (Joint
Institute of Nuclear Research)

SUBMITTED: 19Jan64

DATE ACQ:

ENCL: 00

SUB CODE: NP

NR REF SOV: 004

OTHER: 003

Card 3/3

L 00036-66 EWT(m)/EWP(j)/T/EWP(t)/EWP(b) IJP(c)

JD/RK
UR/0186/65/007/004/0419/0423

ACCESSION NR: AP5020304

542.61:546.799.6 +
546.799.90:54-145.4

AUTHOR: Gvozdev, B. A.; Chuburkov, Yu. T.

TITLE: Extraction of curium and fermium with thenoyltrifluoroacetone

B²⁴²
B²⁵²

SOURCE: Radiokhimiya, v. 7, no. 4, 1965, 419-423

TOPIC TAGS: curium compound, fermium compound, complex compound, solvent extraction

ABSTRACT: The purpose of this work was to study the quantitative extraction of curium and fermium by solutions of thenoyltrifluoroacetone in toluene from acetate solution and solutions of -hydroxyisobutyric acid at different concentrations and pH values and to determine the stability constants of the complexes which are formed. The distribution of fermium and curium between the organic and the aqueous phase was investigated using Cm²⁴² and Fm²⁵² isotopes. Cm and Fm were simultaneously extracted from a 0.1 M solution of acetate buffer at different pH values.. After extraction phases were separated and dried on gold planchets for measurement of the α-activity. In another series of experiments curium was extracted with a 0.2 M solution of thenoyltrifluoroacetone from 0.1 and 0.4 M solutions of α-hydroxy-

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L 00036-66

ACCESSION NR: AP5020304

isobutyric acid. The extraction constants were calculated and also the two phase stability constant (β_{zQ_z}) of curium and fermium complexes with thenoyltrifluoroacetone:

$$K_{Cm} = (2.2 \pm 0.3) \cdot 10^{-9}; K_{Fm} = (2.6 \pm 1.1) \cdot 10^{-8}; (\beta_{zQ_z})_{Cm} = (2.1 \pm 0.3) \cdot 10^{15}$$

$(\beta_{zQ_z})_{Fm} = (2.3 \pm 1.0) \cdot 10^{16}$. Assuming that in the investigated concentration and pH range the curium complex with α -hydroxyisobutyric acid is of the CmB form, a complete stability constant was calculated for this complex, equal to $(2.3 \pm 0.6) \cdot 10^{-6}$. The authors wish to express their gratitude to I. Stary for valuable comments during the discussion of this work; to Ye. D. Donets and G. Kumpf for permitting the use of their α -spectrometer. Orig. art. has: 3 figures and 2 tables.

ASSOCIATION: none

SUBMITTED: 24Dec64

ENCL: 00

SUB CODE: IC, GC

NO REF SOV: 001

OTHER: 007

LW
Card 2/2

L 00037-66 EWT(m) DIAAP
ACCESSION NR: AP5020306

UR/0186/65/007/004/0453/0461

859.379.4 : 846.789.8.02.241 : 846.789.8.02.242 + 846.789.8.02.243

AUTHOR: Dedov, V. B.; Volkov, V. V.; Gvozdev, H. A.; Yermakov, V. A.; Lebedev, I. A.
Razbitnoy, V. M.; Trukhlyayev, P. S.; Chuburkov, Yu. T.; Yakovlev, G. N.

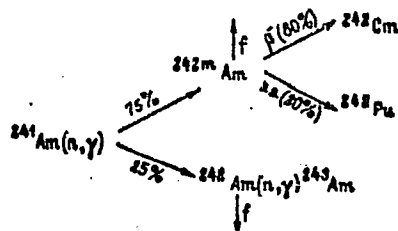
TITLE: Production of Pu-242 and Cm-242 from neutron-irradiated Am-241

79, 25

SOURCE: Radiokhimiya, v. 7, no. 4, 1965, 453-461

TOPIC TAGS: plutonium, curium, americium, extraction, neutron irradiation

ABSTRACT: Irradiation of Am-242 with thermal neutrons produces Pu²⁴², Cm²⁴² and Am²⁴³ which are of great interest in a number of physical and radiochemical investigations. The synthesis scheme is as follows:



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ACCESSION NR: AP5020306

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The thermal neutron cross section of Am^{241} is 900 barn, thus even upon short irradiation with a high density thermal-neutron beam a significant amount of the above isotopes may be produced. It can be seen from the above process that the yield of fission products is small since they are produced mainly during fission of Am^{242} . This facilitates the chemical processing of irradiated substances. Production of Pu^{242} by this process requires much less time than the method which uses Pu^{239} as starting material. The authors describe the chemical separation of Pu^{242} , Cm^{242} and Am^{243} from irradiated Am^{241} . The scheme for the chemical processing was selected to be such that it would produce rapid separation of the products. The main separation steps involved chromatographic and chemical extraction methods. Chromatographic separation was made extremely difficult by high α -activity due to the presence of Cm^{242} . Chemical processing was carried out in a shielded area on a special stand with remote control of all operations. The article indicates some properties of curium oxalate, potassium curium sulfate, curium hydroxide and curium carbonate. Orig. art. has: 5 tables and 3 figures.

ASSOCIATION: none

SUBMITTED: 18Apr64

ENCL: 00

SUB CODE: GC, NP

NO REF SOV: 004

OTHER: 005

Card 2/2 *W*

L 44428-66 EWT(m)

ACC NR: AP6023080 (AN) SOURCE CODE: UR/0367/66/003/004/0609/0613

AUTHOR: Belyayev, B. N.; Gvozdev, B. A.; Gudov V. I.; Kalyamin, A. F.;
Krizhanskiy, L. M.

34
33
B

ORG: none

TITLE: Investigation of the gamma spectrum of Br^{74} isotopes

SOURCE: Yadernaya fizika, v. 3, no. 4, 1966, 609-613

TOPIC TAGS: gamma spectrum, bromine isotope, scintillation spectrometer,
nuclear energy level, radiation intensity, cyclotron

ABSTRACT: The gamma spectra of Br isotopes, obtained in the reaction of $Cu(C^{12}, xn)Br$, have been investigated with the aid of a scintillation spectrometer. Energies and relative intensities have been measured for a number of new γ -lines in bromine with the decay half-times 36 ± 1 and 4 ± 1 min. The energies and relative intensities are given for γ -lines with $T_{1/2} = 36$ min. The isotope with $T_{1/2} = 36 \pm 1$ min was identified as Br^{74} . A diagram of the lower energy levels in Se^{74} is given in the original article. The authors thank G. N. Flerov for his

Card 1/2

L 44428-66

ACC NR: AP6023080

valuable discussions and support of this study, and B. A. Zager and his assistants for maintaining reliable operation of the cyclotron during irradiation. Orig. art. has: 2 figures and 1 table. [Based on authors' abstract] [NT]

SUB CODE: 18/ SUBM DATE: 10Jul65/ ORIG REF: 003/ OTH REF: 010

Card

2/2

KHODANOVICH, I.Ye.; GVOZDEV, B.P.; MANAYEV, V.A.

Quantitative correlation between film liquids and liquids with
suspended drops in gas and condensate flow in a pipeline. Trudy
VNIIGAZ no.13:130-134 '61. (MIRA 14:12)
(Gas, Natural--Pipelines)

GVOZDEV, B.P.; MAMAYEV, V.A.

Method of calculating the height of the settling section of an
oil dust collector. Trudy VNIIGAZ no.13:135-146 '61. (MIRA 14:12)
(Dust collectors)
(Gas, Natural---Transportation)

GVOZDEV, B.P.; ZAYTSEV, V.I.; MITROFANOV, I.A.; SHUSHLYAKOV, N.N.;
CHERNOBYL'SKIY, V.A.

Testing a remodelled vertical oil dust collector in the
"Shosseinaia" gas-distribution station. Gaz. delo no.10:13-18
'63. (MIRA 17:4)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut prirodnogo gaza (for Gvozdev, Zaytsev).
2. Leningradskoye upravleniye magistral'nykh gazoprovodov (for Mitrofanov, Shushlyakov).
3. Gosudarstvennyy proizvodstvennyy komitet po gazovoy promyshlennosti SSSR (for Chernobyl'skiy).

GVOZDEV, B.P.; ZAYTSEV, V.I.; SALTYKOV, A.L.

Similarity criterion for the separation of a drop liquid
from gas flow in the jalousie separating component. Trudy
VNIIGAZ no.21/29:152-162 '64. (MIRA 17:9)

BROKSH, M.M.; GVOZDEV, B.P.; ZAYTSEV, V.I.; ESTRINA, A.A.; SALTYKOV, A.L.

Investigating a full-scale model of a spherical scrubber, a
ball-shaped dust collector. Trudy VNIIGAZ no.21/29:172-182 '64.
(MIRA 17:9)

BROKSH, M.M.; GVOZDEV, B.P.; KVASHUK, V.S.; KOSHELEV, V.A.

Using cermet filters to remove solid impurities from natural
gas. Trudy VNIIGAZ no.21/29:205-217 '64. (MIRA 17:9)

GVOZDEV, B.F.; ZAYTSEV, V.I.

Bench tests of a ball-shaped dust collector and scrubber.
Gaz. prom. 9 no.3:27-30 '64. (MIRA 17:9)