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Roll # 267

Khotov, A.

"APPROVED FOR RELEASE: 06/19/2000 CIA-RDP86-00513R000826710001-6

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Magnetism, Terrestrial-Donets Basin

Magnetic observatory in the Donets Basin for mine surveying work, (Italy), VIMI, 22,19

9. Monthly List of Russian Accessions, Library of Congress, October 1957, Unclassified
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KROTOV, G. A.

"Study of Behavior of Magnetic Declination of the Donbass Meridian".
Nauch tr. Mosk. gorn. in-ta, No. 12, pp 16-37, 1954.

The magnetic orientation of Donbass mines was studied in connection with the variations of the meridional declination. Because the Makeyevskaya Declination Observatory was destroyed during the war, a possibility was established of using the Odessa and Mizhne-Devitskaya Observatory using magnetic measurements carried out at Donbass. A survey showed that variations of declinations at the mines are the same at 1,000 m dept and on the surface. (RZhAstr, No. 1, 1956)

SO: Sum No 884, 9 Apr 1956

KROTOV, Gavriil Alekseyevich; TYUPKIN, S.N., otvetstvennyy redaktor;
SLAVOROSOV, A.Eh., redaktor izdatel'stva; NADEINSKAYA, A.A.,
tekhnicheskiy redaktor

[Underground surveying in drawn and mined workings] Marksheiderskais
s'emenka ochistnykh i nareznykh gornykh vyrabotok. Moskva, Ugletekhnidat
1956. 179 p.
(Mine surveying)

RYZHOV, Petr Aleksandrovich, prof., doktor tekhn.nauk; BUKRINSKIY, Viktor Aleksandrovich, dotsent, kand.tekhn.nauk; GUDKOV, Valentin Mikhaylovich, kand.tekhn.nauk; KROTOV, Gavriil Alekseyevich, dotsent, kand.tekhn.nauk; LYUBMAN, Yerail' Borisovich, assistant; RUDAKOV, Mikhail Lazarevich, prof., doktor; PIKULIN, A.P., kand. tekhn.nauk, ratsenzenz; BUTKEVICH, T.V., red.; PARTSEVSKIY, V.N., red.izd-va; BEKKER, O.G., tekhn.red.

[Mine surveying] Marksheiderskoe delo. Pod nauchnoi red. P.A. Ryzhova. Moskva, Gos. nauchno-tekhn.izd-vo lit-ry po chernoi i tsvetnoi metallurgii, 1958. 463 p. (MIRA 12:1)
(Mine surveying)

KROTOV, G.A.

Ultrasonics and their use in mine surveying. Zap. LGI 37 no.1:
132-149 '58. (MIRA 12:8)
(Mine surveying)
(Ultrasonic waves--Industrial applications)

KAZAKOVSKIY, Dmitriy Antonovich, prof., doktor tekhn.nauk; AVERSHIN,
Stepan Gavrilovich, prof., doktor tekhn.nauk; BKOLOLIKOV,
Antonin Nikolayevich, dotsent, kand.tekhn.nauk; GUSEV, Mikhail
Iosifovich, dotsent, kand.tekhn.nauk; ZDANOVICH, Vyacheslav
Grigor'yevich, prof., doktor tekhn.nauk; KROTOV, Gavriil Alekseyevich,
dotsent, kand.tekhn.nauk; LAVROV, Vladimir Nikolaevich, kand.tekhn.
nauk; LEBEDEV, Kirill Mikhaylovich, assistant; PIATLIN, Mikhail
Petrovich, dotsent, kand.tekhn.nauk; STEPIN, Nikolay Ivanovich,
assistant; BUKRINSKIY, V.A., otv.red.; SFLAVOROSOV, A.Kh., red.izd-va;
ALADOVA, Ye.I., tekhn.red.; KOROVENKOV, Z.A., tekhn.red.

[Mine surveying] Marksheiderskoe delo. Moskva, Ugletekhnizdat,
1959. 688 p. (MIRA 13:11)
(Mine surveying)

KAZAKOVSKIY, D.A., prof., doktor tekhn.nauk; KROTOV, G.A., dots., kand.tekhn.
nauk; GURIN, A.A., kand.tekhn.nauk

Use of acoustical equipment for solving of mine surveying problems.
Nauch.dokl.vys.shkoly; gor.delo no.2:85-91 '59. (MIRA 1?:7)
(Mine surveying)
(Ultrasonic waves--Industrial applications)

KAZAKOVSKIY, D.A., prof.; KHOTOV, G.A., dots.; GURICH, A.A., kand.tekhn.
nauk

Use of sound ranging for the solution of geological and mine
surveying problems. Gor.zhur. no.9:70-71 S '60. (MIRA 13:9)
(Mining geology) (Mine surveying)

KROTOV, G.A., dotsent

Use of the results of a sonar survey to determine the losses in dredging placer deposits. Izv. vys. ucheb. zav.; gor. zhur. 6 no.3:39-46 '61. (MIRA 16:10)

1. Leningradskiy ordena Lenina i ordena Trudovogo Krasnogo Znameni gornyy institut imeni G.V.Plekhanova. Rekomendovana kafedroy marksheyderskogo dela.

KAZANOVSKIY, D.A., prof.; KROTOV, G.A., dotsent; GURICH, A.A., kand.
tekhn. nauk

Use of sound-fixing apparatus in dredge workings. Izv. vys.
ucheb. zav.; gor. zhur. no.6:40-48 '61. (MIRA 16:7)

1. Leningradskiy ordena Lenina i ordena Trudovogo Krasnogo
Znameni gornyy institut imeni G.V. Plekhanova. Rekomendovana
kafedroy marksheyderskogo dela.

(Mine surveying—Equipment and supplies)
(Sound—Equipment and supplies)

ARANOVICH, V.B.; GURICH, A.A.; KROTOV, G.A.; RUDNEV, L.N.

Technical errors in sound ranging measurements in mine
surveying. Zap. LGI 46 no.2:117-130 '63. (MIRA 17:6)

ACC NR: AT6032747

SOURCE CODE: UR/0000/66/000/000/0177/0185

AUTHOR: Krotov, G. A.

ORG: none

TITLE: Use of echo-sounding in solving mining problems

SOURCE: ::: fssr. Institut fiziki Zemli. Geoakustika; ispol'zovaniye zvuka i ul'trazvuka v seismologii, seismorazvedke i gornom del'e (Geoacoustics; the use of sound and ultrasound in seismology, seismic prospecting, and mining). Moscow, Izd-vo Nauka, 1966, 177-185

TOPIC TAGS: mining engineering, seismic prospecting, acoustic method

ABSTRACT: Investigations conducted by the Leningrad Mining Institute in the application of echo-sounding in geological mining problems are discussed. Successful application has been achieved in 1) echo-location surveying and in working placer deposits by dredging, 2) determining the form and size of inaccessible mining faces, and 3) checking the state of underground brine-filled chambers. The research was under the direction of Professor D. A. Kazakovskiy. The article contains echograms showing the bottom deposits of Lake Gladyshevskoye, as well as a diagram depicting the borehole technique used to probe underground salt- and brine-filled cavities.
Orig. art. has: 2 figures.

SUB CODE: 08/ SUBM DATE: 28Mar66/
Card 1/1

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MYSHENKOV, D.P.; KROTOV, I.A.

Modeling equipment made of the AK-1 material. Mashinostroitel'
no.12±23 D '64. (MIRA 18:2)

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CIA-RDP86-00513R000826710001-6"

S/113/60/000/004/006/007
D249/D301

AUTHORS: Myshenkov, D.P. and Krotov, I.A.

TITLE: Elimination of nitroenamel destruction in the zone of welded seams

PERIODICAL: Avtomobil'naya promyshlennost', no. 4, 1960, 35

TEXT: Formation of deposit and destruction of the lacquer paint film in the zone of welded seams was observed on automobiles ЯАЗ (YaAZ). The investigation, carried out in the Central Laboratory of the Yaroslavskiy motornyy zavod (Yaroslavl' motor plant), disclosed that this deposit consists of carbonates of alkali metals. Caustic alkalis destroy the paint film on welded seams and, by absorbing the carbon dioxide from the atmosphere, turn into carbonates which deposit in the form of a white coating. To investigate this process, metal plates were prepared on which seams were welded, using for it the electrodes ОММ-5 and МЦ-7 (MTs-7). The plates were treated by water vapor and carbon dioxide until a white coating appeared on the welded seams. Having established the

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D249/D301

Elimination of nitroenamel...

chemical composition of the coating the investigators applied a number of weak acids solutions (phosphorus, oxalic, chromic) to remove it. The best results were obtained when a weak solution of chromic anhydride was used. When the scale was removed the welded seams were twice washed by a solution containing 0.1% of chromic anhydride, 0.05% potassium bi-chromate and 0.85% water. [Abstractor's note: Rest of the solution not given]. The temperature of the solution was 60-70° C.

ASSOCIATION: Yaroslavskiy motornyy zavod (Yaroslavl' motor plant)

Card 2/2

KROTOV, I.B.

Composition of rust formed on iron in contact with water and air.
Zhur.fiz.khim. 30 no.8:1696-1701 Ag '56. (MIRA 10:1)

U.S.S.R. Akademya nauk SSSR, Institut fizicheskoy khimii, Moskava.
(Iron--Corrosion)

KROTOV, I. N.

"Basic Errors in the Calculations of Water-Supply Systems and Methods for Improvement." Sub 21 Apr 51, All-Union Sci Res Inst of Water Supply, Sewerage, Hydraulic Structures and Engineering Hydrogeology (VODGEC)

Dissertations presented for science and engineering degrees in Moscow during 1951

SO: Sum. No. 480, 9 May 55

~~KROTOV, L.N.~~, KONYUSHKOV, A.M., kandidat tekhnicheskikh nauk nauchnyy
redaktor; GOLUBENKOVA, L.A., redaktor; TOKER, A.M., tekhnicheskiy
redaktor

[Precision methods of calculation for water supply lines] Priemy
uchetnennogo rascheta vodoprovodnykh setei. Moskva, Gos. izd-vo
lit-ry po stroitel'stvu i arkhit. 1954. 103 p. (MIRA 8:4)
(Water supply engineering)

TURK, Vladimir Ivanovich, kandidat tekhnicheskikh nauk, dotsent; ZAMEVSKIY,
M.S., kandidat tekhnicheskikh nauk, dotsent; KHOTOK, I.N., inzhener,
nauchnyy redaktor; SMIRNOVA, A.P., redaktor Izdatel'stva; EL'KINA, E.M.,
tekhnicheskiy redaktor.

[Pumps and pumping stations] Nasosy i nasosnye stantsii. Izd.2-oe,
perer.i dop. Moskva, Gos.izd-vo lit-ry po stroit.i arkhit., 1957.
181 p. (MIRA 10:11)

(Pumping machinery) (Pumping stations)

ABRAMOV, N.N., prof., doktor tekhn.nauk; GENIYEV, N.N., prof., doktor tekhn.nauk [deceased]; PAVLOV, V.I., dotsent, kand.tekhn.nauk [deceased]. Prinimali uchastiye: KLYACHKO, V.A.; KASTAL'SKIY, A.A.; POKROVSKIY, V.N.; MOSHWIN, L.F., prof., retsenzent; MINTS, D.M., prof., retsenzent; ABRAMOV, S.K., dotsent, retsenzent; BONDAR', F.I., inzh., retsenzent; KROTOV, I.N., kand.tekhn.nauk, nauchnyy red.; SMIRNOVA, A.P., red.; MEDVEDEV, L.Ya., tekhn. red.; SOLNTSEVA, L.M., tekhn.red.

[Water-supply engineering] Vodosnabzhenie. Izd.3., perer. Moskva, Gos.izd-vo lit-ry po stroit., arkhit. i stroit.materialam, 1958.
578 p. (MIRA 12:5)

(Water-supply engineering)

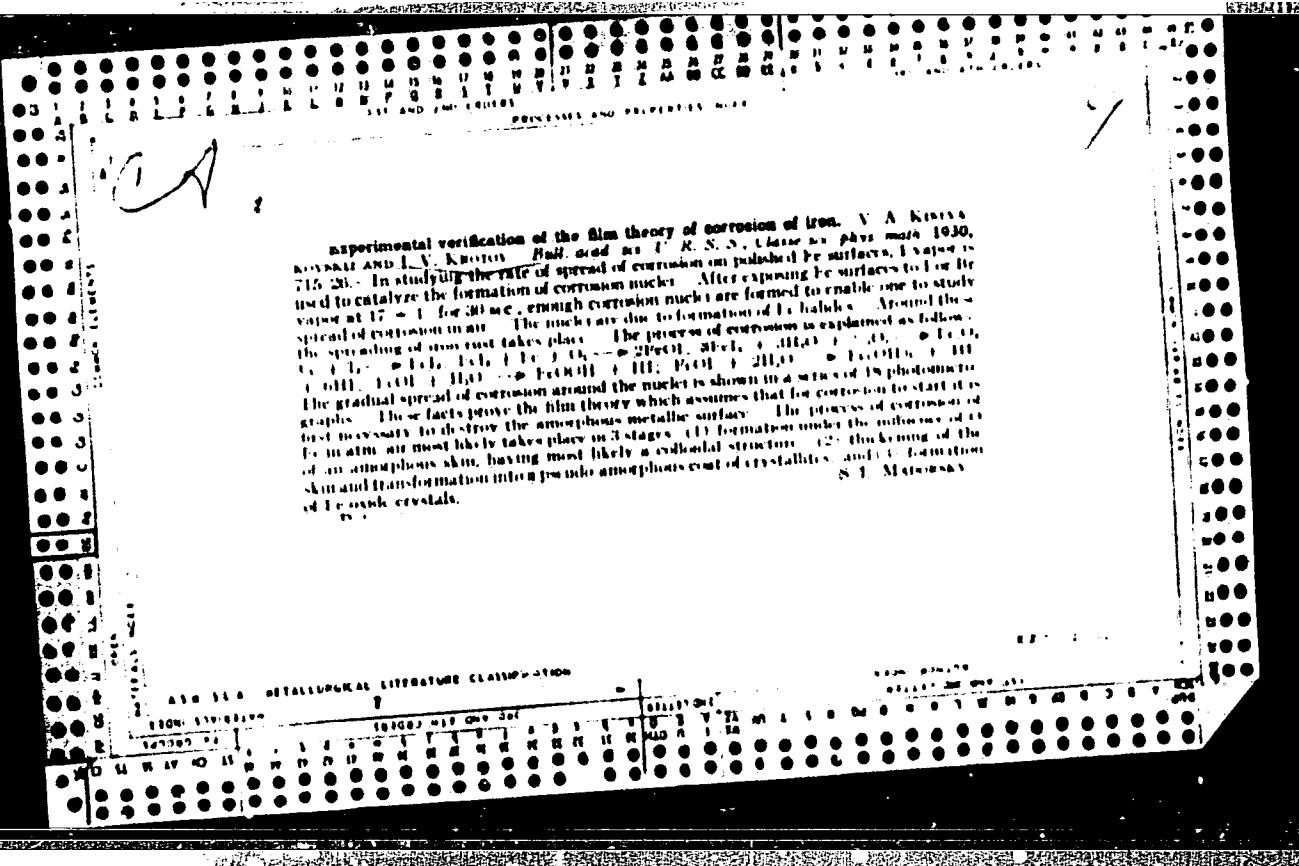
BABAIAK, K.E.; KROTOV, A.V.

Problem of mullet breeding in the littoral lakes and the
Danube Delta. Hidrologia 4:329-336 '63.

KROTOV, I.S.; MAYOROVA, T.A., zootekhnik.

Restore the fame of Siberian butter. Nauka i pered. op. v sel'khoz.
7.no,5:4-5 My '57. (MIRA 10:6)

1. Upravlyayushchiy Novosibirskim trestom "Masloprom" (for Krotov).
2. Cherepanovskiy sovkhoz, Novosibirskoy oblasti.
(Siberia--Butter)



Experimental verification of the film theory of corrosion of iron. V. A. Krasovskii and I. V. Kostyuk. *Bull. Acad. SSSR Classe des sciences math. 1950, 715-20.* In studying the rate of spread of corrosion on polished Fe surfaces, Fe vapor is used to catalyze the formation of corrosion nuclei. After exposing Fe surfaces to Fe or Br vapor at $17 \pm 1^\circ$ for 30 sec., enough corrosion nuclei are formed to enable one to study the spread of corrosion in air. The nuclei are due to formation of Fe halides. Around these the spreading of corrosion takes place. The process of corrosion is explained as follows: $\text{Fe} + \text{H}_2 \rightarrow \text{FeH}$; $\text{FeH} + \text{O}_2 \rightarrow 2\text{FeO}$; $2\text{FeO} \rightarrow \text{Fe}_2\text{O}_3 + \text{O}_2$; $\text{Fe}_2\text{O}_3 + \text{H}_2 \rightarrow \text{Fe(OH)}_2$; $\text{Fe(OH)}_2 + \text{H}_2 \rightarrow \text{Fe(OH)}_3$; $\text{Fe(OH)}_3 \rightarrow \text{Fe}_2\text{O}_3 + \text{H}_2\text{O}$. The gradual spread of corrosion around the nuclei is shown in a series of 15 photomicrographs. These facts prove the film theory which assumes that for corrosion to start it is first necessary to destroy the amorphous metallic surface. The process of corrosion of Fe in air can most likely take place in 3 stages: (1) formation under the influence of O₂ of an amorphous skin, having most likely a colloidal structure; (2) thickening of the skin and transformation into a pseudo amorphous coat of crystallites; and (3) formation of Fe oxide crystals. S. I. Matensky

CA

Inothermal titration microcalorimeter. I. V. Khorov. Bull. Acad. sci. U. R. S. S. Chem. no. 3, 1933, 686-89.—The temp. of the water inside the calorimeter is kept constant to within 0.4° of that in the vessel. The rise of temp. due to the combustion of samples is immediately counterbalanced by the addition of a measured quantity of NH_4NO_3 to the vessel. Intrinsic cooling is corrected by the addition of a measured quantity of NaOH . With a thermometer reading to 0.05° an accuracy of 1% was possible when the evolution of heat was 90 cal.

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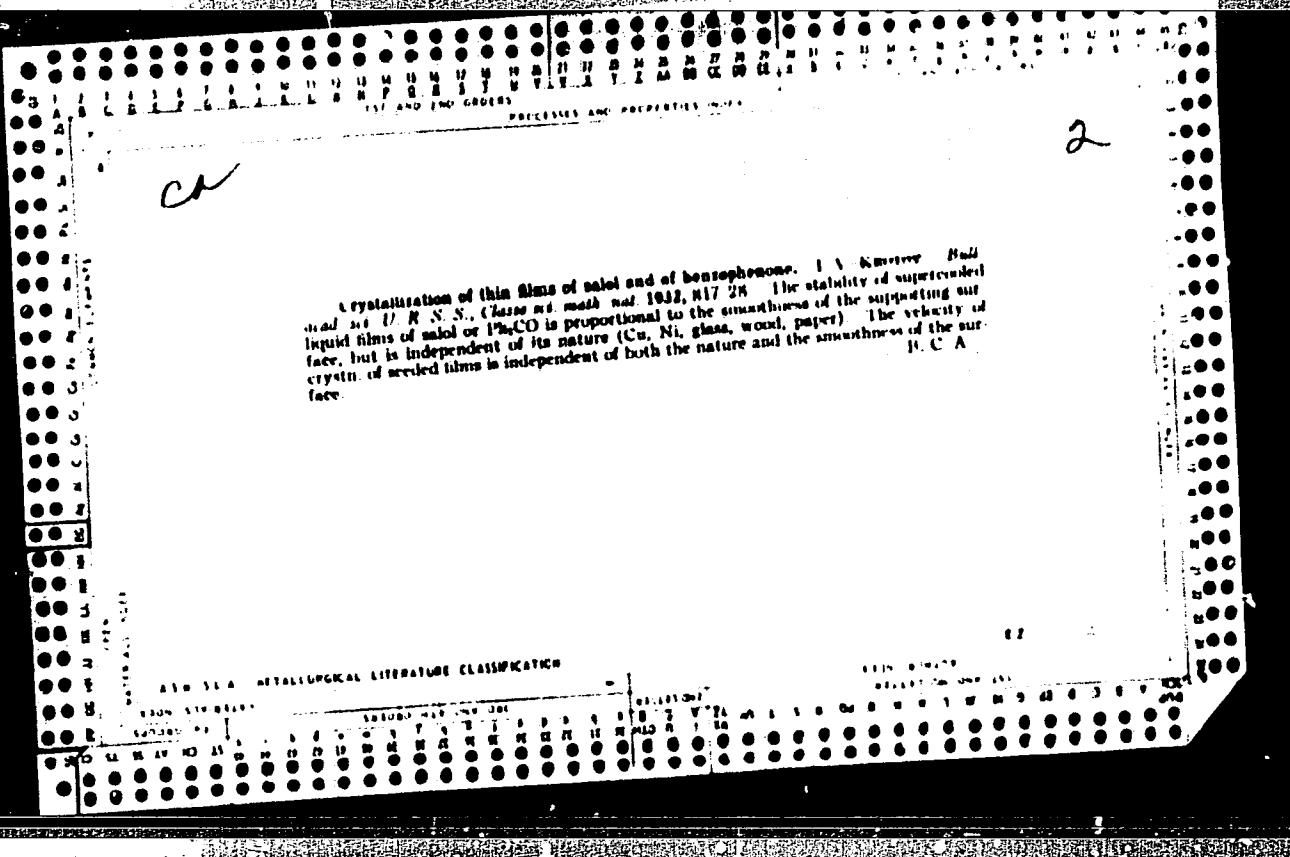
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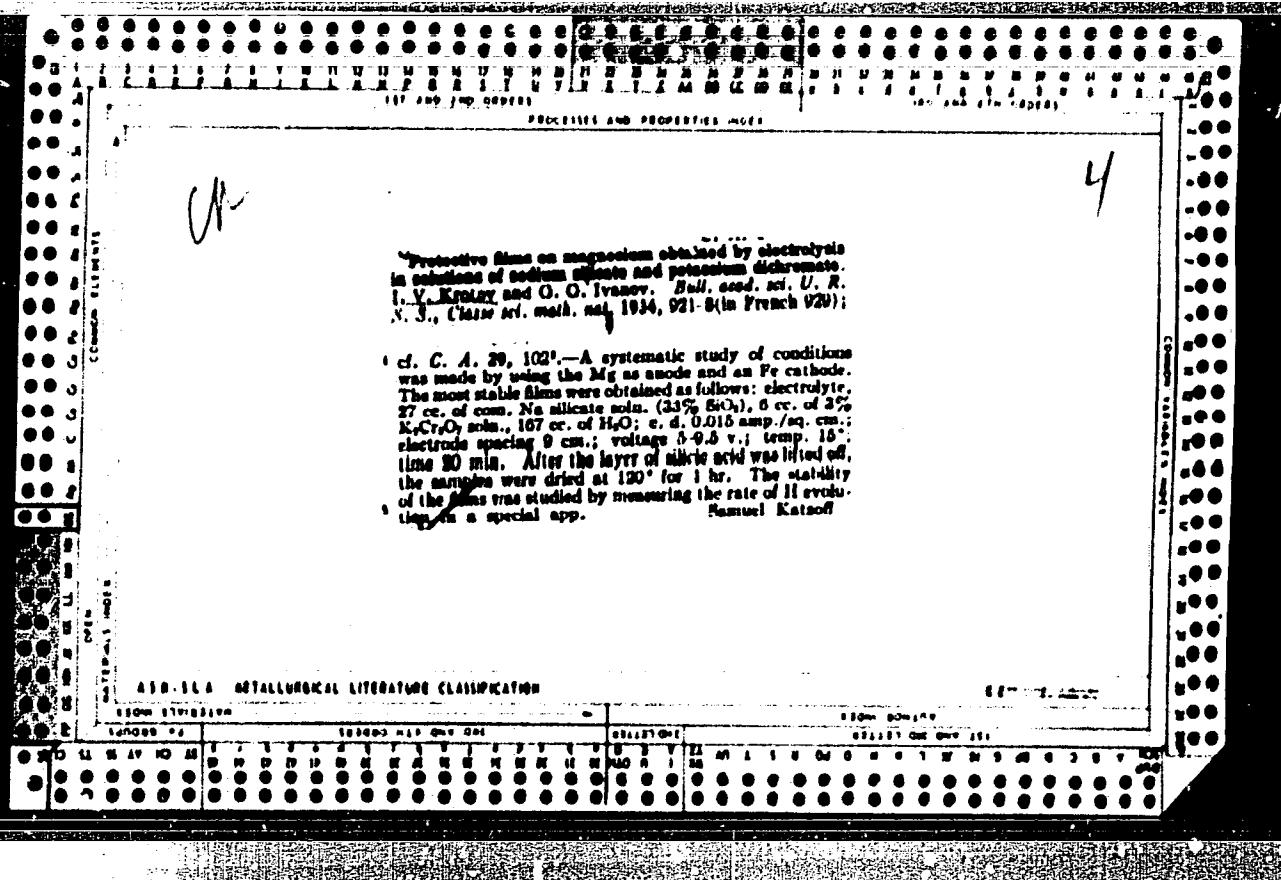
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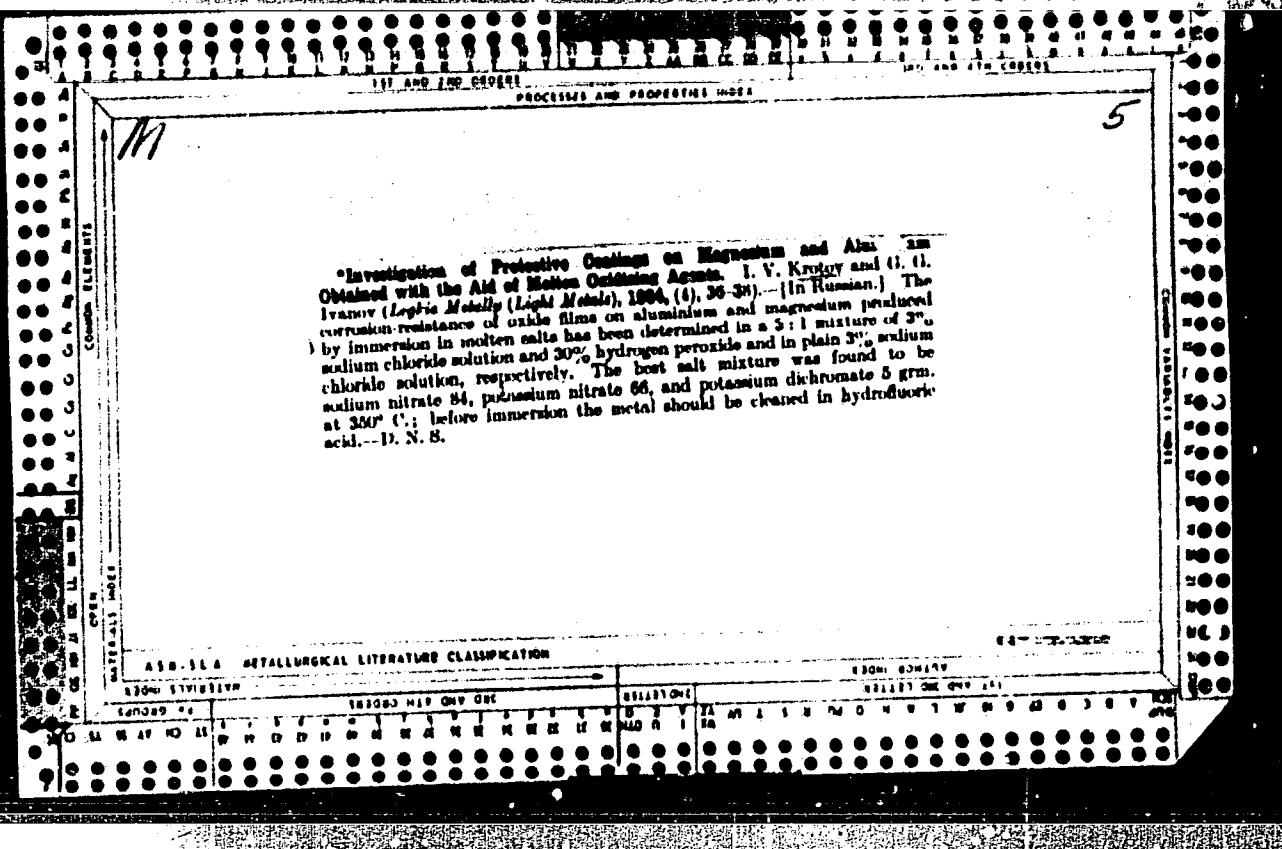


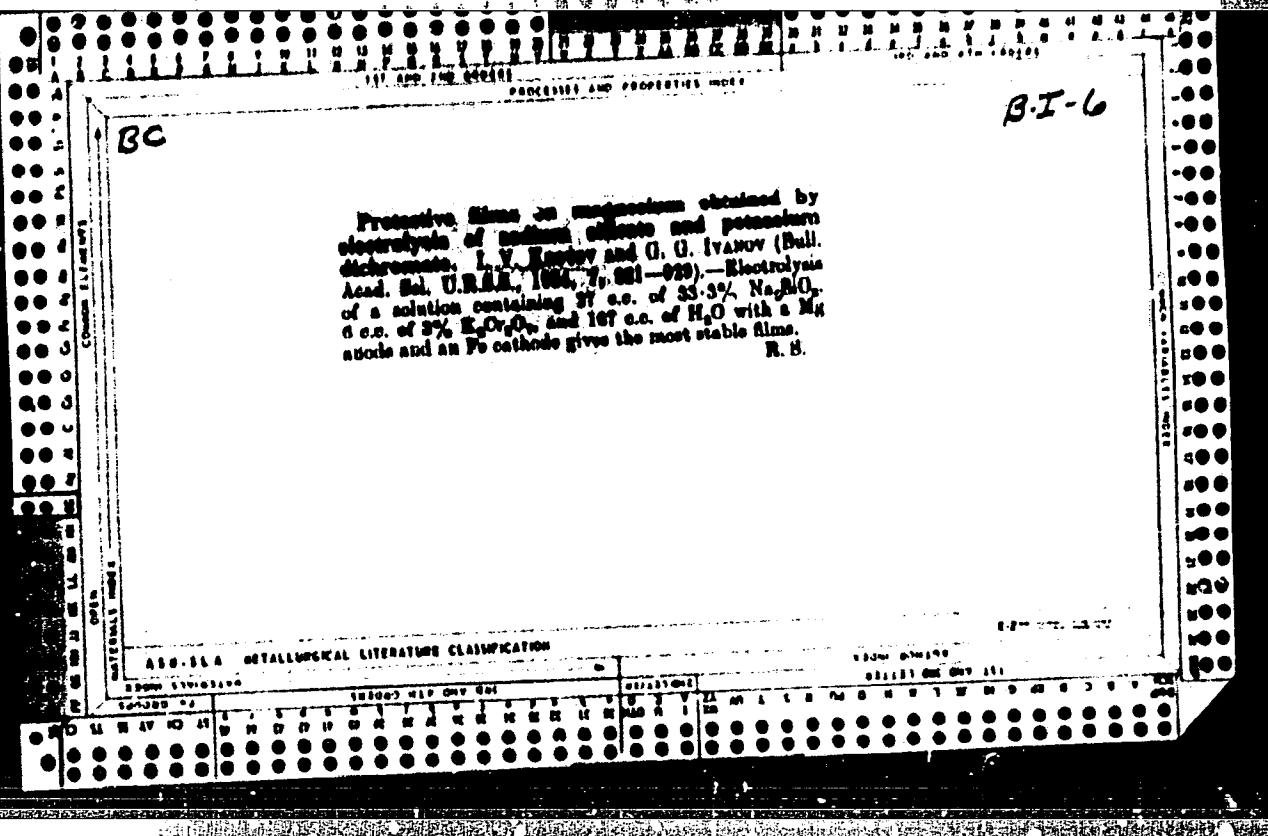
Methods of obtaining and investigating stannic and stannous thiocyanates. I. V. Mulyin. (Chem. and Ind., ref. U. R. S. J. 3, 192 (in English) 1937(1938). Sn(CNS)₂ was obtained from 10 g. of SnCl₂ 2H₂O dissolved in 100 cc. BaOH with the addition of 8.5 g. KCNS in 100 cc. KOH. The KCl ppt. was filtered (yield 5.8 g.) and the filtrate evapd. to about 25 cc. in the absence of O. The Sn content of the light yellow crystals was determined. Fifteen g. K₂SnI was dissolved in 100 cc. 1/4M, 0.2 g. of crystals and a small quantity of perchloric acid were added, the soln. was boiled and 15 g. (CuO)₂ added;

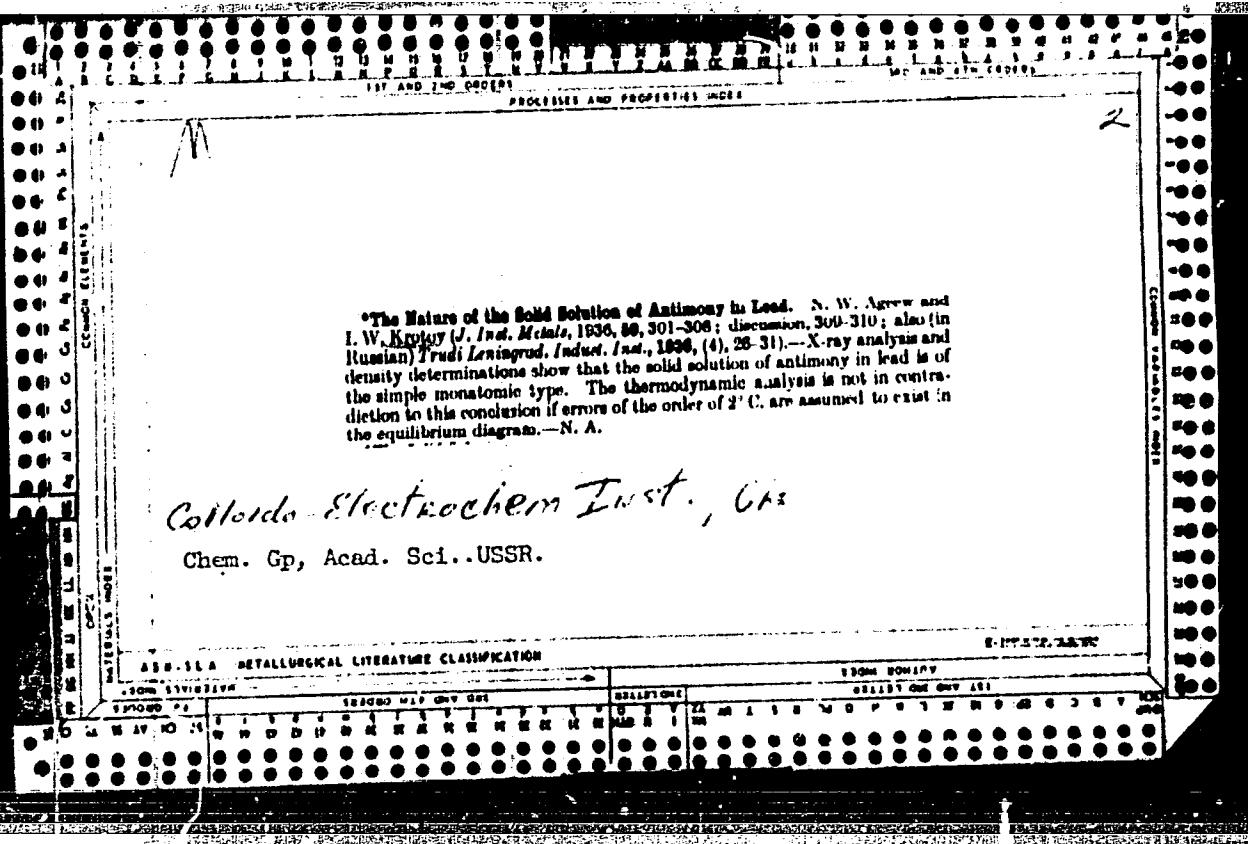
then it was dried, to 176 cc. and electrolyzed for 3 hrs. with 1.75 amp. and 4 v. with Pt electrodes. Sn content of compd. Sn(CNS)₂ (RtOII); calcd., 33.5%; found, 33.0%. The compd. decomposes with change in color on heating. Sn(CNS)₂ was obtained by adding a soln. of 13.6 g. KCNS in 100 cc. KOH to 6 cc. liquid BaCl₂ in 40 cc. BaOH. The KCl ppt. was filtered (yield about 30 g.) and the filtrate evapd. to 15 cc. in the absence of O. Analysis of the evapd. filtrate gave the formula Sn(CNS)₂ (RtOII).

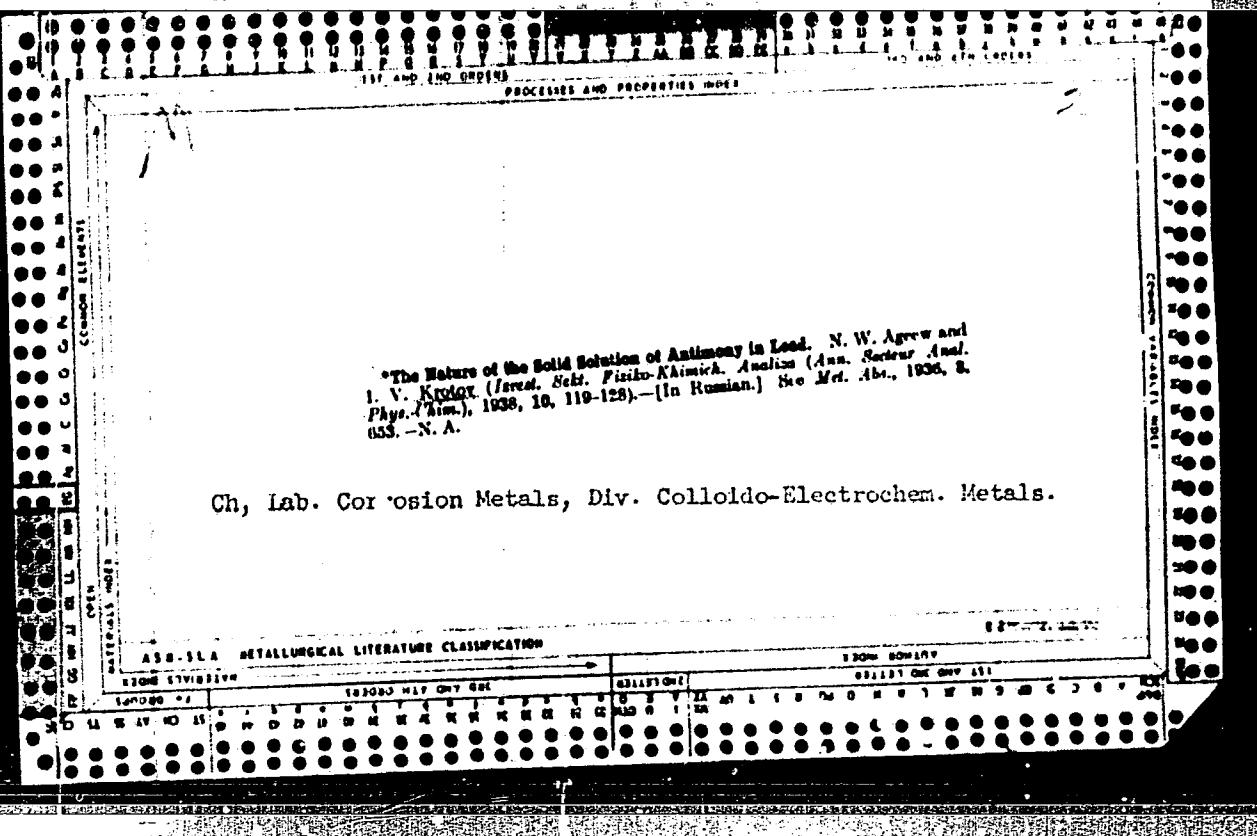
A. S. S.











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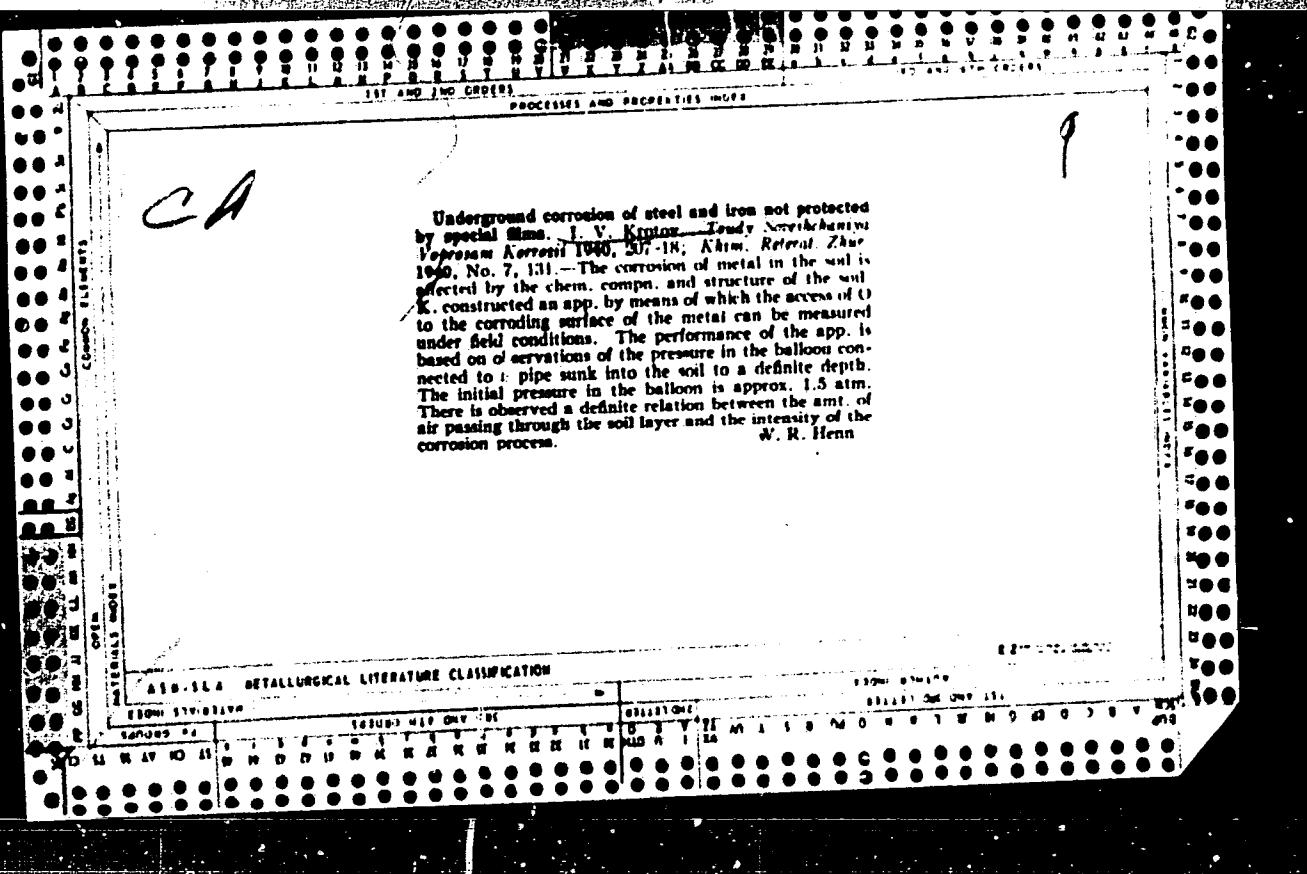
600

1. KROTOV, I.V.

2. USSR (600)

"An Answer to the Criticism by G. V. Akimov of I.V. Krotov's Article 'A Theory of the Passive Condition of Iron!'; 13, No. 4, 1939;

9. [REDACTED] Report U-1613, 3 Jan. 1952.



The passive state of metals. I. V. Kargin. *Trudy Sotsializmogo po Voprosam Korrekta 1920*, 20-4; Khim. Referat. Ser., (No. 7, 1921). An app. was constructed for measuring the Fe electrode potential under conditions of active access of O_2 to the surface of the Fe. The electrode potential of Fe in contact with a drop of KCl soln. and CCl_4 (d. approx. 1) dependent on an Fe plate from a calomel electrode preserved its hemispherical shape. A higher potential produced a drop with a larger diameter. It is noted that the potential of the Fe electrode varied with the previous history of the sample under investigation, with the diam. of the drop of the KCl soln., and with change in the place of contact of the surface and the drop hanging from the tip of the capillary of the calomel electrode. Fe that was very active at the beginning of the measurements often became passive in these because probably more active. The potential of the macroelectrode changed depending on the time of exposure to the air. A method of determining the resistance between the cathode and anode regions of the drop is described. Considering that the concn. of O_2 is proportional to the distance between the boundary and the active and passive zones of the cathode, that part of Fe formed on the cathode which is used up for the reduction of the film, it is concluded that the passivating film on the surface of polarized Fe consists of a blind layer of FeO_2 . This agrees with the data of Trenck and other investigators.

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ASA-11A METALLURGICAL LITERATURE CLASSIFICATION

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SEARCHED	INDEXED	SERIALIZED	FILED

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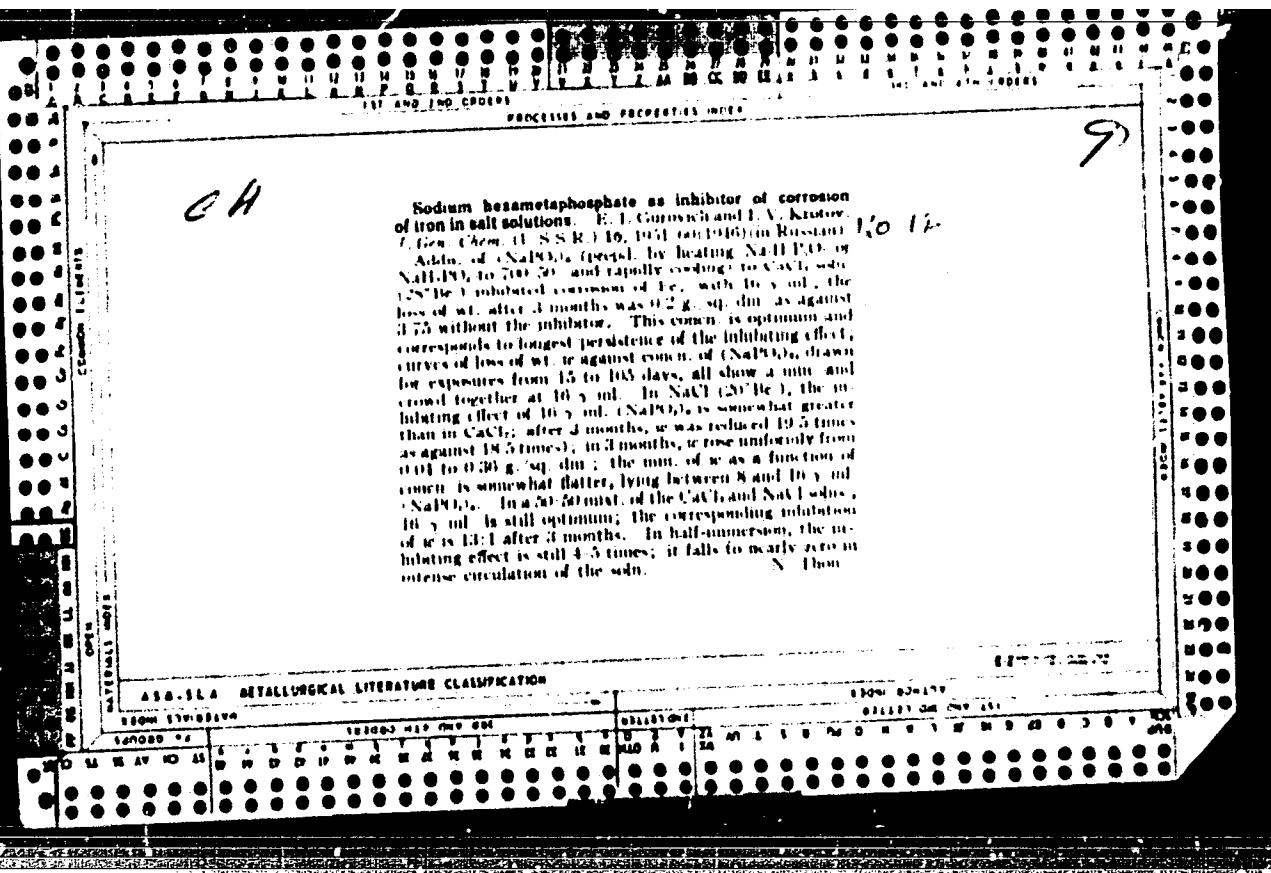
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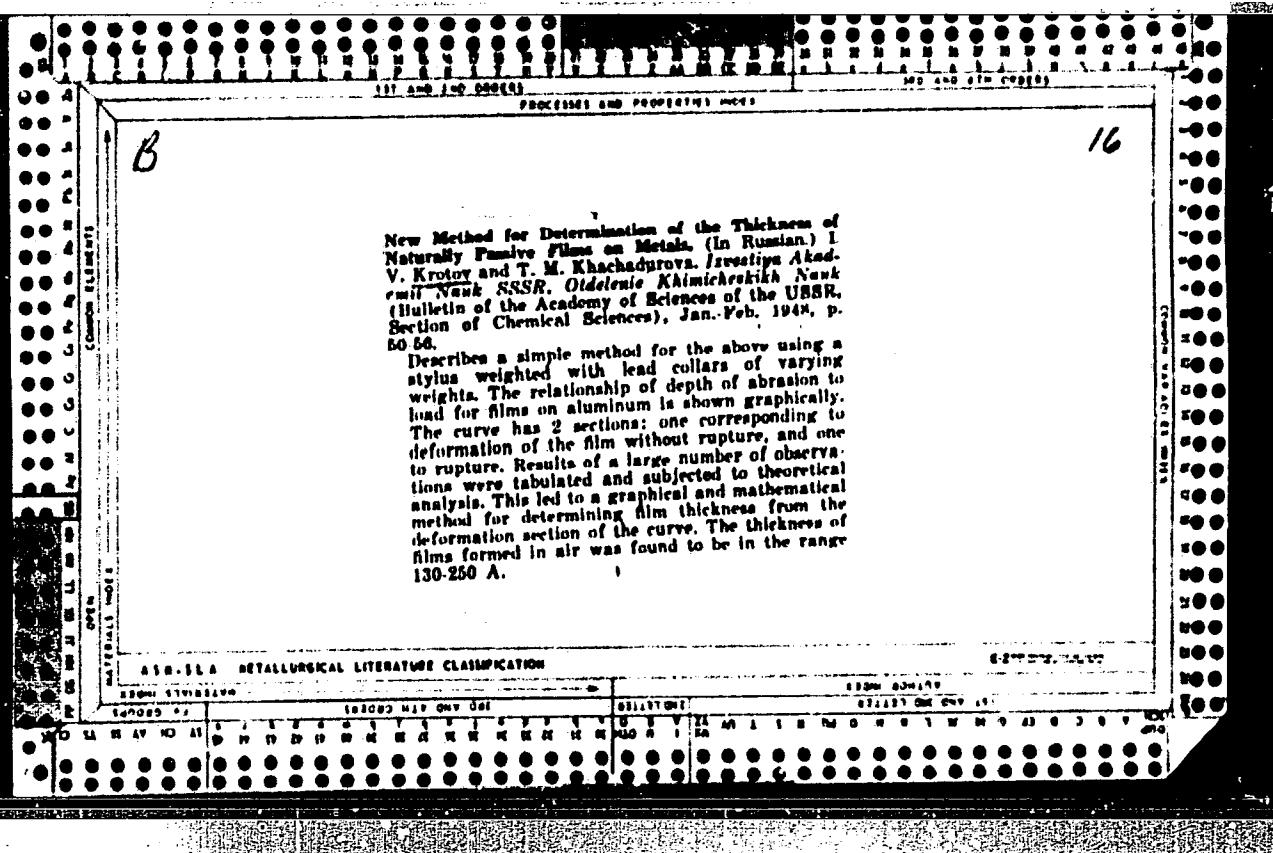
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CA

Application of the reflectometric method to the investigation of corrosion of metals. I. V. Krotov and V. V. Andreeva. Trudy Sovetskogo Topograficheskogo Konservatora, 1940, 319-30; Khim. Referat. Zhur. 1940, No. 7, 136. The reflectometric method is the most simple optical method for investigating metal surfaces. It is less accurate than the polarized-light method, but more simple in its application and it permits to study the kinetics of the corrosion processes. The method is based on the reduction of the intensity of light (detd. by photocells) reflected by the surface of the corroded elements. K. and A. constructed an app. with converging light rays (0.5-1.0 mm. slit) which detns. not only the corrosion processes as a whole, but gives also an insight on local phenomena taking place. Armco steel (3 X 4 cm.) polished with a paste contg. Cr₂O₃ and degreased with benzene and alc. was investigated. (1) in water vapors with and without the access of air and in boiling water and (2) in a boiling 0.1% NaOH soln. and over its vapors (without the access of air). The duration of the expts. was 2, 4, 10 and 30 hrs. Two photograms produced 3 lines of light beams reflected by the galvanometer mirror from the surface of the metal, both polished and corroded. O plays an important role in the corrosion process in water vapors, leading usually to local corrosion which increases with the concn. The total area of corrosion differs only slightly from that obtained with a very small access of air, but its local character is better expressed. Corrosion over vapors of boiling NaOH solns. decreases, but local corrosion remains. Corrosion is of a local character in boiling water. The reflectometric method detns. quantitatively the intensity of corrosion in water, in solns. and in vapors. A more detailed investigation of local corrosion phenomena is possible from decreasing the cross section of the beam of light passing through the slit.

W. R. Henn





C.A

Corrosion of iron in contact with the boundary between an electrolyte solution and an organic liquid. Effect of different cations and anions on the corrosion process. I. V. Kuday. Dokl. Akad. Nauk SSSR 59, 1453-0 (1948). The corrosion was observed in 2 hrs exposures of Fe strips immersed in 0.33 N solns. of electrolytes covered with kerosene. In NiSO_4 , CuSO_4 , FeSO_4 , MnSO_4 , MgSO_4 , and CdSO_4 , corrosion was noticeable mainly only at some distance below the boundary of the 2 liquid phases, whereas in H_2SO_4 and in Al_2SO_4 , there was visible attack along the boundary. The nature of the cation makes a difference only insofar as it lowers the pH through hydrolysis of the salt. In expts. with the same cation (Na^+) and different anions, corrosion began a few mm. below the phase boundary in NaCl and in NaNO_3 . No corrosion was observed in NaCrO_4 down to 0.0417 N, and only sporadic corrosion in 0.0101 N. NaNO_3 prevents corrosion completely down to a concn. of 0.00001 N; in a 0.00016 N soln., corrosion begins a few mm. below the boundary. NaOH protects in the concn. range 0.333-0.0104 N, corrosion appearing at the edges of the sample along the phase boundary at a diln. of 0.001 N. Na_2CrO_4 protects down to a concn. of 0.001 N while Na_2HPO_4 protects down to 0.003 N. There is rusting at pH > 12. The sp. protective action of the anions CrO_4^{2-} and NO_3^- is attributed to plugging of pores in the passivating film.

These are Seven Aviation Inst. in Cetinje

Processes of drop corrosion of aluminum. I. Corrosion of aluminum in contact with drops of salt solutions without addition of free acid. L. V. Krytov and T. M. Khochubina, *Izvest. Akad. Nauk S.S.R., Otdel. Khim. Nauk* 1949, 18-20. — Drops of 0.1 N solns. of NaCl , CaCl_2 , and CuCl_2 deposited on a polished Al surface cause corrosion scattered over the whole area of the drop, and of a depth increasing in the above order. The corrosive action of the salt-solutions increases in the same order, with NaCl , practically ineffective. Nitrates are still less effective. From CuCl_2 , metallic Cu is pptd. densely over the whole area covered by the drop; from CuSO_4 , only a few discrete Cu crystals are pptd.; from $\text{Cu}(\text{NO}_3)_2$, metallic Cu is pptd. mainly along the edge of the drop. Where no drops of CuCl_2 and CuSO_4 do not spread, there is some spreading of drops of $\text{Cu}(\text{NO}_3)_2$ and AgNO_3 , and the tarnish ring adjacent to the drop spreads continuously. In the tarnished zone, free acid can be detected by Congo paper. Drops of mixed solns. of CuCl_2 (or CuSO_4) and $\text{Cu}(\text{NO}_3)_2$ give rise to phenomena similar to those observed with $\text{Cu}(\text{NO}_3)_2$ alone. If the surface is scratched, probably after deposition of the drop, metallic Cu is deposited immediately along the scratch from either $\text{Cu}(\text{NO}_3)_2$ or $\text{Cu}(\text{NO}_3)_2 + \text{CuCl}_2$. In the absence of air, in a CO_2 atm., a drop of $\text{Cu}(\text{NO}_3)_2$, ppt. Cu all over the area covered by the drop, not predominantly along its edge, but the total amt. of Cu pptd. is less than in the presence of air. Pptn. of Cu along scratches, under CO_2 , is much slower than in air. N. Thom

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CH

1. Vladimír Aleksandrovich Kostyukovskii. I. V. Krotov.
Izdat. Akad. Nauk S.S.R., Odzol. Khim. Nauk 1960.
881-4.—Summary of scientific work and biography, with
portrait, on 80th jubilee. G. M. Kosolapoff

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CIA-RDP86-00513R000826710001-6"

CA

Induction period in the displacement of copper from its solutions by passivated iron. I. V. Krotov (Acad. Sci. U.S.S.R., Moscow). Doklady Akad. Nauk S.S.R. 76, 833-836 (1951).—Fe plates, passivated for definite lengths of time t in air at 470-500°, were immersed in solns. of CuSO_4 or of $\text{CuSO}_4 + \text{HgO}$, and the time τ of first appearance of the color of metallic Cu was noted. Induction periods were observed, of a length depending on the thickness y of the oxide film, calcd. from t by the empirical relation $y^2 = h_1 t + h_2$, with the constants h_1 and h_2 detd. experimentally by interference colors. In CuSO_4 0.01 N, τ increases with y , first slowly, then much more rapidly from $y \sim 1000$ Å up. At about 2000-2500 Å, τ decreases sharply, evidently as a result of cracks developed in thick oxide films. In the presence of HgO , τ is, at equal y , considerably shorter than in CuSO_4 alone. At const. concn. of HgO , and varying concn. of CuSO_4 , τ passes through a min. at 0.372 N CuSO_4 , at all y . The induction period is detd. by soln. of the oxide film necessary to bare Fe for the anodic phase of the displacement reaction. In the range of short τ , thin (~ 20 Å) oxide films are dissolved slowly, at about 10 Å./sec., whereas thicker (up to 800 Å.) oxide films are dissolved much faster, at about 180 Å./sec. N. Thom

Ques 25. - A general discussion of the effect of gamma particles as
described in the text, is given in the following section.
It is an easy answer & it can be done in one page.
Ques 26. Structure of metal and metal diffraction
correlation of metals. The film theory of Hirsch et al. for
C.A. 25, 26(83) is analyzed.

APPROVED FOR RELEASE: 06/19/2000 CIA-RDP86-00513R000826710001-6"

KROTOV, I. V.

Chemical Abst.
Vol. 16 No. 9
May 10, 1954
Metallurgy and Metallography

1) 14.5
Corrosion of iron in gasoline, kerosene, and water containing potassium chromate. I. V. Krotov and T. M. Khachaturova. Zhur. Priklad. Khim. 26, 605-71 (1953).—Fe immersed one half in a layer of aq. electrolyte, the other half in supernatant liquid fuel (aviation or cracked gasoline or kerosene) is uniformly and appreciably corroded in contact with aq. phase only. Adding K_2CrO_4 to the latter reduces corrosion sharply. With dilut. H_2O_2 , 0.5N concn. of K_2CrO_4 suppresses corrosion completely. Adding K_2CrO_4 to electrolytes transforms corrosion from general to localized. Corrosion is inhibited by K_2CrO_4 only in weak aq. electrolytes not contg. significant amt's. of dissolved salts. The type of liquid fuel affects corrosion only slightly. R. M. R.

Spontaneous
Klobots (1) eff. in which most spontaneous ignition of
fuels are observed in the layer for 1 minute, 10 feet
and half in an oil layer for 1 minute, exposed to 50°F
water at the heat source, the onset of ignition depends on
the rate of 0.33 g/s for gasoline and 0.15 g/s for kerosene
and 0.4 g/s for kerosene. The onset of ignition depends on
the following concn. of ammonia: 0.1% for kerosene
0.5% for gasoline and 0.5% for cracked gasoline. For spontaneous
ignition to occur, the temperature must be at least 100°F.

Electrochemical Ignition

USSR/Chemistry Physical chemistry

Card : 1/1 Pub. 147 - 22/25

Authors : Krotov, I. V.

Title : Composition and mechanism of formation of a film on iron

Periodical : Zhur. fiz. khim. 28/7, 1327 - 1330, July 1954

Abstract : The composition and mechanism of formation of an oxide film, formed on an iron surface during alkaline burnishing, were investigated electrochemically. Chemical-analytical and thermographic studies of the composition of such an outside film on a burnished iron surface led to the conclusion that this film consists of Fe_3O_4 plus a small amount of adsorbed moisture. It was established that the entire process of alkali burnishing of iron and steel surfaces is a purely electrochemical process. Four references: 3 USSR and 1 German (1920 - 1944). Graphs.

Institution : Acad. of Sc. USSR, Institute of Physical Chemistry, Moscow

Submitted : December 25, 1953

KROTOV, I. V.

USSR/Chemistry - Physical chemistry

Card 1/1 Pub. 147 - 4/27

Authors : Krotov, I. V.

Title : Theory about the process of formation of an anodic film on aluminum

Periodical : Zhur. fiz. khim. 28/9, 1550-1554, Sep 1954

Abstract : The existence of $\text{Al}(\text{OH})_3$ and AlOOH in the composition of an anodic film formed on aluminum in sulfuric acid solutions, was established by chemo-analytical and thermographic methods. It is also shown that the components of the anodic film on Al form directly during the anodic process and that the anodic solution of the Al in the pores of the film offer a protective measure against the solution of the film in acid. It was established that the process of anodic film formation of Al in sulfuric acid solutions is an electro-chemical and colloidal-electrochemical process as well. Eleven references: 4-USSR; 1-Japanese; 1-German; 5-USA (1932-1951). Graph.

Institution : Academy of Sciences USSR, Institute of Physical Chemistry, Moscow

Submitted : July 8, 1953

KRISTOV, IV.

proved that the layer of phosphates represents a solid scale of phosphates of Fe and Mn. Minute admixture of ferric phosphates could also be separated out to give the same result.

The process gradually becomes hydrolytic as the reaction continues.

APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000826710001-6"

✓ 3083* Problems of Investigating Corrosion of Sintered-Iron Products. K voprosu o metodike issledovaniia korroziia pereshirovogo shlekaza. (Russian.) I. V. Kravov. Zhurnal prikladnoi khimii, v. 29, no. 12, Dec. 1955, pp. 1802-1807.

Corrosion-testing difficulties caused by non-homogeneity and porosity. Quantitative determination of rust by cathodic removal in a 2% solution of sulfuric acid. Diagrams, 4 ref.

Krotov I.V.
USSR/Kinetics - Combustion. Explosions. Topochemistry. Catalysis. B-9

Abs Jour : Referat Zhur - Khimiya, No 6, 1957, 18610

Author : I.V. Krotov.

Title : Composition of Rust Developing of Iron in Contact with
Water and Air.

Orig Pub : Zh. fiz. khimii, 1956, 30, No 8, 1696-1701

Abstract : The composition of not freshly formed rust produced by
Fe powder ground in a whirling grinder and rusting in air
(previously wetted with distilled water) was studied by
the chemico-analytical and thermographic methods. As it
seems, rust consists of γ -FeOOH, which loses the absor-
bed water when heated to 149°. In the author's opinion,
 γ -FeOOH transforms into δ -Fe₂O₃ at a further temperatu-
re rise. This process discontinues at about 290°. The
process of transition from δ -Fe₂O₃ into α -Fe₂O₃ ends
at about 676°.

Card 1/1

- 254 -

..... is attributed by the author to
the strong bonding between H₂O and Al₂O₃ in this variety
of aluminum hydroxide.

Card 1/1

APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000826710001-6

AUTHOR:

Krotov, I. V. (Moscow)

74-27-5-5/6

TITLE:

On the Use of radioactive Isotopes in the Investigation
of Corrosion Processes and the Passivation of Metals
(Применение радиоактивных изотопов в исследовании
процессов коррозии и passivirovaniya metallov)

PERIODICAL:

Uspekhi Khimii, 1958, Vol. 27, Nr 5, pp 643-667 (USSR)

ABSTRACT:

In carrying out investigations with marked atoms, β -particles, γ -rays or combinations of both types of radiation are in most cases used. In the introduction the author emphasizes that 2 characteristic features of radioactive substances must be taken into consideration when using it: the half-life and the energy of radioactive radiation. The use of radioactive isotopes emitting particles of a very low energy, as for example the hydrogen-isotope (tritium), requires much more sensitive methods of radiation recording. After further statements on the advantages of the method of radioactive isotopes section 2 of the paper deals with the exchange mechanism between marked ions in the solution and the surface of a metal. The author discusses the investigations

Card 1/3

On the Use of Radioactive Isotopes in the Investigation 74-27-5-5/6
of Corrosion Processes and the Passivation of Metals

of the exchange mechanism (between ions and metals) in solutions which contained ions of radioactive cobalt (works by Simnad and Ruder, reference 8), the investigations by von Hevesy and Blitz (reference 9) and the experiments with nickel samples in the absence of oxygen (where it was proved that nickel attains a much lower activity than cobalt). After further concrete statements the investigation of the purification processes of metals on the basis of the use of marked atoms is discussed in section 3 of the paper. In this connection the published results by Hensley and King (references 14, 15) and the methods of application by Es'um, Fabian and Neut'm (reference 16) are mentioned. Section 4 deals with the use of stable and radioactive isotopes in the investigation of some corrosion processes. In section 5 the use of marked atoms in the investigation of the mechanism of action of the corrosion inhibitors is discussed. In this connection the works by Brasher and Stove (reference 24), Powers and Hackerman (reference 25), Van Hong, Eisler, Brotzin and Harrison (reference 28) are referred to. In section 6 the author deals with the investigation of

Card 2/3

On the Use of Radioactive Isotopes in the Investigation 74-27-5-5/6
of Corrosion Processes and the Passivation of Metals

phosphate coatings on metals by means of marked atoms. In this connection some works and investigation results are discussed, such as those by Boyd, Galan, Markowitz (reference 29), Pryor, Brown, and Cohen (reference 30), Eisler and Doss (reference 36). At the end the author of the present report deals with the employment of the method of autoradiography in investigations of the properties of metal surfaces. There are 5 figures and 42 references, 9 of which are Soviet.

1. Corrosion--Radiographic analysis
2. Radicisotopes--Application

Card 3/3

KROTOV, I.V.; GRININA, V.V.; ZAPOL'SKAYA, N.A.

Formation of aluminum phosphate and chromium films on aluminum
and its alloys. Zhur. prikl. khim. 31 no.1:33-40 Ja '58.

(MIRA 11:4)

(Thermal analysis) (Aluminum alloys) (Metallic films)

3/03/00/01/01/01
100/100/1

AUTHORS: Krotov, I. V., Grinina, V. V.

TITLE: / The Densimetric Method for Determining the Thickness of Protective
Films on Metals

PERIODICAL: Zhurnal prikladnoy khimii, 1960, Vol. 33, No. 11, pp. 2606-2617

TEXT: The following methods are described in literature for determining the thickness of protective films on metals: - The gravimetric method (Refs. 1-3), the electrochemical method (Refs. 4-6), the optical method (Refs. 7-9), the spectral method (Ref. 10), the roentgenographic method (Ref. 11), the volumetric weight method (Ref. 12), the determination by means of a double Lipnik's microscope (Refs. 13-15) for transparent films, the determination by means of the spark-over voltage, and the determination by means of the visible volume (Ref. 19). The following method is proposed here: a sample of sheet material covered with a protective film weighing A g and with a total area of S cm² is used. This sample is placed into a pycnometer filled with a liquid of density d₁ g/cm³ and is weighed. The obtained weight will be C g. The weight of the pycnometer with the liquid will be B g. The average density of the sample with protective film is

Card 1/2

S/080/60/033/011/013/014
A003/A001

The Densimetric Method for Determining the Thickness of Protective Films on Metals

designated as d_{av} g/cm³, the thickness of the protective layer as h cm, the density of the film as d_f g/cm³, the thickness of the metal layer in the sample with the film as h_{Me} cm, the quantity of the free metal in the sample with the film as g. The following formula was found: $C = B + A - \frac{A}{d_{av}} d_1$, from which follows: $d_{av} = \frac{Ad_1}{B+A-C}$. The value g is determined by Ref. 19 as $g = Sh_{Me} d_{Me}$.

where d_{Me} is the tabular value. $A = (2h_f + h_{Me}) Sd_{av}$, from which follows: $h_f = \frac{A - h_{Me}}{2 Sd_{av}}$. $A = 2 Sh_f d_f + g$, from which follows: $d_f = \frac{A - g}{2 Sh_f}$.

[Abstractor's note: Subscripts l(liquid), av (average) and f(film) are translations from the Russian zh ('zhidkost'), sr ('sredniy') and pl ('plenka').] The error of the method described is approximately 10% for films with a thickness of 10 μ . There is 1 table, 1 figure and 22 references: 9 Soviet, 10 English, 3 Italian.

SUBMITTED: March 7, 1960

Card 2/2

TIKHONOV, Mikhail Konstantinovich; KROTOV, I.V., doktor khim. nauk,
prof., otv. red.; GORSHKOV, G.B., red.izd-va; ZUDINA, V.I.,
tekhn. red.; GUS'KOVA, O.M., tekhn. red.

[Corrosion and the protection of concrete and reinforced-concrete hydraulic structures] Korroziia i zashchita morskikh sooruzhenii iz betona i zhelezobetona. Moskva, Izd-vo Akad. nauk SSSR, 1962. 119 p. (MIRA 15:3)

(Hydraulic structures--Corrosion)
(Concrete construction--Corrosion)

KAZNACHEYEV, Yu.I.; KROTOV, I.V.; GRININA, V.V.; KOLESNIKOVA, N.A.

Producing a film on a wave-guide in order to protect it from corrosion and ensure small losses on centimeter and millimeter radio waves. Zhur.prikl.khim. 35 no.12:2684-2687 D '62.

1. Institut fizicheskoy khimii AN SSSR i Institut radiotekhniki i elektroniki AN SSSR.
(Protective coatings) (Radio waves)
(Electronic apparatus and appliances--Corrosion)

(MIRA 16:5)

PUTILOVA, Iya Nikolayevna; LEVANT, Grigoriy Yefimovich; RAYTSYN,
Genrikh Aleksandrovich; MENKOVSKIY, Mikhail Abramovich;
KROTOV, Ivan Vasil'yevich; LOSEV, Boris Ivanovich;
STUKOVNIN, N.D., red.

[Course in general chemistry] Kurs obshchei khimii. [By]
I.N.Putilova i dr. Moskva, Vysshiaia shkola, 1964. 444 p.
(MIRA 18:1)

KROTOV, L. B.

"Experience of Preventive Maintenance Tests of 825-v Traction Network Cables,"
"Operation of Cable Networks" (Eksploatatsiya kabeley i kabel'nykh setey), Gosenergoizdat,
1949, 384 pp.

KROTOV, L.B., inzh.

Special circuits used in rail systems to reduce stray currents,
Vest. TSNII MPS no. 5:52-53 J1 '58. (MIRA 11:8)
(Electric railroads--Wires and wiring)
(Electric currents, Leakage)

- K U T B V L . G

SOF/2246

PHASE I BOOK EXPLOITATION

18(7167)

Zashchita podzemnykh metallicheskikh struktur ot korrozii i
korrektsiya. (Proteksiya of Underground Metal Structures from
Corrosion). (Manual) Moscow, Izdatelstvo inostrannoi literatury,
1959. 705 p. Arista slip inserted. 6,000 copies printed.

M.: P.I. Ryabtsev, Ed. or Publishing House: V.G. Akstres, Tech.
M.: Yu. S. Petrovskaya.

PURPOSE: This collection of articles is intended as a manual on
corrosion protection of underground metal structures.

CONTENTS: The book is divided into four parts. The first part
gives information on the characteristics of underground metal
structures and sources of stray currents. The second part deals
with the theory of soil corrosion of metals and the theory of
corrosion of metals by stray current. The third part deals with
the problems of combating leakage from sources of stray current,
methods and devices for investigating corrosion and the funda-
mentals of planning corrosion prevention. The fourth part ex-
plains measures for preventing corrosion of equipment
and structures and gives the basic operating principles of equipment
and structures. No personalities are mentioned. References follow
Card 1726.

Protection of Underground Metal (Cont.)

SOV/2246

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Card 13/26

Protection of Underground Metal (Cont.) SOV/2246

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Card 6/26

ALEKSAGHEV, V.P. (Moskva); BYKOV, Ye.I. (Moskva); ZIMLYANAYA, Ye.A. (Moskva);
ZIMLYANAYA, L.B. (Moskva); KURTSINA, T.T. (Moskva); OSKINOV, L.G. (Moskva);
SAKOLICH, A.A. (Moskva); YUDITSKIY, S.B. (Moskva)

New semiconductor rectifiers for subway traction substations.
Elektrichestvo no.1:42-48 Ja '65. (MIRA 18:7)

L 11551-66 EWT(1)/EWT(m)/EEC(k)-2/T/EWP(t)/EWP(b)/EWA(h) IJP(c) JD
 ACC NR: AP6005026 SOURCE CODE: UR/0105/65/000/001/0042/0048
 AUTHOR: Aleksashkin, A. A.; Bykov, Ye. I.; Zemlyanaya, Ye. A.; Krotov, L. B.;
 Kurtsina, Z. T.; Poselenov, L. B.; Sakovich, A. A.; Yuditskiy, S. B.
 ORG: none
 TITLE: New semiconductor rectifiers for the rectifier substations of subways
 SOURCE: Elektrichestvo, no. 1, 1965, 42-48
 TOPIC TAGS: semiconductor rectifier, electric engineering, electric substation equipment
 ABSTRACT: Computations are presented to prove the feasibility and economy of replacing the six IVS-500/2 type sealed mercury-arc rectifiers with stacks of VK-200/4A type silicon rectifier cells (average current 200 amp; operating peak inverse voltage [PIV] 400 v; rated PIV 600 v) in the rectifier substations that supply 825 volts d-c for subway traction. The computed six-phase rectifier unit has six parallel branches per phase, with six series-connected cells per branch. The number of parallel branches is computed on the basis of peak load and surge current, taking the circuit-breaker interrupting time into consideration. The number of series-connected cells is computed on the basis of the PIV's, with allowances for variations in the supply voltage. The overall efficiency of the rectifier unit is 98.9 percent. It is assembled from modular stacks (12 cells and one fan per module) and fits into two cabinets 800 by 600 by 2000 mm. Although at present silicon rectifiers are more expensive than mercury-arc
 Card 1/2 27 UDC: 621.314.632.4:621.311.44:625.42

"APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000826710001-6

L 11551-66

ACC NR: AP6005026

rectifiers (due to the high cost of single-crystal silicon and the low level of automation), conversion of subway substations to such rectifiers pays for itself in five to six years, and the economy of building new rectifier substations of this type is even greater. Orig. art. has: 4 figures, 23 formulas, and 1 table.

SUB CODE: 09 / SUBM DATE: 08Jun64 / ORIG REF: 002

H W
Card 2/2

APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000826710001-6"

KUZNETSOV, V.N.; KROTOV, L.F.

Production of antifriction, heat-resistnat cast iron by the alloying
of ordinary gray cast iron in the ladle. Lit. proizv. no.5:39 My '62.
(Cast iron—Metallurgy) (MIRA 16:3)

KROTOV, L.N., inzhener.

Spacing for air drying of Siberian larch lumber. Der.prom. 5 no.9:
18 S '56.
(MLRA 9:10)

1.Sibirskiy lesotekhnicheskiy institut.
(Lumber--Drying) (Larch)

KROTOV, I. N.: Master Tech Sci (diss) -- "Investigation of atmospheric drying
of sawdust from the lumber of the Siberian larch". Leningrad, 1958. 12 pp
(Min Higher Educ USSR, Leningrad Order of Lenin Forestry Engineering Acad im
S. M. Kirov), 150 copies (KL, No 7, 1959, 124)

KROTOV, MODEST. A.

Rodnaia IAkutiia My Yakutia.
IAkutsk, 1952. 218 p.

SO: Monthly List of Russian Accessions, Vol. 6 No. 7 October 1953

KROTOV, Modest Alekseyevich; GOLUBKOVA, V.A., red.; ROZEN, E.A., tekhn.red.

[In the Far North] Na severo dal'nem. Moskva, Izd-vo "Sovetskaja Rossiia," 1957. 213 p.
(MIEA 11:5)
(Yakutia)

ARGUNOV, I.A., red.; VASIL'YEV, S.N., red.; KORYAKIN, P.I., red.; KROTOV,
M.A., red.; LUKONIN, G.A., red.; TOMSKIY, S.K., red.; CHERSKII,
N.V., red.; CHIRYAYEV, G.O., red.; SOLOV'YEVA, Ye.P., tekhn.red.

[Forty years of the Yakut A.S.S.R.] 40 let Iakutskoi ASSR.

Iakutsk, Iakutskoe knishnoe izd-vo, 1962. 189 p.

(MIRA 16:1)

(Yakutia—Economic conditions) (Yakutia—Culture)

KROTOV, N., inzh.

Show interest and concern in our integrated brigades. Mor. flot
23 no.3:7-8 Mr '63. (MIRA 16:3)

1. Otdel truda i zarabotnoy platy Potiyskogo porta.
(Wages—Longshoremen)

5 (2)
AUTHORS:

Rode, Ye. Ya., Krotov, N. A.

SOV/78-4-8-13/43

TITLE:

The Physico-chemical Investigation of Silico-12-Tungstic Acid
(Fiziko-khimicheskoye izucheniiye kremne-12-vol'framovoy
kisloty)

PERIODICAL:

Zhurnal neorganicheskoy khimii, 1959, Vol 4, Nr 8,
pp 1782 - 1793 (USSR)

ABSTRACT:

The authors give the publication data which are available on :
the compound mentioned in the title (Refs 1-19). This com-
pound shows cis-trans-isomerism. On the basis of radiographic
analysis its formula is $H_4[SiW_{12}O_{40}] \cdot nH_2O$. The compound was
investigated in solid crystalline state at different degrees
of hydration. Figures 1 - 3 show the thermographic analysis,
figure 4 shows the polythermal lines of the preparations of
different hydration, figure 5 the isothermal lines of decompo-
sition, figure 6 the diagram composition - temperature and
figure 7 the isothermal lines of vapor pressure for prepara-
tions of different hydration. The results may be summarised as
follows: at 40 - 44° a liquefaction and decomposition of the
hydrate with $n = 33H_2O$ takes place. The hydrate is formed with

A

SU1
Car.

Card 1/2

- references,
- the General and Inorganic Chemistry Institute im. N. S. Kurnakova
of the Academy of Sciences, USSR)

RODE, Ye.Ya.; KROTOV, N.A.

Tetrasubstituted lithium salt of silico-12-tungstic acid.
Zhur.neorg.khim. 8 no.4:939-949 Ap '63. (MIRA 16:1)

1. Institut obshchey i neorganicheskoy khimii imeni M.S.Kurnakova
AN SSSR.
(Lithium compounds) (Silicotungstic acid)

KROTOV, N.A.; RODE, Ye.Ya.

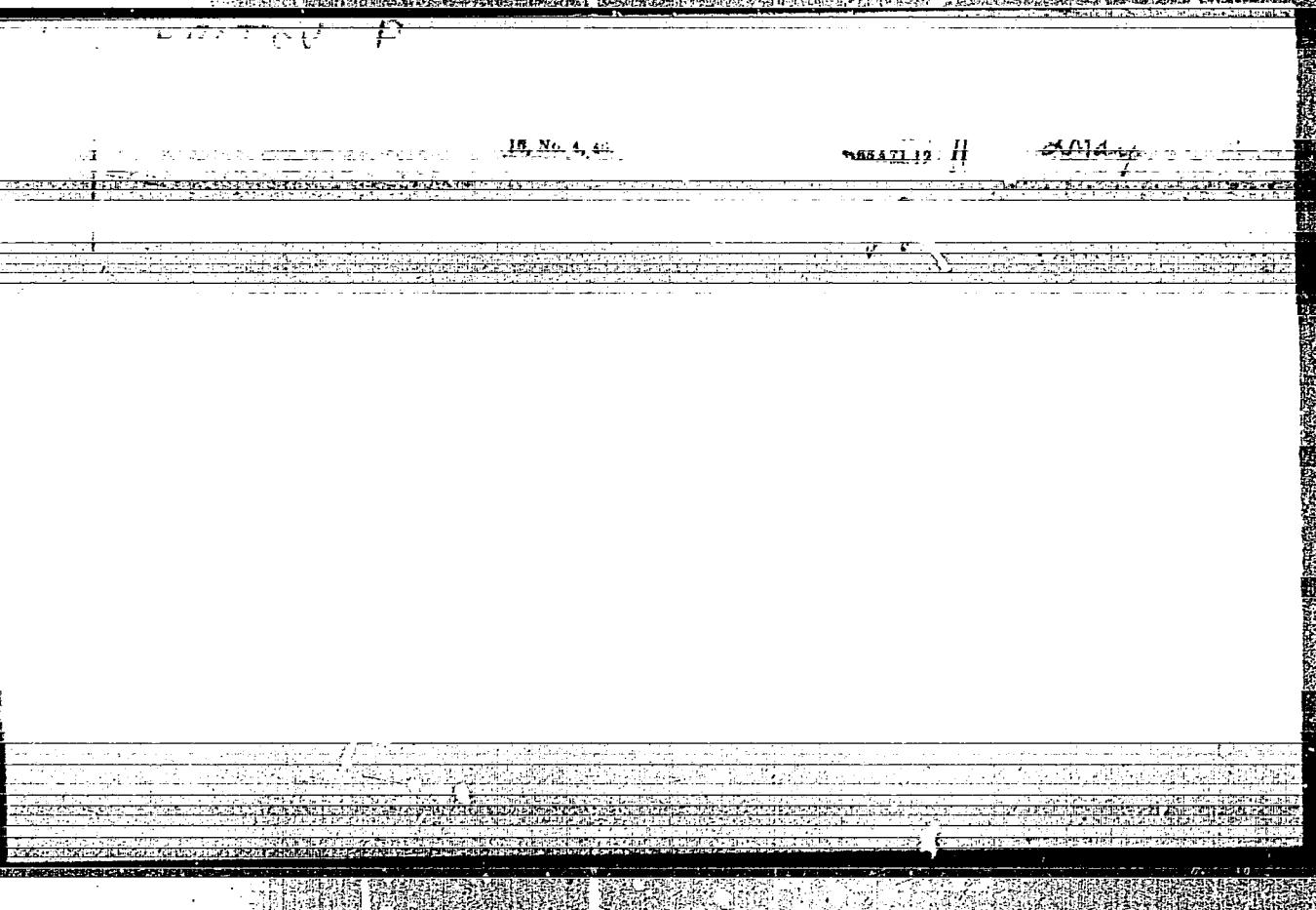
Acid lithium salts of silico-12-tungstic acid. Zhur. neorg. khim. 8 no.7:1722-1736 Jl '63. (MIRA 16:7)

1. Institut obshchey i neorganicheskoy khimii imeni N.S. Kurnakova AN SSSR.
(Silicotungstic acid) (Lithium salts)

POLYANIN, D.V.; ZOTOV, G.M.; GRYAZNOV, E.A.; MENZHINSKIY, Ye.A.; RUBININ, A.Ye.; CHEBOTAREVA, Ye.D.; ZAKHMATOV, M.I.; OKUNEVA, L.P.; SHMELEV, V.V.; STULOV, A.A.; POKROVSKIY, A.N.; SHIL'DKRUT, V.A.; IVANOV, A.S.; NABOROV, V.B.; FINOGENOV, V.P.; KUR'YEROV, V.G.; KHRAZTSOV, B.A.; BATYGIN, K.S.; BOGDANOV, O.S.; KROTOV, O.K.; GONCHAROV, A.N.; KRESTOV, B.D.; LYUBSKIY, M.S.; SOKOL'NIKOV, G.O.; KAMENSKIY, N.N.; YASHCHENKO, G.I.; SABEL'NIKOV, L.V.; GERCHIKOVA, I.N.; FEDOROV, B.A.; STEPANOV, G.P.; BORODAYEVSKIY, A.D.; INGATUSHCHENKO, S.K.; VARTUMYAN, E.L.; KAPELINSKIY, Yu.N., red.; MAYOROV, B.V., red.; NABOROV, V.B., red.; SOLODKIN, R.G., red.; DROZDOV, A.G., red.; ROSSHIMA, L. red.; SOLOV'YEVA, G., red.; CHEPELEVA, O., red.

[The economy of capitalist countries in 1961; economically developed countries] Ekonomika kapitalisticheskikh stran v 1961 godu; ekonomicheski razvitye strany. Pod red. Iu.N.Kapelinskogo. Moskva, Sotskgiz, 1962. 447 p. (MIRA 16:2)
(Economic history)

"APPROVED FOR RELEASE: 06/19/2000 CIA-RDP86-00513R000826710001-6



APPROVED FOR RELEASE: 06/19/2000 CIA-RDP86-00513R000826710001-6"

KROTOV, P.V., starshiy nauchnyy sotrudnik

Single-phase electric locomotive with silicon rectifiers. Trudy
TSNIIME no.34:41-51 '62, (MIRA 16:1)
(Electric locomotives) (Electric current rectifiers)

KROTOV, P.V., starshiy nauchnyy sotrudnik; BOLDOV, M.Ye., starshiy nauchnyy sotrudnik; SHVIONOV, I.V., mladshiy nauchnyy sotrudnik

Studying silicon rectifiers. Trudy TSNIIME no.34:53-69 '62.
(MIRA 16:1)
(Electric locomotives) (Electric current rectifiers)

KBOTOV, P.V., Instn.

Electrification of narrow-gauge railroads. Thel. dor. transp. 46
no. 10:73-77 6 '64. (MIRE 17:11)

KUOTOV, S.

36046 MeKhanizirovanny maslozavod iz stornykh konstruktsiy. Moloch prom-st',
1949, No. 11, S. 12-17

SO: Letopis' Zhurnal'nykh Statey, Vol. 45, 1949

KRUTOV, S.

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Dairy Plants

Constructing a creamery with prefabricated sections. Mol. prom. 13
No. 6, 1952.

9. Monthly List of Russian Accessions, Library of Congress, September 1952, Unc1.

KROTOV, S. A., KOLPAKOV, A. S.

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Buildings prefabricated

Large panel construction of industrial enterprises. Stroi. prom. 30,
No. 3, March 1952.

9. Monthly List of Russian Accessions, Library of Congress, August 1952 ~~1953~~, Unc1.

Kirov S.A.
Sep 53
USSR/Engineering - Prefabricated
Construction

"Experience of Constructing Industrial Buildings of
Large Panels," Engr S. A. Krotov, Tsentrmaslostroy

Bgull Stroi Tekh No 14, pp 1-4

States Tsentrmaslostroy trust has designed a cream-
ery plant of prefabricated reinforced-concrete
panels made at the trust's mfg plants. First exptl
prefab creamery erected 1950 at Svecha stantsiya,
Kirovskaya Oblast. Mass production of prefab parts

270r85

organized 1952 at plant in Vologda, and group of pre-
fab creameries erected in Voronezhskaya and Kurskaya
Oblast. Wall panels are 4.5 X 1.0 and 4.5 X 1.6 m
in dimension; corresponding wts are 1.7 and 2.5 tons.
Describes construction of panels, with dimensions,
and buildings.

270r85

PA 27/49T29

USSR/Chemistry - Aluminum, Corrosion of Jan/Feb 49
Chemistry - Corrosion of Aluminum, by
Saline Solutions

"Processes of Drop Corrosion of Aluminum: I, Corro-
sion of Aluminum in Contact With Drops of Saline
Solutions Without the Addition of Free Acid," V. Krotov
T. M. Khashadurova, Moscow Avn Inst imeni S.
Ordzhonikidze, 9 pp

"Iz Ak Nauk SSSR, Otdel Khim Nauk" No 1

Makes several new observations on processes of drop
corrosion in aluminum. Shows characteristic role of
atmospheric oxygen in these processes. Considers
results of experiments from thermodynamic standpoint.

27/49T29

USSR/Chemistry - Aluminum, Corrosion of Jan/Feb 49
(Contd)

"Calculates characteristics and changes of the oxide
film on aluminum. Submitted 3 Jan 48.

27/49T29

KROTOV, V.

KROTOV, V., delegat XXI s"yesta Kommunisticheskoy partii Sovetskogo Soyusa.

Heavy industry is the basis of our country's might. Voen.
znan. 35 no.6:3-5 Je '59. (MIRA 12:12)

1. Direktor Ural'skogo zavoda tsvetnoy metalostroyeniya
im. S.Ordzhonikidze.
(Sverdlovsk--Machinery industry)

33219

S/141/61/004/006/005/017
E032/E114

9,9441

AUTHORS: Gorskiy, S.M., and Krotov, V.A.

TITLE: Some characteristics of atmospheric noise in the
range 2 - 25 cps

PERIODICAL: Izvestiya vysshikh uchebnykh zavedeniy,
Radiofizika, v.4, no.6, 1961, 1025-1028

TEXT: The aim of this work was to investigate the intensity
and the spectral distribution of the vertical magnetic component
of the electromagnetic field associated with atmospheric noise in
the frequency range 2-25 cps. The measurements were carried out
in the Crimea. The atmospheric noise was received by a $5 \times 10^4 \text{ m}^2$
horizontal frame antenna. The block diagram of the arrangement
is given in Fig.1. The amplifier had a symmetric input and an
asymmetric output. The maximum amplification coefficient was
 2.5×10^5 and could be reduced by factors of 2.5 and 10
respectively. The amplified signals were recorded on magnetic
tape and could also be inspected visually on the screen of an
oscilloscope. The minimum detectable signal was 2×10^{-10} oe.
It was found that the rms fluctuation in the magnetic component

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Some characteristics of atmospheric... S/141/61/004/006/005/017
E032/E114

during October was 5×10^{-9} oe. The signals were also analysed with an optical Fourier analyser with a resolution of 0.1 cps (V.A. Zverev, Ye.F. Orlov, Ref.8; Pribory i tekhnika eksperimenta, in press).

Fig.4 shows a typical spectrogram of atmospheric noise. The spectrum is not flat; its intensity increases at lower wavelengths and there is a sharp line at 9 cps (roughly in the middle of the picture). This line is ascribed to a cavity resonance, in accordance with the suggestion put forward by W.O. Schumann and H. König (Ref.9: Naturwiss., v.41, 183 (1954)). It is pointed out that the line was not observed after sunset. Fig.5 shows the spectral density of atmospheric noise averaged over 25 sets of observations for September and October as a function of frequency. The slight minimum between 5 and 10 cps is interpreted as the boundary between the atmospheric noise spectrum and the geomagnetic micropulsation spectrum. Acknowledgments are expressed to V.A. Zverev and M.M. Kobrin for their valuable suggestions.

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33219

Some characteristics of atmospheric... S/141/61/004/006/005/017
E032/E114

There are 5 figures and 9 references; 3 Soviet-bloc and 6 non-Soviet-bloc. The four most recent English language references read as follows:

Ref.3: E.F. Pierce. J.Res.Nat.B.St., v.64-D, 4 (1960).
Ref.4: A.D. Watt, J.Res.Nat.B.St., v.64-9, 4 (1960).
Ref.6: W.H. Campbell, J.Res.Nat.B.St., v.64-D, 4 (1960).
Ref.7: Obayashi Tatsuzo, Rept.Ionos. Res. Japan, v.12, 3 (1958).

ASSOCIATION: Gor'kovskiy nauchno-issledovatel'skiy radiofizicheskiy institut pri Gor'kovskom universitete
(Gor'kiy Scientific Research Radiophysics Institute at Gor'kiy University)

SUBMITTED: April 14, 1961

Card 3/8 3

RUMYANTSEV, G.N., redaktor; BORISOV, N.I., redaktor; BUYANTUYEV, B.B.,
redaktor; KROTOV, V.A., redaktor; RAZUMOV, I.M., redaktor;
KHADALOV, P.I., redaktor; SHNIPER, R.I., redaktor; AKHANOV,
Ts.B., tekhnicheskiy redaktor.

[Studies on the production forces of the Buryat-Mongolian
A.S.S.R.] Materialy po isucheniiu proizvoditel'nykh sil
Buriat Mongol'skoi ASSR. Ulan-Ude, Buriat-Mongol'skoe kn-vo.
no.1. 1954. 425 p. (MLRA 9:5)
(Buryat-Mongolia--Economic geography)

POPOV, S.D., ovt.red.; BORISOV, N.I., red.; BUYANTUYEV, B.R., red.; GALAKTIONOV, I.I., red.; KROTOV, V.A., red.; OZNOBIN, N.M., red.; PAVLOVSKIY, Ye.V., red.; TARASOV, G.L., red.; SHNIPER, R.I., red.; AKHANOV, TS.B., tekhn.red.

[Studies on the production forces of the Buryat-Mongolian A.S.S.R.]
Materialy po izucheniiu proizvoditel'nykh sil Buriat-Mongol'skoi ASSR.
No.2. Ulan-Ude, Buriat-Mongol'skoe knishnoe izd-vo. 1955 507 p.

(MIRA 12:4)

1. Akademiya nauk SSSR. Vostochno-Sibirskiy filial. 2. Sovet po izucheniiu proizvoditel'nykh sil AN SSSR (for Popov, Galaktionov, Tarasov).
3. Zamestitel' predsedatelya Soveta Ministrov Buryat-Mongol'skoy ASSR (for Borisov). 4. Vostochno-Sibirskiy filial AN SSSR (for Buyantuyev).
5. Institut ekonomiki AN SSSR (for Oznobin). 6. Gosplan Buryat-Mongol'skoy ASSR (for Shniper).

(Buryat-Mongolia—Geography, Economic)

PHASE I BOOK EXPLOITATION 822

Krotov, V.A.

'Irkutskaya oblast', ekonomiko-geograficheskiy obzor (Irkutskaya oblast'; an Economic and Geographical Survey) [Irkutsk] Irkutskoye knizhnoye izd-vo, 1956. 32 p. 15,000 copies printed.

Ed. (title page): Silinskiy, P.P.; Ed. (inside book): Shafirova, A.S.; Tech. Ed.: Trushkina, T. M.

PURPOSE: This booklet is intended for the general public and students interested in the geography and economics of Eastern Siberia.

COVERAGE: The material presented in this booklet is an expanded version of the text of a public lecture sponsored by the Irkutsk section of the Obshchestvo po resprostraneniyu politicheskikh i nauchnykh znanii (Society for the propagation of Political and Scientific Knowledge) and by the Vostochno-Sibirskiy filial (East Siberian Branch) of the USSR Academy of Sciences. The author discusses briefly the history and geography of Irkutskaya oblast', its weather conditions and natural resources, its population, industry, agriculture and transportation. He foresees a bright future for Irkutskaya oblast' insofar as its industrial potential is concerned. Its planned industrial development is based primarily on the existing and future supply of hydroelectric energy. The Bratsk hydroelectric

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Irkutskaya oblast'; an Economic and Geographical Survey 822

plant alone, which is being constructed on the Angara River, will have a 3,200,000 kw hr. capacity. Its annual output, estimated at 22,000,000,000 kw. hr., will equal the combined output of the Kuybyshev and Stalingrad Hydroelectric stations and will provide energy for the operation of a number of planned industrial plants. Irkutskaya oblast' is rich in gold, coal, muscovite, salt, gypsum, furs and timber. Its timber amounts to 12 percent of the total timber available in the USSR. An oil pipeline is planned from Omsk and Govosibirsk to an Irkutsk refinery now under construction. A number of existing and planned industrial plants and hydroelectric power stations are mentioned. The machine-building industry centers are Irkutsk, Usol'ye, Angarsk and Cheremkhovo. The author also mentions the fact that there is no cement factory in Irkutskaya oblast' and that its network of railroads and paved roads is one of the poorest in the USSR. A chemical industry based on the available minerals and sawmill wastes will become one of the most important local industries. No personalities are mentioned. There are no references.

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Irkutskaya Oblast'; an Economic and Geographical Survey 822

Transportation
Internal Diversity

27
30

Maps:

1. Administrative rayons
2. Principal mineral products and industrial centers
3. Chart of areas under cultivation

AVAILABLE: Library of Congress

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"APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000826710001-6

KROTOV, V.

V. KROTOV, Chairman of the Presidium of the East Siberian Branch, Acad. Sci. USSR,
author of article "Increase the Role of Science in Developing the Eastern
Areas of the USSR,"
SO: Pravda, 4 May 1956.

APPROVED FOR RELEASE: 06/19/2000

CIA-RDP86-00513R000826710001-6"

KROTCV, V.A.

Development and distribution of productive forces of Eastern Siberia in
the sixth five-year plan and some problems of geographical research. Izv.
AN SSSR. Ser. geog. no. 4:76-84 Jl-Ag '56. (MIRA 9:10)

1. Vostochno-Sibirskiy filial Akademii nauk SSSR.
(Siberia, Eastern--Economic policy)