CIA-RDP86-00513R001857810005-6

S/194/61/000/006/050/077 D201/D302

AUTHOR: _____Tyutin, A.A.

TITLE: Radioelectronic method of laminar and volume radiography

PERIODICAL: Referativnyy zhurnal. Avtomatika i radioelektronika, no. 6, 1961, 6, abstract 6 E36 (Elektronika v meditsine, M.-L., Gosenergoizdat, 1960, 361-376)

TEXT: A method is considered of interpreting a radiogram obtained from X-raying a certain object by plane X-ray in a certain prescribed manner. The block diagram of a computer arrangement for this purpose is suggested. The X-ray photograph is scanned by a light beam from a low after-glow CRT. The receiver is a photoelectron multiplier. The output of the latter is applied to the input of an analogue computer with an integrator as its basic element. New diagnostic possibilities of this method are considered. [Abstracter's note: Complete translation]

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THE COURSE OF CALLS STRAN 31981 s/142/61/004/004/002/018 E192/E382 Determination of the X is the column matrix of the dependent variables (unknown vector) and is the square matrix of the order n of the W generalized parameters of the network. If m independent coordinates are separated from n independent coordinates and n-m components of the drive vector Q are equal to zero, the matrix equation of the system is in the following form: (21) $Q^{0} = W^{0} X_{T}$ is a new drive vector identically equal to Q , since where Q٩ only m of the n components of the vector Q are not equal to zero, is an m-dimensional vector whose components are to be xι determined, is the normalized matrix which can be determined from W⁰ the formula: Card 2/4

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31981 S/142/61/004/002/018 E192/E382 secondly, the parameters of the equivalent quadripole are evaluated from the normalized matrix. There are 2 figures and 3 Soviet references. ASSOCIATION: Kafedra radioperedayushchikh ustroystv Kiyevskogo ordena Lenina politekhnicheskogo instituta (Department of Radio-transmitting Equipment of Kiyev Order of Lenin Polytechnical Institute) SUBMITTED: July 1, 1960	Determination of the E192/E382 secondly, the parameters of the equivalent quadripe evaluated from the normalized matrix. There are 2 figures and 3 Soviet references. ASSOCIATION: Kafedra radioperedayushchikh ustroys ordena Lenina politekhnicheskogo in (Department of Radio-transmitting E Kiyev Order of Lenin Polytechnical	ripole are roystv Kiye vs ko instituta g Equipment of		
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THE REPORT OF THE REPORT OF THE REPORT TYUTIN, A.A. Determination and use of secondary parameters in the method of subsidiary circuits. Mat. mod. i elek. tsepi no.1:226-237 (MIRA 16:11) 163.

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REMARKS REPORT (1 REALD AND NUSE IN A REPORT PROPERTY AND A DESCRIPTION OF A DESCRIPR **动用的名称**的 [35] TURKIN, A.N., insh.; TYUTIN, Yo.V., inzh. Operation of feed pumps with superhigh pressures and hydraulic clutches. Elek. sta. 33 no.7:21-27 J1 '62. (MIRA 15:8) (Electric power plants--Equipment and supplies) (Pumping machinery, Electric)

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Tyulin, A.O. 821:36 \$/102/60/000/02/03/005 16.6800 c_{111}/c_{222} AUTHOR: Tyutin, A.O. 68 Computing Amplifier for a Specialized Integrator With Periodization TITLE: of the Solution PERIODICAL: Avtomatika, 1960, No., pp.50-61 TEXT: For the experiments carried out according to the method of Tetel'baum, S.I., Corresponding Member of the Academy of Sciences Ukr.SSR (Ref. 1) the author uses a wide-band computing amplifier described by D.M. Mac-Kay (Ref.2) and P. Davis (Ref. 3). In the present paper the author gives a theoretical investigation of the device. The author's investigation method bases on the consideration of the transient responses as well as on the consideration of the especial problems of the device, e.g. performance of mathematical operations at video signals. The method of V.S. Sigorskiy (Ref. 4) is used and recommended to be especially effective. The author proposes a scheme for the calculation of the device for a prescribed linear range of variations of the output voltage and the boundary frequency of the working range. Card 1/2· 注意之意:] 《]] [] []] []] [] []] [] []] [] []] [] []] [] []] [] []] [] [] []] [] []] [] []] [] [] []] [] [] []] [] [] []] [] [] []] [] [] []] [] []] [] [] [] []] [] [] [] []] [] [] [] []] [] 的影子

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Computing Amplifier for a Specialized Integrator With Periodization of the Solution	82436 S/102/60/000/02/03/005 C111/C222
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Some preliminary findings on the opening of the gasified span of experimental gas producer No. 1 at the Lisichansk "Podzemgaz" Station. Podzem.gat.ugl. no.4:11-15 '57. (MIRA 11:1)

1.Vsesoyuznyy nauchno-issledovatel'skiy institut Podzemgaz. (Lisichansk--Coal gasification, Underground)

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	Changes in the mechanical strength of coal along stopes as con- nected with gas drainage from seams. Ugol' 34 no.4:48-52 Ap '59. (MIRA 12:7) (CoalTesting) (Mine gases)



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AUTHOR:	Tyutin, F. G. (Moscow)	SOV/24-58-8-30/37
TITLE:	Variability of the Mechanical on the Time they were Allowed Conditions of Free Gas Emissi mekhanicheskoy prochnosti ugl vremeni degazatsii ikh pri sv	, to Become Degassed under on (Izmenyayemost' .ev v zavisimosti ot
PERIODICAL	L: Izvestiya Akademii Nauk SSS Nauk, 1958, Nr 8, pp 147-149	SR, Otdeleniye Tekhnicheskikh (USSR)
ABSTRACT:	the phenomenon has not been a The formulae deduced in the w (Ref 6) and that of A. A. Nik account for the fact that in outburst of gas and coal woul it is necessary to take into that characterize the variable on its degasification. Conse (Refs 6 and 7) could not suff natural weological and indust	ince, the quantitative side of sufficiently investigated. Forks of S. A. Khristianovich some instances the sudden and not occur at all and that consideration the parameters thity of the strength of coal equently laboratory experiments fice as they did not reflect trial pit-work conditions.
Card 1/3	In this paper the author pres	sents the results of experiment

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SOV/24-58-8-30/37 Variability of the Mechanical Strength of Coals Depending on the Time they were Allowed to Become Degassed under Conditions of Free Gas Emission

obtained under natural conditions in shafts and conducted according to the method proposed by M.M. Protodyakonov (Ref 4). In Tables 1 and 2 are given the names of shafts and the aggregates of the seams worked. The specimens of coal taken directly from the ccal face are termed "fresh", the others, taken after a lapse of time, differing in each particular instance are termed "degasified". Mathematical relations are derived for calculating the mechanical strength coefficients and graphs are included expressing the strength variation as a function of the degassing time. Eq.(6), p.149, expresses the relation between the strength coefficient of coal, the degassing time and the gas saturation. An increase in the strength of the coal with the progress of degassing is due to a weakening of the wedging effect of the molecules of the free and of the adsorbed gases on the walls of the fine cracks in the coal. With increasing gas removal the molecular forces of cohesion between the walls of the micro-cracks become stronger

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SOV/24-58-8-30/37 Variability of the Mechanical Strength of Coals Depending on the Time they were Allowed to Become Degassed under Conditions of Free Gas Emission and this may lead to the closing up of these cracks. There are 5 figures, 2 tables and 8 references, all of which are Soviet. ASSOCIATION: Institut gornogo dela Akademii nauk SSSR (Institute of Mining, Ac.Sc., USSR) SUBMITTED: July 27, 1957 1. Coal---Mechanical properties 2. Coal--Degasification 3. Coal--Sampling 4. Coal--Test_regults. 5. Mathematics Card 3/3

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Method of calculating pressure hydraulic coal conveying recommended by the All-Union Scientific Research and Design and Construction Institute for Hydraulic Coal Mining. Trudy VNIIGidrouglia no.4:66-72 '64. (MIRA 18:3)

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ALATORTSEV, S.A., prof., doktor tekhn.nauk; ANDREYEV, A.V., kand.tekhn. nouk; ANCHAROV, I.L., inzh.; BALINSKIY, S.I., inzh.; BELOUSOV, V.G., inzh.; VINNITSKIY, K.Ye., kund.tekhn.neuk; VLASOV, V.M., inzh.; VORONTSOV, N.P., kand.tekhu.nauk; GIPSMAN, M.K., inzh.; GLUZMAN, I.S., kend.tekhn.nauk; GJR YEV, S.V., kand.tekhn.nauk [deceased]; DEMIN, A.M., kand.tekhn.nauk; YEGURNOV, G.P., kand. tekhn.nauk; YEFIMOV, I.P., inzh.; ZHUKOV, L.I., kand.tekhn. nauk; ZEL'TSER, N.M., inzh.; KOSACHEV, M.N., kand.tekhn.nauk; KOTOV, A.F., inzh.; KUDINOV, G.F., inzh.; LAPOVENKO, N.A., kand. tekhn.nauk; MAZUROK, S.F., inzh.; MEL'NIKOV, N.V.; MUDRIK, N.G., inzh.; NIKONOV, G.P., kand.tekhn.nauk; ORLOV, Ye.I., inzh.; POTAPOV, M.G., kand.tekhn.nauk; PRISEDSKIY, G.V., inzh.; RZHEVSKIY, V.V., prof., doktor tekhn.nauk; RYAKHIN, V.A., kand. tekhn.nauk; SIMKIN, B.A., kand.tekhn.nauk; SITNIKOV, I.Ye., inzh.; SOROKIN, V.I., inzh.; STASYUK, V.N., kend.tekhn.nauk; STAKHEVICH, Ye.B., inzh.; SUSHCHENKO, A.A., inzh.; TYUTIN, I.F., inzh.; TYMOVSKIY, L.G., inzh.; FISENKO, G.L., kand. tekhn. nauk; FURMANOV, B.M., inzh.; SHATAYEV, M.G., inzh.; SHESHKO, Ye.F., prof., doktor tekhn.nauk; TERPIGOREV, A.M., glavnyy red. [deceased]; (Continued on next card)

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KIT, I.K., zamestitel' glavnogo red.; SHESHKO, Ye.F., zamestitel' otv.red.; BUGOSLAVSKIY, Yu.K., red.; BYKHOVSKAYA, S.N., red.; DIONIS'YEV, A.I., kand.tekhn.nauk, red.; KOZIN, Yu.V., red.; SOKOLOVSKIY, M.M., red.; YASTREBOV, A.I., red.; DEMIDYUK, G.P., kand.tekhn.nauk, red.; KRIVSKIY, M.N., kand.tekhn.nauk, red.; LYUBIMOV, B.N., inzh., red.; MOLOKANOV, P.L., inzh., red.; REISH, A.K., inzh., red.; RODIONOV, L.Ye., kand.tekhn.nauk, red.; SLA-VUTSKIY, S.O., inzh., red.; TRAKHMAN, A.I., inzh., red.; TRYMOV-SKIY, L.G., inzh., red.; FIDELEV, A.S., doktor tekhn.nauk, red.; SHUKHOV, A.N., kand.tekhn.nauk, red.; Continued on next card)

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ACCESSION NR: AP5000339	D(t)/SSD/AFWL/ASI(a)-5 S/0056/64/047/005/1826/1828
THORS: <u>Gaponov, Yu. V.; Tyuti</u> TITLE: Inelastic scattering of	n V. neutrinos by deuterons
30URCE: Zhurnal eksperimental'n no. 5, 1964, 1826-1828	oy i teoreticheskoy fiziki, v. 47,
differential cross section	scattering, inelastic scattering,
ents dealing with the possible rent in the universal weak-inter with a Hamiltonian to estimate t	eal feasibility of low-energy experi- existence of a neutral neutrino cur- raction Hamiltonian, the authors use theoretically the cross section for

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16506 -65 ACCESSION NR: AP5000339 approximations of the allowed and forbidden transitions, resulting from the expansion of the exponential in the matrix element, are the same as for β decay. Plots of the differential cross section as a function of the relative energy of motion and tables of the cross sections as functions of the total reaction energy are presented. "The authors thank I. S. Shapiro for interest and for valuable advice. One of the authors (Yu. G.) is grateful to L. A. Mikaelyan, V. G. Vaks, and A. I. Larkin of the IAE im. I. V. Kurchatova for interesting discussions." Orig. art. has: 2 figures, 3 formulas, and 2 tables. ASSOCIATION: None 00 ENCL: SUEMITTED: 30Apr64 003 OTHER: NR REF SOV: 006 SUB CODE: NP 2/2Card 調整さ
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21 June 995-17 INDUCED RADIATIVE PROCESSES IN CLASSICAL AND QUANTUM THEORIES (USSR) Sobel'man, I. I., and I. V. Tyutin. Uspekhi fizicheskikh nauk, v. 79, no. 4, S/053/63/079/004/002/004 Apr 1963, 595-616. It is shown that although induced emission is not a pure quantum effect, in the quasi-classical limit the transition from quantum formulas is quite unique. In particular, there is no classical analogue to the quantities determining the induced emission power and the absorption power. The authors point out that their analysis of induced emission in terms of classical theory is warranted by the recent interest in masers and because a general classical theory for the interaction of radiation of nonlinear oscillators with the field was never fully developed. The article deals with a systematic development of the classical theory of radiative processes, compares in detail the classical theory with the quantum theory, and shows that under certain conditions classical [CS] systems can amplify incident radiation. Card 1/1

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1.	TYUTIN, M. G.	
2.	USSR (600)	7
4.	Sweet Potatoes	•
7.	The sweet potato, a forgotten crop. Sad i og. no. 11, 1952.	
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9.	Monthly List of Russian Accessions, Library of Congress, <u>March</u> 1953, Uncl.	



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UTHOI	S: Arkhangel'skaya, V. A.; Nikitinskaya, T. I.; Tyutin, N	<u>1. S.</u>	
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ITLE: rysta	Effect of oxygen on the lonic conductivity of fluorite I	3	
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uctiv BSTR/	TAGS: calcium fluoride, electric conductivity, impurity co ity, crystal lattice vacancy, temperature dependence CT: The authors report an observed change in <u>ionic conductiv</u>	vity	
f Cal rocec ne of 961).	crystals when 0 ² ions are introduced in the lattice. The ure and research apparatus were described in earlier papers the authors (Nikitinskaya, FTT v. 1, 835, 1959; v. 3, 3224 The investigations were made at temperatures 350 650K, on where the conductivity of CaF ₂ is sensitive to structure.	by	
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1, 11,122-66 ACC NR: AP6000889 The temperature dependence of the ionic conductivity of CaF with or without rare-earth impurities (Er³⁺, Ho³⁺, Sm³⁺, and Dy²⁺), in concentrations up to 1.0 mol. ε , can be described by the relation $\sigma = \sigma_0$ exp (-u/kT), where u = 23 kcal/mole = 1.0 ev for all the investigated samples. The fact that impurities do not affect the variation of the conductivity indicates that in the temperature region 350 -- 600K the conductivity is governed by anion vacancies, and not by interstitial anions. These vacancies are in all probability produced by the 0^{2-} introduced in the crystal. The presently available data are insufficient to explain all the results. Orig. art. has: 2 figures and 1 formula. SUBM DATE: 10Ju165/ ORIG REF: 005/ OTH REF: 005 SUB CODE: 20/

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"A Few Observations on the GAM Tyutin, L. Reznik. Engineers;	2-51 Automobile;" N. Żip	
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Therebis no doubt that the GA than those manufactured in fo similar purposes. Truck oper However, individual parts nee formance of axles, particular must be improved.	reign countries for ates well as a unit. 1 improvement. Per-	
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C. B. Frank



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Use of a petrographic method for the study of mineral particles in animal tissues and urine after the introduction of quartzcontaining dust. Arkh.pat. 23 no.4:54-61 '61. (MIRA 14:6)

1. Iz kafedry patofiziologii (zav. - prof. Ya.G. Uzhanskiy)
Sverdlovskogo meditsinskogo instituta (dir. - prof. A.F.
Zverev) i Berezovskoy opytnoy stantsii po bor'be s silikozom
(dir. N.N. Liberman [deceased]) Instituta gigiyeny truda i
profzabolevaniy AMN SSSR (dir. - deystvitel'nyy chlen AMN SSSR
prof. A.A. Letavet).

(FOREIGN BODIES) (LUNGS .- DUST DISEASES)

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ord peaked replaced restrictions in restriction and the restriction of the restriction of the restriction of the TYUTIN, P.I., petrograf Using the petrographic method to determine the miner composition, dispersion, and specific surface of dusts. Sbor. rab. po. silik. no.3: 147-153 161. (MIRA 15:10) 1. Berezovskaya opytanaya nauchno-issledovatel'skaya stantsiya po bor'be s siltkozom Instituta gigiyeny truda i professional'nykh zabolevaniy AMN SSSR. (Mine dusts--Analysis) 11 用的计算机和计算机是

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TYUTIN, P.I., petrcgraf Composition of mine dusts and ash from the lungs of silicotuberculosis patients. Sbor. rab. po silika no.3:129-146 161. 1. Berezovskaya nauchno-issledovatel'skaya stantsiya po bor'be s silikozom instituta gigiyeny truda i professional.'nykh zabolevaniya AMN SSSR.

(TUBERCULOSIS) (LUNGS-DUST DISEASES) (MINE DUSTS-ANALYSIS)

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TERENT'YEV, M.L.; OSAD'KO, M.P.; BRAGINSKIY, B.I.; SLOBODIN, V.M.; FISHMAN, Z.A.; LEVIN, I.Ye.; TSYNKOV, M.Yu.; BADIR'YAN, G.G.; <u>TYHTH, Y.A.;</u> ABRAMOV, V.A.; FRAYNR,S.V.; KOBCHIKOVA, I.A.; KARMAUKHOVA, Ye.I.; OBOLENSKIY, K.P.; IL'IN,S.A.; GAVRILOV, V.I.; FREYDMAN, S.M.; KALASHNIKOVA, V.S., redaktor; IAPIDUS, M.A., redaktor; RAKITINA, Ye.D., redaktor; FEDOTOVA, A.F., tekhnicheskiy redaktor

[Manual for students of collective farm economy] V pomoshch[†] izuchaiushchim ekonomiku kolkhozov. Moskva, Gos. izd-vo selkhoz. lit-ry, 1956. 423 p. (MIRA 10:1) (Collective farms)

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BADAR'YAN, G.G.; TYUTIN, V.A.; CHEREAUSHKIN, S.D.; ZUZIK, D.T.; KHODASEVICH, B.G.; FRAYER, S.V.; GUSAROV, Ye.I.; KAZANSKIY, KHODASEVICH, B.G.; FHAYEH, S.V.; GUSAHOV, YE.I.; KAZANSKIY, A.M.; KASSIROV, L.N.; KARAYEV, S.A.; AFRAMOV, V.A.; VASIL'YEV, N.V.; BUGAYEV, N.F.; SAPIL'NIKOV, N.G.; KASTORIN, A.A.; RUDNIKOV, V.N.; YAKOVLEV, V.A.; PEREMYKIN, V.I.; ISAYEV, A.P.; KUZ'MICHEV, N.N.; IL'IN, S.A.; FROMIN, V.A.; LUK'YANOV, A.D.; SHAKHOV, YA.K.; IL'ICHEV, A.K., kand. sel'-khoz. nauk; KOGAN, A.Ya.; TSYNKOV, M.Yu.; BABIY, L.T.; CORBUNOV, T.T., KOVALEV, A.M., ROMANCHENKO, G.R., FRODSKAVA GORBUNOV, I.I.; KOVALEV, A.M.; ROMANCHENKO, G.R.; ERODSKAYA, M.L., red.; IVANOVA, A.N., red.; GUREVICH, M.M., tekhn. red.; TRUKHINA, O.N., tekhn. red.

[Economics of agriculture]Ekonomika sotsialisticheskogo sel'skogo khoziaistva; kurs lektsii. Moskva, Sel'khozizdat, 1962. (MIRA 15:10) 710 p.

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AREF'YEV, T.I., kand. ekon. nauk; BRASLAVETS, M.Ye., prof., doktor ekon. nauk; BROZGUL', M.M.; VLASOV, N.S., prof., doktor ekon. nauk; DUBROVA, P.F., doktor ekon. nauk; YESAULOV, P.A., kand. sel'khoz. nauk; ZAL'TSMAN, L.M., prof., doktor sel'khoz. nauk; KAL'M, P.A., dotsent, kandidat sel'skokhoz. nauk; KOSTSELETSKIY, N.A., kand. ekon. nauk; KRYLOV, V.S., kand. sel'khoź. nauk; LIEKIND, A.S., dots., kand. ekon. nauk; MAKAROV, N.P., prof., doktor ekon. nauk; OGLOBLIN, Ye.S., kand. sel'khoz. nauk; POLOVENKO, S.I., kand. ekon. nauk; POPOV, S.A., dots., kand. ekon.nauk; SAPIL'NIKOV, N.G., doktor ekon. nauk; TISHCHENKO, G.A., prof., kand. ekon. nauk; TYUTIN, V.A., prof., doktor ekon. nauk; YANYUSHKIN, M.F., kand. ekon. nauk; PYLAYEVA, A.P., red.; FREYDMAN, S.M., red.; SOKOLOVA, N.N., tekhn. red. [Organization of socialist agricultural enterprises] Organizatsiia sotsialisticheskikh sel'skokhoziaistvennykh predprijatii; kurs lektsii. Moskva, Sel'khozizdat, 1963. 662 p. (MIRA 16:8) 1. Zaveduyushchiy otdelom ekonomiki Vsesoyuznogo nauchnoissledovatel'skogo instituta sakharnoy svekly (for Aref'yev). 2. Odesskiy sel'skokhozyaystvennyy institut (for Braslavets),

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AREF'YEV, T.I.- (continued). Gard I.

3. Moskovskaya seliskokhozysystvennaya skademiya im. K.A.Timiryazeva (for Viasov), 4. Zaveduyushchiy otdelom ekonomiki i organizatsii Nauchno-issledovatel'skogo instituta sadovodstva im. I.V. Michurina (for Dubrova). 5. Moskovsk.y Gosudarstvennyy universitet im., M.V.Lomonosova (for Zalitsman, Polovenko). 6. Zaveduyushchiy kafedroy organizatsii sel'skokhozyaystvennogo proizvodstva Leningradskogo sel'skokhozyaystvennogo instituta (for Kal'm), 7, Zaveduyushchiy otdelom ekonomiki Nauchno-issledovatel skogo instituta ovoshchnogo khozyaystva (for Kostseletskiy), 8. Vsesoyuznyy nauchnoissledovatel'skiy institut ptitsevodstva (for Krylov). 9. Moskovskiy ekonomiko-statisticheskiy institut (for Libkind). 10. Vsesoyuznyy sel'skokhozyayetvenniy institut zaochnogo obrazovaniya (for Makarov). 11. Zaveduyushchiy otdelom ekonomiki Krasnodarskogo nauchno-issledovatel'skogo instituta sel'skogo khozyaystva (for Ogloblin). 12. Kafedra organizatsii sel'skokhozyaystvennogo proizvodstva Leningradskogo sel'skokhozyaystvennogo instituta (for Popov). 13. Zaveduyushchiy kafedroy Sovetskoy ekonomiki Vysshey partiynoy shkoly (for Sapil'nikov). 14. Voronezhskiy sel'skokhozyaystvennyy institut (for Tishchenko). 15. Leningradskiy sel'skokhozyaystvennyy institut (for Tyutin). 16. Direktor Severo-Kavkazskogo filiala Vsesoyuznogo nauchnoissledovatel'skogo instituta ekonomiki sel'skogo khozyaystva (for Yanyushkin).

(Agriculture---Economic aspects)

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TYUTINA, K.M.

TYUTINA, K.M.: "Investigation of electrode processes in the precipitation of lead-nickel slag". Moscow, 1955. Min Higher Education USSR. Moscow Order of Lenin Chemicotechnological Inst imeni D.I. Mendeleyev. (Dissertations for the Degree of Candidate of Technical Sciences).

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SO: Knizhnaya letopis' No 45, 5 November 1955. Moscow.

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 AUTHORS Tyutina, K. M., and Kudryavtsev, M. 1. TITLE Note on the Electrolytic Deposition of a Tin-Nickel Alloy from Chloride-Fluoride Solutions (Elektroliticheskoye osazhdeniye splava olovo-nikel' iz khlorid-ftoridnykh rastvorov). PERIODICAL Doklady Akademii Nauk, 1957, Vol. 115, Nr 3, pp. 580-582 (USSR). ABSTRACT This method of deposition on the basis of tin are of considerable interest to electroplating. Coatings with such alloys are distinguished interest to electroplating. Coatings with such alloys are distinguished ustry with good results as protective and protective and decorative coatings on steel products. This new method, mentioned in the title, where the alloy contains up to 65 % of tin, possesses a practical as wall as a theoretical importance. Apart from its resistance to diluted mineral acids, the coating emerges from the trough with a glossy sur- face and a nice pink shade, making any polishing unnecessary. The coating consists of 35% Ni and 65% NS and forms an intermetallic compound, which can only be produced by electrolysis. The composi- tion of the electrolyte and the electrolysis conditions are given: 2,1 = 2,5 N NiCl₂ + 0,4 = 0,5 N SnCl₂ + 7 N NAF + 1,0 N NH_bF; 	דטץ	INA, K. AL. 20-3-44/59	
 Chloride-Fluoride Solutions (Horid-ftoridnykh rastvorov). PERIODICAL Doklady Akademii Nauk, 1957, Vol. 115, Nr 3, pp. 580-582 (USSR). ABSTRACT This method of deposition on the basis of tin are of considerable interest to electroplating. Coatings with such alloys are distinguished by a number of valuable properties, some of which are utilized in ineby a number of valuable properties, some of which are utilized in the coatings on steel products. This new method, mentioned in the title, where the alloy contains up to 65% of tin, possesses a practical as well as a theoretical importance. Apart from its resistance to diluted mineral acids, the coating emerges from the trough with a glossy surface and a nice pink shade, making any polishing unnecessary. The coating consists of 35% Ni and 65% SN and forms an intermetallic compound, which can only be produced by electrolysis. The composition of the electrolyte and the electrolysis conditions are given: 2,h = 2,5 N NiCL₂ + 0,4 = 0,5 N SnCL₂ + 7 N NAF + 1,0 N NH₄F; 	AUTHORS	Trutina, K. M., and Kudryavtsev, N. 1.	
PERIODICAL Doklady Akademii Nauk, 1957, Vol. 115, Nr 3, pp. 580-582 (USSR). ABSTRACT This method of deposition on the basis of tin are of considerable interest to electroplating. Coatings with such alloys are distinguished by a number of valuable properties, some of which are utilized in in- dustry with good results as protective and protective and decorative coatings on steel products. This new method, mentioned in the title, where the alloy contains up to 65% of tin, possesses a practical as well as a theoretical importance. Apart from its resistance to diluted mineral acids, the coating emerges from the trough with a glossy sur- face and a nice pink shade, making any polishing unnecessary. The coating consists of 35% Ni and 65% SN and forms an intermetallic compound, which can only be produced by electrolysis. The composi- tion of the electrolyte and the electrolysis conditions are given: 2, l = 2, 5 N NiCl ₂ + 0, $l = 0, 5$ N SnCl ₂ + 7 N NaF + 1,0 N NH ₁ F;	TITLE	Note on the Electrolytic Deposition of a Tin-Nickel Alloy Hour Chloride-Fluoride Solutions (Elektroliticheskoye osazhdeniye chloride-fluoride i z khlorid-ftoridnykh rastvorov).	
interest to electropiating, contains, some of which are utilized in in by a number of valuable properties, some of which are utilized in in dustry with good results as protective and protective and decorative coatings on steel products. This new method, mentioned in the title, where the alloy contains up to 65% of tin, possesses a practical as well as a theoretical importance. Apart from its resistance to diluted mineral acids, the coating emerges from the trough with a glossy sur- face and a nice pink shade, making any polishing unnecessary. The coating consists of 35% Ni and 65% SN and forms an intermetallic compound, which can only be produced by electrolysis. The composi- tion of the electrolyte and the electrolysis conditions are given: 2,h = 2,5 N NiCl ₂ + $0,h = 0,5$ N SnCl ₂ + 7 N NAF + 1,0 N NH ₄ F;	PERIODICAL	Doklady Akademii Nauk, 1957, Vol. 115, Nr 3, pp. 580-582 (USSR).	
coatings on steel produces in 65% of tin, possesses a practical as where the alloy contains up to 65% of tin, possesses a practical as well as a theoretical importance. Apart from its resistance to diluted mineral acids, the coating emerges from the trough with a glossy sur- face and a nice pink shade, making any polishing unnecessary. The coating consists of 35% Ni and 65% SN and forms an intermetallic compound, which can only be produced by electrolysis. The composi- tion of the electrolyte and the electrolysis conditions are given: 2,h = 2,5 N NiCl ₂ + $0,h = 0,5$ N SnCl ₂ + 7 N NAF + 1,0 N NH ₄ F;	ABSTRACT	interest to electropiating properties, some of which are utilized in in	
the electrolyte 45 - 55 to our out		dustry with good results. This new method, mentioned in the croise, coatings on steel products. This new method, mentioned in the croise, where the alloy contains up to 65 % of tin, possesses a practical as well as a theoretical importance. Apart from its resistance to diluted mineral acids, the coating emerges from the trough with a glossy sur- face and a nice pink shade, making any polishing unnecessary. The coating consists of 35 % Ni and 65 % SN and forms an intermetallic compound, which can only be produced by electrolysis. The composi- compound, which can only be produced by electrolysis are given;	:
Card $1/3$ pH = 4,0 - 4,5. Temperature of the electrony co 4,5 million of the electrony co 4,		$2 = 2 \leq N \text{ Nicl}_{+} + 0.4 = 0.5 \text{ N SnCl}_{2} + 7 \text{ N Nar} + 2.5 \text{ N SnCl}_{4}$	
	Car d 1/3	pH = 4,0 - 4,5. Temperature of the erection, to 4,5.	

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Note on the Electrolytic Deposition of a Tin-Nickel Alloy from Chloride-Fluoride Solutions.

density 0,5 - 4,0 A/dm². The production efficiency with respect to current amounts to 96 - 98%. Anode surface relation. $S_{Sn} \stackrel{\circ}{\cdot} \stackrel{\circ}{S_{Ni}}$

1: 20. The average anodic current density 0.5 - 1.0 A/dm^{ik} over the total surface of the anodes. By the addition of flucric salts of sodium and of ammonium to the chlorous electrolyte the cathodic potentials of the tin separation reach more negative values than those of nickel in the identical solution. This apparently takes place because of the formation of solid complex ions SnF_{1}^{μ} and $SnCl_2F_2$,

which need a higher activation energy for discharge. The presence of fluoride effects an approach of the potentials of tin and nickel, which enables them to be deposited simultaneously at the cathode. This is proved by the so-called decomposition curves (fig. 1). A considerable depolarization takes place which, apparently is dependent upon a diminuition of the free energy at the formation of the chemical compound, that is the Sn-Ni- alloy. The investigations of the authors have shown, that the composition of the alloy is little dependent on the Sn- and Ni- concentration in the electrolyte and on the character of the cation of the fluoric acid, which, however, is of great importance in the modification of the physico-chemical properties of the deposits. The range of current density, where glossy deposits are

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"APPROVED FOR RELEASE: 04/03/2001 CIA-RDP86-00513R001857810005-6 - TO MENDER BUILDER STRUCTURE STRUCTURE 20-3-44/59 4 Note on the Electrolytic Deposition of a Tin - Nickel Alloy from Chloride - Fluoride Solutions. produced, depends on the temperature and on the pH-value of the electrolyte. The interval of permissible current densities shrink with the increase of both. Operating codes were worked out for a diminished Sn- content and for a.c.coatings and for current reversion. The two latter factors had almost no influence on the composition of the deposit, they impaired, however, its quality to a considerable extent. There are 2 Slavic references and 1 figure. ASSOCIATION: Moscow Chemical-Technological Institute imeni D. I. Mendeleyev (Moskovskiy khimiko-tekhnologicheskiy institut im. D. I. Mendeleyeva) PRESENTED: By A. N. Frumkin, Academician, January 23, 1957. SUBMITTED: November 20, 1956. AVAILABLE : Library of Congress.

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AUTHORS:	Kudryavtsev, N. T., Tyutina, K. M. SOV/156-58-3-8/52
TITLE:	The Cathodic Polarization in the Electrolytic Separation of a Tin-Nickel Alloy (Katodnaya polyarizatsiya pri elektro- osazhdenii splava olovo-nikel')
PERIODICAL:	Nauchnyye doklady vysshey shkoly, Khimiya i khimicheskaya tekhnologiya, 1958, Nr 3, pp. 435 - 438 (USSR)
ABSTRACT:	The presence of sodium and ammonium fluoride increases the cathode potential of tin to the point where it approaches the separation potential of nickel from chloride solutions. The potential in the simultaneous separation of tin and nickel at the cathode is a higher positive quantity than the potentials in the individual separation of tin and nickel. The course of the polarization curves in the separation of the alloy and its constituent parts from chloride-fluoride solutions was traced: based on an analysis of these curves it was found that the simultaneous deposition of tin and nickel on the cathode is accompanied by a considerable de- polarization.
Sard 1/2	The nature of the polarization was investigated according to various methos (suggested by Vagramyan and Gorbachev). It

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	turned out that mainly chemical polarizations were concerned. This is due to the fact that the complex ions of the deposited metals are very stable. The experimental results obtained are illustrated in diagrams. They show the cathodic polarization in the electric separation of tin, nickel and the Sn-Ni-alloy (Diagrams 1 and 2); furthermore, the change of the cathodic potential with the period of electrolysis (Diagram 3), and the dependence of the lg J on $\frac{1}{T}$ at constant polarization
	values (Diagram 4). There are 4 figures and 4 references, 3 of which are Soviet.
SSOCIATION:	Kafedra ekhnologii elektrokhimicheskikh proiz- vodstv Moskovskogo khimiko-tekhnologicheskogo instituta im. D.I.Mendeleyeva(Chair for the Technology of Electrochemical Industries at the Moscow Chemical and Technological Institute imeni D.I.Mendeleyev)
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FURPOSE: This book is intended for themical and electrical engi-neers, physicists, metallurgists and researchers interested in neers, physicists, metallurgists and researchers interested in overlaus aspects of electrochemistry. CURRINGE: The book contains 127 of the 138 reports presented at coverdary of Chemical Sciences and the institute of Thysical Anstains Acadeay of Sciences, USSR. The collection pertains to different attivants processes in metal electrochemistry sponsored by the bepart-sciences of sciences in setal electrochemical intering and published in periodical literature. A personalities and published in periodical literature. A personalities are published in periodical literature. UTivti īÝ 410 404 11 \$21 427 435 111 154 Yesin, Zditorial Boardi A.N. Frunkin (Rasp. Ed.) Academician, O.A. Yesi Froctessor; S.I. Zhdanov (Resp. Secretary). S.N. Kabanov, Fro-ressor, S.I. Zhdanov (Resp. Secretary). S.N. Kabanov, Fro-ressor, S.I. Zhanov (Resp. Secretary). S.N. Kabanov, Fro-rukortesv, Frofessor; Z.A. Solovyeva, V.V. Secnder, Frofessor; Lukortesv, Frofessor; Z.A. Solovyeva, V.V. Secnder, Frofesor; and G.M. Fluzanovich. Ed. of Publishing House: N.O. Yagorov; Tech. Ed. I., Fruskova. Sutragiza, A.A., and K.N. Gorbunova (Institute of Physical Chamistry, Academy of Sciences, USSR). Some Regularities of the Electrocrystallization of Metals Under the Influence of an Alternating Current Trudy...; [abornik] (Transactions of the Pourth Conference on Liec rochemistry: Collection of Articles) Moscow, Izd-vo AN SSL, 2059. 868 p. Errata allp interted. 2.500 copies printed. Aponsoring Agency: Akademiya nauk SSSR. Otdeleniye khimicheskikh nauk Rheyfets, Y.L., and A.L. Rotinyan (Froyekiny) in nuchno-issiedowate) skip institut nikeleyy, kob21 tovoy 1 oluvy-mony promyhlennosti-Planning and Scientific Research Institute of the Nickel, Cobalt and Tin Industry). Joint Bischneg of Ions and the Problem of Obtaining Metals of High Purity Polukarov Xu, M., and X.M. Gorbunova (Institute of Physical Chemistry Acadeay of Scinces USCN: Some Theoretical Problems on the Electrocrystalization of Alloys <u>[Aishevy</u>-R. Kinetics of Nuclei Formation During the Electro-deposition of Metals . Mudryawisev, N.T., and K.M. Tyutina (Institute of Chemical Technology Tment D.T. FARGelevev). Cathodic Polarization During the Electrodeposition of a Tin-Mickel Alloy Levin, A.I. Role of a Cathode Surface Discharge and the Fur-- mation of Passive Films in the Process of Metal Electro-deposition <u>Aydubert</u> R. (Deceased) (France). Machanism of Anode Dis-solution Irmaylov, A.V. (Institute of Chemical Technology imeni D.I. Mendelevey). Kinetics of Calnodic Processes During the Restrockeposition of Metula From Aqueous Solutions SOV/2216 Baymakny...Yw.-Y.. Kinetics of the Joint Discharge of Ions During the Electrolytic Deposition of Metals PHASE I BOOK EXPLOITATION SOV/2216 Transactions of the Fourth Conference (Cont.) . Squeshchaniye po elektrokhimii. 4th, Moscow, 1956. Card 18/34 5(±) ł . U.S.

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25(1) PHASE I BOOK EXPLOITATION SOV/3161 Nauchno-tekhnicheskoye obshchestvo mashinostroitel'noy promyshlennosti, Kiyevskoye oblastnoye pravleniye Zashchitno-dekorativnyye i spetsial'nyye pokrytiya metallov (Protective, Decorative, and Special Coatings for Metals) Kiyev, Mashgiz, 1959. 291 p. 4,200 copies printed. Editorial Board: P. K. Lavorko, N. I. Litvak, and A. P. Eychis (Resp. Ed.); Ed. of Publishing House: M. S. Soroka; Chief Ed. (Southern Division, Mashgiz): V. K. Serdyuk, Engineer. . FURPOSE: This book is intended for technical personnel in the field of protective coatings for metals. COVERAGE: The papers in this collection, presented at a conference of the NTO Mashprom held in Odessa, deal with the mechanization and acceleration of metal-coating and plating processes performed by spraying, electrolytic, and other methods. Quality control of protective coatings is also discussed. No personalities are mentioned. References follow several of the papers. Card 1 TA BE [1] "我们就是我们的问题。" [1] "我们就是我们的问题。" 计注意 网络拉马拉马 关于"齐国王"。

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JiBA S/539/61/000/032/013/017 D204/D301 AUTHORS: Kudryavtsev, N.T., Tyutina, K.M., Chvankin, I.V. and Tsupak, T.Ye. Electrodeposition of a Sn-Ni alloy from alkaline cyanide solutions SOURCE: Moscow. Khimiko-tekhnologicheskiy institut. Trudy, no. 32, 1961. Issledovaniya v oblasti elektrokhimii, 283,288 TEXT: A study of the joint deposition of Ni and Sn from stannate solu- tions containing additions of complex Ni cyanide. The influence of Ni concentration in the solution and of current density, D _R , on the quality, composition and current efficiency of the alloy were studied at 65°, 75° and 85°C, depositing the metals on brass or Ti plates. Cathode potential at various D _k 's were measured during the deposition of the alloy and of sn alone. Alloys with 10-26% Ni could be obtained from solutions con- tianing 0.12 ~ 0.66 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1; (alloys of Card 1/3			
AUTHORS: Kudryavtsev, N.T., Tyutina, K.M., Chvankin, I.V. and Tsupak, T.Ye. TITLE: Electrodeposition of a Sn-Ni alloy from alkaline cyanide solutions SOURCE: Moscow. Khimiko-tekhnologicheskiy institut. Trudy, no. 32, 1961. Issledovaniya v oblasti elektrokhimii, $283 \neq 288$ TEXT: A study of the joint deposition of Ni and Sn from stannate solu- tions containing additions of complex Ni cyanide. The influence of Ni concentration in the solution and of current density, D_{R^2} on the quality, concentration in the solution and of the alloy were studied at 65° , 75° if and 85° C, depositing the metals on brass or Ti plates. Cathode potentials at various D_k 's were measured during the deposition of the alloy and of sn alone. Alloys with 10-26% Ni could be obtained from solutions con- taining 0.12 - 0.66 g Ni/1, 53 $\leq Na_2Sn0_3/1$ and 10 g NaOH/1; (alloys of	x		
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tions containing data solution and of current denotity R^{1} (R^{2}) concentration in the solution and of current denotity R^{2} (R^{2}) composition and current efficiency of the alloy were studied at 65°, 75° (R^{2}) and 85°C, depositing the metals on brass or Ti plates. Cathode potentials and 85°C, depositing the metals on brass or Ti plates. Cathode potentials at various D_{R}^{13} s were measured during the deposition of the alloy and of the various D_{R}^{13} s were measured during the obtained from solutions common solutions common solutions common solutions R^{2} (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, 53 g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (alloys of taining 0.12 \simeq 0.6 g Ni/1, (alloys 0.12 \simeq 0.6 g Ni/1, (b = 0.12 + 0.12) g Na ₂ SnO ₃ /1 and 10 g NaOH/1, (b = 0.12) g Na ₂ SnO ₃ /1 g Na ₂ SnO ₃ /1 g Na ₂ SnO ₃ /1 g Na ₃ SnO ₃ SnO ₃ /1 g Na ₃ SnO	SOURCE	Moscow. Khimiko-tekhnologicheskiy institut. Trudy, Hor (09) 1961. Issledovaniya v oblasti elektrokhimii, 2837288	
Card 1/3	tions contr concentrat: compositio and 85°C, at various Sn alone. taining 0.	ion in the solution and of current density, $r_{\rm k}$,	
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人,这些人的人身上,这些人的人,你就是你的人,你们还是你的人,你们还是你没能够没有你的人?""你们还是你是你的,你们,你还是你们,这么 <u>不是你,我就是你们不知道你的,你不能没有我的</u>			

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Electrodeposition of a Sn-Ni

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20-26% Ni were bright), but the current efficiency fell sharply on increasing the Ni and lowering the Sn content in the electrolyte and at lower temperatures. Thus on increasing Ni from 0.06 to 0.6 g/l at 75°C₃ at $D_{k} = 1 \text{ amp/dm}^{2}$, the current efficiency decreased from 65 to 8%. A proportion of Ni in the deposit rose with increasing Ni content of the solution, but was practically unaffected by changes in temperature or D. Cathode polarization in the deposition of the alloy was more pronounced^k than during the deposition of Sn alone. The results are discussed in terms of the polarization curves derived for the various processes taking place, concluding that the joint deposition of Ni and Sn facilitates the evolution of H₂ by reducing its overvoltage on the cathode. Passivated Sn anodes or anodes of Sn and an insoluble metal were found suitable and the following conditions are recommended for the deposition of an alloy cortaining 5-12% Ni: electrolyte composition - St (as Na₂SnO₃) 30g/l₁, Ni (as Ni(CN)₂) 0.06 - Ql2 g/l₁ NaOH 10 g/l, NaCN 0.25 g/l; temperature 75°C; D_k equal to 1 amp/dm². Analyses of the electrolyte and of the

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	D204/D301	
AUTHORS :	Kudryavtsev, N.T., Tyutina, K.M. and Baraboshkina, N.K.	
TITLE:	The effects of organic additives on the cathode process in the electrolysis of Sn-Ni alloys	
SOURCE:	Moscow. Khimiko-tekhnologicheskiy institut. T _r udy ₉ no ₀ 32 ₉ 1961 ₀ Issledovaniya v oblasti elektrokhimii, 289-292	
vestigation could influ trolytes co phonic acid cally pure	inuation of earlier work, inspired by Soviet and Western in- s which showed the addition of surface active organic compounds ence the composition and quality of alloys deposited from elec- ntaining more than 1 metallic salt. Additions of p-phenyl sul- (I) prepared from (a) freshly distilled phenol, (b) chemi- synthetic phenol, (c) technical phenol and (d) crude carbolic made to the electrolyte consisting of 300 g/l NiCl ₂ .6H ₂ O ₅	
50 g/l SnCl	$_{2}^{6H_{2}O}$ and 60 g/l NH ₄ F, to investigate their effects on the	
quality and	composition of the deposit and on the cathodic polarization.	
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The additions were made at $50-65^{\circ}C_{2}$ at pH 4.5. It was found that the cathodic polarization, quality and composition of the deposits depended on the purity of the phenol from which I was prepared. Additions of I prepared from freshly distilled or chemically pure phenol had practically no effect. 0.5-0.7 moles/l of I prepared from technical phenol displaced the electrodeposition potentials in the electronegative direction by 100-200 mv and lowered the Sn content in the alloy by 11.14% (to 51.54%), with current densities of $0.5-4.0 \text{ amp/dm}^2_8$ The deposits were shiny and elastic. Raising the temperature to 55-65 C further decreased the Sn to 49-50%. The effect of I prepared by the sulphonation of crude carbolic acid increased the cathodic polarization by 400 mv during the deposition of S_n and thereby, lowered the latter to 20% in the alloy; the deposite were gray when the current density exceeded 0.2 amp/dm². The addition of 1-1.5 g/1 of di-2-dimethylamino-5-pyridyl methans increased the cathodic polarization for both S_n and N_i and had, therefore, little effect on the alloy composition. The deposits were elastic, light in color and contained up to 60% Sno There are 2 figures and 5 references: 4 Soviet-bloc and 1 non-Soviet-bloc.

Card 2/2

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	37868 S/080/62/035/005/006/015 D204/D307
19.8310 Authors:	Kudryavtsev, N. T., Tyutina, K. M. and Firger, S. M.
TITLE:	Protecting steel articles against corrosion with an electrolytic zinc-nickel coating
PERIODICAL:	Zhurnal prikladnoy khimii, v. 35, no. 5, 1962, 1035- 1043
	a stand the stand studied from evanide
(I) and ammo 0.5 - 2% Ni from the sec production of Zn (as a com- cyanide) 0.0 (D) - 1 - 3 80 - 96\% and	dic codeposition of Ni and Zn was studied from cyanide niacal (II) solutions, obtaining alloys containing from the first electrolyte and 10 - 30 (or more)% Ni ond. The following conditions are recommended for the f high quality coatings ($\sim 2\%$ Ni) from I: Electrolyte - mplex) 32 g/l, NaCN 75 - 100 g/l, NaOH 70 g/l, Ni (as 25 - 0.75 g/l; temperature - 20 - 25°C; current density amp/dm ² . The corresponding current efficiencies were t the anodes were pure Zn. The Ni content of the depo- ed with increasing Ni concentration in the electrolyte th increasing D. Electrolyte II was investigated in

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Protecting steel articles ...

greater detail since deposits containing 10 - 20% Ni were found to be the best protection against corrosion. It was found that the Ni content in the alloy increased with temperature and pH and depended on the Ni concentration in II and on D as in the case of I. Best quality deposits containing 10 - 18% Ni were obtained at 40° C and at pH 6.5 - 6.8 from a solution containing 15 g ZnO, 12 - 36 g NiOl₂.6H₂O, 250 g NH₄Cl and 20 g H₃BO₃ per liter, at a cathode current density of 1 - 3 amp/dm². The cathode current efficiency was 95% under these conditions and the anodes were Zn or Zn/Ni. Solution II was sufficiently stable w.r.t. composition and showed good buffer properties at pH 6 - 10. Pitting of the deposits could be partially alleviated by the addition. of various organic compounds to the electrolyte, but was best avoided by stirring the solution during electrolysis. There are 2 figures and 2 tables.

SUBMITTED: April 13, 1961

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- THERE ARE AND A DESCRIPTION OF THE AREA **1** s/123/62/000/006/011/018 A004/A101 Kudryavtsev, N. T., Tyutina, K. M. AUTHORS: Electrodeposition of tin-nickel alloys PERIODICAL: Referativnyy zhurnal. Mashinostroyeniye, no. 6, 1962,43, abstract 6B214 (V sb. "Elektrolit. osazhdeniye splavov". Moscow, Mashgiz, 1961, 76-93) The authors analyze the composition of electrolytes and operating conditions for electrodeposition of tin-nickel alloys from acid chloride-fluoride and alkaline electrolytes. The authors recommend the following composition for Sn-Ni alloy coatings from chloride-fluoride electrolytes containing 65% Sn (in g/l): nickel chloride - 250 - 300 (2.1 - 2.5 n); tin chloride SnCl₂·2H₂O g/1: ficker chloride - 200 - 000 (2.1 - 2.9 H); off chloride onoig engoting 45 - 50 (0.4 - 0.5 n); sodium fluoride NaF - 30 (0.7 n); ammonium fluoride NH₄F - 35 - 38 (1.0 n), with a pH-value of 4.5 - 5, at a temperature of 45 - 55°C and a cathode current density of 0.5 - 4.0 amp/dm². The anodes are of tin and and a cathode current density of 0.5 - 4.0 amp/dm². nickel with a surface ratio from 1:10 to 1:20. The mean anode current density amounts to $0.5 - 1.0 \text{ amp/dm}^2$ of the total anode surface. For Sn-Ni alloy coatings with a reduced tin content (approximately50% Sn) the authors suggest Card 1/2

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