

1ST AND 2ND ORDERS										3RD AND 4TH ORDERS									
PROCESSES AND PROPERTIES INDEX																			
BC										B-I-7									
Electrolytic parting of the noble metals. F. Voss. (Zem.-German. Voss: Metall. Tech., 1938, No. 11-12, 21-23).—The Ag content of a Au-Ag alloy is preferably increased by fusion with a Ag-rich alloy. For electrolytic refining of Ag, the Siemens & Halske process is preferred to that of Dittol. Cu. Ann.																			
ASM-AIA METALLURGICAL LITERATURE CLASSIFICATION 100 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31 32 33 34 35 36 37 38 39 40 41 42 43 44 45 46 47 48 49 50 51 52 53 54 55 56 57 58 59 60 61 62 63 64 65 66 67 68 69 70 71 72 73 74 75 76 77 78 79 80 81 82 83 84 85 86 87 88 89 90 91 92 93 94 95 96 97 98 99 100																			

111 AND 112 ORDERS

PROCESSES AND PROPERTIES INDEX

4A

Electrolytic parting of the noble metals. Pr. Vogel. *Russko-Germanski Vestnik Nauk i Tech.* 1932, No. 11/12, 21-8; *Chem. Zvest.* 1932, 11, 2519. --From a crit. consideration of the methods of sepn. of Au and Ag, V. concludes that the most satisfactory process is to fuse a poor Au-Ag alloy with a rich Ag alloy, i. e., increase the Ag content. The Siemens and Halske process for electrolytic Ag refining is better commercially than that of Dietzel.

W. A. Moore

COMMON ELEMENTS

ASME-SLA METALLURGICAL LITERATURE CLASSIFICATION

111 AND 112 ORDERS

111 AND 112 ORDERS

1ST AND 2ND ORDERS										PROCESSES AND PROPERTIES INDEX										100 and 4th ORDERS									
7 Electrolytic Parting of the Noble Metals. Fr. Vogel (<i>Russko-Nemotinski Vestnik Nauk i Tech.</i>, 1932, (11/12), 21-28; <i>Chem. Zentr.</i>, 1932, 103, 11, 2519; <i>C. Abstr.</i>, 1933, 27, 5649). From a critical consideration of the methods of separation of gold and silver alloy with a rich silver alloy, i.e. increase the silver content. The Siemens and Halske process for electrolytic refining of silver is better than that of Dietzel.—S. G.																													
ASTM-SLA METALLURGICAL LITERATURE CLASSIFICATION																													
SECTION 100-1000										SECTION 100-1000										SECTION 100-1000									
SECTION 100-1000										SECTION 100-1000										SECTION 100-1000									

HEYM, E.; KRAUSE, H.; LIESegang, E.; VOGEL, G.; WESTPHAL, W.

Model experiments on influencing the irritability of skeletal muscles in various functional states of the superior central areas. Acta physiol. hung. 9 no.1-3:179-192 1956.

1. Institut for Veterinar-Physiologie der Humboldt-Universitat, Berlin.

(Muscles, physiol.
eff. of various indirect stimulations on chronaxy in
skeletal musc. of frogs (Ger))

VOGEL, I.; SEGAL, A.

"We shall build, but how?"

p. 18 (Drumul Belsugului) No. 8, Aug. 1957
Bucharest, Rumania

SO: Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 4,
April 1958

VOGEL, I.; SEGAL, A.

"Let us establish a base of building materials for our constructions."

p. 8 (Drumul Belsugului) No. 9, Sept. 1957
Bucharest, Rumania

SO: Monthly Index of East European Accessions (EEAI) LC. Vol. 7, no. 4,
April 1958

7/1

10

High-Intensity X-Ray Spectrograph with a Bent Crystal. N. D. Borisov and J. M. Vogel (*Zhurnal Tekhnicheskoy Fiziki* (*J. Tech. Physics*), 1937, 7, (2), 164-170).—[In Russian.] The instrument described requires no adjustments during use in the relative positions of crystal holder and plate holder. The operation of the instrument is described, and photographs obtained with it are shown.—N. A.

ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

SEARCHED SERIALIZED INDEXED

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NOV 1937

U.S. DEPT. OF COMMERCE

LIBRARY OF CONGRESS

General & Specific
Chem - 2

CA

The diagram of state iron-iron phosphide-aluminum phosphide-aluminum. Rudolf Vogel and Helmut Kloss (Univ. Göttingen, Ger.). *Arch. Eisenhüttenw.* 23, 287-91(1952).—The diagram of state of the system Fe-Al-P was investigated more closely in the Fe corner by thermal and structural exams. The P content of the ternary solid soln. range decreases rapidly with increasing Al content from 2.8 to 1.2% and then remains const. In the peritectic and eutectic quaternary reaction one ternary α -solid soln. in each participates. An abs. difference in affinity of Fe and Al to P does not exist. Whether P is bound to Fe or Al depends on the ratio of Fe to Al. M. Hartenhein
Possibility of finding models for the study of condensed systems. T. A. Popova (State Sci. Research Inst. Glass, Moscow). *Zhur. Fiz. Khim.* 26, 879-81(1952); cf. C.A. 43, 5322b.—A binary system (C and D) melting at low temp. can serve, as far as the phase diagram is concerned, as a model for a system of A and B melting at high temps. If there is a definite relation between the ratios T_1M_1/T_2M_2 and T_3M_3/T_4M_4 : T_1, T_2, T_3 , and T_4 , and M_1, M_2, M_3 , and M_4 are the m.p.s. and the mol. wts. of A, B, C, and D, resp. Then, system PbOH-cresol can be a model for TiBr-TiI system, etc. J. J. Bikerman

CA

The structure of α -glucosan and similar compounds.
H. Vogel (Bludenz, Austria). *Osterr. Chem. Zig.* 52, 21-6
(1951).—The glucosan prepd. by Pictet (C.A. 14, 3226)
on heating glucose *in vacuo* is not a pure compd. but is con-
taminated with the starting material and other by-products.
The main portion, however, consists of 2-hydroxy-~~glucose~~
glucal(1,5). The latter was prepd. by heating glucose with
anhyd. piperidine. Dorit L. Noether.

VOGEL, J.; KRYSPIN, J.

New method of measuring of tissue resistance in ohms.
Physiol. bohém. 5 no.3:381-384 1956.

1. Polarographisches Institut der Tschech. Akademie d.
Wissenschaften, Prag, und Neurochirurgische Klinik der VLA,
Hradec Kralove.

(ELECTROPHYSIOLOGY,
measurement of tissue resist. in ohms (Ger))

VIGEL J

VOGEL, J.; KRYSPIN, J.

New method of electrical resistance of tissue in ohms.
Cesk. fysiolog. 5 no.2:240-243 23 June 56.

1. Polarografický ústav CSAV, Praha a Neurochirurgická klinika
VLA, Hradec Králové. Předneseno na schůzi Fysiolog. sekce společnosti
J. E. Purkyne, odbočka Hradec Králové dne 2. XI. 1954.
(ELECTROPHYSIOLOGY,
determ. of electrical resist. of tissue in ohms. (Cz))

VOGEL, Jiri

CZECH

Acetylation of vinylidene chloride. (Oto Wichterle and Jiri Vogel) (Vysoka skola chem. Prace, Chem. Listy 48, 1225-31; Collection Czechoslov. Chem. Commun. 19, 1197-1224 (1954) (in English).—A new modification of the prepn. of $\text{CH}_2=\text{CCl}_2$ (I) from $\text{CH}_2\text{ClCCl}_3$ (II) is described: II was added dropwise to the top of a column filled with granulated Zn. Water was boiled in a flask connected to the lower end of the column. The top of the column was fitted with a reflux condenser fed with 30° warm water. The condenser returned II and H_2O to the column while I passed to a condenser and receiver cooled with ice water; b.p. of I, 31-35°. Adding to a stirred mixt. of 50 g. AcCl and 100 g. AlCl_3 73 g. I at 0 to -5°, decomp. the mixt. with ice, and steam distg. gave 83 g. $\text{MeCOCH}:\text{CCl}_2$ (III), stable when wet and covered with H_2O , b. 153-6°, d_4^{20} 1.3008, n_D^{20} 1.4928. Dry III decomp. slowly, releasing HCl . III phenylhydrazones, m. 77-8°; semicarbazone, m. 169-70° (from MeOH). Refluxing III (5 g.) in 30 ml. EtOH with 6 g. Zn dust (added in three 2-g. portions), dis. the cooled mixt. with H_2O , extg. the oil with Et_2O , and distg. the ext. gave $\text{MeCOCH}:\text{CHCl}$, b. 125-30°, semicarbazone, m. 127° (from EtOH). Adding 8.8 g. III in 35 g. Et_2O to EtMgBr prep. from 4.8 g. Mg and 22 g. EtBr gave 5.5 g. $\text{MeEtC(OH)CH}:\text{CCl}_2$, b. 77-8°. Treating a mixt. of 78 g. AcCl and 66.6 g. AlCl_3 with 48.6 g. I at 30° gave, in

to the title
 addn. to 50 g. III, 12 g. of a compd., $C_8H_7ClO_2$ (IV), b. 220-230°, b.p. 97-8°, m. 85-6° (from H_2O). IV is considered to be either 2-methyl-4-chloro-4H-pyran-4-one, or 6-methyl-4-chloro-2H-pyran-2-one, since its hydrogenation over PtO_2 in MeOH gave 65% $EtMeCH(CH_3)CO_2Me$, b. 160-2°. Adding the ester (1.7 g.) to a mixt. prepd. by the reaction of $EtNH_2$ (2.12 g.) with $EtMgBr$ (from 2.81 g. $EtBr$ and 0.43 g. Mg) folded $Me(CH_2)_5CONHPh$, m. 64-7°. Treating 65.5 g. III with 5 g. concd. H_2SO_4 at 60-70° 2 hrs., removing the HCl liberated with a stream of air, raising the temp. after 1 hr. to 100°, adding 1 g. H_2SO_4 , pouring the mixt. after 2 hrs. at 100° into H_2O (the product solidified), steam distg. 3 g. of III, dissolving the residu. in Et_2O , and distg. the ext. in vacuo gave 37 g. of a compd., $C_{10}H_{11}ClO_2$ (V) (3,6-dichlorovinyl-4-methyl-2H-pyran-2-one) (V), b. 143-6°, m. 65-6° (from MeOH), mol. wt. 209.2. Bromination of V in CCl_4 gave a compd., $C_{10}H_{11}BrClO_2$, m. 143-3°. Hydrogenation of 25 g. V in 200 ml. MeOH over 0.1 g. PtO_2 gave, after 20 hrs. (consumption 4.2 moles H_2) 7 g. $EtMeCH(CH_3)CO_2Me$ (VI), b. 65.5-7°, d. 0.832, n. 1.416. Hydrolysis of VI gave 80% $EtMeCH(CH_3)CO_2H$ (VII), b. 115°, d. 0.9103, n. 1.4220. Adding 4.03 g. VI to 0.665 g. $LiAlH_4$ in Et_2O , refluxing the mixt. 1 hr., and decoring the mixt. with 2% HCl gave 3.24 g. $MeEtCH(CH_3)OH$ (VIII), b. 184-5°. Reducing 3.24 g. VIII, 11.3 g. anisotropic H_2O , and 2.9 g. concd. H_2SO_4 5 hrs. yielded 2.13 g. $EtMeCH(CH_3)Br$ (IX), b. 185-9°. Treating 2.13 g. IX in Et_2O with 0.205 g. Mg , and decompz. the Grignard agent with H_2O gave 0.4 g. $EtMeCH(CH_3)Me$, b. 114-16°

5

EtO

The identity of VII was checked by comparison with a synthetic compd. prepd. as follows: MeCHBrEt (69.5 g.) and 80 g. $\text{CH}_3(\text{CO}_2\text{Et})_2$ gave 70 g. $\text{MeEtCHCH}(\text{CO}_2\text{Et})_2$, the hydrolysis of the ester and decarboxylation of the free acid yielding 85% $\text{EtMeCHCH}_2\text{CO}_2\text{H}$, b. 104-8°. Treatment with SOCl_2 and subsequent treatment with EtOH gave 93% $\text{EtMeCHCH}_2\text{CO}_2\text{Et}$, b. 158-9°, the reduction of which with Na in EtOH -PhMe yielded 44% $\text{EtMeCHCH}_2\text{CH}_2\text{OH}$, b. 151-4°. Transforming the alc. with PBr_3 at 10° to 70% $\text{EtMeCHCH}_2\text{CH}_2\text{Br}$, b. 145-8°, treating the bromide with Na and $\text{CH}_3(\text{CO}_2\text{Et})_2$, hydrolyzing the ester and decarboxylating the free acid at 200° yielded 67% VII, b. 228-32°; *Me ester*, prepd. with CH_3N_3 , b. 180-5°; *anilide*, m. 40-40.5°, identical with that prepd. from VI.

M. Hudlicky

VOGEL, J

CZECH

✓ Apparatus for rapid oscillographic quantitative analysis.
P. Valenta and J. Vogel (Polarografický ústav CSAV,
Prague). *Chem. Listy* 49:581-8 (1955).—A simple adapting
unit for the cathode-ray oscilloscope "Křizik N-523" was
designed. The device enables the method of saw-tooth
voltage impulse to be applied in quant. oscillo-polarographic
analysis. The wiring diagrams are included. F. S.

AP

BI

VOGEL, J.

Czech

CA: 47:11030

Central Polarographic Inst., Prague

"A new apparatus for oscillographic polarography."

Sborník Mezinárod. Polarog. Sjezdu Praze, 1st Congr. 1951, Pt. III, Proc., 731-6
(in Czech), 736-8 (in Russian), 739-42 (in German)

THE CZECHOSLOVAKIA

VOGEL, J. J.

J. Heyrovsky Institute of Polarography of the Czechoslovak
Academy of Sciences, Prague

Prague, Collection of Czechoslovak Chemical Communications,
No 10, 1965, pp 3523-3525

"A Note on the Theory of Polarographic Currents."

C A

Addition of nitroso compounds to conjugated systems.
 O. Wichterle and J. Vogel. *Collection Czechoslov. Chem. Commun.* 14, 270-18 (1949) (in English); cf. C.A. 42, 556. Present work confirms the results of Arbuzov (C.A. 43, 654) that the reaction of PhNO and $\text{Me}_2\text{C}=\text{CH}\text{NOCl}$ gives 2-phenyl-3,6-dihydro-1,2-oxazine (I) rather than 1-phenyl-3,4-oxidopyrrolidine. The action of various Grignard reagents on I results in the splitting of the O-N bond and hydrogenation to $\text{PhNHCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ (II), b. 150-60°; the alkyl radicals formed from the Grignard either join or disproportionate into an olefin and a paraffin. Dehydration of II by H_2PO_4 gives 1-phenyl-3-pyrrolone (III); methiodide, m. 138° (from EtOH). Ozonization of III and subsequent hydrogenation of the ozonide with PtO_2 gives 4-phenylmorpholine, m. 58.3-9°. EtMgBr (IV) and 2-phenyltetrahydro-1,2-oxazine give 61° 4-phenylamino-1-butanol (V), b. 151°, n_D^{20} 1.5629, d_4^{20} 1.0508, which yields an acid oxalate, white needles from MeOH, m. 124.8-5° (decomp.); hydrogenation of II also yields V. Dehydration of V by H_2PO_4 gives 1-phenylpyrrolidine; picrate, m. 115-15.5°. IV did not react with either $\text{PhNMeOCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$ or $\text{PhNMeOCH}_2\text{CH}_2\text{CH}_2\text{CH}_2\text{OH}$. P. M. Downey

POLAND / General Problems of Pathology. Immunity. U

Abs Jour : Ref. Zhur - Biologiya, No. 3, 1959, 13425

Author : Manski, W.; Vogel, M.; Zylberberg, A.

Inst : Polish Academy of Sciences

Title : Complement Inactivation in vivo and in vitro.

Orig Pub : Bull. Acad. polon. sci., 1957, Cl.2, 5, No. 9, 287-293, XXXVIII

Abstract : Some polysaccharides and their sulfonated derivations were investigated with the purpose of finding the substances with the greatest anti-complementarity and least toxicity for its possible use in inhibition of immune hemolysis in hemolytic diseases. In vitro the complement (C) of rabbit, guinea pig and man was inactivated: soluble (depolymerized) inulin (I), heparin (II),

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POLAND / General Problems of Pathology. Immunity.

U

Abs Jour : Ref. Zhur - Biologiya, No. 3, 1959, 13425

Congo red, aqua blue in a concentration of 4.5-5 mg/ml; germanin (III), II, agar -3 mg/ml; zymozane (IV) - 0.6 mg/ml; 1-0.5 mg/ml; sulfonated dextran (V) - 0.075 - 0.1 mg/ml. They inactivated C'3. In experiments on rabbits and guinea pigs, C was inactivated by V (8mg/ml), III (310 mg/ml), I and IV (100 mg/ml). The duration of complete inactivation of C is larger in guinea pigs (for III 40 min., for V 70 min.), than in rabbits (15 min.); restoration of the initial titer of C is faster in guinea pigs (after 70-125 min.) than in rabbits (after 220 min.). The least toxic in vivo are V and III even with frequent multiple introduction. The most toxic is I. Since the SO₃H group enters

Card 2/3

POLAND / General Problems of Pathology. Immunity.

U

Abs Jour : Ref. Zhur - Biologiya, No. 3, 1959, 13525

into the composition of II, V, III, Congo red and aqua blue, which inactivate C, it is assumed that it determines the anticomplementary and anticoagulative properties of the given compounds. Therefore it follows that C'3, similar to prothrombin, is a protein which has in all probability many alkaline reactive groups. -- F. L. Buch

Card 3/3

4

VOGEL, M.

VOGEL, M.

VOGEL, M. The German tourist paths are waiting for Polish tourists. p. 12.

Vol. 28, no. 10, Oct. 1956

TURYSTA

Poland

So: East European Accession, Vol. 6, No. 5, May 1957

MANSKI, Wladyslaw; VOGEL, Marian; ZYLBERBERG, Aurelia

Effect of various polysaccharides, sulfonic derivatives of polysaccharides & aromatic salts of sulfa-acids on the complement. Med. dosw. mikrob. 10 no.1:53-60 1958.

1. Z Instytutu Immunologii i Terapii Doswiadczałnej im. Ludwika Hirszfelda O. A. N. we Wrocławiu Dział Biochemii Kierownik: doc. dr W. Manski.

(POLYSACCHARIDES, eff.

on complement (Pol))

(COMPLEMENT, eff. of drugs on

polysaccharides, sulfur-containing polysaccharides & aromatic salts of sulfa-acids (Pol))

COUNTRY	:	CZECHOSLOVAKIA	H
CATEGORY	:	Chemical Technology. Chemical Products and Their Uses. Part 3. Processing of Solid*	
ABS. JOUR.	:	RZKhim., No. 1 1960, No. 2406	
AUTHOR	:	Vogel, O.	
INST.	:		
NOTE	:	Long-Distance Gas Supply in Czechoslovakia. Part III. Brown-Coal Gas	
ORIG. PUB.	:	Sklar a korenik, 1959, 9, No 2, 42-44, 46	
ABSTRACT	:	The modern method of gasification of brown coals, particularly Czechoslovak ones, with a view to production of gas suitable for long-distance gas supply of industrial and private consumers, as well as prospects of development of the production of such gas, were examined. Bibl. 20 titles. Part II, see RZKhim., No 17, 1959, No 62236.	
		*Fossil Fuels	
CARD:		1/1	

VOGEL, O.

An interesting case of a serious breakdown in gas supply. p.203.
(Sklar A Keramik, Vol. 7, No. 7, July 1957, Praha, Czechoslovakia)

SO: Monthly List of East European Accessions (EEAL) LC. Vol. 6, No. 9, Sept. 1957. Uncl.

VOGEL, O.

"Long-distance transportation of gas in Czechoslovakia." P. 10.

SKLAR A KERAMIK. (Ministerstvo lehkeho prumyslu). Praha, Czechoslovakia,
Vol. 9, No. 1, Jan. 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 8,
August 1959.
Uncla.

VOGEL, O.

"Long-distance distribution of gas in Czechoslovakia." Pt. 3. Lignite gas. P. 42.

SKLAR A KERAMIK. (Ministerstvo lehkeho prumyslu). Praha, Czechoslovakia, Vol. 9, No. 2, Feb. 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 8, August 1959.
Uncla.

VOGEL, O.

"Long-distance distribution of gas in Czechoslovakia." Pt. 4.
"Natural gas. (To be contd.) P. 73.

SKLAR A KERAMIK. (Ministerstvo lehkeho prumyslu). Praha, Czechoslovakia,
Vol. 9, No. 3, Mar. 1959.

Monthly list of East European Accessions (EEAI), LC, Vol. 8, No. 6,
August 1959.
Uncla.

VOGEL, O.

"Long-distance gas distribution in Czechoslovakia. Pt. 5. Comparison of long-distance gases with a producer gas." P. 105.

SKLAR A KERAMIK. (Ministerstvo lehkeho prumyslu). Praha, Czechoslovakia, Vol.9, No. 4, Apr. 1959.

Monthly list of East European Accessions (EEA1), LC, Vol. 8, No. 8, August 1959.
Uncla.

VOGEL, O.

Loss of slag by insufficient burning, p. 174, SKLAR A KERAMIK (Ministerstvo lehkého průmyslu) Praha, Vol. 5, No. 8, Aug. 1955

SOURCE: East European Accessions List (EEAL) Library of Congress,
Vol. 45, No. 12, December 1956

VOGEL, G.

"Remarks on Gas Analyses by the Orsat Apparatus." p. 158 (Paliva, Vol. 33, no. 7/8, July/Aug. 1953, Praha)

SO: Monthly List of East European Accessions, Vol. 3, no. 2, Library of Congress, Feb. 1954, Uncl.

VOGEL, Theo

Experiences and possibilities of using vacuum spectrometers.
Magy kem folyoir 68 no.12:541-542 D '62.

1. ARL, Inc., Lausanne, Svajc.

[illegible]

... 017, 0045, 0046

ADT: 100

TITLE: Optical glass. No. 2826

SOURCE: Steklo i keramika, no. 1, 1964, 45

TOPIC TAG: tertiary system, refractive index

with the B_2O_3 - ZnO - WO_3 (or MoO_3) ternary system, and the B_2O_3 - ZnO - WO_3 - MoO_3 quaternary system. The B_2O_3 - ZnO - WO_3 system has been studied by [1, 2] and the B_2O_3 - ZnO - WO_3 - MoO_3 system by [3].

~~Nb₂O₅, 15-25 La₂O₃, 10 TiO₂, and in place of oxygen is a structure~~

ASSOCIATION: VEB Jenaer Glaswerk Schott u. Gen (VEB Jena Glassworks, Schott and Gen)

SUBMITTER: JENAS

ENCL: 00

SUB CODE: MT. 17

NO REF SOV: 000

OTHER: 000

End 17

SOURCE CODE: PO/0015/66/000/008/0217

ACC NR: AP6034884

AUTHOR: Vogel, W. (Doctor); Gerth, K.

ORG: VEB Jena Glassworks Schott Gen. Jena Glassworks Schott Gen. Jena, (NRD)
(VEB Jenaer Glaswerk Schott Gen. Jena) (NRD)

TITLE: Fluorophosphate optical glass with low light refractive index and high Abby number. GE Pat. No. 1/00. 49610 [announced by VEB Jena Glassworks Schott Gen. Jena (NRD) (VEB Jenaer Glaswerk Schott Gen. Jena (NRD))]

SOURCE: Szklo i ceramika, no. 8, 1966, 219

TOPIC TAGS: optic glass, fluorophosphate, light refraction index, fluorosilicate glass, metal fluoride glass

ABSTRACT: The patent deals with fluorophosphate optical glass which is colorless and which because of the high content of metal fluorides and fluorosilicates has very low viscosity in the melted state. Evaporation of the components of the glass during the melting process takes place to a degree sufficient for further industrial utilization. By properly varying the chemical composition, glass with a light refractive index from 1.45 to 1.53 and Abby number from 55 to 80 can be obtained. Because of these properties, the glass is suitable for use in micro-optical instrumentation and, by virtue of its good resistance to humidity, may be employed in external lens systems. This glass

Card 1/2

ACC NR: AP6034884

is characterized by a composition consisting of $Me(I)_F - Me(II)SiF_6 - Al(PO_3)_3$, where Me(I) indicates lithium, sodium, or potassium, and Me(II) indicates manganese, calcium, strontium, barium, cadmium, zinc, lead, or aluminum. Further details on the composition are given.

SUB CODE: 11/ SUBM DATE: none

Card 2/2

1. The following is a list of the
materials used in the preparation of the
samples.

A. In the preparation of the samples

the following materials were used:

1. Lead phosphate

2. Lead phosphate
3. Lead phosphate
4. Lead phosphate

ABSTRACT: A series of samples of lead phosphate
which are prepared by the reaction of lead phosphate
with various acids. The glass may contain up to the following
proportions of other compounds: PbO , 10 wt%; Al_2O_3 , 5 wt%;
and/or KPO_3 , 12 wt%; InF_3 , 50 wt%; Sb_2O_3 , 40 wt%; LaF_3 , 5 wt%.

ADDITIONAL INFORMATION: The samples are prepared by the reaction of lead phosphate

SUBMITTED: 1984-11-14

NO REF DATA
Card 1/1

ATL PRESS: 3140

1.17001-65 15/10/11 011/000/011/0045/0045 011/000/011/0045/0045 03/04/

03/04

ACCESSION NO: AP404-004

011/000/011/0045/0045

AUTHOR: Vogel, W.

TITLE: Fluoride-phosphate optical glass. No. 404

SOURCE: Steklo i keramika, No. 11, 1964, 41

REF: Table: Monophosphate, diphosphate, triphosphate, tetrakisphosphate compounds, refractive index

ABSTRACT: Glasses of various compositions of the $\text{BaF}_2\text{-BaF}_2\text{-Sr PO}_3$ system or Ba PO_3 where the x compound are CaF_2 , CaCl_2 , TiF_4 , ThF_4 , LaF_3 .

L 47331-65 EWP(e)/EWT(m)/EWP(i)/EWP(b) Pg-4 WH

Z/056/63/020/003/004/005
E073/E135

AUTHOR: Vogel, Z.

TITLE: Cutting properties of austenitic steels

PERIODICAL: Hutnictví a strojírenství. Přehled technické a
hospodářské literatury, v.20, no.3, 1963, 150,
abstract HS 63-1843. (Mechanik, Warsaw, v.35, no.9,
1962, 495-500)

TEXT: A study of the factors which lower the cutting
properties of austenitic steels and of the methods of improving
them so as to increase the service life of the tool cutting edge.
An analysis is made of the influence of stresses, hardness,
quantity of martensite, plastic deformation, etc. The cutting
properties are also studied with regard to surface smoothness.
10 figures. 5 tables. 17 references.

[Abstracter's note: Complete translation.]

Card 1/1

VOGEL, Zbigniew, mgr inz.

The machinability of austenitic steel. Mechanik 35
no.9:495-500 '62.

1. Politechnika Slaska, Gliwice.

Z. VOGEL

"Distribution of the Aesculapian snake (*Elaphe longissima*) in the territory of Czechoslovakia. p. 8. (CASOPIS;ODDIL PŘÍRODOVEDNÝ, Vol. 121, no. 1, 1952, Prague, Czechoslovakia)

SO: Monthly List of East European Accessions, L.C. Vol. 2 No. 7, July 1953, Uncl.

"Nature and its protection in Bulgaria.", p. 3, (COHEANA PRIRODY, Vol. 3, #1, Mar. 1953, Czechoslovakia)

SO: Monthly List of East European Accessions, Vol. 2, #2, Library of Congress, August 1953, Uncl.

CH

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ASB-SLA METALLURGICAL LITERATURE CLASSIFICATION

HIGH STRENGTH

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COLLECTION

HIGH BOMBY

HEAVY ONE ONLY 181

Combined distillation and evaporation apparatus, particularly for fermented molasses. Wilhelm Vogeltuch. Austrian 140,840, Feb. 28, 1935 (Cl. 62); - Addn. to 127,706 (C. A. 26, 4410).

1ST AND 2ND EDITIONS

PROCESSES AND PROPERTIES

3

24

The Raman spectra of some derivatives ClSO_2X of chlorosulfonic acid. Ruth Vogli-Hogler. *Ann. Acad. Sci. Wien., Math.-naturw. Klasse 81*, No. 7 8, 31 5 (1911). The Raman spectra and depolarizations of the compds. ClSO_2X , where $\text{X} = \text{OH}, \text{OCH}_3, \text{OC}_2\text{H}_5, \text{OC}_3\text{H}_7$ (*n*-propyl ester), OC_4H_9 (*n*-butyl ester), and Cl , are reported. The bond between S and O has a force constant between 9 and 10×10^9 dynes per cm. and corresponds to a double bond. H. L. Bernstein

ASTM S.L.A. METALLURGICAL LITERATURE CLASSIFICATION

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117 AND 119 (0010)										140 AND 141 (0001)									
PROCESSES AND PROPERTIES INDEX																			
<i>SA</i> 710. Negative Polarisation of Fluorescence bands of Adsorbed Dye-Substances at Low Temperatures. P. Fringsheim and H. Vogel. Acta Physica Polonica, 4.4. pp. 341-344, 1935. In German-- Experiments are described which establish the existence of negative polarisation (i.e., secondary vector perpendicular to the primary vector) in some fluorescence bands of some dye-substances both adsorbed on cellophane and when embedded in isotropic media (e.g., glycerin). <i>A 54 C</i>																			
ASD-51A METALLURGICAL LITERATURE CLASSIFICATION FROM SYMBOLS FROM SYMBOLS FROM SYMBOLS																			

VOGELTANZ, E.

The Brewery Museum in Plzen. Kvasny prum 9 no.9:222 S '63.

1. Plzenske pivovary, n.p., Plzen.

VOGENHUBER, R.

Problems of Austrian Machine Tool Industry
Huber (Metall, 1946, 7, 112, 161, 610-611) - R. V.

VOGINOV, S., inzh.

Mechanized swine-fattening farm for 2400 head. Sel'. stroi
no.12:4-5 D '62. (MIRA 16:1)

(Bryansk Province—Swine houses and equipment)

VOGINOV, S.

Economize everywhere and in everything. Sel'. stroi. 13 no.4:5-7
Ap '59. (MIRA 12:6)

1. Zamestitel' nachal'nika Bryanskogo oblastnogo upravleniya sel'skogo
khozyaystva.

(Bryansk Province---Farm buildings)

VOGINOV, S.

Year dedicated to new work methods. Sel'.stroi. 13 no.3:14-15
Mr '59. (MIRA 12:5)

1. Zamestitel' nachal'nika Bryanskogo oblastnogo upravleniya
sel'skogo khozyaystva.
(Dubrovka District--Building)

VOGINOV, S. [Vohynov, S.], inzh.; LOPUKHIN, M., inzh.; NOVALIKHIN, G.
[Novalykhin, H.], inzh.

Installing water-supply systems on farms without using metal
pipes. Sil'.bud. 9 no.10:10-14 0 '59. (MIRA 13:3)

1. Upravleniye stroitel'stva Bryanskogo oblastnogo upravleniya
sel'skogo khozyaystva RSFSR.

(Bryansk Province--Water supply, Rural)

VOGINOV, S.

Improving pipeless water-supply systems. Sel'.stroitel'. 15
no.6:5-6 Je '60. (MIRA 13:8)

1. Nachal'nik otдела stroitel'stva Bryanskogo oblastnogo
upravleniya sel'skogo khozyaystva.
(Bryansk District--Water supply, Rural)

VOGINOV, S.

Feeders with pipeless water supply for cattle. Sel'.stoi.
13 no.11:18-19 N '58. (MIRA 11:12)

1. Zamestitel' nachal'nika Bryanskogo oblastnogo upravleniya
sel'skogo khozyaystva.
(Cattle--Watering)

VOGINOV, Semen Mefod'yevich; BALAKIN, V.M., red.; SAYTANIDI, L.D.,
tekhn.red.

[Automatic watering of cattle without the use of metal pipes]
Ustroistvo avtomaticheskogo poeniia dlia skota bez primeneniia
metallicheskikh trub. Moskva, Izd-vo 14-va sel'.khoz.RSFSR.
1960. 21 p. (MIRA 14:1)
(Cattle--Watering)

VOGL, B.

Journal of
Vol. 187

and Steel Ind. Abstr.

The Styrian Ore Mines--The Deposits and the Present State
of the Workings. B. Vogl. *(Glasgow)*, 1953, 88, Sept., 937-
951). The postwar development of the Styrian iron ore
deposits, which include a high degree of mechanization, is
discussed in detail.--L. D. H.

VOGL, Ferenc

Complex mechanization and automation in the hosiery industry.
Magy textil 15 no.9:435-439 S '63.

FRANK, Tibor; SZASZ, Marton; MARK, Ferenc; BOSNYAK, Tamas; LUGOSI, Karoly;
FEKETE, Istvan; TOMPOS, Karoly; KABDEBO, Kornel; JAVOR, Bela; SCHEFTSIK,
Jeno; VOGL, Ferenc; REITER, Gyorgy

Conference on the current tasks of the light industry workers. Munka
szemle 5 no.3:5-7 Mr '61.

1. Textilipari Muszaki Tudomanyos Egyesulet Ipargazdasagi Szakosztalya
titkara (for Frank).
2. Kisepesti Textilgyar munkaugyi osztaly vezetoje
(for Mark).
3. Konnyuipari Miniszterium Munkaugyi es Oktatasi Onallo Osz-
talya vezetoje (for Szasz).
4. Ujpesti Gyapjuszovogyar munkaugyi osztalya
vezetoje (for Bosnyak).
5. Kender Juta es Textil Ipar munkaugyi osztaly
vezetoje (for Lugosi).
6. Kobanyai Textilgyar munkaugyi osztalya vezetoje
(for Fekete).
7. Konnyuipari Miniszterium Pamutipari Igazgatosaga mun-
kaugyi osztaly vezetoje (for Tompos).
8. Magyar Pamutipar munkaugyi osz-
talya vezetoje (for Kabdebo).
9. Majus 1 Ruhagyar munkaugyi osztalya
vezetoje (for Javor).
10. Konnyuipari Miniszterium Len-Kenderipari Igaz-
gatosaga munkaugyi osztalya vezetoje (for Scheftsik).
11. Ruhaiipari
Tervezo Vallalat (for Vogl).
12. Goldberger Textilmuvek munkaugyi foosz-
taly vezetoje (for Reiter).

VOGL, Ferenc

Experiences with organizing up-to-date, complex and large belts in the clothing industry. Magy textil 13 no.3:109-110 Mr '61.

1. Ruhaipari Tervezo Vallalat.

NEDOMA, W.; PAV, J.; VOGL, J.

Influence of the grate shape on cinder removal from a pressure-
generator model. Paliva 43 no.5:129-133 My '63.

1. Ustav pro vyzkum paliv, Bechovice.

VOGL, J.

Safety first.

P. 151, (Zeleznicar) No. 6, June 1957, Praha, Czechoslovakia

SO: Monthly Index of East European Acessions (EEAI) Vol. 6, No. 11 November 1957

VOGL, K.

Short-range hydrologic forecasts and their reliability. p. 65. Vol. 60, No. 2
Mar/Apr. 1956. IDOJARAS. Budapest, Hungary.

SOURCE: East European List, (EEAL) library of Congress Vol. 6, No. 1
January 1956.

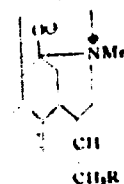
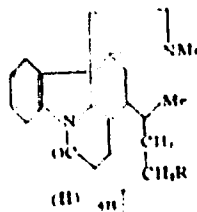
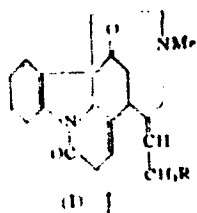
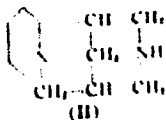
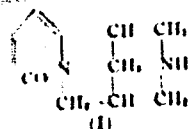
VOGL, M.

See: FOLDVARI-VOGL, Marie

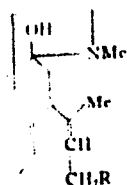
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Organic Chemistry 10

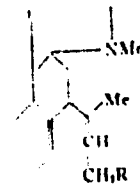
Synthesis of tetrahydrodeoxystrychnine. F. Galimovsky,
O. Vogl and W. Moras (Univ. Vienna). *Monatsh.* 83, 1



2H



2H
-H2O



over

242 8(1942) -- The structure of cytosine (I) was proved by synthesis of the closely related tetrahydrooxycytosine (II). 1,3-Dicarbethoxy-4H-quinolizin-4-one, m. 133-3° (from alc.) [prept. in 60% yield from Kt 2-pyridine acetate and $\text{EtOCH}_2\text{C}(\text{CO}_2\text{Et})_2$ (5.0 g.) in 35 ml. dioxane and 180 ml. alc. hydrogenated 22 hrs. at 20° and 730 mm. with 0.93 g. PtO_2 in 30 ml. alc. gave a colorless soln. from which was isolated a viscous oily mist. (III) of stereoisomeric 1,3-dicarbethoxyoctahydro-1H-quinolizin-4-one, partly crystg. on standing. Trituration with Et_2O and petr. ether and recrystn. from Et_2O gave 1 racemate, m. 80-1°. Crude III (10.04 g.) in 250 ml. Et_2O , added dropwise to 5 g. Li-AlH₄ in 250 ml. Et_2O and refluxed 8 hrs., yielded 6.3 g. (92%) octahydro-1,2-bis(hydroxymethyl)-1H-quinolizine (IV), b. 166-78°. IV with *N*-acetoxy gave a salt, m.

157-0°. By heating 4.7 g. IV and 100 ml. aq. HBr, satd. at 0°, in a sealed tube 10 hrs. on the steam bath, followed by evapn. *in vacuo*, the oily octahydro-1,2-bis(hydroxymethyl)-quinolizine-HBr (V) was isolated. The free di-HBr compd from V, b. 145-50°, could be obtained as a colorless oil but from V in 100 ml. abs. alc. satd. at room temp. on standing. Crude V in 100 ml. abs. alc. satd. at 0° with dry NH_3 , heated 4 hrs. on the steam bath in a sealed tube, the product extensively concd. *in vacuo*, acidified with HCl, the soln. evapd. to dryness, the residue, treated with KOH, steam-distill., and the distillate acidified and concd. gave an HCl salt which yielded 1.00 g. (38% based on IV) of a basic oily mist, b. 125-35°, of the 2 racemates of II, in 100 ml. alc. with 4.10 g. picric acid in 1.50 g. of the mist, immediately pptd. an oily picrate, and further heating gave a single racemic II picrate (VI), m. *in vacuo* 227-0° (decompn.) (from alc.-H₂O), which depressed the m.p. (231-2°) of the picrate of II derived from natural I. VI (2.0 g.) decoupled, with the exact amt. of 2 N HCl and the free base extd. and distd. twice, gave 0.633 g. of 1 race. mate of II, which, treated with natural II *D*-tartrate deposited ml. MeOH and recdled with natural II *D*-tartrate, m. 223-6° (decompn.) (from MeOH), identical with the natural II *D*-tartrate, as shown by mixed m.p. and $[\alpha]_D^{25}$ 37.9° (in H₂O). Free synthetic (+)-II (94 mg.) was prepd. from 180 mg. synthetic *D*-tartrate and proved by the following to be the same as the II derived from natural I: $[\alpha]_D^{25}$ 36.0° (abs. alc.); *dipicrate*, m. 230-2° (decompn.) (from alc.-H₂O); *dipicrate*, m. 230-40° (decompn.) (from alc.-H₂O); *N*-Ac deriv., b. 130-40°, m. 70-1° (from abs. alc.). Marie Mercury Roth

VIGL, J

✓ The structure of laburnine. P. Galinovsky, O. Vogt, and H. Nevadba (Univ. Vienna). *Monatsh.* 85, 949-19 (1954). To confirm the previously proposed (C.A. 44, 1484) structure of laburnine (I) as 1-hydroxymethylpyrrolizidine, it is converted by 2 methods into α -pyrrolidolactone (II). First, the *p*-toluenesulfonate of I is prepared by allowing 0.5 g. I in 2 ml. CH_2Cl_2 to stand overnight at 0° with half of a solution of 0.4 g. *p*-Toluenesulfonic acid in 3.7 ml. dry CH_2Cl_2 , and to stand 48 additional hrs. at room temp. with the remainder, distg. off the CH_2Cl_2 in *vacuo*, acidifying the aq. soln. of the residue with 10% NaOH , extg. with ether, making the aq. layer alk., drying its concd. ether ext. over Na_2SO_4 , adding 60 ml. abs. ether and refluxing the soln. 7 hrs. with 0.25 g. LiAlH_4 (Karrer, *et al.*, C.A. 46, 6591d) to give, after the usual decompn. and purification, 50-60% II, b.p. 65-75° (air bath temp.); $[\alpha]_D^{25}$ 17.1° (c 5.2, EtOH); picrate, m. and mixed m.p. 234-5° (decompn.); picrolonate, m. and mixed m.p. 164-5° (decompn.). Second, 0.27 g. I treated at 0° with 20 ml. satd. aq. HBr was heated 6 hrs. at 100° in a sealed tube, the solvent distd. off *in vacuo*, and the residual 1-bromomethylpyrrolizidine-HBr hydrogenated over PtO_2 in aq. soln. made weakly alk. with NH_3 to give from the ether ext. after 6 hrs. II, identical to the preceding. A similar procedure changed lupinine through its *p*-toluenesulfonate into lupanine, b.p. 80-5° (air bath temp.), $[\alpha]_D^{25}$ -8.8° (c 8.06, MeOH) (Karrer and Vogt, C.A. 25, 990); picrate, m. 185-6°; picrolonate, m. 236-7° (decompn.). A previous

prepn. (G., *et al.*, *loc. cit.*) of pyrrolizidine involved the oxidation of I to laburnic acid (III), and its immediate decarboxylation without isolation. The oxidation of I is here repeated to give almost 100% crude III, purified by sublimation in *vacuo* and recrystn. from EtOH-Et₂O to give III H_2O , m. 234-5° (c 4.44, MeOH); picrate, m. 235-6° (c 4.44, MeOH); picrolonate, m. 236-7° (c 4.44, MeOH). This work shows the confirmation of the relationship between I and II establishes not only the structure of I but also its configuration as having the CH₂OH group on the right of the methine group. Leonard and Fellay, C.A. 44, 6056h. H. S. French

(2)

11/4

S/035/62/000/011/035/079
A001/A101

AUTHOR: Vogl, W.

TITLE: Geodetic instruments at the technical fair at Leipzig

PERIODICAL: Referativnyy zhurnal, Astronomiya i Geodeziya, no. 11, 1962, 3,
abstract 11G21 ("Bergakademie", 1962, v. 14, no. 6, 465, German)

TEXT: The author notes that in addition to instruments manufactured by
the enterprises of the German Democratic Republic, the Leipzig 1962 fair (see
RZhAstr, 1962, 10G230) showed also instruments of the Hungarian optical enter-
prise MOM (1" theodolite Te-B1, small optical theodolite Te-E4, universal theo-
dolite Te-C1, level instrument Ni-B3, and plane table set MA) and Soviet instru-
ments (small optical theodolite TOM and theodolite with verniers).

V. M.

[Abstracter's note: Complete translation]

Card 1/1

VOGLAR, K.,
G. ZIGUNER, Monatsh. 82, 847-55 (1951)

VOGLER, Valter

The twelve-channel system Z 12 F. PTT zbor 16 no.1/2:41-42 F '62.

VOGIAN, D.,
V. GONCHAROV, Ognepory 4, 552-3 (1936)

1ST AND 2ND LETTER																										3RD AND 4TH LETTER																									
AUTHOR NAME																										TITLE																									
R Goncharov, V., Feodorov, K., and Vogman, D., U.S.S.R. <i>Proc. Acad. Sci. USSR Div. Chem. Phys. Sci., 4, 552 (1969).</i> With 10 to 20% of a spinel binder obtained by adding fused magnesite and Al_2O_3 to the batch, a highly refractory (over 1620°) material, resistant to deformation under load (beginning at 1730°), was obtained under laboratory conditions.																																																			

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VOGMAN, D.

Goncharov, V., Fendol'ev, K., and Vokmar, D. Use
OF A SPINEL BINDER FOR CHROMITE REFRACTORIES. *Ogne-
pory*, 4, 622-23 (1930).—With 10 to 20% of a spinel binder
obtained by adding fused magnesite and Al_2O_3 to the batch,
a highly refractory (over 1920°) material, resistant to de-
formation under load (beginning at 1730°), was obtained
under laboratory conditions.

VOGMAN, D.,
V. GONCHAROV, Ogneupory 4, 552-53 (1936)

BOLDYREV, G.P.; VOGMAN, D.A.; NOVOKHATSKIY, I.P.; VERK, D.L.; DYUGAYEV, I.V.; KAVUN, V.M.; KURENEO, A.A.; UZBEKOV, M.R.; ARSEN'YEV, S.Ya.; YEGORKIN, A.N.; KORSAKOV, P.P.; KUZ'MIN, V.N.; STRELETS, B.A.; PATKOVSKIY, A.B.; BOLESLAVSKAYA, B.M.; INDENBOM, D.B.; FINKEL'SHTEYN, A.S.; SHAPIRO, I.S.; LAFIN, L.Yu.. Prinimali uchastiye: NEVSKAYA, G.I.; FEDOSEYEV, V.A.; KASPILOVSKIY, Ya.B.; ZERNOVA, K.V.. BARDIN, I.P., akademik, otv.red.; SATPAYEV, K.I., akademik, nauchnyy red.; STRUMILIN, akademik, nauchnyy red.; ANTIPOV, M.I., nauchnyy red.; BELYANCHIKOV, K.P., nauchnyy red.; YEROFEYEV, B.N., nauchnyy red.; KALGANOV, M.I., nauchnyy red.; SAMARIN, A.M., nauchnyy red.; SLEDZYUK, P.Ye., nauchnyy red.; KHLIBNIKOV, V.B., nauchnyy red.; STREYS, N.A., nauchnyy red.; BANKVITSER, A.L., red.izd-va; POLYAKOVA, T.V., tekhn.red.

[Iron ore deposits in central Kazakhstan and ways for their utilization] Zhelezorudnye mestorozhdeniya Tsentral'nogo Kazakhstana i puti ikh ispol'zovaniia. Otvetstvennyi red. I.P.Bardin. Moskva, 1960. 556 p. (MIRA 13:4)

1. Akademiya nauk SSSR. Mezhduevedomstvennaya postoyannaya komissiya po zhelezu. 2. Gosudarstvennyy institut po proyektirovaniyu gorn'nykh predpriyatiy zhelezorudnoy i margantsevoy promyshlennosti i promyshlennosti nemetallicheskih iskopayemykh (Giproruda) (for Boldyrev, Vogman, Arsen'yev, Yegorkin, Korsakov, Kuz'min, Strelets, (Continued on next card)

BOLDYREV, G.P.--(continued). Card 2.

3. Institut geologicheskikh nauk AN Kazakhskoy SSR (for Novokhatskiy).
 4. Tsentral'no-Kazakhstanskoye geologicheskoye upravleniye Ministerstva geologii i okhrany nedr SSSR (for Verk, Dyugayev, Kavun, Kurenko, Uzbekov).
 5. Nauchno-issledovatel'skiy institut mekhanicheskoy obrabotki poleznykh iskopayemykh (Mikhanobr) (for Patkovskiy).
 6. Gosudarstvennyy institut proyektirovaniya metallurg.zavodov (Gipromez) (for Boleslavskaya, Indenbom, Finkel'shteyn, Nevskaya, Fedoseyev, Karpilovskiy).
 7. Mezhdunarodnaya postoyennaya komissiya po zhelezu AN SSSR (for Shapiro, Zernova, Kalganov).
 8. Gosplan SSSR (for Lapin).
- (Kazakhstan--Iron ores)

VOGMAN, D.,
GONCHAROV, V.V., (No Journal)

VOG-MAN, D.A.

POSPELOV, G.L., starshiy nauchnyy sotrudnik; LAPIN, S.S.; BELOUS, N.Kh.;
 KLYAROVSKIY, V.M.; KINE, O.G.; VAKHRUSHEV, V.A.; SHAPIRO, I.S.,
 starshiy nauchnyy sotrudnik; KALUGIN, A.S.; MUKHIN, A.S.; GARNETS,
 N.A.; SPEYT, Yu.A.; SELIVESTROVA, M.I.; RUTKEVICH, V.G.; BYKOV, G.P.;
 NIKONOV, N.I.; SAKOVICH, K.G.; MEDVEDKOV, V.I.; ALADYSHKIN, A.S.;
 PAN, F.Ya.; RUSANOV, M.G.; YAZBUTIS, E.A.; ROZHDESTVENSKIY, Yu.V.;
 SAVITSKIY, G.Ye.; PRODANCHUK, A.D.; LYSENKO, P.A.; LEBEDEV, T.I.;
 KAMENSKAYA, T.Ya.; MASLENNIKOV, A.I.; PIPAR, R.; DODIN, A.L.;
 MITROPOL'SKIY, A.S.; LUKIN, V.A.; ZIMIN, S.S.; KOREL', V.G.;
 DERBIKOV, I.V.; BARDIN, I.P., akademik, nauchnyy red.; GORBACHEV,
 T.F., nauchnyy red.; YEROFEYEV, N.A., nauchnyy red.; NEKRASOV, N.N.,
 nauchnyy red.; SKOBNIKOV, M.L., nauchnyy red.; SMIRNOV-VERIN, S.S.,
 nauchnyy red. [deceased]; STRUMILIN, S.G., akademik, nauchnyy red.;
 KHLIBNIKOV, V.B., nauchnyy red.; CHINAKAL, N.A., nauchnyy red.;
 SLEDZYUK, P.Ye., red.tome; SOKOLOV, G.A., red.tome; BOLDYREV, G.P.,
 red.; VOG-MAN, D.A., red.; KASATKIN, P.F., red.; KUDASHEVA, I.G.,
 red.izd-va; KUZ'MIN, I.F., tekhn.red.

[Iron-ore deposits of the Altai-Sayan region] Zhalezorudnye mesto-
 rozhdenia Altae-Saianskoi gornoj oblasti. Vol.1. Book 1. [Geology]
 (Continued on next card)

POSPELOV, G.L.---(Continued) Card 2.

Geologiya. Otvetstvennyi red. I.P. Bardin. Moskva. 1958. 330 p.
(MIRA 12:2)

1. Akademiya nauk SSSR. Mezhdovedomstvennaya postoyannaya komissiya po zhelezu. 2. Postoyannaya mezhdovedomstvennaya komissiya po zhelezu Akademii nauk SSSR (for Pospelov, Shapiro, Sokolov). 3. Zapadno-Sibirskiy filial Akademii nauk SSSR (for Vakhrushev, Pospelov.) 4. Zapadno-Sibirskoye geologicheskoye upravleniye (for Sakovich). 5. Krasnoyarskoye geologicheskoye upravleniye (for Pan). 6. Zapadno-Sibirskiy geologo-razvedochnyy trest Chernetrazvedka (for Prodanchuk). 7. Sibirskiy geofizicheskyy trest (for Pipar). 8. Vsesoyuznyy geologicheskyy nauchno-issledovatel'skiy institut (for Dodin). 9. Gornaya ekspeditsiya (for Mitropol'skiy). 10. Gornoye upravleniye Kuznetskogo metallurg.kombinata (for Lukin). 11. Tomskiy politekhnicheskyy institut (for Zimin). 12. Sibirskiy metallurg.institut (for Korel'). 13. Trest Sibneftegeofizika (for Derbikov). (Altai Mountains--Iron ores) (Sayan Mountains--Iron ores)

POZIN, Yu.M.; VOGMAN, M.Sh.; GAMASKIN, Ye.I.; BONDARENKO, O.I.

Preparing an electrode ribbon from cadmium oxide by rolling
powder composition materials between rolls. Porosh. met. 5
no.8:103-107 Ag '65. (MIRA 18:9)

1. Nauchno-issledovatel'skiy akkumulyatornyy institut.

L 1702-66 EWT(d)/EWP(e)/EWT(m)/EPF(c)/EMP(v)/EWP(t)/EWP(k)/EWP(h)/EWP(z)/EWP(b)/
EWP(1) IJP(c) JD/HW
ACCESSION NR: AF5020777 UR/0226/65/000/008/0103/0107 45

AUTHOR: Pozin, Yu. M.; Vogman, M. Sh.; Gamaskin, Ye. I.; Bondarenko, O. I.

TITLE: Producing an electrode strip from cadmium oxide by rolling powder compositions in rollers

SOURCE: Poroshkovaya metallurgiya, no. 8, 1965, 103-107

TOPIC TAGS: electrode, rolling mill, cadmium oxide, nickel compound, powder metallurgy

ABSTRACT: The general method for preparing the powder composition is as follows: cadmium oxide is mixed successively with solar oil and with a solution of nickel sulfate and is then passed through a 0.5x0.5 mm sieve and mixed with an aqueous solution of polyvinyl alcohol and then passed again through the same sieve. The finished electrode has dimensions of 35x70x1.9±0.2 mm, a weight of 15.0±0.5 grams, a porosity of 30%, and contains 1.9-2.1 grams cadmium/cm³. The present article considers methods of producing continuous electrode strips with better characteristics (thinner with a higher volumetric cadmium content, that is, more dense). The rolling unit did not differ from the standard type. To

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increase the friability of the composition, the cadmium oxide, before mixing with the other components, was rolled on rollers with a diameter of 60 mm, ground in a ball mill, and then passed through a 5x5 mm sieve. 10-20% of an aqueous solution of sodium-carboxy methyl cellulose was introduced into the composition, which was then dried to a residual moisture content of 3.0%. It is established that additions of nickel hydroxide and sodium-carboxymethyl cellulose improve the pressability of the composition. The strip can be obtained with different thicknesses and densities. The article also considers various mechanical methods for rolling and for cutting the strip into individual electrodes. Orig. art. has: 3 figures and 2 tables

ASSOCIATION: Nauchno-issledovatel'skii akkumulyatornyi institut (Scientific Research Institute for Accumulators)

SUBMITTED: 06Aug64

ENCL: 00

SUB CODE: MM

NR REF SOV: 003

OTHER: 000

Cord 2/2

VOGNAREK, J.

SURNAME (in caps); Given Names

Country: Czechoslovakia

Academic Degrees: [not given]

Affiliation: Internal Clinic III of Masaryk University (III. vnitřní klinika MU), Brno; Chief (Přednosta): Prof Dr Jaroslav Pojer

Source: Brno, Vnitřní Lékařství, Vol VII, No 8, August 1961, pp 841-848

Data: "Hepatic Changes in Obesity"

Authors:

POJER, J
MASTIK, C
DRAŽILOVA, V

VOGNAREK, J., Technical collaborator

DOLEZAL, Bohuslav, MDr., C.Sc.; VOGNAROVA, Irena, inz.

Comparison of natural and artificial casings for meat products.
Prum potravin 13 no.12:628-630 D '62.

1. Vyzkumny ustav pro maso, Brno.

DVORAK, Zdenek; VOGNAROVA, Irena

Effect of the Maillard reaction and smoking on the biologic properties of meat and meat products. Prum potravin 16 no.4: 172-176 Ap '65.

1. Research Institute of Meat, Brno. Submitted September 30, 1964.

VOGNIC, Hubert

Automatic control and recording of mold heating in thermoset pressing.
Energetika Cz 12 no.9: 78-479 S '62.

1. Lisovny novych hmot, n.p., Vrbno pod Pradadem.

~~VANECHNIK,~~
~~VOGNOUTS,~~

VANECHNIK, Yu.; KYUKHEL', O.; VOGNOUT, S.; REGLING, S.

Control of completeness of isolation of organs in situ using tagged atoms; so-called humoral isolation of the vascular system. Farm. i toks. 19 no.5:44-50 S-O '56. (MLRA 10:3)

1. Kafedra farmakologii (zav. - dotsent Ye. Bashkova) pediatricheskogo fakul'teta Karlova universiteta v Prage, III klinika po vnutrennim zabolevaniyam (zav. - akademik Kharvat) meditsinskogo fakul'teta Karlova universiteta v Prage i Endokrinologicheskii institut v Prage (zav. - dotsent Shilink)

(PERFUSION,

radio-isotope labeled substances in control of completeness of isolation of perfused organs in situ (Rus))

(ISOTOPES,

same)

VOGRADSKIY, F.; SAFAR-ALIYEVA, I. [translator]

Prevention of infectious atrophic rhinitis in swine. Veterinariia
41 no.5:112-113 May '64. (MIRA 18:3)

VOGRALID, V. G.

20086 VOGRALID, V. G. O vzaimootnoshenii mekhan bronkhial'noy astmy i gipertonicheskoy boleznyu. Vracheb. delo, 1949, No. 6, str. 521-24.

SO: LETOPIS ZHURNAL STATEY, Vol. 27, Moskva, 1949.

VOGRALIK, M.V.

Oxygen tension and conditions of the blood supply in transplanted
tumors. Vop. onk. 11 no.3:73-77 '65. (MIRA 18:6)

1. Iz kafedry patofiziologii (zav. - dotsent V.Ya. Batunina)
Gor'kovskogo meditsins'ogo instituta imeni Kirova (rektor -
dotsent I.F. Matyushin).

VOGRALIK, M.V., student VI kursa

Vascular crisis in hypertension. Terap.arkh. 30 no.9:52-56 S'58
(MIRA 11:10)

1. Iz gosspital'noy terapevticheskoy kliniki (zav. - prof. V.G.
Vogralik) Gor'kovskogo meditsinskogo instituta.
(HYPERTENSION, compl.
vasc. crisis (Rus))

VOGRALIK, M.V., aspirant

Change in skin sensitivity and peripheral vegetative reactivity
in acupuncture. Uch. zap. GMI no.8:13-18 '59. (MIRA 14:9)

1. Iz kafedry patofiziologii (ispolnyayushchiy obyazannosti
zaveduyushchego kafedroy - dotsent V.Ya. Batunina) i gosital'noy
terapevticheskoy kliniki (zav. - prof. V.G.Vogralik).
(ACUPUNCTURE) (NERVES, PERIPHERAL)

VOGRALIK, M.V.

Technique of double stimulation in electroencephalography as a method to evaluate the reactivity of the central nervous system in man. Terap. arkh. 35 no.1:78-83 Ja '63. (MIRA 16:9)

1. Iz kafedry patofiziologii (zav. - dotsent V.Ya. Batunina) i kafedry gospiatal'noy terapii (zav. - prof. V.G.Vogralik) Gor'kovskogo meditsinskogo instituta imeni S.M.Kirova.
(ELECTROENCEPHALOGRAPHY) (BRAIN)

VOGRALIK, M.V.

Materials on a further study of the dynamics of the bio-electric activity of the brain in local stimulation of the receptors and nerves of the body integuments (dynamics of the rhythm adaptation). Sbor. trud. GMI no.9:29-35 '62.
(MIRA 17:2)

1. Iz kafedry patologicheskoy fiziologii (zav. - dotsent V.Ya. Batunina) i gosspital'noy terapevticheskoy kliniki (zav. - V.G. Vogralik) Gor'kovskogo meditsinskogo instituta imeni S.M. Kirova.

VOGRALIK, M.V. (Gor'kiy)

Clinical use of sulfanilamide preparations for the treatment
of diabetes mellitus. Klin.med. 36 no.1087-95 0'58(MIRA 11:11)

1. Iz kafedry gosital'noy teranii (zav. - prof. V.G. Vogralik)
Gor'kovskogo meditsinskogo instituta imeni S.M. Kirova.
(ANTIDIABETICS, ther. use
carbutamide & tolbutamide (Rus))