





Card	:	1/1 Pub. 116 - 9/20	1
Authors	:	Krasovitskiy, B. M., Glinov, V. A., Matskevich, R. M. and Slavina, O. S.	
Fitle	:	On the substantiveness of dyes - benzanilide derivatives.	
Periodical	1	Ukr. khim. zhur. 20, Ed. 4, 392 - 395, 1954	
bstract	1	The effects of CO-NH grouping and amide grouping, having a non-substitut- ed H on the substantiveness of dyes - benzanilide derivatives -, were investigated. The material, necessary for the synthesis of the dyes, is described. The sharp drop in dye selectivity, due to the absence of the H-atom at the N-amide grouping, was determined on the basis of graphs. Four references: 2-USA; 1-German and 1-Italian (1921-1940)	
Institution	:	The A. M. Gorkiy State University and K. E. Voroshilov Scient Research Institute of Organ. Semi-Products and Dyes, Kharkov	·
ubmitted	:	December 21, 1953	

"APPROVED FOR RELEASE: Monday, July 31, 2000

TSUKERMAN, S.V.; KEASOVITSKIY, B.M.
Condensation of phenanthrenediamine-9,10 with certain aromatic peri-di- and peri-tetracarboxylic acids. Ukr.khin.shur. 20 no.5: 50:3-500 %54. (MIRA 8:1)
1. Shar'kovskiy gosudarstvennyy universitet in. A.M.Gor'kogo. (MIRA 8:1)
Chenanthrenediamine) (Acids. Organic)
2. Depose Depiet EASE: Mendary luby 21 2000. CLA DEPES 00E12D00020



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C. There is a strategy and the strategy	
r Asovits	KIY, B.M.
USSR/ Chemis	try - Dyes
Card 1/1	: Pub. 22 - 18/44
Authors	Krasovitskiy, B. M., and Pereyaslova, D. G.
Title	About the effect of spatial structure on the color of benzidine bisazodyes
Periodical	2 Dok. AN SSSR 98/1, 71-74, Sep 1 1954
Abstract	* Various 3,3'-, 2,2'- and 2-substituted benzidine dyes, were investigated to determine the effect of spatial structure on the color of these bisazo-benzidine dyes. The position of the biphenyl nucleus was replaced by different electro-donor and electro-acceptor substitutes for better estimation of the absorption maximum. Data on the non-planar structure of dyes derived from non-substituted benzidine and its 3,3'-di-sub- stitutes with the biphenyl nucleus as a base are included. Thirteen references: 7-USSR; 3-USA; 2-German and 1-Scandi- navian (1930-1953). Tables.
Institution	: The A. M. Gorkiy State University, Kharkov
Presented by	: Academician B. A. Kazanskiy, May 13, 1954

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KK	ALOVITCHIY, EM
USSR/Organic O	Chemistry - Synthetic Organic Chemistry, E-2
Abst Journal:	Referat Zhur - Khimiya, No 19, 1956, 61502
Author:	Krasovitskiv, B. M., Pereyaslova, D. G.
Institution:	None
Title:	Influence of Steric Factors on Properties of Dyes Containing the Biphenyl Nucleus. 2. Investigation of Bis-Azodyes. Derivatives of Biphenyl, Fluorene, Fluorenone, Carbazole and Phenanthrene Quinone
Original Periodical:	Ukr. khim. zh., 1955, 21, No 1, 71-75
•	Invéstigation of the influence of the structure of the third ring formed on bridging bond formation between 2,2'-positions of ben- zidine (I) or diphenylene (II), as concerns the color and substan- tive dye properties of disazo dyestuffs produced from such con- densed diaminos. There is presented a determination and comparison of absorption spectra and substantiveness of dyestuffs from 1,2- methylbenzidine (III), 2,7-diamino fluorene (IV), 2,7-diamino-
Card 1/2	

USSR/Organic Chemistr Synthetic Organic Chemistry, E-2 Abst Journal: Referat Zhur - Khimiya, No 19, 1956, 61502 Abstract: fluorenone (V), 2,7-diamino carbazole (VI), 2,7-diamino phenanthraquinone (VII) and 2,5-diamino fluorene (VIII): the azo-component is naphthal-l-sulfonic acid- (IX), or H-acid (X). For dyes derived from X the following results were obtained (listed are the diazocomponent, λ, in water, ε, substantiveness in 4: 4, 580, 55470, 67; III, 560, 39000, 57; IV, 590, 47760, 55; V, 570, 40470, 55; VI, 595, 51220, 65; VII, 560, 37500, 42; II, 515, --, 12; VIII, 535, ..., 50. It was confirmed that substantiveness of dyestuffs containing a biphenyl nucleus in the condensed system, is associated with coplanar disposition of the benzene rings. For dyes derived from IX (listing diazocomponent and λ_{max}); I, 540; II, 505; IV, 550; V, 530; VI, 555; VIII, 520. Communication I, see Referat Zhur - Khimiya, 1956, 4914. Card 2/2

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$\sum f (f - 1) = f$	$(\sqrt{2})^{\frac{1}{2}} = \frac{1}{2} \left(\frac{1}{2} \left(\frac{1}{2} + \frac{1}{2} \right) \right)$
USSR/Organic (Chemistry - Synthetic Organic Chemistry, E-2
Abst Journal:	Referat Zhur - Khimiya, No 1, 1957, 866
Author:	Krasovitskiy, B. M., Pereyaslova, D. G., Kovalenko, O. D., and Sherbakova, L. I.
Institution:	None
Title:	Effect of Steric Factors on the Properties of Dyes Containing the Bi- phenyl Nucleus. III. Investigation of Disazo Dyes Derivatives of Biphenol, Phenanthrene, Phenazone, and Phenanthridone
Original Periodical:	Ukr. khim. zh., 1955, Vol 21, No 5, 614-618
Abstract:	A comparative study has been made of the properties of disazo dyes (DAD) obtained from 2,7-diaminophenanthrene (I), 2,7-diaminophenazone (II), and 2,7-diaminophenanthridone (III) as the disazo constituent. The products obtained were compared with previously investigated DAD produced from benzidine (IV), 2,7-diaminofluorene (V), 2,7-diamino- phenanthraquinone (VI), 2,7-diaminodiphenyl ketone, and other 2,7- diamines (see Communication II, Referat Zhur - Khimiya, 1956, 61502).
Card 1/3	

USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 866

Abstract: I was prepared in 81% yield by the heating (5 hours at 225-230°) of 2,7-dibromophenanthreme with concentrated HNO_3 in the presence of Cu₂Cl₂ and powdered Cu in a sealed tube. II was prepared by the reduction of 2,2'-dinitrobenzidine with Na-Hg, while III was prepared by the saponification of the diacetyl derivative with 0.5 N KOH (refluxing 4 hours). The DAD were obtained by coupling the disazotized I-III with H-acid (VII), 1-naphthol-4-sulfonic acid (VIII), or 1-amino-8-naphthol-2,4-disulfonic acid (IX) in alkaline medium. DAD obtained from I or II with VII (I \rightarrow VII, λ_{max} 565 mm, directness 55%) are close in color to DAD obtained from VI \rightarrow VII and are considerably more intense than those from IV or V with VII; this is apparently due to the electron-acceptor properties of the ethylenic bridge in I and of the azo-group in II. In directness the dyes from I and II occupy an intermediate position between those from IV and VI with VII. The transition from the DAD from II to those from 3,3'diaminoazobenzene (X) is characterized by the absence of diphenyl bonds, a sharp decrease in directness, and an intensification in the color (the composition of the dye, λ_{\max} in mµ, ε_{\max} , and directness in percent are indicated in that order): $II \rightarrow VII$, 560, 40,000,

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USSR/Organic Chemistry - Synthetic Organic Chemistry, E-2

Abst Journal: Referat Zhur - Khimiya, No 1, 1957, 866

Abstract: 54; II \rightarrow VIII, λ_{max} 525 mµ; X \rightarrow VII, 525, 45,000, 31; X \rightarrow VIII, λ_{max} 490 mµ. A comparison of DAD from III with the corresponding DAD from 3,3'-(XI) and 4,4'-diaminobenzanilides (XII) and from IV shows that the introduction of the CONH-group has no marked effect on the directness, whereas the appearance of the biphenyl boxd sharply increases the directness (the composition of the dye, λ_{max} in mµ, and directness in percent are indicated in that order): IV \Rightarrow IX, 575, 59; XI \rightarrow VII, 520, 13; XI \rightarrow X, 520, 9; XI \rightarrow VIII, λ_{max} 490 mµ; XII \rightarrow VII, 550, 63; XII \rightarrow IX, 550, 56; II \rightarrow VIII, 570, 61; III \rightarrow IX, 575, 54; III \rightarrow VIII, λ_{max} 540 mµ. Trisazo dyes derived from 2,4,4'-triaminobiphenyl (VIII) / sic/ occupy an intermediate position in color and directness between the DAD from IV and from biphenyl (as above): XIII \rightarrow VII, 560, 39; XIII \rightarrow VIII, 520, --.

Card 3/3





1.Khar'kovskiy gosudarstvennyy universitet imeni A.M.Gor'kogo. (Butyl alcohol) (Amines)

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"APPROVED FOR RELEASE: Monday, July 31, 2000 KRASOVITSKIY, B. M. VEffect of space factors on the properties of dyes containing the biphenylyl radical. IV. Study of monoato dyes. B. M. Krasovitskil and T. A. Serova (A. M. Gor'kil State Univ., Kharkov). Ubrain Khim. Zhur. 22, 70-5(1950)(in Russian); cf. C.A. 50, 0021d.—The effect of substituent X (X stands for NO, Br, CH., CH.O, CH.S, HO, H.N) on the dyeing properties in mois. of the type XCH4CHAN:NR was found to be less pronounced than in the corresponding dyes of the XCH4N:NR structure. Introduction of methyl groups into 2.2° positions markedly decreased the effect of the substituent in the 4° position along the con-jugated double bonds. This prevents the plane adherence of the dyeing mol. on the fiber resulting in an increased dye-ability. A. P., Kotloby... 2 1111 . 14 2010 Constant in

KRASCVITSKIY, B.M.; KHOTINSKAYA, Ye.Ye.

Condensation of naphthalic anhydride and its derivatives with aromatic amines. Part 3: Aso dyes derived from phenylimides of naphthalic acid. Uch.zap. KHGU 71:145-154 '56. (MIRA 10:8) (Azo dyes)

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USSE/ Chemi	R (str	asovitskiy, B.M. y - Dyes	-
Card 1/2		Pub. 22 - 19/43	
Authors	ŧ	Krasovitskiy, B. M.; Ostrovskaya, B. I.; and Pereyaslova, D. G.	
Title	8	Relation between structure, color and substantiveness of benzanilide dyes	•
Periodical	t	Dok. AN SSSR 106/1, 72-75, Jan 1, 1956	
Abstract		The relation between the structure, color and substantiveness of benzanilide dyes was investigated. The increase in the color and drop in substantiveness of these dyes due to the absence of conjugated double bond chains in their molecules, are discussed. The possibility that the atructure of the dyes is less favorable for the formation of hydrogen bonds between the auxochromes of the dyes and cellulose is analyzed. It	a. U
Institution	:	Kharkov State University im. A. M. Gorkiy	4
Presented by	:	Academician B. A. Kazanskiy, July 15, 1955	

APPROVED FOR RELEASE: Monday, July 31, 2000 CIA-RDP86-00513R000826210(

A CONTRACTOR OF THE OWNER Card 2/2 Pub. 22 - 19/43 Dok. AN SSSR 106/1, 72-75, Jan 1, 1956 Periodical : Abstract was found that the reaction between benzanilide dyes results in formation • of hydrogen bonds not only between cellulose and auxochrome but also between cellulose and the amide group of the dye. The effect of the axo-group on the substantiveness of the dye is explained. Nine references: 4 USSR, 3 Germ., 1 USA and 1 French (1914-1954). Tables.



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AUTHOR3:	Krasovitskiy, B. M., Moryganov, P. V., 20-3-21/46 Titarenko, N. I., Mel'nikov, B. N.
TITLE:	A Comparative Study of the Affinity of Direct Azodyes - of the Diphenyl and p - Terphenyl Derivatives - to Cellulose Cotton Fibre (Sravnitel'noye issledovaniye srodstva pryamykh azokrasiteley - proizvodnykh difenila i para-terfenila - k tsellyuloznomu khlopkovomu voloknu).
PERIODICAL:	Doklady AN SSSR, 1957, Vol. 116. Nr 3, pp. 425-428 (USSR)
ABSTRACT :	The question of the relation between the structure or the azo dyes and their affinity to cellulose fibre, attracts since long the attention of many researchers. To enable a dye to express its substantive properties, the presence of a long chain of conjugated double bonds in its molecules is required. Then the molecules become unsaturated and can easily be fixed on the cellulose fibre. An essential condition of the dye is that the substantive properties are correlated to a large extent with the planear structure of their molecules, or respectively with the assumption of such a structure in the case of an interaction with the
Card $1/4$	cellulose fibre. An essential condition of the properties

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A Comparative Study of the Affinity of Direct Azodyes 20-3-21/46 - of the Diphenyl and p - Terphenyl Derivatives - to Cellulose Cotton Fibre. is, after all, the capacity of forming at least two hydrogen horde between the molecules of the dye and the hydroxil

bonds between the molecules of the dye and the hydroxil groups of the cellulose. In spite of antithetical assertions, Robinson has proved that distances between the groups of molecules of dyes able to form hydrogen bonds, must not be approximated to the identity period of the cellulose

(= 10.3 Å). Hydrogen bonds may occur at almost every place of the cellulose chain. After quoting further references the authors state that the investigation of the p-terphenyl derivatives allows to trace a successive agglomeration of benzene-rings. In this way the influence of the chain prolongation of the conjugated double bonds on the affinity of dyes to the cellulose fibre can be traced too. Further it can be stated in his context how far the mutual position of the groups able to form hydrogen bonds with cellulose, and the distance between them is of importance. Since this method was available to the authors, they compared some benzidine dyes with corresponding p-terphenyl derivatives. The azo-component to which very little attention is paid,

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"APPROVED FOR RELEASE: Monday, July 31, 2000

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A Comparative Study of the Affinity of Direct Azodyes 20-3-21/46 - of the Diphenyl and p - Terphenyl Derivatives - to Cellulose Cotton Fibre.

together with its influence on the activity of the dyes, should be investigated simultaneously. Properties of benzidine dyes with & various azo components and of p terphenyl derivatives with 3 azo-components were investigated. The data in table 1 show that the introduction of an additional benzene-ring in the molecule of the dye increases in all cases the affinity of the dyes to cotton cellulose fibre. In the case of dyes with the azo components Chicago SS and E -acids the affinity grows more at 800 than at 100°. This difference can be well explained by a greater tendency to aggregation in the case of decline of temperature with these dyes. The affinity is thus correlated also with the great entropy changes at the transition of an individual dye molecule into aggregates since the fibre is able to absorb also the later ones. In the coloring process this phenomenon is presented by the fact that the fibre is able to absorb a larger quantity of dyes than provided by the nature of the forces acting between the dye and the fibre. Thus the prolongation of the

Card 3/4

Comparative of the Diphe llulose Cott	Study of the Affinity of Direct Azodyes 20-3-21/46 enyl and p - Terphenyl Derivatives - to con Fibre.
	chain of the conjugated double bonds lead to an enlargement of the hydrophobic surface and to an intensification of polarization of the molecules of the dyes. By this, the affinity to the cellulose fibre increases regardless of the fact that the distance between the groups forming hydrogen compositions with cellulose, does not agree with the identity period of the cellulose. Concluding, further comparisons between the benzidine dyes with various azo components are quoted and conjectures about differences between them enounced. There are 1 table, and 21 references. 10 of which are Slavic.
SSOCIATION:	Khar'kov State University, imeni A. M. Gor'kogo. Ivanovo Chemical Technological Institute (Khar'kovskiy gosudarstvennyy univer- sitet im. A. M. Gor'kogo.Ivanovskiy khimiko-tekhnologicheskiy institut).
RESENTED:	May 15, 1957, by B. A. Kazanskiy, Academician
SUBMITTED:	May 14, 1957.
AVAILABLE:	Library of Congress
Card $4/4$	



KRASOVITSKIY, B.M.; TITARENKO, N.I.

Effect of space factor on the properties of dyes containing a biphenyl ring. Part 11: Comparative investigation of azo dyes, the derivatives of biphenyl, n-terphenyl and n-quaterphenyl. Ukr. khim. zhur. 24 no.4:481-486 '58. (MIRA 11:10)

1. Khar'kovskiy gosudarstvennyy universitet im. A.M. Gor'kogo. (Azo dyes)



5(3)	80V/63-4-2-32/39
AUIHORS:	Krasovitskiy, B.M., Pirogova, I.N., Tsarenko, S.V.
TITLE:	Vat Dyes Made From Pyrenic Acid
PERIODICAL:	Khimicheskaya nauka i promyshlennost', 1959, Vol 4, Nr 2, pp 282-283 (USSR)
ABSTRACT :	The vat dyes were prepared by the condensation of pyrenic acid with ortho-phenylene-diamine and 1,8-naphthylene-diamine. The separation of the dyes into cis- and trans-isomers is not possible, which shows their homogeneity. One dye is an orange powder soluble in concentrated sulfuric acid, pyridine and aniline, the other a dark-green powder soluble in the same media.
ASSOCIATION:	Booddal booddal y all versiter imeni A M Gor Kogo ikhen kor
SUBMITTED:	State University imeni A.M. Gor'kiy) September 15, 1958
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5(3)	<i>SC1/63-4-2-36/39</i>
AUTHORS :	Blinov, V.A., Krasovitskiy, B.M., Khotinskaya, Ye.Ye.
TITLE:	On the Light Resistance of Some Monoazo-Dyes Which are Derivatives of Benzanilide and I-Acid
PERIODICAL:	Khimicheskaya nauka i promyshlennost', 1959, Vol 4, Nr 2, pp 285-286 (USSR)
ABSTRACT:	The tested azo-dyes were used in dyeing cellophane. The azo-component of the dyes was I-acid. The resistance to light and light-weather was studied in the usual way employed by colorists. All dyes showed con- siderable resistance to light. In the light-weather test the dyes with- out substitutes in the benzanilide grouping had the lowest resistance. The dimethylamino-group and the carbethoxy-group increase the resistance. The introduction of a second benzene ring increases also the light re- sistance of the dye.
Card 1/2	There is 1 table and 1 Soviet reference.



· (3) PHORS:	SOV/79-29-8-62/81 Litvinenko, L.M., Levchenko, N.F., Krasovitskiy, B.M., Titarenko, N.I.
CLE:	Spatial Structure and Reactivity. XIV. On the Interaction of the Atom Groups Separated by One, Two, or Three Benzene Nuclei According to the Investigation Data of the Reaction Kinetics of Aromatic Amines With Picrylchloride
CORIODICAL:	Zhurnal obshchey khimii, 1959, Vcl 29, Nr 8, pp 2724-2729 (USSR)
ABSTRACT:	Recently Litvinenko and collaborators succeeded in determining, during the investigation of the acylation kinetics of 4-amino- biphenyl (I) and 4-amino-4-nitrobiphenyl (II) as well as aniline and p-nitroaniline, that the effect of the nitro group upon the reactivity of the aromatic amino group in the molecular system of biphenyl is many hundreds of times weaker than it is in the benzene system (Ref 1).
Card 1/3	$ \underbrace{ \begin{array}{c} \\ \end{array}}_{- \underbrace{ \\ }} \underbrace{ \\ \\ \\ \\ \\ \end{array}}_{- \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ $

Spatial Structure and Reactivity. XIV. On the SOV/79-29-8-62/81 Interaction of the Atom Groups Separated by One, Two, or Three Benzene Nuclei According to the Investigation Data of the Reaction Kinetics of Aromatic Amines With Picrylchloride

The ratio of the rate constants for the reactions of the amino and aminonitro derivatives may serve the quantitative evaluation

of this effect, e.g. $\frac{K_{I}}{K_{II}}$ (factor f)(Refs 2-5). In the present

paper the authors dealt with the problem of how the effect of the NO₂ group upon the reactivity of the NH₂ group occurs when the

same kinetic method is used in the case that these groups are separated from one another by a system of three benzene nuclei. The reaction of the aromatic amines with picryl chloride in a benzene solution was taken as an example, since it proved to be highly sensitive to structural changes in the amins molecule (Ref 6) and can therefore be successfully used for the quantitative characterization of the influence of the slightest differences in the structures of the named compounds upon the reactivity of the aromatically bound amino group. On the basis

Card 2/3

 S_ratial Structure and Reactivity. XIV. On the SOV/79-29-8-62/81 Interaction of the Atom Groups Separated by One, Two, or Three Benzene Nuclei According to the Investigation Data of the Reaction Kinetics of Aromatic Amines With Picrylchloride

> of these considerations the kinetics of the reaction of compounds (III) and (IV) with picryl chloride in benzene was investigated and compared to the data of the kinetics previously obtained for the reaction of picryl chloride with aniline-4-amino-biphenyl and 4-amino-4-nitrobiphenyl (Ref 7). It was shown that the nitro group has a very strong effect upon the reactivity of the amino group which occupies the para-position in the same benzene nucleus. This effect is reduced in the binuclear molecular system of biphenyl and disappears almost completely in the system of n-triphenyl. These phenomena are due to a specific structural spatial arrangement. There are 4 tables and 16 references, 14 of which are Soviet.

ASSOCIATION: Khar'kovskiy gosudarstvennyy universitet (Khar'kov State University)

SUBMITTED: July 10, 1958

Card 3/3

5(3)	50V_80-32-3-35/42
NTTTORC:	Pobunichia, M.C., Kroceitacijo, J.M., Matskevich. J.M., Blinev, V.A., Vischhime, J.Ya.
TITLD:	Linear Dis-Azo Dyes Which and Durivatives of Gasdiagel and Thic- diapol (Gryamyye disazokranitali - preisvodnyse akas indiazola i tiodiazola)
PERIODICAL:	Weurnal prikladnoy kházii, 1959, Vol XXXII, Mr 3, př. 354-667 (USDR)
ADSTRACT:	Linuar are dyes containing heterocyclic model connected with the chain by conjugated double bonds are investigated here. The nuxochromes are also connected by conjugated double bords to the chain. An cradinzol and thiodized ring is introduced to the same chain. The dyes prepared are: 2,5-bis-(4-mitrophenyl)- 1,5,4-accdinzol, 2,5-bis-(4-aminophenyl)-1,5,4-oxadistol, 1,5-11-(4-mitrophenyl)-1,5,4-bis-(4-amirophenyl)- 1,5,5-bis-(4-mitrophenyl)-1,5,4-bis-(4-amirophenyl)- 1,5,4-bis-(4-amirophenyl)-1,5,5-bis-(4-amirophenyl)- 1,5,4-bis-(4-amirophenyl)-1,5,5-bis-(4-amirophenyl)- 1,5,4-bis-(4-amirophenyl)-1,5,5-bis-(4-
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MEL'NIKOV, 3.N.; KRASNOVITSKIY, B.M.; MORYGANOV, P.V.; ZAKHAROVA, T.D. Relation between the structure of axo dyes (ora- and thiodiazol derivatives) and the rate of their diffusion in corper rayon fibers. Izv.vys.ucheb.zav.; tekh.tekst.prom. no.6:120-124 '60. (MIRA 14:1) 1. Ivanovskiy khimiko-tekhnologicheskiy institut 1 Khar 'kovskiy gosudarstvennyy universitet imeni A.M. Gor 'kogo. (Dyes and dyeing--Rayon) (Axo dyes)

s/073/60/026/001/011/021 B004/B054 Krasovitskiy, B. M., Pereyaslova, D. G., and Titarenko, N.I. Effect of Steric Factors on Properties of Dyes Containing a AUTHORS: Biphenyl Ring. XIV. Comparative Study of Color and Affinity to Cotton of Some Azo Dyes Which Are Derivatives of Biphenyl, TITLE: Dibenzyl, Trans-stilbene, Tolan, and Azobenzene Ukrainskiy khimicheskiy zhurnal, 1960, Vol. 26, No. 1, PERIODICAL: pp. 73-77 TEXT: The authors studied the effect of conjugated double bonds on the affinity of dyes to cotton. A μ (Ash) acid in alkaline medium was used as azo component [Abstracter's note: This acid is not defined in the present paper]. The following were used as diazo components: benzidine (I); paper]. The following were used as draw compension of VI); 4,4'-diamine-4,4'-diamino-dibenzyl (V); 4,4'-diamino-trans-stilbene (VI); 4,4'-diaminetolan (VII); 4,4'-diamino-azobenzene (VIII); 4,4'-diamino-p-terphenyl (IX); and 3,31-diamino-azobenzene (X). The absorption spectra were taken by an $C\Phi-2M(sF-2M)$ spectrophotometer. The affinity to cotton was determined at 80° and 100° C by measuring the decrease in dye concentration of the solution Card 1/3

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Effect of Steric Factors on Properties of Dyes Containing a Biphenyl Ring. XIV. Comparative Study of Color and Affinity to Cotton of Some Azo Dyes Which Are Derivatives of Biphenyl, Dibenzyl, Trans-stilbene, Tolan, and Azobenzene

by an $\Phi + K - M$ (FEK-M) photocolorimeter. The following data are given in Tables 1-3: Diazo component λ_{max} , m μ 80°C 100°C which the saigunstion

Jidzo component	Λ_{\max}, μ	80-0	100°C
I	595	4.45	3.73
V	556	2.97	1.45
VI	600	5.75	4.70
VII	58 0	4.83	3.97
VIII	609	5.12	4.19
IX	566	6.00	5.16
X_	540	3, 18	2 7

affinity. Z. V. Oleynikova assisted in the experiments. There are 3 tables and 26 references: 12 Soviet, 6 British. 1 French, 4 German, 1 Italian, and 1 Swiss.

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of double bonds is interrupted (V, X), the affinity to cotton decreases. A replacement of the ethylene bridge of VI by the acetylene tridge of







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L 12908-65 EWT(m)/EPP(c)/T/EWP(j) Pc-4/Pr-4 RPL/AFWL/AFUC(b)/AS(mp)-2/ ASD(a)-5/BSD/ESD(ga)/ESD(t) RM/JW ACCESSION NR: AP4047177 s/0051/64/017/004/0558/0564 Krasovitskiy, B. M .; Smelyakova, V. B .; Nurmukhametov, R. N. AUTHORS : Absorption and fluorescence spectra of certain azomethine TITLE: derivatives of benzidine and its 2,2' and 3,3' dichlorosubstitutes SOURCE: Optika i spektroskopiya, v. 17, no. 4, 1964, 558-564 TOPIC TAGS: absorption spectrum, fluorescence spectrum, benzidine ABSTRACT: For comparison with similar tests on salicylal aniline and its derivatives (DAN SSSR v. 143, 1145, 1962; ZhFKh v. 37, 2432, 1963), the authors investigated the absorption spectra of the condensation products of aniline, ortho-, meta-chloranilines, benzidine, and its 2,2' and 2,2' dichloroderivatives with salicylic and 2-oxy-1-naphthoic aldehydes in dimethylformamide, and the fluorescence spectra of frozen solutions (77K) of these substances in the same solvent. The doubling of the salicylal aniline molecule causes Card 1/3

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L 12908-65 ACCESSION NR: AP4047177 \mathbf{O} a bathochromic shift of the absorption and fluorescence bands, evidencing appreciable conjugation between the two halves of the disalicylal benzidine molecule. Disalicylal benzidine and its 3,3" dichloroderivative differ very little in their absorption spectra: their fluorescence spectra are also of like character, but the fluorescence intensity of the latter is much larger than that of the former. At the same time, the absorption and fluorescence bands of 2,2' dichloroderivative of disalicylal benzidine are less intense, owing to the spatial difficulties in the grouping of the biphenyl, and are shifted towards the short-wave end of the spectrum compared with the disalicylal-benzidine and disalicylal-3, 3'-dichlorobenzi-The rules characteristic of the derivative of salicylic dine. aldehyde hold true also for the absorption and fluorescence spectra of the products of condensation of the foregoing amines with 2-oxy-1-naphthaldehyde. Plots of the various spectra and of the time variation of the fluorescence intensity are presented. The azomethine derivatives of the diamines of the benzidine series are Card 2/3

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L 62828-65 EWT(m)/EPF(c)/EWP(1) Pc-4/Pr-4/Ps-4/Peb DIAAP WW/JAJ/ ACCESSION NR: AP5019048 UR/0286/65/000/012/ 621.039 678.746.22	,
AUTHOR: Chernobay, A. V.; Gunder, O. A.; Dmitriyevskaya, L. I.; Krasovi	39 tskiy,
TITLE: A method for producing plastic scintillators. Class 39, No. 1720	04015
SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 12, 1965, 75	
TOPIC TAGS: scintillator, block polymerization, plastic	
ABSTRACT: This Author's Certificate introduces a method, for producing pl scintillators by thermal block <u>polymerization of styrene</u> ^b in the presence tillating additives which are capable of copolymerization with styrene. output of the scintillators is increased by using <i>n</i> -vinylterphenyl as the lating additive.	of scin-
ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut monokristall Union Scientific Research Institute of Single Crystals)	ov (AlI-
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L 62828-65 ACCESSION NR: AP5019048		
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ACCESSION NR: AP5021556		UR/0286/65/000/013/0021/0021 547.652.1.787.002.2	
AUTHOR: <u>Krasovitskiy</u> , B. M. TITLE: Liquid scintillatore	19, 50	ornaya, L. M. 422 14,55 B	· •
SOURCE: Byulleten' izobrete	niy i tovarnykh znakov, n	0. 13. 1965. 21	
TOPIC TAGS: scintillator, co			
		1	
ABSTRACT: An Author Certif liquid <u>scintillators</u> for stu on vaseline oil with added n oxazole.	licate has been issued for dying cosmic radiation. The aphthalene or 1-methylnaph 7,44		
liquid <u>scintillators</u> for stu on vaseline oil with added n	aphthalene or 1-methylhaph 7,44	The scintilators are based hthalene and 2,5- <u>diphenyl-</u> [BO]	
liquid <u>scintillators</u> for stu on vaseline oil with added n oxazole. // ASSOCIATION: Vsesovnznyv na	aphthalene or 1-methylhaph 7,44	The scintilators are based hthalene and 2,5- <u>diphenyl-</u> [BO]	
liquid <u>scintillators</u> for stu on vaseline oil with added n <u>oxazole</u> . (ASSOCIATION: Vsesoynznyy na (All-Union Scientific Resear)	aphthalene or 1-methylhaph 7,44 uchno-issledovatel'skiy in ch Institute of Single Cry 44, 55	The scintilators are based hthalene and 2,5- <u>diphenyl-</u> [BO] nstitut monokristallov (stals)	

L 5294-66 EWT(m)/EWP(j)/EWG(v)RM ACC NR: AP5024998 SOURCE CODE: UR/0286/65/000/016/0061/0062 AUTHORS: Krasovitskiy, B. M.; Pereyaslova, D. G.; Fodiman, I. V.; Tatsiy, ORG: none TITLE: A method for obtaining daylight fluorescent pigments. ¹⁵Class 22, No. 173867 announced by All-Union Scientific Research Institute of Single Crystals (Vsesoyuznyy nauchno-issledovatel'skiy institut monokristallov)/ SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 16, 1965, 61-62 TOPIC TAGS: pigment, dye, resin, single crystal, n toluolsulfamide, melamine, formaldehyde ABSTRACT: This Author Certificate presents a method for obtaining daylight fluorescent pigments based on a resin of n-toluolsulfamide, melamine, and formaldehyde, to which a dye is added. To increase the fastness of colors in daylight, cation pigments are used as dyes. Their general formula is: Card 1/2 UDC: 666.291.3:668.819.45 19010541

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	L 63957-65 ENT(1)/EWT(m)/EPF(c)/EWP(j)/T/EWA(c) IJP(c)/RPL JW/RM	
	ACCESSION NR: AP5020955 UR/0073/65/031/008/0828/0834 547.97 AUTHOR: <u>Krasovitskiy</u> , <u>B. M.</u> ; <u>Mal'tseva</u> , <u>N. I.</u> ; <u>Nurmukhamatov</u> , <u>R. N.</u>	
1	TITLE: I Investigation is of azomethine bases. II, The effect of con- jugation on <u>color</u> and <u>fluorescence</u> of <u>bisazomethine</u> derivatives of some aromatic diamines	
	SOURCE: Ukrainskiy khimicheskiy zhurnal, v. 31, no. 8, 1965, 828- 834	
	TOPIC TAGS: azomethine, conjugation, fluorescence, spectroscopy, hydrogen bond, aromatic diamine, restricted rotation	
	ABSTRACT: The purpose of this work was to investigate the optical properties of a number of bis-azomethines with continuous or inter- rupted chains of conjugation between the nitrogen atoms. The com- pounds under investigation were obtained by condensation of benzidine.	
	p-phenylenediamine, 4,4"-diamino-p-terphenyl and similar diamines with benzaldehyde, salicylaidehyde, a-naphthaidehyde, and S-hydroxy- a-naphthaidehyde in dimethylformamide. It was found that inter-	
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	ACCESSION NR: AP5020955	
	ruption of the chain of conjugation, e.g., by one or more methylene groups separating two benzene rings, results in a hypsochromic shift benzene rings between the nitrogen atoms produced no noticeable changes, as compared to the parent compounds, except in the case of chromic shift is observed. Some spectrophotometric evidence was found that there exists some hydrogen bonding in salicylal derivatives rotation of the benzene rings. The relatively high stability of alence of the keto form, making proton transfer less probable. Com- pounds prepared for the first time are tabulated together with their sfigures and 1 table.	
	ASSOCIATION: Veera Herena	
	ASSOCIATION: Vseso uznyy nauchno-issledovatel'skiy institut mono- p kristallov (All-Union Scientific Research Institute of Monocrystals); Fiziko-khimicheskiy institut im. Kernors (Dhudiation);	
	Fiziko-khimicheskiy institut im. Karpova (Physical Chemistry Institute)	44
1999 - 1 99	Card 2/8	



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	P(j)/EWA(h)/EWA(1) Pc-4/Pel	
ICCESSION IR: AP5015280	UT:/04	286/65/000/009/0064/0064
NUTHORS: Grachev, N. M.; D. Grasovitskiy, B. M.	ykhanova, A. S.; Gunder, U. I	A.; Bezuglyy, V. D.; B
TITLE: A method for obtain	ing film scintillators./5 Class	ss 39, No. 170650 15
SOURCE: Byulleten' izobret	eniy i tovarnykh znakov, no.	9, 1965, 64
TOPIC TAGS: scintillator, lzation, tetramethylstyrene	thermal stability, light omis	ssion, polymor, polymer-
lators based on <u>polystyrene</u> mission of a scintillator,	ificate presents a method for . 5 To increase the heat resi a polymer obtained during the is used as the styrene polym	stance and the light he polymerization of
lators based on <u>polystyrene</u> emission of a scintillator, 2,3,5,6-tetramethylstyrene ASSOCIATION: Vsesoyuznyy n	. 15 To increase the heat resi a polymer obtained during the	stance and the light he polymerization of er. titut monokristallov
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lators based on <u>polystyrene</u> mission of a scintillator, 2,3,5,6-tetramethylstyrene ASSOCIATION: Vsesoyuznyy n (All-Union Scientific Resea	. 5 To increase the heat resi a polymer obtained during to is used as the styrene polym auchno-issledovatel'skiy ins rch Institute of Single Crys	stance and the light he polymerization of er. titut monokristallov tals)

L 60266-65 EPF(c)/IMP(j)/EWA(c)/EWT(m) Pc-4/Pr.4 RPL JAJ/RM ACCESSION NR: AP5018600 UR/00/79/65/035/007/1243/1246 547.553.1 : 543.426 B AUTHOR: Pereyaslova, D. G.; Bondarenko, V. Ye.; Krasovitskiy, B. M. TITLE: Influence of conjugation on optical properties of alkaline solutions of N,N'-di'(2,3-oxynaphthoy1) derivatives of certain aromatic diamines SOURCE: Zhurnal obshchey khimii, v. 35, no. 7, 1965, 1243-1246 TOPIC TAGS: conjugation, aromatic, diamine, N,N'-di-(2,3-oxynaphthoy1) derivative ABSTRACT: Optical properties of N,N'-di-(2,3-oxynaphthoy1) derivatives of meta-	
AUTHOR: <u>Pereyaslova</u> , <u>D.</u> G.; <u>Bondarenko</u> , <u>V. Ye</u> , <u>Krasovitskiy</u> , <u>B. M.</u> TITLE: Influence of conjugation on <u>optical properties</u> of alkaline solutions of N,N'-di'(2,3-oxynaphthoyI) derivatives of certain aromatic diamines SOURCE: Zhurnal obshchey khimii, v. 35, no. 7, 1965, 1243-1246 TOPIC TAGS: conjugation, aromatic, <u>diamine</u> , <u>N,N'-di-(2,3-oxynaphthoy1)</u> derivative	
N,N'-di'(2,3-oxynaphthoy1) derivatives of certain aromatic diamines SOURCE: Zhurnal obshchey khimii, v. 35, no. 7, 1965, 1243-1246 TOPIC TAGS: conjugation, aromatic, diamine, N,N'-di-(2,3-oxynaphthoy1) derivative	
TOPIC TAGS: conjugation, aromatic, diamine, N,N'-di-(2,3-oxynaphthoy1) derivative	
동안 이 제 그 밖에는 지난 것이 가지 않는 것을 가장한 것이 있는 것이 나라가 있는 것이지 못했다. 정말 가지 않는 것이 나라 나라 나라 나라 나라 나라 나라 나라.	Ĵ.
ARCTRACT. Ontical properties of N.Nº-di-(2.3-oxypaphthoyl) derivatives of meta-	
and para- phenylenediamine, benzidine, and 2,2'- and 3,3'-disubstituted benzidine derivatives were investigated by UV- spectroscopy. For comparison UV- spectra of an anilide and ortho-aniside of 2,3-oxynaphthoic acid were taken. Absorption and fluorescence maxima (in mµ) were measured in a 5% NaOH solution. Absorption and fluorescence spectra of anilide of 2,3-oxynaphthoic acid and of N,N'-di-(2,3-oxy- naphthoyl)-benzidine are shown in fig. 1 of the Enclosure. Out of the three bands only the middle one (v320 mµ) is affected by conjugation within the molecula. Doub- ling of the molecule of anilide of 2,3-oxynaphthoic acid results in a bathochromal	
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	intensification of the middle absorption band. A decline in conjugation between the amide groups of the di-oxynaphthoyl diamide derivatives always results in a shift of the middle absorption band toward the short wave length region. Maxima of fluo- rescence of the doubled molecules occur in a shorter wave region than those of the "Half-molecules". The effect of conjugation on displacement of the fluorescence maxima is greater for the single than for the doubled molecules. Maxima of fluores- cence of ortho-oxybenzoyl derivatives of benzidine occur in a shorter wave length region than those of the corresponding 2,3-oxynaphthoyl derivatives of benzidine. The effect of conjugation on displacement of fluorescence maxima is greater in the former case. Orig. art. has: 1 figure and 1 table.	
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KRABOVITCKTY, B.M.; IEREYASLOVA, F.G.

신전 전화

Fluorescence method of determining small amounts of motion ins in organic media. Zhur. VKHO 10 no. 62702 (55 (Fruit 1991)

1. Vsecoyusoyy nauchno-isoledovatel'skdy testimit o cobrisingler. Submitted March 9, 1965.

L 15321-66 EWT(m)/EWP(j) RM	- and
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AUTHORS: Krasovitskiv, B. H. Shari	~
AUTHORS: Krasovitskiy, B. M.: Showshowka, B.	
AUTHORS: <u>Krasovitskiy, B. M.;</u> Shevchenko, E. A.; Pereyaslova, D. G.	
0RC: none 15 25	
BINTING B	
TITLE: A method for obtaining phosphorogen. Class 12. No. 176200	
Union Scientific Research Institute for Single Crystals (Vassourced by All-	
Union Scientific Research Institute for Single Crystals (Vsesoyuznyy nauchno-	+
SOTTROF. Duran i	
SOURCE: Byulleten' izobreteniy i tovarnykh znakov, no. 22, 1965, 29	
TOPTC TACS	
TOPIC TAGS: crystal phosphor, phosphorescent material, phosphorescence, luminophor ABSTRACT: This Author Cortificant	
ABSTRACT: This Author Certificate presents a method for obtaining a phosphorogen derived from 1,8 naphthoylene-1',2' benzimidazol. To increase the method for obtaining a phosphorogen	e1
derived from 1.8 markly	
derived from 1,8 naphthoylene-1',2' benzimidazol. To increase the variety of phosphorogens with fluorescence in the yellow-green spectral part	
anhydride or its destants in the yellow-green spectral more as	
phosphorogens with fluorescence in the yellow-green spectral region, naphthalene anhydride or its derivatives are condensed with corresponding phenylenediamine	
planty inerviewediamine	
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Card 1/1/20	
UDC: 547.785.5.07.1621.3.032.35	

ACC NR AP5021423 SOURCE CODE: UR/0413/66/000/011/0022/0022 INVENTOR: Grigor'yeva, V. I.; Krasovitskiy, B. M.; Mil'ner, R. S. ORG: None TITLE: A method for producing luminescent monomers, Class 12, No. 182162 [announced by the All-Union Scientific Research Institute of Single Crystals (Vsesoyuznyy nauchno-issledovatel'skiy institut monokristallov)] SOURCE: Izobreteniya, promyshlennyye obraztsy, tovarnyye znaki, no. 11, 1966, 22 TOPIC TAGS: monomer, luminescent material ABSTRACT: This Author's Certificate introduces a method for producing luminescent $R - C - C_{0}H_{1} - CH - CH_{1}$ where R is an aromatic radical. 2-[bromomethylphenyl]-5-aryloxazole is interacted with triphenylphosphine, paraform and lithium methylate. SUB CODE: 07, 11/ SUBM DATE: 15Mar65 Card 1/1 UDC; 547,787,1153.024.07 and a second second

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	AP6021424	SOURCE	CODE: UR/041376	6/000/011/0022/0022	•••••
INVENTOR:	Krasovitskiy, B. 1	M.; Podzhaylo, V. F.;		0,000,011,0022/0022	
ORG: None	•	· · · · · · · · · · · · · · · · · · ·	Derevyanko, L.	N.	
	sledovatel'skiy ins	g liquid scintillator ntific Research Insti titut monokristallov)]	rystals (Vsesoyuznyj	y
SOURCE: I	zobreteniya, promys	hlennyye obraztsy, to	-		
TOPIC TAGS:	scintillator, lu	minescent material	varnyye znaki, n	0. 11, 1966, 22	
ABSTRACT: lators by u 1,3-oxazole scintillato	This Author's Cert using a base and act . The luminescence rs is produced by u	ificate introduces a structure distributions diaryl des yield is increased a using dicumylmethane a	nethod for productivatives of 1,3 and a wider selection the back	cing liquid scintil- 4-oxadiazole and tion of liquid	,
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SOV/84-58-11-50/58 AUTHORS: Arzhanov, Yu., Krasovitskiy, M. Difficult Road (Trudnyy put!) TITLE: PERIODICAL: Grazhdanskaya aviatsiya, 1958, Nr 11, pp 36-37 (USSR) ABSTRACT: The authors describe the arrival of two Super Aero planes at Vnukovo airfield from Czechoslovakia on their way to the Mongolian Peoples Republic. Of the two Soviet navigators on board serving as guides, one Rotislav Gatovskiy spoke Czech which he had learned while confined in prison camps during World War II. Vadim Margorin was the second navigator. There is one photograph of Rotislav Gatevskiy. Card 1/1

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