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1.... 27901 s/079/61/031/010/001/010 53610 D227/D304 5.4500 Bezuglyy, V.D., and Shimanskaya, N.P. AUTHORS: Polarographic study of some oxozoles TITLE: Zhurnal obshchey khimii, v. 31, no. 10, 1961, PERIODICAL: 3160-3177 The investigations were concerned with phenyl-, diphenyl-, and naphthyl substituted oxozoles, used as luminophores in a scintillator to establish the relation between their polarographic and optical properties. The measurements were conducted with the Geyrovsky-Shikal-polarograph using a saturated, high specific re-sistance solution of $N(C_{CH_5})_A$ I in 92% methanol. The solutions of oxozoles used in the investigation were prepared using 60:40 methanol:dioxane mixtures. The experiments showed that phenyl-, naphthyl-, and biphenyl- substituted oxozoles undergo reduction at the cathode giving unique polarographic waves for different substituents. 2-Methyl-5-phenyl-oxozole did not reduce under the experimental conditions used, but 2,5-diphenyl oxozole gave two waves on the polarograph, whose half wave potentials corresponded to Х

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Polarographic study ...

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-2.10 and -2.33V, referred to the standard calomel electrode. Substitution of the 2-phenyl radical with $2 \propto$ -naphthyl displaces the first half-wave potential to -1.85V; B-naphthyl substitution displaces $E_{1/2}$ of the first wave to -2.02V. $E_{1/2}$ of the second wave for 1-naphthyl derivative is -2.23 and for 2-naphthyl derivative -2.23, i.e. very near the value for the second wave of 2,5-diphenyloxozole. A greater effect may be obtained by introducing into position 2 of the diphenyloxozole radical, when $E_{1/2}$ of the first wave = -1.77 with $E_{1/2}$ for the second wave -2.18V and $E_{1/2}$ for the third wave -2.28V. This compound may be considered composed of two 5-phenyloxozole groups joined by a phenyl radical forming a bridge between the electron interaction of the two groups. If the bridge is provided by -CH=CH-, a group containing \mathcal{T} electrons, $E_{1/2}$ is displaced towards less negative values and is equal to -1.36V. Introduction of a -CH2-CH2- bridging group causes the loss of the polarographic activity of 1,2 - di K.

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27901 s/079/61/031/010/001/010 D227/D304 Polarographic study ... (5 phenyloxozole)-ethane. To investigate the effect of 5-position substituted derivatives, 2,5-di-(4-biphenyly1) oxozole and 2-(4-biphenyly1)-5-(1-naphthy1)- had $E_{1/2}^1 = -2.00V$, $E_{1/2}^2 = 2,22V$ and $E_{1/2}^1 = -1.97V$, and $E_{1/2}^2 = 2,22V$ respectively. From the experimental results it follows that the 5-membered oxozole ring undergoes reduction, under specific conditions, at the mercury dropping cathode and the ease of reduction depends on the substituent, and its position in the ring. Substitution of phenyl instead of methyl group in 2-position gives rise to unique polarographic waves. Substituents in position-5 have a smaller effect and the introduction of 1-naphthyl in place of biphenyl in the 5-position (substituents in position-2 remaining the same) changes the halfwave potential very little. It follows then that the most readily reducible is the C=N- bond and only after its reduction can the lend the between position-2 substituents and >C=C < bond (between 4th and 5th C) be broken; the substituent has practically no effect on the $E_{1/2}$ of the second wave. The electro-negative effect of X Card 3/6

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Polarographic study ...

which explains the absence of second waves. A table also gives the wave lengths in the region of maximum absorption and it may be seen that the character of Λ_{max} , in general, corresponds to the variation of $E_{1/2}$ values. The fact also confirms the dependence of polarographic results on the character of the substituent. It was also interesting to compare scintillation effectiveness of the compounds with the polarographic results which shows a certain correlation between these properties. It may be concluded that the polarographic method may be used for determining the effectiveness of a given substance as a scintillator, this effectiveness being higher for less negative $E_{1/2}$ values. There are 1 table, 10 figures and 9 references: 5 Soviet-bloc and 4 non-Soviet-bloc. The references to the English-language publications read as follows: F. Hayes, L. King, J. Am. Chem. Soc. 74, 1106 (1952); E. Hartnell, C. Bricker, J. Am. Chem. Soc. 70, 3385 (1948).

Card 5/6

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"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001549510009-7 27901 S/079/61/031/010/001/010 D227/D304 Polarographic study ... Khar'kovskiy filial vsesoyuznogo nauchno-issledo-vatel'skogo instituta khimicheskikh reaktivov ASSOCIATION: (Khar'kov branch of All-Union Scientific Research Institute for Chemical Reagents) September 10, 1960 SUBMITTED: Ł, Card 6/6 2 - S . 15

APPROVED FOR RELEASE: 08/23/2000

L 13353-63 EWP(j)/EPF(c)/EWT(m)/BDS AFFTC/ASD Pc-4/Pr-4 RM/WM ACCESSION NR: AP3002622 8/0079/63/033/006/1726/1732 AUTHOR: Shimanskaya, N. P.; Bezugly#/, V. D. 66 TITLE: Polarographic investigation of oxadiazole derivatives 65 SOURCE: Zhurnal obshchey khimii, v. 33, no. 6, 1963, 1726-1732 TOPIC TAGS: polarographic investigation, oxadiazole derivative, half-wave potential, polystyrene ABSTRACT: The polarographic properties of oxadiazole derivatives such as phenyl, ABSTRACT: The polarographic properties of oxadiazole derivatives in plastics were	
ACCESSION NR: AP3002022 AUTHOR: Shimanskaya, N. P.; Bezugly, V. D. TITLE: Folarographic investigation of <u>oxadiazole</u> derivatives SOURCE: Zhurnal obshchey khimii, v. 33, no. 6, 1963, 1726-1732 SOURCE: polarographic investigation, oxadiazole derivative, half-wave TOPIC TAGS: polarographic investigation, oxadiazole derivative, half-wave motential, polystyrene	
TITLE: Polarographic investigation of <u>oxadiazola</u> and a second s	
TITLE: Polarographic investigation of <u>Oxadiazola</u> activation SOURCE: Zhurnal obshchey khimii, v. 33, no. 6, 1963, 1726-1732 TOPIC TAGS: polarographic investigation, oxadiazole derivative, half-wave motential, polystyrene	
TOPIC TAGS: polarographic investigation, managements	
metential, polysurcus	
malarographic properties of oxadiazole derivatives in plastics were	
biphenyl, naphting as found that substituents in the ring. The polarographic data	
advantageously the some optical properties. The latter wave potential (diffusion	-
and catalytic) and the polarographic data, the possible schedule properties are basis of the polarographic data, the polarographic properties are ubstances can be determined. The polarographic properties as an experimental.	
influenced by the effect of conjugation. 1010-11 substance. Orig. art. has: 4 figures and 1 table. 15	
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TAUS <u>plastic scintillator</u>, electrochemical property, optical property

TRANGATION. The authors evolved a procedure for the polarographic action mineric scintillators. This enabled them to define the electrochemical proper-A and oxarole-1.3, as well as



HEZUGLYY, V.D.; SHIMANSKAYA, N.P.; PERESLENI, 70.M.

Mechanism of reduction of 1,3-oxazole and 1,3,4-oxadiazole derivatives. Zhur. ob. khim. 34 no.11:3540-3545 N *64 (MIRA 18:1)

1. Vsesoyuznyy nauchno-issledovatel'skiy institut monokristallov, Khar'kov.

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CC NRI AP6031380	SOURCE CODE: UR/0079/66/0	36/009/1601/1603
UTHOR: Shimanskaya,	N. P., Malkes, L. Ya.; Bezuglyy, V. D.	50B
RG: All Union Scien	tific Research Institute of Monocrystals, Scin	tillating Materials
nd High Purity Chemi	cals (Vsesoyuznyy naudno-issledovatel'skiy ins materialov i osobo chistykh khimicheskikh vesh	titut monokristalky,
	1 1	cnestv)
TLE: Reaction rate	of thermal decomposition of some azines	
URCE: Zhurnal obsh	chey khimii, v. 36, no. 9, 1966, 1601-1603	
PIC TAGS: the t	hermal decomposition, reaction rate constantly c	DRGANIC AZINE
STRACT: Polarograp	hic studies were made of the effect of the strate constants of their thermal decomposition a	t 300C.
	$\begin{array}{c} R-CH=N-N=CH-R \xrightarrow{\prime_{1}} R-CH=CH-R+N_{1}. \end{array}$	
a manulta abound th		
on. Reaction rate e given in the tabl	at the decomposition of azines is a first order constants calculated for some azines from the e e:	experimental data
-		
rd 1/2	UDC: 57.8524	

- International States		11 A - 1
SHART	ANSKAYA, N-S. SHIMANSKAJA, N.O.	
	CARD 1 / 2 PA - 1805	
SUBJECT AUTHOR	USSR / PHYSICS GORŠKOV,G.V., SIMANSKAJA,N.S. On Calorimetric Measurements of Preparations of Naturally Radio-	
TITLE	On Calorimetric Louis active Families. Atomnaja Energija, <u>1</u> , fasc.5, 86-93 (1956)	
PERIODICAL		
stances whi of self-abs measurement committed (tions by co On the occ: the author active con (1953). Th gies of ga are no ist	Asomnays 1 / 1957 Issued: 1 / 1957 Ations themselves can be liquid or mixed with other non-active sub- ations themselves can be liquid or mixed with other non-active sub- ations themselves can be liquid or mixed with other non-active sub- ations the some cases, are highly absorbent. The effect of absorption and ach, in some cases, are highly absorbent. The effect of absorption and sorption can mostly not be estimated, for which reason the accuracy of sorption can mostly not be estimated, for which reason the accuracy of the mostly does not go beyond from 2 to 3%. Even greater errors are on the occasion of the determination of the radioactivity of prepara- on the occasion of the determination of different origin. Somparison with a gauging preparation of different origin. asion of calorimetric measurements of naturally radioactive preparations asion of calorimetric measurements of naturally radioactive preparations (e.g. J.HOLLANDER, J.PERLMAN, G.SEABORG, Rev.Mod.Phys. 25, 429, istants (e.g. J.HOLLANDER, J.PERLMAN, G.SEABORG, Rev.Mod.Phys. 25, 429, and their relative and absolute intensities. Furthermore, there are concerning the number of conversion electrons and the average ener- a concerning the number of conversion electrons and the average ener- a spectra. It is for this reason that the authors carefully analyzed as spectra. It is for this reason that the authors carefully analyzed ing data on energies and on the radiation yields of the elements of the ingline tradioactive families. By the critical investigation of a great experimental works it was possible to determine the energy E ₁ for each	

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		PA - 1803
Atomnaja Energij	E , <u>1</u> , fasc. 5, 86-93 (1956) CARD 2 / 2	
element as well a one act of decay equilibrium. For	as to determine the total "thermal" energies of the preparations of Ra, MsTh, RaTh, and the radium preparations also the correction of caused by the accumulation of RaE and Po	ons for the increase 210 was computed.
Knowledge of all	these quantities made calorimetric measured	active families
possible. For th A table shows th	is purpose double calorimeters of the set e measuring results of some Ra-, RaTh- and	Au-preparations. the calorimetric
method: The rel curie for the Ra	ations between the milligiam equilibrit Th- and Ac-isotopes which are in equilibrit	um; these relations With the conditions
usual in the USS standard chamber	R for ionization measuring (lead inclusion $SGM-1$) the following results are obtained $SGM-1$) the following results are 10.0 ± 0.5 mil	: 1 mg-equ RaTh = licurie Ac), the con-
content of Ra an	d MsTh in radium-mesothorium preparations.	The calorimetric ith unknown time of
production ("age	"), but it is well surrou for the dotter	ation of the compo-
sition of "young	g" preparations.	
INSTITUTION:		
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SHIMANSK	CAPA, N.S. CARD 1/2 PA - 1759
SUBJECT _nthor TITLE	USSR / PHISICS SIMANSKAJA, N.S., JASUGINA, E.A. SIMANSKAJA, N.S., JASUGINA, E.A. 227 by the calorimetrical
PERIODICAL	Determination of Method. Atomnaja Energija, <u>1</u> , fasc.5, 133-133 (1956) Issued: 1 / 1957 Nors had a weighable quantity of pure Ac ²²⁷ available, they attempted its half-life calorimetrically. The calorimetrical measuring of half-
to measure life of lon heat genera tope. The f The prepara ties in the tained no preparatio together w out after	Issued: $1/1997$ available, they attempts hors had a weighable quantity of pure Ac ²²⁷ available, they attempts its half-life calorimetrically. The calorimetrical measuring of half- its half-life calorimetrically. The calorimetrical measuring of half- its half-life calorimetrically. The calorimetrical measuring of the half-life is to the calorimeter by a known quantity of the radioactive iso- nated in the calorimeter by a known quantity of the radioactive iso- formula for the computation of the half-life is written down. formula for the computations was first chemically purified. ation used here for investigations was first chemically purified. ation ($Ac_2^{227}O_3$) had a weight of 2,01 \pm 0,02 milligrams. The impuri- te preparation are quantitatively mentioned. The preparation of the radioactive impurities. On the occasion of the purification of the ation is isotope, RdAc, a daughter-product of Ac ²²⁷ , was deposited on also its isotope, RdAc, a daughter-product of Ac ²²⁷ , was deposited with Th. Therefore all basic calorimetrical measurements were carried with Th. Therefore all basic calorimetrical measurements were carried on. Measurements were carried out in a double static calorimeter, which on. Measurements were carried out in a double static calorimeter, which inetrical measuring of radioactive substances. The thermal efficiency inetrical measuring of radioactive substances. The thermal efficiency experation was 23,7 milliwatts ($\pm 0,5\%$). Then determining its activity
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SIMANSKAYA, N.S.

CHRISTER CONTRACTOR CONTRACTOR OF CONTRACTOR OF

 SHBJJOT USER (PHYSICS CARD 1/2 PA - 1532 SHBJJOT USER (PHYSICS STMANSKAJA, N.S. TIFLE The Determination of the Bifurcation Ratio in the Decay Scheme 210 of Po PERIODICAL Zurn.eksp.i teor.fis, <u>51</u>, fasc.2, 174-177 (1955) Issued: 5.10.1956 At first several previous works dealing with the same subject are discussed. When this bifurcation ratio is determined the number of <i>f</i>-quanta emitted by the Po²¹⁰ source per time unit must be estimated, and absolute activity must the Po²¹⁰ source committed when measuring a-intensity with the help of occurring grave errors committed when measuring a-intensity with the help of control was determined by comparison with a Co⁶⁰ source of known intensity Cond was determined by comparison with a Co⁶⁰ source of known intensity is a standarl-β-counter with an additional 2 mm aluminium filter, out by means of a standarl-β-counter with an additional 2 mm aluminium filter, and the results of two measured as well as their absclute activities N, and N_a for the intensity of the <i>f</i>-radiation of the gauged Co⁶⁰ source and of the source network, the number n_f of the 800 keV-<i>f</i>-quanta emitted by the Po²¹⁰ per and the results of the determined. The correction coefficients yet to be intro- uct of decay can be determined. The correction coefficients yet to be intro- 	2//////				
 SUBJECT USER / PHYSICS MUTHOR SIMANSKAJA,N.S. THE Determination of the Bifurcation Ratio in the Decay Scheme TITLE The Determination of the Bifurcation Ratio in the Decay Scheme PERIODICAL Zurn.eksp.i teor.fis,<u>31</u>,fasc.2, 174-177 (1955) Issued: 5.10.1956 At first several previous works doaling with the same subject are discussed. At first several previous works doaling with the number of <i>J</i>-quanta exitted by When this bifurcation ratio is determined the number of <i>J</i>-quanta exitted by When this bifurcation ratio is determined is drawn to the frequently be measured with sufficient accuracy.Attention is drawn to the frequently be measured with sufficient accuracy.Attention for <i>J</i>-quanta emitted per ionization chambers, impulse chambers and luminescence counters. ionization sused a pure Po²¹⁰ Preparation. The number of <i>J</i>-quanta emitted per cond was determined by comparison with a Co⁰⁰ source of known intensity t cond was determined by comparison with a Co¹⁰⁰. 			CARD 1 / 2	PA - 1532	
of Po ⁻¹⁰ . PERIODICAL Zurn.eksp.i teor.fis, <u>31</u> ,fasc.2, 174-177 (1955) Issued: 5.10.1956 At first several previous works doaling with the same subject are discussed. At first several previous works doaling with the number of γ -quanta exitted by When this bifurcation ratio is determined the number of γ -quanta exitted by the Po ⁻²¹⁰ source per time unit must be estimated, and absolute activity must the Po ⁻²¹⁰ source per time unit must be estimated, and absolute activity must the Po ⁻²¹⁰ source per time unit must be estimated, and absolute activity must be measured with sufficient accuracy.Attention is drawn to the frequently be measured with sufficient accuracy.Attention is drawn to the help of occurring grave errors committed when measuring a-intensity with the help of ionization chambers, impulse chambers and luminescence counters. ionization chambers, impulse chambers and luminescence of χ -quanta emitted per The authors used a pure Po ⁻²¹⁰ preparation. The number of χ -quanta emitted per the cond was determined by comparison with a Co ⁻¹⁰⁰ source of known intensity the cond was determined by comparison with a Co ⁻¹⁰⁰ source of aluminium filter,	AUTHOR	SIMANSKAJA,	the Bifurcation Ratio	in the Decay Scheme	
PERIODICAL Juin discription of the second and second and the secon	TITLE	of Po	31. fasc. 2, 174-177 (1	955)	i.
the Po ²¹⁰ source per time unit must be estimated, and used the frequently the Po ²¹⁰ source per time unit must be estimated, and used the frequently be measured with sufficient accuracy. Attention is drawn to the frequently occurring grave errors committed when measuring a-intensity with the help of ionization chambers, impulse chambers and luminescence counters. ionization chambers, impulse chambers and luminescence counters. ionization chambers, impulse chambers and luminescence of <i>f</i> -quanta emitted per the authors used a pure Po ²¹⁰ preparation. The number of <i>f</i> -quanta emitted per the determined by comparison with a Co ⁵⁰ source of known intensity the cond was determined by comparison with a Co ⁵⁰ source of aluminium filter,	PERIODICAL	Luin 6 10, 1956	line with the same	subject are discussed.	
	the Po sub be measured occurring g ionization to authors t cond was	ource per time unit mus with sufficient accura- rave errors committed v chambers, impulse chamb s used a pure Po ²¹⁰ pre- determined by comparis	t be estimated, and icy.Attention is draw when measuring a-inte- bers and luminescence paration. The number on with a Co ^O source per second. These me	of frequently insity with the help of counters. of f-quanta emitted per of known intensity easurements were carried of 2 mm aluminium filter,	

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507/120-58-2-23/37

AUTHOR: <u>Shimanskaya, N. S.</u>
 TITLE: Tap Determination of Half-Lives of Radioactive Isotopes
 Using a Double Static Calorimeter (Opredeleniye periodov
 Using a Cal

(USSR) ABSTRACT: The calorimeter is shown in Fig.l and consists of two identical internal calorimetric cylinders A and B placed inidentical internal calorimetric T₀ is kept constant. If

side the persen 6 whose temperature of the radioactive sample is placed, for example, in A then the radioactive sample is placed, for example, in A then ofter a certain interval of time a temperature difference is established between A and B. This temperature difference is a function of the heating power of the radioactive sample is a function of the heating power of the radioactive sample only, for a given calorimeter system. The calorimeter is only, for a given calorimeter system. The calorimeter is only for a given calorimeter A, A particularly useful placel in the same cylinder A. A particularly useful placel in the same cylinder A and the heater in the other backle is placed in the cylinder A and the heater in the other

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STERN TREAMS AND A MANAGEMENT AND A MANAGEM

S0V/120-58-2-23/37 · The Determination of Half-Lives of Radioactive Isotopes Using a Double Static Calorimeter. cylinder B . The heater is adjusted so that the temperatylinder B. The heater is adjusted so that the tempera-ture difference is reduced to zero. The device was used to measure the half-life of Rn²². The half-life was found to be $T = 3.83 \pm 0.03$ days, which is in good agreement with the accepted value which is $T = 3.825 \pm 0.005$ days. There are 2 figures, no tables and 8 references, 4 of which are Exclicitly and 3 Sources English, 1 French and 3 Soviet. ASSOCIATION: Radiyevyy institut AN SSSR (Radium Institute of the Academy of Sciences of the USSR) SUBMITTED: July 2, 1956. 1. Radioisotopes--Half life 2. Half life--Measurement 3. Calorimeters--Applications Card 2/2

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	Sov/48-22-7-19/26 Seigor', 7, 6. T., Zuznetwov, B. S., Shimanskaye, N. T.,
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the states	Setermination of the Setio $1/2$ in 157 and 57^{115} and 60^{119} Setermination of the Setio $1/2$ in 197^{119} and 57^{119} . To Set: stion 6. the Transmutation Shergies of 57^{119} \longrightarrow To and 17^{119} \longrightarrow 10^{119} (Opredeleniye of nonheniya $1/2$ dive 157^{119} \longrightarrow 165^{110} isotsenka chergii perekhodov 57^{159} \longrightarrow Th ¹⁵⁹ 165^{119} \longrightarrow 165^{119}
	$105 \longrightarrow 10^{10}$
6 8 ML:	i frankademii nauk SSSR, Seriya fizicheskay, 1959, Tol. 20, Nr 7, pp. 850-860 (USSR)
, <u>a</u> ∾3.,C ⊄ :	The decay energy \mathcal{E} of radioactive isotopes, which are defined jected to an electron capture can be determined by 5 different methods. They are described. From the evidence given it is methods, that the 5th method, that utilizing the ratio L/K concluded, that the 5th method, that utilizing the ratio L/K is very convenient in the determination of small transmutation is very convenient in the determination of small transmutation energies (<200 keV) in isotopes with a relatively simple energies (shown which do not exhibit a considerable converting
e rd 1/4	decay scheme, mution. The application of this method theory cascade y-redictions fill inherent in the modern theory ed by the imperfections still inherent in the modern theory

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301/43-22-7-1)/26 Leterministion of the Ratio L/K in \mathbb{T}^{159} and \mathbb{T}^{165} onlian Estimation of the Dranssutation inergies of $\mathbb{T}^{159} \longrightarrow \mathbb{T}^{159}$ and $\mathbb{T}^{165} \longrightarrow \mathbb{H}^{165}$ of K-capture and by the incomplete knowledge of the qualitative an quantitative rules governing the processes of the rearrangement of the cleatron shell of the atom. L/K was detarrangement of the cleatron shell of the atom. L/K was detarrangement of two isotopes of rare earths, Dy159 and Er165, both having a neutron deficit. Proceeding from the results the transmutation energies of the processes Dy 157 -> Tr 157 and Er¹⁶⁵ --- Ho¹⁶⁵ were estimated. A y-spectrometer combined with a proportional counter was used for measuring the energies and the intensities of an X-ray K- and L-radiation. The proportional counter (Ref 20) permitted to measure the γ - and X-ray radiation of small energies, which is quite impossible with other with other with a cylindrical aluminum cathode and its circuit diagram is leccribed. The recording power of the counter for 7- and X-ray-todiction of varying energy is computed according to the known absorption coefficients for this radiation in argon and heryllium (Ref 22), taking into account the geometry of the experimental arrangement. The electronic circuit diagram Card 2/4

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SOV/48-22-7-1)/26 Determination of the Natio L/R in Dy^{159} and Er^{165} and an Estimation of the Transmutation Energies of $Dy^{159} \rightarrow Tb^{159}$ and $Er^{165} \rightarrow Ho^{165}$ 9 . and the calibration of the device is described. The Dy¹⁵⁹ source was obtained from a tantalum target, which was irradiated in the synchrocyclotron of the "United Institute of Muchar Research" with 660 MeV protons. The ratio L/K was computed according to formula (3). It is shown that the transition $Dy^{159} \longrightarrow Tb^{159}$ must be classified as being superforbidden. Marshak's formula was used, giving an energy value of 79^{+10}_{-5} keV for this transition. The lowest level of Tb¹⁵⁹ at 57 keV is apparently not excited in the decay of y^{159} . In estimation of the quantity ft on the basic of the decay onergy of 79 keV and a half-life of 156 days furnishes a value for 1g ft of about 6,2. According to the classification of King (Ref 32) this value.agrees with the essumption, that this transmutation is a superforbidden one. The Ξr^{165} -sources were also obtained from tantalum irradiat-ed with fast protons ($\mathcal{E}_0 = 660 \text{ MeV}$). The X-ray radiation Card 3/4

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 $\begin{array}{c} \mathrm{S07/48-22-7-19/76}\\ \text{Determination of the Ratio L/K in Dy^{159} and Er^{165} and an Estimation of the Transmutation Unergies of Dy^{159} \longrightarrow \mathrm{Vb}^{159}$ and Er^{165} \longrightarrow Ho^{165} of a series of tantalum targets irradiated for different periods was becaused. The ratio $\mathrm{I_L/I_K}$ (for the intensities of these rediations) was equal to 0,40. From this value for L/K a result of 1,2 \pm 0,4 was obtained. Using Marghak's formula and the experimentally fount value of L/K (Ur^{165}) as 2⁺¹⁰ keV were found for the transmutation energy of the process $\mathrm{Tr}^{165} \longrightarrow \mathrm{Ho}^{165}$. The value of 1g ft was 3,1 with a half-life of 10,5 hours, which is in agreement with the permitted character of the transmutation. There are 9 figures, 1 table, and 35 references, 3 of which are Soviet. ASSOCIATION: Radium Institute iteni V. G. Khlopin, AS USSR) Card 4/4

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*	SOV/56-34 -3-39/55	- 8
AJTHORS:	Gorshkov, G. V., Gritchenko, Z. G., Shimans.aya, N. S.	
TITLE:	The Calorimetric Determination of the Half-Life of Ra ²²⁶ (Kalorimetricheskoye opredeleniye perioda poluraspada Ra ²²⁶)	
PERIODICAL:	Zhurnal Eksperimental'noy i Teoreticheskoy Fiziki, 1958, Vol. 34, Nr 3, pp. 756 - 757 (USSR)	
ABSTRACT: Card 1/3	First, brief reference is made to some previous works deal- ing with the same subject. The authors of the present re- port carried out careful calorimetric measurements on 3 equilibrated radium preparations which were liberated from possible contaminations by means of additional crystalliza- tion. The purity of these preparations was controlled by means of the spectroscopic method. The results of the imme- diate weighing of the radium preparations prior to their sealing, their radium-content and the results of the calori- metric measurements carried out by means of a double static calorimeter, are contained in a table. The last column of the table contains the values found here for $Q_{t+\beta}/p$ - the	

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The Calorimetric Determination of the Half-Life of Ra 226 . thermal effect of the α - and β -radiation of 1 g radium. Calculating these values, the absorption of the γ -rays in the preparation itself (self-absorption), in the protective container, in the glass of the ampules and within the walls of the calorimetric cylinder, were taken into consideration. Also the increase of the thermal effect due to the accumulation of Po^{210} and RaE in the preparations was taken into account. $\boldsymbol{\epsilon}$ (the energy liberated in the calorimeter in a process of decay) was calculated on the basis of the last experimental data on the ∞ - and β -spectra of the elements of the radium-series for an equilibrated preparation of Ra^{226} . This energy amounted to 25,335 MeV (\pm 0.3 %). Utilizing this value, the authors found the value T = 1577+9years for the half life of Ra^{226} . Hence results the value $z = 3.71 \pm 0.02.10^{10}$ decay-processes/sec.g. for the specific activity. Firther measurements of these important values z and T for Ra^{226} with the methods discussed here and also by other methods, would be desirable. There are 1 table and 10 references, 4 of which are Soviet. Card 2/3

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"APPROVED FOR RELEASE: 08/23/200 CIA-RDP86-00513R001549510009-7 SOV/56-34-3-39/55 The Calorimetric Determination of the Half-Life of Ra²²⁶ ASSOCIATION: Radiyevyy institut Akademii nauk SSSR (Radium Institute AS USSR) SUEMITTED: December 6, 1957

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sov/115-59-3-24/29 Gorshkov, G.V., Karavayev, F.M., and Shimanskaya, 21(8) AUTIIORS: N.S. The Determination of the Radium Content in Radium Compounds (Ob opredelenii soderzhaniya radiya v TITLE: raliyevykh preparatakh) Izmeritel'naya tekhnika, 1959, Nr 3, pp 52-53 (USSR) PERIODICAL: The radium content of radium compounds is mainly determined by the ionization method, or more exactly, its gamma equivalent is determined. The ioniza-ABSTRACT: tion effect of the radiation of the compound under investigation is compared to that of a standard with a known radium content. At VNIIM, two state standards, X and XI, are used, whose radium content was set equal (for 1957) to 29.37 and 14.27 mg radium elements. The self-absorption of the gamma radiation within the radiation source itself is not considered sufficiently. Although lead filters are used, which are 2 cm thick at VNIIM, whereby the soft gamma radiation is eliminated, the error can attain a consider-Card 1/3

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The Determination of the Radium Content in Radium Compounds

able magnitude, if the differences of self-absorpticn are not taken into consideration. The authors determined the accuracy of contemporary ionization methods used for determining the radium content. For this purpose, three pure radium compounds were available which were to be used for the calorimetric determination of the radium half decay period (Ra²²⁶). The results of these investigations and measurement results of VNIIM and the Radiyevyy institut AN SSSR -RIAN- (Radium Institute AS USSR) are shown in one table. The calculations performed by the authors show that the difference of the self-absorption of the gamma radiation of radium in 15 mg RaCl, and 150 mg RaBr, is of a considerable magnitude. The effective self-absorption in standard XI was found to be 0.9% while it was 1.7% in 150 mg RaBr, whereby the difference was 0.8%. The authors recommend to establish new standards in the USSR with a radium content of 1, 5, 10, 25, 100, 200, 500 mg, whereby the error

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507/89-6-4-14/27 21(8) Gorshkov, G. V., Shimanskaya, N. S. AUT HORS : Total Energy of the Radioactive Radiation of a Radium Preparation (Ra²²⁶) in Equilibrium (Polnaya energiya radio-TITLE: aktivnogo izlucheniya ravnovesnogo preparata radiya (Ra²²⁶) Atomnaya energiya, 1959, Vol 6, Nr 4, pp 474-475 (USSR) PERIODICAL: In 1935 I. Zlotovskiy calorimetrically measured the total energy of all radioactive radiations radiated from a radium ABSTRACT: preparation in equilibrium. This value was now checked with the help of 3 sources the exact radium content (Ref 4) of which was known. This measurement was carried out with the static γ -calorimeter (Ref 5) the tungsten walls of which absorbed ~93% of the y-radiation of Ra-Ra(B+C). For q (total energy) the value 138.9 ± 0.7 cal/h.1g Ra was measured. It is by 0.7% lower than that obtained by Zlotovskiy. This lower value agrees well with expectations. Individual data, from which q was calculated, are shown by a table. Ye. K. Smirnova produced the radium preparations. Yu. S. Martynov took part in the measurements. There are 1 table and 6 references, 4 of which are Soviet. Card 1/2 N7959 1-18

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	8 7959 S/115/60/000/012/013/018 B019/B056
Legend to Table 1: 1) Day of measurement. 2) sealing and measuring. 3) Composition of the various days of measurement. 3a) Ra, millicu- of radium. 3c) Millicuries. 3d) Ratio betwee $T_t = (\lambda N_t) RdTh/(\lambda N_t) MsTh$, where λ is the de number. 3e) Millicuries. 3f) Milligram-equi- γ -equivalent, milligram-equivalent of radiu $4c$) J/J_0 calc. 4d) J/J_0 exp. 5) Relative ne 5b) A_n exp. 5c) A_n/A_n calo. 5d) A_n/A_n exp. 6b) Q_{exp} . Legend to Fig. 3: The curves denoted by 1, relative neutron yield and the relative rad different original compositions of the Ra+ 60% Ra + 40% MsTh; 65% Ra + 35% MsTh; 70% a) = years.	en the number of decaying atoms: ecay constant and N _t the atom valent of radium. 4) Radium- im. 4a) J _{calc} . 4b) J _{exp} . eutron number. 5a) A _n calc. 6) Q value, cal/h; 6a) Q _{calc} . 2, 3, 4, and 5 stand for the dium-y-equivalent for five
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Decay of Dy 159

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relative to gamma decay of Gd ¹⁵⁹. Apart from the $\gamma58$ line, a weak line with an energy of 350 kev was also observed (Ref. 2). The intensity of this line amounts to 2.10^{-5} quanta per decay event. Shorter wave lines in the energy range up to 2 Mev were no more observed, or at least not any such with an intensity exceeding 10^{-4} to 10^{-5} quanta per decay event. Simultaneous measurements of the two Dy¹⁵⁹ sources in the 4 π scintillation counter and in the 4 π gas counter gave the following ratios between the intensities of the LX and KX radiation and the intensities of the corresponding LX - LX and KX - KX coincidences:

 $\frac{I_{KX}}{I_{KX-KX}} = 6.56^{\pm}0.18, \quad \frac{I_{LX}}{I_{LX-LX}} = 48.1^{\pm}4.1, \quad \frac{I_{KX-KX}}{I_{LX-LX}} = 37.1^{\pm}5.8, \quad \frac{I_{LX}}{I_{KX}} = 0.21^{\pm}0.01.$

One may calculate therefrom the ratio L_1/K_1 for the transition to the first excited 58-kev level of Tb¹⁵⁹ and the amount \mathcal{X} of the bifurcation. If the value $\overline{\omega} = 0.18\pm0.02$ is assumed for the L fluorescence yield of Tb, one obtains $L_1/K_1 = 0.58$ and $\mathcal{X} = 0.32\pm0.08$. The article under consideration is the reproduction of a lecture delivered at the 10th All-Union Conference on Nuclear Spectroscopy, which took place in Moscow

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Decay of Dy ¹⁵⁹	89253 S/048/61/025/001/019/031 B029/B060	
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S/048/62/026/002/008/032 B101/B102 Biryukov, Ye. I., and Shimanskaya, N. S. AUTHORS : K/β^+ ratio for Pr^{140} TITLE: Akademiya nauk SSSR. Izvestiya. Seriya fizicheskaya, PERIODICAL: v. 26, no. 2, 1962, 215 - 216 TEXT: A direct measurement was made of K/β^+ of a Pr^{140} preparation obtained from the reaction $Pr^{141}(n,2n)Pr^{140}$. Pr_{203} was applied in a thin layer (1.3 mg/cm^2) to a polyethylene film, and bombarded with 14-MeV neutrons (intensity 10^{10} neutrons cm⁻² sec⁻¹). Check tests showed that the emission from polyethylene and oxygen was negligibly small. K/β^+ was determined with a 4π scintillation gamma spectrometer with 40.40 cm CsI(T1). For measuring the annihilation radiation spectrum, the source was shielded with a lead filter (1000 $mg \cdot cm^{-2}$) and placed into a channel (6 mm in diameter) bored through the crystal axis. The K-radiation Card 1/8

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 K/β^+ ratio for Pr^{140}

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spectrum was then measured without a filter. Two AM -100 (AI-100) multichannel analyzers were used for the measurement. From the two spectra, the mean value of K/β^+ was found to be 0.75 ± 0.03 , which is in good agreement with the Pr^{140} decay scheme published earlier (Izv. AN SSSR, Ser. fiz., 24, 1135 (1960)). By extrapolating K/β^+ for allowed transitions one obtains 2480 kev as the limit of the Pr^{140} positron spectrum. L. Zyryanova and K. Gromov are mentioned. There are 2 figures and 5 references: 3 Soviet and 2 non-Soviet. The two references to English-J., Martin, D., Phys. Rev., <u>85</u>, 146 (1952); Handley, T., Olson, E., Phys. Rev., <u>96</u>, 1003 (1954).

ASSOCIATION: Radiyevyy institut im. V. G. Khlopina Akademii nauk SSSR (Radium Institute imeni V. G. Khlopin of the Academy of Sciences USSR)

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"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001549510009-7 3 BIRYUKOV, Ye.I.; SHIMANSKAYA, N.S. Decay of Md¹⁴¹. Izv. AN SSSR. Ser. fiz. 27 no.11:1402-1407 (MIRA 16:11) N '63.



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ACCESSION NH: AP4014696 AUTHORS: Sirota, N. N.; Shimanskaya, V. P. TITLE: Lattice constant of zinc sulfide-cadmium sulfide solid solution films SOURCE: AN BSSR. Dokiady*, v. 8, no. 1, 24-25

TOPIC TAGS: zinc sulfide, cadmium sulfide, ZnS CdS solid solution, lattice constant, ZnS CdS phase composition, ZnS CdS structure, vaporized coating, x ray apparatus URS 501

ABSTRACT: Films of the binary system 2nS-CdS have been subjected to x-ray analysis in order to determine its phase composition, its structure, and its crystal lattice constant. Experiments were conducted directly after film deposition and also after a heat treatment of films which were produced by sublimating a pressed mixture of ZnS and CdS of a definite composition onto glass and quartz plates. The process was carried out in a vacuum of no less than 10^{-4} mm Hg. For CdS the temperature of the plates was held at 90C and for ZnS at 200C. After the deposition the samples were held in vacuum at 300C. Microscope inspection and x-ray analysis proved that the films were either polycrystalline or monocrystalline, the latter ranging from 0.5 to 1.6μ in thickness. They were monophase in type and

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Investigation of the critical state of substances by Toepler's
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near the critical point. Ukr.fiz.zhur. 4 no.6:769-788 H-D '59.
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ACCESSION NR: AR4014689 S/0271/64/000/001/B028/B029 SOURCE: RZh. Avtomatika, telemekhanika i vy*chislitel'naya tekhnika, 1964, no. 1. Abs. 18211 AUTHORS: Meshkov, N. V.; Mesterov, P. V.; Smirnov, V. I.1 and Shimanskiy, A. M. TITLE: Memory unit for a multidimensional analyzer with 16,000 channels CITED SOURCE: Tr. 5-y Nauchno-tekhn, konferentsii po yadern, radio-elektronike. T. 2., Ch. 2. M., Gosatomizdat, 1963, 62-71 TOPIC TAGS: multidimensional analyzer, memory unit, ferrite core memory, high capacity memory, multidimensional analysis, information sorter TRANSLATION: Multidimensional analysis requires a memory of great capacity. A 16000-cell ferrite tore memory has been built which can be used directly as a memory for a multidimensional analyzer or as a sorter of information from a magnetic tape. In many cases the high capacity makes it possible to rewrite all the information from one or several tapes in one run and obtain graphic picture of the spectrum. The memory is intended for use with various units. A number o. auxiliary functions added to the addressing system and the arithmetic block of the 1/2 Card .

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"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001549510009-7 ACCESSION NR: AR4014689 memory converts the system to a specialized computer capable of highly simplified operations in reducing information. The memory capacity is 128 X 128 sixteendigit bits. The dead time of the memory is 200 microsec. The ferrite cores used in the memory are type K-260, measuring 2 X 1 and 3 X 1 mm. Selection half-current is 240 ms. The circuit includes 200 type 6N3P tubes, 100 6N6P tubes, 1200 semiconductor diodes, and 200 transistors. Orig. art. has 4 figs. and 4 refs. O. B. ł SUB CODE: CP ENCL: 00 DATE ACQ: 19Feb64 Card 2/2

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ACCESSION NR: AR4022434 S/0058/64/000/001/A028/A028 SOURCE: RZh. Fizika, Abs. 1A268 AUTHORS: Meshkov, N. V.; Nesterov, P. V.; Smirnov, V. I.; Shimanskiy, A. M. TITLE: Memory unit for multidimensional 16000 channel analyzer CITED SOURCE: Tr. 5-y Nauchno-tekhn. konferentsii po yadern. radioelektronike. T. 2, Ch. 2. M., Gosatomizdat, 1963, 62-71 TOPIC TAGS: memory unit, ferrite core memory, multidimensional analyzer, magnetic tape data reduction, ferrite core matrix, reading amplifier, magnetic commutator, address current TRANSLATION: A 16,000 address ferrite-core memory has been developed. This memory can be used in a multidimensional analyzer or serve as a sorting block for the reduction of information from a magnetic Cqrd/2

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CCESSION NR: AR4022434 ape. The memory block is made in the form of a 128 x 128 x 16 atrix of K-260 ferrite cores measuring 2 x 1.3 x 1 mm. To increase he reliability, the matrix, the reading amplifiers, and the mag- etic commutator for the address currents are placed in a container here a (35 ± 1) C temperature is maintained. The memory capacity s 128 x 128 16-digit binary numbers. The dead time is 200 micro- econds. The circuit contains 300 vacuum tubes and approximately 00 transistors. The operating principle is analyzed and block iagrams of the main units of the memory are presented. Yu. Semenov. ATE ACQ: 03Mar64 SUB CODE: CP, SD ENCL: 00							
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