

FEDELESOVA, M.; ZIEGELHOPFER, A; HUEKA, M.; Technicka spolupraca:  
CEENUSAKOVA,M.; HROCHOVA, L.; BRICHTOVA, A.

A study of the changes of various substrates and of enzyme  
activity in mitochondria of the isolated dog heart after  
hypothermic storage. Bratisl. lek. listy 45 no.5:265-272  
15 Mr '65

1. Ustav experimentalnej chirurgie Slovenskej akademie ved  
(riaditeľ: akademik K. Siska).

ZIEGELHOFFER, A.

Use of an antimony electrode for measurement of the blood pH. Bratisl.  
lek. listy 42 no.4:199-202 '62.

1. Z oddelenia experimentalnej chirurgie (veduci clen korespondent  
CSAV K. Siska) Ustavu experimentalnej mediciny SAV, riaditeľ clen  
koresp. SAV J. Antal.

(HYDROGEN ION CONCENTRATION blood) (ANTIMONY)

HUBKA, M.; SUJANSKY, E.; SILVAY, J.; FEDELESOVA, M.; ZIEGELHOPFER, A.

Current status of the problem of artificial asystoles. Bratisl.  
lek. listy 43 Pt. 2 no.4:185-189 '63.

1. CSAV - Oddelenie experimentalnej chirurgie Ustavu experimentalnej mediciny SAV v Bratislave, veduci akademik CSAV K. Siska.

(HEART ARREST) (HEART SURGERY)  
(HEART, MECHANICAL) (HYPOTHERMIA, INDUCED)  
(POTASSIUM) (MAGNESIUM SULFATE) (NEOSTIGMINE)

HUBKA, M.; PEDELESOVA, M.; ZIEGELHOFER, A.; SUJANSKY, E.; SILVAY, J.

Changes in glycide and energy metabolism of the myocardium  
during artificial asystoles under experimental conditions.  
Bratisl. lek. listy 43 Pt. 2 no. 4:189-196 '63.

1. CSAV - Oddelenie experimentalnej chirurgie Ustavu experimen-  
talnej mediciny SAV v Bratislave, veduci akademik CSAV  
K. Siska.

(HEART ARREST) (HEART MECHANICAL) (MYOCARDIUM)  
(ENERGY METABOLISM) (HYPOTHERMIA, INDUCED)  
(CARBOHYDRATE METABOLISM) (GLUTATHIONE)  
(ASPARTATE AMINOTRANSFERASE)  
(ADENINE NUCLEOTIDES)

HUBKA, M.; ZIEGELHOFFER, A.; FEDELESOVA, A.; SILVAY, J.; SUJANSKY, E.

Changes in the acid-base equilibrium and concentration of cations in artificial asystoles under experimental conditions.  
Bratisl. lek. listy 43 Pt. 2 No.4:197-204 '63.

1. CSAV - Oddelenie experimentalnej chirurgie Ustavu experimentalnej mediciny SAV v Bratislave, veduci akademik CSAV K. Siska.

(ACID-BASE EQUILIBRIUM) (HEART ARREST)  
(HYPOTHERMIA, INDUCED) (OXIMETRY) (SODIUM)  
(POTASSIUM) (CALCIUM) (HEART, MECHANICAL)

*10 GELHOFER, A.*

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1. "Experience With A Percutaneous Transabdominal Cholecystectomy," R. KOMORNÍK, presumably of the Research Institute for Gastroenterology, University of Bratislava, Corresponding Member of SAV, Corresponding Member of the Academy of Sciences, V. ALMOSO, Director of the János Bajcsy and J. ŠIMIČ, MD, Interna Klinik, Department of Internal Medicine Faculty of Medicine Comenius University, Bratislava, Prof. T. H. LIND, MD, director (see also) pp 153-158.

2. "Applicability of an Autogenous Electrode for the Measure of Blood PH." A. ČIKLÍKOVÁ, of the Department of Experimental Surgery (Graduate experimental clinic), Corresponding Member of the Academy of Sciences, Institute of Experimental Medicine, SAV (see also) pp 159-160. English summary.

3. "Tumourcharacteristics of the Carotid Artery, Participating in the Development of Tumours in the Zora Clinic of Surgery," J. ČERNÝ and R. KUDACKO, of the Zora Clinic of Surgery, University of Charles University, Faculty of Medicine (see also) pp 171-176. English summary.

4. "Some Problems of Radical Radiotherapy of Bronchial Carcinoma," B. KURNAJAD, Faculty of Medicine (Radiation Institute), P.J. Šafárik University, Bratislava, Prof. B. KURNAJAD, MD, director; pp 211-220. English summary.

5. "Electrophysiological Examination of the Muscular Tonus During Control of Clinical Phenomena," P. KOVÁČIK, of the Department of Clinical Physiology, J. ČERNÝ, Corresponding Member of the Academy of Sciences, Director of the Institute of Experimental Medicine, SAV, Corresponding Member of the Academy of Medical Sciences (Slovenakia) and J. ŠIMIČ, MD, Interna Klinik, Department of Internal Medicine Faculty of Medicine Comenius University, Bratislava, Corresponding Member of SAV, Director; pp 221-226.

SPANAR, Eugen, MUDr.; ZIEGELHOFFEROVA, M.

Anabolism of testosterone by proteins and its use in  
asthenia. Cas. lek. cesk. 91 no.29:861-867 18 July 52.

1. Z Endokrinologickeho liecebneho ustavu v Lubochni.  
(TESTOSTERONE, therapeutic use,  
asthenia, anabolism by proteins.)  
(ASTHENIA, therapy,  
testosterone, anabolism by proteins.)  
(PROTEINS,  
anabolism of testosterone in ther. of asthenia.)

SPANAR, E.; VARGA, I.; KELLEN, J.; DUBAJ, J.; ZIEGELHOFFIREOVA, M.

An attempt to evaluate the chromatographic differentiation of  
17-ketosteroids in pulmonary tuberculosis. Bratisl.lek.listy 35  
no.6:321-336 31 Mar 551

1. Z endokrinologickeho liecbevneho ustavu v Lubochni prednosta dr.  
Eugen Spanar, a z plucneho oddelenia nemocnice v Ruzomberku, pred-  
nosta dr. Imrich Varga.

(URINE,

17-ketosteroids, chromatographic differentiation in pulm.  
tuberc.)

(TUBERCULOSIS, PULMONARY, urine in,

17-ketosteroids chromatographic differentiation)

SPANAR, E.; KELLEN, J.; DUBAJ, J.; ZIEGELHOFFEROVA, M.

Studies on pathogenesis of asthenia. Bratisl. lek. listy 34  
no.4:377-389 Ap '54.

1. Z Endokrinologickeho liecebneho ustavu v Labochni, prednosta  
dr. E.Spanar.  
(ASTHMA, etiology and pathogenesis.)  
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*CA**10*

A simple synthesis of multi-substituted ethane, I  
Tetra(*p*-hydroxyphenyl)ethane, R. Ziegler and W. Kleim  
entschits (Univ., Graz, Austria); *J. Macromol. Sci.*, B1, 413-49  
(1968). — Heating of 120 g. PhONa and 270 cc.  $\text{CH}_2\text{Cl}_2\text{CHCl}_2$   
4 hrs. in an autoclave at 175-80°, treatment with  $\text{H}_2\text{O}_2$ , and  
fractional distn. *in vacuo* yields 30 g. dichloromethyl Ph  
ether (I) (Swiss 249,117; C.A. 43, 5008b), m.p. 6°. Hydro-  
genation of I in the presence of PtO<sub>2</sub> gives PhOH. Treatment  
of I (20 g.) with HCl gas for 30 hrs. under irradiation  
yields after fractional distn. *in vacuo*, dichloromethyl Ph  
ether (II), m.p. 118-20°. Heating 2 g. I with 2 g. PhOH 4  
hrs. yields 60% 1,1,2,2-tetrakis(*p*-hydroxyphenyl)ethane  
(III), m. 200-300°. III (60%) is also obtained from II  
with PhOH. Methylation of III with Me<sub>3</sub>SiO yields 1,1-  
2,2-tetrakis(*p*-methoxyphenyl)ethane (IV) (Bergmann and  
Fujie, C.I. 23, 912); ethylation yields the 1,10 compd.  
[Gattermann, *Ber.*, 28, 2875 (1895)]; acetylation gives  
1,1,2,2-tetrakis(*p*-acetoxyphenyl)ethane (V), m. above 300°;  
reaction with NaNO<sub>2</sub> yields probably 1,1,2,2-tetrakis(3,  
nitro-4-hydroxyphenyl)ethane, yellow rods, m. above 300°;  
and oxidation with CrO<sub>3</sub> gives ( $\Delta$ -MeOC<sub>2</sub>H<sub>3</sub>)<sub>2</sub>CO and anisic  
acid. III, IV, and V had no estrogenicity.

Peter Betzfeld

*1951*

CA

*Organic Chemistry - 10*

Cleavages with diazonium compounds. XI. Phenolphthalein. Relationship between constitution and biologic action. H. Ziegler and H. Toppier (Univ. Graz, Austria). *Z. Naturforsch.* **7B**, 122-4 (1952); cf. *C.A.* **45**, 10210e.—Phenolphthalein (**I**) and 2 mols. *p*-O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>Cl reacted in 5% NaOH, with sufficient NaOH added during the reaction to maintain the alky., to yield 10% 2,4-(*p*-O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>Cl)<sub>2</sub>OH, m. 218°; *acetate*, yellow-red needles, m. 160-70°. *p*-Cresolphthalein and 2.2 mols. PhNiCl in 4 mols. 5% NaOH gave a product yielding on recrystn. from AcOH 3.51-[2(PhN<sub>2</sub>)<sub>2</sub>C<sub>6</sub>H<sub>4</sub>(OH)<sub>2</sub>], m. 116-17°, and the AcOH filtrate contained an unknown substance, m. 217-8°. The relationship between the reactive groups of **I** and its pharmacol. action is discussed. A. Dietz

CA

Syntheses and conversions of tertiary aliphatic-aromatic alcohols of the ethylene series II. Action of sulfuric acid on methylphenylvinylcarbinol and methyliobenzylvinylcarbinol. A. I. Lebedeva (A. A. Zhdanov State Univ., Leningrad). J. Gen. Chem. U.S.S.R. 20, 431-6 (1950) (Engl. translation).—See C.A. 44, 7732g.

Cleavage with diazonium compounds. VI. Hydroxybenzalamines and their *N*-acyl derivatives. G. Ziegner, E. Ziegler, F. Aspern, and E. Wiesenberg (Univ. Graz, Austria). Monatsh. 81, 181-7 (1950); cf. C.A. 45, 1486. Phenols and  $\alpha$ -Carboxylic acids (ROH) in  $\text{ClCH}_2\text{CH}_2\text{CONHCH}_2\text{OH}$  (R'OH) in concd. HCl form R- and R'-CONHCH<sub>2</sub>OH (R'OH) radicals and hydrolysis of 2,4-R-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>NH<sub>2</sub>HCl (R') radicals gives 2-HOCC<sub>6</sub>H<sub>3</sub>CONHCH<sub>2</sub>OH (I), m. 137°, which, coupled with  $\rho$ -O<sub>2</sub>NCH<sub>2</sub>NiCl (III), gives 2,4-Me<sub>2</sub>O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH (IV), m. 107°. I is similarly prep'd. from  $\rho$ -MeOC<sub>6</sub>H<sub>4</sub>OH and ROH. Hydrolysis of I gives 2,4-R''-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH (V), m. 106°. V coupled with Ph<sub>2</sub>N<sub>2</sub>Cl (VI) gives 2,4,6-R'''-Me<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, m. 106°.  $\alpha$ -RC<sub>6</sub>H<sub>4</sub>OH and III give 2,4-R''-O-N<sub>2</sub>H<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, m. 230-1°. The groups are replaced by diazonium cumps. in alk. soln. from the para position of phenols and from the 1-position of naphthols. They are not split out from the ortho position if there is a  $\beta$ - or  $\alpha$ -H to be replaced. Thus,  $\rho$ -MeOC<sub>6</sub>H<sub>4</sub>OH and ROH give 4,2-R-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH (II), m. 107°, which, coupled with  $\rho$ -O<sub>2</sub>NCH<sub>2</sub>NiCl (III), gives 2,4-Me<sub>2</sub>O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH (IV), m. 106°. I is similarly prep'd. from  $\rho$ -MeOC<sub>6</sub>H<sub>4</sub>OH and ROH. Hydrolysis of I gives 2,4-R''-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH (V), m. 106°. V coupled with Ph<sub>2</sub>N<sub>2</sub>Cl (VI) gives 2,4,6-R'''-Me<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, m. 106°.  $\alpha$ -RC<sub>6</sub>H<sub>4</sub>OH and III give 2,4-R''-O-N<sub>2</sub>H<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, m. 230-1°. The

following were prep'd. (phenol, coupling agent, product, and m.p. of product, resp.): 1,2- $\text{Cl}_2\text{C}_6\text{H}_4\text{OH}$ , V, 1,2-PPh<sub>2</sub>, 1,2-C<sub>6</sub>H<sub>4</sub>OH, (VI), m. 130°; 1,2-R''-C<sub>6</sub>H<sub>4</sub>OH, V, VII, 2,4,6-Me<sub>2</sub>R-C<sub>6</sub>H<sub>3</sub>OH (m. 116°); V, 2,3,6-HO<sub>3</sub>C<sub>6</sub>H<sub>2</sub>N<sub>2</sub>Ph (VII), m. 80°; 2,4,6-Me<sub>2</sub>R-C<sub>6</sub>H<sub>3</sub>OH (m. 230°); V, VIII, 2,4,6-Me<sub>2</sub>R-C<sub>6</sub>H<sub>3</sub>OH (m. 103-4°); III, 2,4,6-Me<sub>2</sub>R'-O-N<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH (m. 216-17°); 2,4,6-Me<sub>2</sub>R'-C<sub>6</sub>H<sub>3</sub>OH, m. 224-222° (decomp.); V, 2,4,6-Me<sub>2</sub>R'-PhN<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, m. 224-5°; 4,2,6-Me<sub>2</sub>R'-C<sub>6</sub>H<sub>3</sub>OH (m. 106°); V, 4,2,6-Me<sub>2</sub>PPh<sub>2</sub>R'-C<sub>6</sub>H<sub>3</sub>OH, m. 265°; V, 4,2,6-Me<sub>2</sub>PPh<sub>2</sub>R'-C<sub>6</sub>H<sub>3</sub>OH, m. 106°; V, 4,2,6-HOMe<sub>2</sub>C<sub>6</sub>H<sub>3</sub>N<sub>2</sub>Ph (VIII), m. 92°; 2,4,6-HOMe<sub>2</sub>C<sub>6</sub>H<sub>3</sub>N<sub>2</sub>Ph, 2,4-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, and HCl, with a few drops conc. HCl give 2,3,6-HO<sub>3</sub>C<sub>6</sub>H<sub>2</sub>N<sub>2</sub>Ph-C<sub>6</sub>H<sub>3</sub>OH (VIII), m. 182°; the free base of VIII gives a deep red color upon coupling with III. An alk. soln. of  $\beta$ -HOCH<sub>2</sub>CH<sub>2</sub>ONH<sub>2</sub>HCl coupled with III gives a deep red color.  $\beta$ -HOCH<sub>2</sub>CH<sub>2</sub>ONH<sub>2</sub>HCl coupled with III gives a deep red color.  $\beta$ -HOCH<sub>2</sub>CH<sub>2</sub>ONH<sub>2</sub>HCl, m. 213°, and 2,4,6-( $\rho$ -O<sub>2</sub>NCH<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>3</sub>OH, m. 273°) were isolated from the reaction. VII. Mechanism for the action of phenolic purgatives. E. Ziegler, G. Ziegner, and F. Zeidler. *Scientia Pharm.* 17, 37-42 (1949).—Bis( $\rho$ -hydroxyphenyl)methanes are cleaved by diazonium salts; their di-Me ethers do not react, and the mono-Me ethers of some couple in the normal manner. There is a correlation between tendency to be cleaved and purgative activity in this series. From 3,3-bis( $\rho$ -hydroxyphenyl)-2-indolone is obtained 4,4-*p*

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[3,5,4-Me<sub>2</sub>(p-O<sub>2</sub>NCH<sub>2</sub>NH<sub>2</sub>)C<sub>6</sub>H<sub>3</sub>]OH, m. 213-14°, or 2,4-(p-O<sub>2</sub>NCH<sub>2</sub>NH<sub>2</sub>)C<sub>6</sub>H<sub>3</sub>OH (I), m. 278°, [3,5,4-Me<sub>2</sub>(p-O<sub>2</sub>NCH<sub>2</sub>NH<sub>2</sub>)C<sub>6</sub>H<sub>3</sub>]OH, yields 2,6,4-Me<sub>2</sub>(p-O<sub>2</sub>NCH<sub>2</sub>NH<sub>2</sub>)C<sub>6</sub>H<sub>3</sub>OH, m. 173-5°. The compd. described by Braun (C.A. 23, 4087) is presumably 1,3-bis(p-hydroxyphenyl)-1,3-dimethylethaneobutane, since it cleaves, forming I. The compd., which Meyer, *et al.* (C.I. 8, 2878), obtained from [2,4-(p-N<sub>2</sub>S)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>N<sub>2</sub>]SO<sub>3</sub>, m. 310-17° and analyzed for C<sub>11</sub>H<sub>14</sub>N<sub>2</sub>Na<sub>2</sub>. The following compds. are also reported: 4,2,6-Me(*p*-MeOC<sub>6</sub>H<sub>3</sub>CH<sub>2</sub>Cl)<sub>2</sub>ONa, decomp. 299-300°, from *p*-cresol, *p*-MeOC<sub>6</sub>H<sub>3</sub>CH<sub>2</sub>Cl, and Na<sub>2</sub> in C<sub>6</sub>H<sub>6</sub>; Di-Me ether, m. 81.5-82°, and mono-Me ether, m. 87-8°, of [3,5,4-Me<sub>2</sub>(HO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>]OH, 4-(*p*-MeOC<sub>6</sub>H<sub>3</sub>CH<sub>2</sub>Cl)<sub>2</sub>ONa, m. 83-4°; *p*-NO<sub>2</sub>C<sub>6</sub>H<sub>3</sub>N<sub>2</sub> deriv., m. 149-50°; 4-Methoxy-2'-hydroxy-3'-(*p*-nitrophenylazo)-5'-methyl-diphenylmethane, m. 160.5°. VIII. *p*-Cresolphthalein, E. Ziegler and G. Zigeuner, *Ibid.* 113-17 — *p*-MeC<sub>6</sub>H<sub>3</sub>N<sub>2</sub>Cl and *p*-cresolphthalein (I) in 10% NaOH form 2-[3,4-Me<sub>2</sub>(HO)<sub>2</sub>C<sub>6</sub>H<sub>3</sub>CO]<sub>2</sub>C<sub>6</sub>H<sub>3</sub>CO<sub>2</sub> (II), m. 223°, 6,2,4-Me<sub>2</sub>(*p*-MeC<sub>6</sub>H<sub>3</sub>N<sub>2</sub>Cl)<sub>2</sub>OH (III), m. 163° (acetate, m. 147°), and mono-*p*-tolylazo-*p*-cresolphthalein (IV), m. 273°. *p*-O<sub>2</sub>NCH<sub>2</sub>NH<sub>2</sub>Cl (V) (2.5 mols.) and I in 3% NaOH form only the analog of IV, m. 237°. In 10% NaOH, 4 mols. of V cleave I completely to II and the analog of III, m. 235-7°.

John Howe Scott

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Cleavage with diazonium compounds. II. Phenolsulfone compounds. R. Ziegler, O. Ziegner, K. Wiesenthaler, and A. Kainzner (Univ. Graz), Monatsh., 62, 234-44 (1931); cf. C.A. 43, 6170a. — Compds. of the types  $(\rho\text{-HOC}_2\text{H}_5)_2\text{CH}_2\text{OH}$  (I), and  $\rho\text{-HOC}_2\text{H}_5\text{CH}(\text{OH})_2\text{Ph}$  are cleaved by diazonium salts in alc. soln.;  $\rho\text{-HOC}_2\text{H}_5\text{CH}_2\text{Ph}$  and  $(\rho\text{-HOC}_2\text{H}_5)_2\text{CO}$  (II) are not. Compds. of the type  $(\rho\text{-HOC}_2\text{H}_5)_2\text{S}$  resemble I in being cleaved by  $(\rho\text{-O}_2\text{NC}_2\text{H}_5)_2\text{SO}_2$  (III), but the related sulfones and sulfoxides can not differ as much from III as is often assumed because, like II, they can not be cleaved.  $\rho\text{-MeC}_2\text{H}_5\text{OH}$  and  $\text{SOC}_2$  form  $[3,4\text{-Me}(\text{HO})\text{C}_2\text{H}_5]_2\text{S}\text{Cl}$ , decomp. 231°, which also can not be cleaved.  $(\rho\text{-HOC}_2\text{H}_5)_2\text{S}$  and III give 2,4-( $\rho\text{-O}_2\text{NC}_2\text{H}_5)_2\text{N}$ ) $\text{C}_2\text{H}_5\text{OH}$ , m. 278°. [3,4-Me( $\text{HO}\text{C}_2\text{H}_5)_2\text{S}$ ] (IV) and III form 6,2,4-Me( $\rho\text{-O}_2\text{NC}_2\text{H}_5)_2\text{C}_2\text{H}_5\text{OH}$ , m. 261-6°. Oxidation of IV with 30%  $\text{H}_2\text{O}_2$  in HOAc forms the sulfoxide, m. 177.5°. In an attempt to prep. a hydroxybenzyl mercaptan, 3,4,5,2-BrMe( $\text{HO})\text{C}_2\text{H}_5\text{CH}_2\text{Br}$  was treated with  $\text{KSC}_2\text{OEt}$  in  $\text{Me}_2\text{CO}$ , forming the ethyl xanthogenate (V), m. 87.5°. Hydrolysis of V with alc.  $\text{NaOH}$  formed (3,4,5,2-BrMe( $\text{HO})\text{C}_2\text{H}_5)_2\text{S}$  (VI), m. 94.5°. Similarly 3,2,6,4-BrMe( $\text{HO})\text{C}_2\text{H}_5\text{CH}_2\text{Br}$  (VII) formed the  $\beta$ -analogs of V and VI (VIII), m. 60° and 154°, resp. VII and NaSH in eq.  $\text{Me}_2\text{CO}$  formed VIII. 6,2,3-Me( $\text{HO})\text{C}_2\text{H}_5\text{CH}_2\text{Cl}$  (IX) likewise gave the analogs of V and VI, m. 66° and 175°, resp. However IX differs from VII in forming with NaSH the benzyl mercaptan, m. 104°. 2,6,4-( $\text{C}_2\text{H}_5)_2\text{MeC}_2\text{H}_5\text{OH}$  treated similarly formed the 2,5-( $\text{BrOCS}_2$ ) compd., m. 67.5°, which on hydrolysis formed 3,3,5- $\text{HSCH}_2\text{HO}(\text{MeC}_2\text{H}_5)_2\text{SCH}_2\text{HO}(\text{MeC}_2\text{H}_5)_2\text{SCH}_2\text{C}_2\text{H}_5\text{Me}(\text{OH})\text{CH}_2\text{SH}$ -5,2,3, m. 157°. These thioethers are cleaved by diazonium salts. Structure of

phenolphthalein in alkaline medium. M. Ziegler, H. Tappeler, and M. Sotocik. *Schenk. Pharm.* 19, 81-93 (1961). —  $\text{PAn}_2\text{Cl}$  (X) and phenolphthalein (XI) (3 equivs.:1) do not react in 4 hrs. contg. 10 equivs. of  $\text{NaOH}$ . In one contg. 12 moles, they form 10%  $\rho\text{-Ph}_2\text{N}\text{C}_2\text{H}_5\text{OH}$  (XII), m. 151°, 6% 2-( $\rho\text{-HOC}_2\text{H}_5)_2\text{CO}\text{C}_2\text{H}_5\text{CO}_2\text{H}$  (XIII), m. 210-12°,

and 25%  $\alpha\text{-C}_2\text{H}_5\text{CO}_2\text{C}(\text{C}_2\text{H}_5\text{OH})\text{N}(\text{Ph}-4\text{-Cl})_2$  (XIV); in one contg. 8 moles, 17%; XII, 17%; XIII, and 45% XIV; in one contg. 3 or 4 moles, 80-100% XIV; in one contg. 2 moles, if carefully buffered, 70% 2-( $\rho\text{-HOC}_2\text{H}_5)_2\text{C}_2\text{H}_5\text{OH}$ , m. 133-3°. XIV, m. 242-3°, was shown to be the sym. compd. by reaction with  $\text{H}_2\text{NOH}$  in KOH soln., formation of the 2-hydroxyindolinone (XV), m. 184-5°, and cleavage with alc.  $\text{H}_2\text{SO}_4$  to (4,3-HO( $\text{Ph}_2\text{N}\text{C}_2\text{H}_5\text{OH})_2\text{C}_2\text{H}_5\text{CO}_2\text{H}$ ) (XVI), m. 130-3°, and 2-(4,3-HO( $\text{Ph}_2\text{N}\text{C}_2\text{H}_5\text{OH})\text{C}_2\text{H}_5\text{CO}_2\text{H}$ ) (XVII), m. 185°. Similar treatment of dinitrophenolphthalein showed that it also is sym. The analogs of XV, XVI, and XVII m. 208.5-7°, 129°, and 180° (Et ester, m. 111-13°), resp. Use of a 1:1 ratio of X and XI did not form a monosubstituted XII similar to XIV. These results are interpreted as meaning that the colored form of XI can not have a permanently fixed quinonoid ring in its formula. Z. favors the usual resonance formulation involving both rings as partially quinonoid and an intermediate carbocation. The two colorless forms present in dil. and concd. alkali are of type I or closely resemble it and so undergo cleavage. For them the customary lactone and triphenylcarbinol structures are adopted. The failure of X and XI to react in very concd. alkali is not clearly understood since III is able to react under these conditions and  $\text{PbO}_2$  and X couple under these conditions also. J. H. Scott

CH

10

**Cleavage with the help of diazonium compounds. IV.**  
**Hydroxybenzenophenones.** G. Zigeuner and H. Ziegler (Univ., Graz, Austria). Monatsh. 80, 340-43 (1949); Oester. Akad. Wiss., Math.-Naturw. Klasse, Sitz.-Ber. Abt. IIb, 158, 350-63 (1949); cf. C.A. 44, 1404e.—Like  $\sigma$ -( $p$ -HOC<sub>6</sub>H<sub>4</sub>CO)<sub>2</sub>C<sub>6</sub>H<sub>4</sub>COOH,  $\beta$ -HOC<sub>6</sub>H<sub>4</sub>COPh fails to couple with diazonium compds. ( $p$ -O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>Cl) (I), 2,4-(O<sub>2</sub>N)<sub>2</sub>C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>Cl). ( $p$ -HOC<sub>6</sub>H<sub>4</sub>CO) (II) is likewise inert.  $\rho$ -HOC<sub>6</sub>H<sub>4</sub>CH(OH)Ph in NaOH with I is cleaved into 4-hydroxy- $\sigma$ '-nitroazobenzene, m. 213° (from PhMe or PhNO<sub>2</sub>), and BiPh (2,4-dinitrophenylhydrazone, m. 233°).  $\rho$ -Benzylphenol with I gives 2-( $p$ -nitrophenylazo)-4-benzyl-phenol, Cu-colored crystals from AcOH, m. 172-3°. The oxime of II, m. 206-7° (decompn.) [cf. Spiegler, Monatsh. 5, 100 (1884)], on coupling with  $\rho$ -NO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>Cl, (?; no doubt misprint for I) gives a dark brown, cryst. compd., m. 244-5° (no analysis given), assumed to be a diazoimine (C.A. 1, 2250). ( $\sigma$ -HOC<sub>6</sub>H<sub>4</sub>)<sub>2</sub>CO with I couples normally to 2,2'-dihydroxy-5,5'-bi( $p$ -nitrophenylazo)benzophenone, orange needles from PhNO<sub>2</sub>, m. 230°. 4,5,2-Me<sub>3</sub>(HO)<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CHO, m-O<sub>2</sub>N<sub>2</sub>C<sub>6</sub>H<sub>4</sub>SO<sub>3</sub>Na, and 10% aq. NaOH, boiled for 1 hr., give 4,5,2-Me<sub>3</sub>(HO)<sub>2</sub>C<sub>6</sub>H<sub>4</sub>CHO, m. 70° [cf. Gattermann, C.A. 2, 830], which does not react with I. 5,2,3-Me<sub>3</sub>(HO)(HOCH<sub>2</sub>)C<sub>6</sub>H<sub>4</sub>CHO (2,4-dinitrophenylhydrazone, m. 210-40° (decompn.)) is likewise inert to I. V.  $\sigma$ -Cresolphthaloin, H. Ziegler and G. Zigeuner. Monatsh. 80, 313-14 (1949).—The reaction of  $\sigma$ -cresolphthaloin (I) in alk. soln. with  $p$ -Me<sub>2</sub>C<sub>6</sub>H<sub>4</sub>N<sub>2</sub>Cl has been formulated by Leandri (C.A. 42, 6246) as giving a mono- and a bis( $p$ -tolylazo) deriv. of I, m. 273° and 166°, resp. In contrast, phenolphthalein has been found by Z. and Z. (C.A. 44, 1464a) to be cleaved with diazonium compds. into azophenols and 2-( $p$ -HOC<sub>6</sub>H<sub>4</sub>CO)-CO<sub>2</sub>H (II). It has now been found that I too is cleaved to the extent of 25% into II, and that L.'s compd., m. 160°, is 4,6(or 3,5)-bis( $p$ -tolylazo)- $\sigma$ -cresol (acetate, m. 147°). No exptl. details are given. U. Weiss

CA

10

2,5-Dimercapto-1,3,4-thiadiazole. R. Ziegler and N. Kreisel (Univ. Graz, Austria). Monatsh. 81, 614 (1950).—The possibility that 2,5-dimercapto-1,3,4-thiadiazole (I) might form a resin with HCHO was investigated. I (8 g.) [Ber. 27, 2518 (1894)] in 16 cc. 10% NaOH treated with 3.6 g. of 37% HClO, the mixt. warmed 2 hrs. at 50°, and acidified gives 90% 2,5-bis(hydroxymethylmercapto)-1,3,4-thiadiazole (II), needles, m. 122° (from PhCl or H<sub>2</sub>O), which by the Schotten-Baumen method is converted to I dianzoate, m. 185°, and HCHO. II (1 g.) boiled with 3 cc. PhNH<sub>2</sub> forms 2-mercapto-5-(*p*-aminobenzylmercapto)-1,3,4-thiadiazole (III), m. 220°; benzoylation of III gives I dibenzoate. Diazoized III couples with PhOH and 2-HOC<sub>6</sub>H<sub>4</sub>H to deep red dyes. II shows no signs of resinifying.

David Toid

CA

Products of the condensation of 3,4-dimethylphenol with formaldehyde. G. Ziegler, E. Orges, W. Schaden, and E. Weissenberger. (Mitt. Chem. Österreich. Akad. Wiss., 81, 320 (1970); cf. U.S. 4,140,416, 2,3,4,5-tetra-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>)C<sub>6</sub>H<sub>5</sub>OH (I) (1 g.) and 2,3,4,5-tetra-Me<sub>2</sub>C<sub>6</sub>H<sub>3</sub>)C<sub>6</sub>H<sub>5</sub>OH (I) (1 g.), heated 4 hrs. on a water bath, the cooled, brown, cryst. mass extd. with petr. ether, and the residue recrystd. from dil. EtOH, gave 2,2'-dihydroxy-3,3'-dibromo-4,4',5,5'-tetramethylidiphenylmethane (III), waxy prisms, m. 185°; mixed in. p. with 2,2'-dihydroxy-3,3'-dibromo-4,4',5,5'-tetramethylidiphenylmethane (III), 122°; mixed in. p. with 2,6-bis(2-hydroxy-3-bromo-4,5-dimethylbenzyl)-3,4-dimethylphenol (IV), 138°. 1 (1.8 g.) and 1.6 g. 3,4-DiMeC<sub>6</sub>H<sub>3</sub>OH in 25 ml. EtOH, heated 30 min. on the water bath with 15 ml. 20% H<sub>2</sub>O<sub>2</sub>, the soln. evapd., and the residue recrystd. from dil. EtOH, gave 2,2'-dihydroxy-3-bromo-4,4',5,5'-tetramethylidiphenylmethane (V), rods, m. 167° (decompn.). To 0.72 g. V in 25 ml. CCl<sub>4</sub>, an equiv. amt. of Br in 10 ml. CCl<sub>4</sub> was added dropwise - while heating on a water-bath, the orange-yellow product treated with benzoyl chloride, filtered, evapd., and the residue recrystd. from methylcyclohexane or ac. EtOH to give III, m. 143°. Bromination in the same manner of 2,2'-I-III. Bromination of 2,6-bis(2-hydroxy-4,5-dimethylbenzyl)-3,4-dimethylphenol gave IV, m. 161°. Treatment of 2,2'-dihydroxy-3,3'-dihydroxymethyl-4,4',5,5'-tetramethylidiphenylmethane with HBr in CCl<sub>4</sub> gave 2,2'-dihydroxy-3,3'-bis(bromomethyl)-4,4',5,5'-tetramethylidiphenylmethane (VI), m. 170° (decompn.) (from C<sub>6</sub>H<sub>6</sub>). VI in Et<sub>2</sub>O, satsd. with HCl, reduced with Zn dust, the Et<sub>2</sub>O soln. washed with water, neutralized with NaHCO<sub>3</sub>, dried with CaCl<sub>2</sub>, the Et<sub>2</sub>O evapd., and the oil recrystd. from methylcyclohexane or ac. EtOH, gave 2,2'-dihydroxy-3,3',4,5,5',6'-hexamethylidiphenylmethane, colorless needles, m. 140°.

J. P. Danchy

CR

An electron-transfer effect in 4,4'-dihydroxybenzophenone. R. Schauerer, R. Ziegler, and W. Bernitt (University of Graz, Austria). *Makromol.*, **13**, 95 (1952); cf. C.A., **44**, 14631, 1950; **14**, 19318. —The ultraviolet absorption spectra are given of PhAc (I), PhCOH(COPh) (II), PhC(R)OH(COPh) (III), BaPb (IV), (4-PO<sub>2</sub>C<sub>6</sub>H<sub>4</sub>)<sub>2</sub>Cl (V) in EtOH-HCl of pH 2(a); and V in EtOH-NaOH of pH 13 (b). For all the compds., the CO max. is at about 3010 Å. It disappears on conversion to the oximes. I, II, III, and IV have an intense max. at 3500 Å, and a more intense max. at about 4000 Å, characteristic of the aromatic ring. The OH in Va effect the merging of the CO max. and the longer-wave of the 2 aromatic max. into a single band of rather great width and intensity at 3350 Å (a max. at about 4000 Å is absent). The effect of pH on the dimension of the phenolic OH groups is shown by the curves for Va and Vb. The phenolate ion in Vb has an inflection in the

3400-Å. region; Vb also has the 2nd aromatic max. at 4000 Å. The p-OH groups induce conjugation between CO and an aromatic ring through intramol.  $\pi$ -electron transfers resulting in a resonating system. . . . Herman Skolnik

HUNGARIAN TECHNICAL ABSTRACTS  
1958, Vol 10, Nr 2

6/11 K. Illés, K. Zsoldos. D. es műszaki elektromotorok — Egyszerűen ismertetve.  
Budapest, 1957. Támasz Kiadó, 445 p., Ft 40.—

3

DORIN, Pavel, allamdijas, a műszaki tudományok doktora, egyetemi tanár;  
ZIEGLER, Karoly, gépeszmérnök [translator]

New methods for determination of the characteristics of  
reaction turbines. Hidrologiai kozlony 38 no.2:102-109 Ap'58.

1. Bukaresti Muegyetem (for Dorin). 2. Vizugyi Tervezo Iroda  
(for Ziegler).

ZIEGLER, Karoly, dr.

"A new pumping installation for increasing the value of  
reserve energies" by M. Wenger. Reviewed by Karoly Ziegler.  
Energia es atom 16 no.10/11 468 01/63.

1. Vizugyi Tervezo Iroda.

ZIEGLER, K.

Alkali metals; achievements and prospective possibilities in  
industrial chemistry. Tehnika Jug 18 no.10:Supplement: Hem-  
industrija 17 no.10:1929 0'63.

ILLEI, Vilmos; KOVATS, D.; Geza; MATRAI, Istvan; ZIEGLER, Karoly;  
RASONYI, Gyorgy

Efficiency of production and utilization of water power.  
Energia es atom 14 no.4/5:190-195 My '61.

1. Vizugyi Tervezo Iroda. 2. "Energia es Atomtechnika"  
szerkeszto bizottsagi tagja (for Illei).

ZIEGLER, Karoly, dr., okleveles gépész mérnök, a műszaki tudományok kandidátusa

Loss reduction at pumping stations by means of back-pressure closing devices with forced opening. Vizugyi közl. no. 3:460-463 '62.

1. Vizugyi Tervező Iroda gépészeti osztályának vezetője.

PAPP, Ferenc, dr.; BOZSONY, Denes; PICHLER, Janos; ZIEGLER, Karoly;  
ERODY, Bela; DEGEN, Imre; HETENYI, Endre; NEMETH, Endre, dr.h.c.,  
a muszaki tudomanyok doktora, műegyetemi tanár.

An account of the annual meeting of the Hungarian Hydrological  
Society arranged on February 24, 1961. Hidrologiai kozlony 41.  
no.4: 356-360 Ag'61

1. Magyar Hidrologiai Tarsasag elnöke; "Hidrologiai Kozlony"  
szerkeszto bizottsagi elnöke (for Papp). 2. Magyar Hidrologiai  
Tarsasag fotitkara; "Hidrologiai Kozlony" szerkeszto bizottsagi  
tagja (for Bozsony). 3. Orszagos Mezogazdasagi Minosegvizsgalo  
Intezet (for Erody). 4. Orszagos vizugyi foigazgato (for Degen).  
5. Magyar Tudomanyos Akademia Vizgazdalkodasi es Hidrologiai  
Foosztalya; Epitoipari es Kozlekedesi Muszaki Egyetem I. Vize-  
pitestani Tanszeke, Budapest (for Nemeth).

"APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5

ZIEGLER, Karoly

"Water power plants" by Dr. Emil Mosonyi. Reviewed by Karoly Ziegler.  
Energia es atom 13 no.1/2:65 Ja-F '60.

APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5"

ZIEGLER, Karoly, okleveles mernok

International water resources management and Hungary's agreements  
on water right. Vizugyi kozl no.4:411-419 '61.

1. Nyugalmazott orszagos vizugyi foigazgatohelyettes.

CSAJAGHY, Gabor; BOZSONY, Denes; PICHLER, Janos; KASSAI, Ferenc;  
GYORGY, Istvan; SZABO, Pal Zoltan; DEVENY, Istvan (Szeged);  
KIRALY, Lajos (Miskolc); ZIEGLER, Karoly; PAPP, Szilard;  
SCHMIDT, Eligius Robert; GALLI, Laszlo; VAIJDA, Jozsef;  
RONAI, Andras; ILLES, Gyorgu; OLLOS, Geza; FINALY, Lajos;  
MOSONYI, Emil; PAPP, Ferenc

Minutes of the December 19, 1958 general meeting arranged by  
the Hungarian Hydrological Society, Hidrologiai kozlony 39  
no.5:39A 401-404 0 '59.

1."Hidrologiai Kozlony" szerkeszto bizottsagi tagja (for  
CsaJaghy, Gyorgy, Szilard, Papp, Ferenc Papp, Schmidt and  
Galli). 2. Orszagos Vizugyi Foigazgatosag (for Ziegler).

ZIEGLER, K., Jr.

Automatization of hydraulic machinery and hydraulic devices. p. 406.  
GEP, Budapest, Vol. 6, no. 8/9, Aug./Sept. 1954.

SO: Monthly List of East European Accessions, (HEAL), LC, Vol. 4, no. 10, Oct. 1955,  
Uncl.

MADAS, Andras, dr.; STELCZER, Karoly; OROSZLANY, Istvan, dr., tanszékvezető docens; MATRAI, Istvan, főmérnök; MANTUANO, József; KARASZI, Kalman; ZIEGLER, Karoly; BARNA, Aladar

Remarks about the lecture by Dr. Ede Kertai entitled "Water resources development in Hungary." Hidrologiai kozlony 43 no.2:95-98 Ap '63.

1. Orszagos Terüthivatal Mezegazdasagi Főosztalyanak vezetője (for Madas).
2. Vizgazdalkodasi Tudomanyos Kutato Intezet igazgatőja (for Stelczer).
3. Gödöllői Agrartudomanyi Egyetem; "Hidrologiai Kozlony" szerkeszeti bizottsági tagja (for Oroszlany).
4. Vizugyi Tervezo Vallalat (for Matrai).
5. Melyepitesi Tervezo Vallalat osztalyvezetője (for Mantuano).
6. Kezepdúmantuli Vizugyi Igazgatesag igazgatója (for Karaszi).
7. "Hidrologiai Kozlony" szerkeszeti bizottsági tagja (for Ziegler).

CZECHOSLOVAKIA

ZIEGLER, K. i. Veterinary Research Institute, Department of Parasitology (Vyzkumny Ustav Veterinarniho Lekarstvi, Odd. Parasitologie), Brno - Medlanky.

"Vaccination of Chickens Against Syngamosis."

Prague, Veterinarni Medicina, Vol 11, No 9, Sep 66, pp 569-578

Abstract /Author's English summary modified/: The larvae of *Syngamus trachea* must receive an irradiation of 8000 r to destroy their ability to penetrate into the trachea and develop there. Experiments were conducted within the range of 4 to  $30 \times 10^3$  r; best results were obtained with a vaccine that received an irradiation of 20000 r; this vaccine produced immunity for 30 days following vaccination. 3 Figures, 3 Tables, 3 Western, 1 Czech, 1 Russian reference. (Manuscript received 30 Dec 65).

1/1

- 79 -

S/282/63/000/002/005/005  
A059/A126

AUTHORS: Ziegler, Ladislav, Medek, Vlastimir, Jelinek, Tomáš

TITLE: Agitator for epoxy resin compounds

PERIODICAL: Referativnyy zhurnal, otdel'nyy vypusk, 47. Khimicheskoye i khologenicheskoye mashinostroyeniye, no. 2, 1963, 63, abstract 2.47.379 P (Czech. pat. 39 a, 19/07, no. 100806, September 15, 1961)

TEXT: An agitator is described consisting of two drums which are disposed one over the other: a vertical and a horizontal one equipped with thermostats to maintain the given temperatures. After the epoxy resin has been agitated with the filler in the stationary drum with rotating shovels, the mixture obtained is fed to the horizontal rotating drum with bevel bottoms. After a hardener has been added to the mixture, the horizontal drum is hermetically sealed and air is evacuated from its internal cavity through the channels of the driving shaft. After the required evacuation has been reached, the valve in the vacuum line is closed and the horizontal drum rotated. The components are mixed with the aid of a stationary perforated mixer and a scraper kept in the

Card 1/2

Agitator for epoxy resin compounds.

S/282/63/000/002/005/005  
A059/A126

vertical position in the drum with a counterweight. The technique indicated permits to obtain a homogeneous mixture without any bubbles within a short time. There are 2 figures.

K. Onosovskiy

[Abstracter's note: Complete translation]

Card 2/2

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CIA-RDP86-00513R002065110010-5"

Ziegler, M.A.

HUNGARY/Theoretical Physics - Classical Electrodynamics. Classical B-3  
Field Theory.

Abs Jour : Ref Zhur - Fizika, No 4, 1958, No 7570

Author : Marx, G., Ziegler, M.A.

Inst : \*R. Eotvos University\*\* Central Research Institute of Physics;  
Budapest, Hungary.

Title : Relativistic Two-Body Problem in the Classical Theory of  
Meson Field

Orig Pub : Acta phys. Acad. sci. hung., 1957, 7, No 1, 125-133

Abstract : The authors consider the motion of two particles of equal mass,  
interacting with each other through a classical scalar meson  
field. In this examination a count is taken of the delay.  
The solution of the equations of motion is obtained by numeri-  
cal integration. At small distances, the absolute value of  
the meson potential reaches the values of the rest energy of  
the particle  $mc^2$ , i.e., the acceleration at this point reverses  
its sign and "relativistic repulsion" takes place. It is noted  
that analogous results were obtained in the work by Kunin and  
Taksar, Verlet, Marx & Samossy, who have considered the motion  
of a single particle in a potential field of a particle at rest.

Card : 1/1

HUNGARY/Nuclear Physics

C-4

Abs Jour : Ref Zhur - Fizika, No 5, 1957, 11140

one assumes the value  $r_0 = 1.2 \times 10^{-13}$  cm, corresponding to the experimental data on the scattering of electrons, it becomes necessary to assume that the volume occupied by the protons in the nucleus is less than the volume occupied by the neutrons. With this, there appears in the semi-empirical formula for the binding energy a term  $a_4(A - 2Z)$ , which takes into account the difference in the mean kinetic energy of the neutrons and protons, and one also obtains

$$\Delta E = -2a_4 + a_5 A^{2/3}. \quad \text{When } r_0 = 1.2 \times 10^{-13}$$

cm and the experimentally-determined value of E is used, the value obtained for  $a_4$  is 0.64 Mev.

Card 2/3

HUNGARY/Nuclear Physics

C-4

Abs Jour : Ref Zhur - Fizika, No 5, 1957, 11140

The resultant semi-empirical formula

$$E = -15.6A + 17.1A^{2/3} + 21.2 \frac{(A-Z)^2}{A} + \\ + 0.6(A-Z) + 0.72Z^2 A^{-1/3} + \delta$$

is in agreement with the experimental data over the entire interval of variation of A.

Card 3/3

Ziegler, M.A.

3

7270 Calculation of pairing energy by means of the  
classification of the Yukawa potential. M. A.  
ZIEGLER. Note in *Acta phys. Hung.* 4, No. 3, 1951.

If a strong spheroidal lens is used in the radial  
equation, pairing energies (which depend on the radial  
eigenfunctions) are obtained such that in orbits of the  
same azimuthal quantum number, that of higher  
angular momentum has the higher pairing energy.

O. E. HEDBERG

ZIEGLER, R., Dipl. Ing.

Influence of C, Si, Mn, P, and S on some properties of lamellar graphite cast iron. Slevarenstvi 11 no.8/9:308-315 Ag '63.

1. Giesserei-Institut, Leoben Austria.

ZIEGLER, V.

ZIEGLER, V. Gyorgy Gerle's Beruhazasok gazdasagimuszaki tervezese (Economic-Technical Planning of Investments); a book review. p. 3C.

No. 24, Dec. 1955.

MUSZAKI ELET.

TECHNOLOGY

Budapest, Hungary

So: East European Accession, Vol. 5, No. 5, May 1956

ZIEGLER, V.

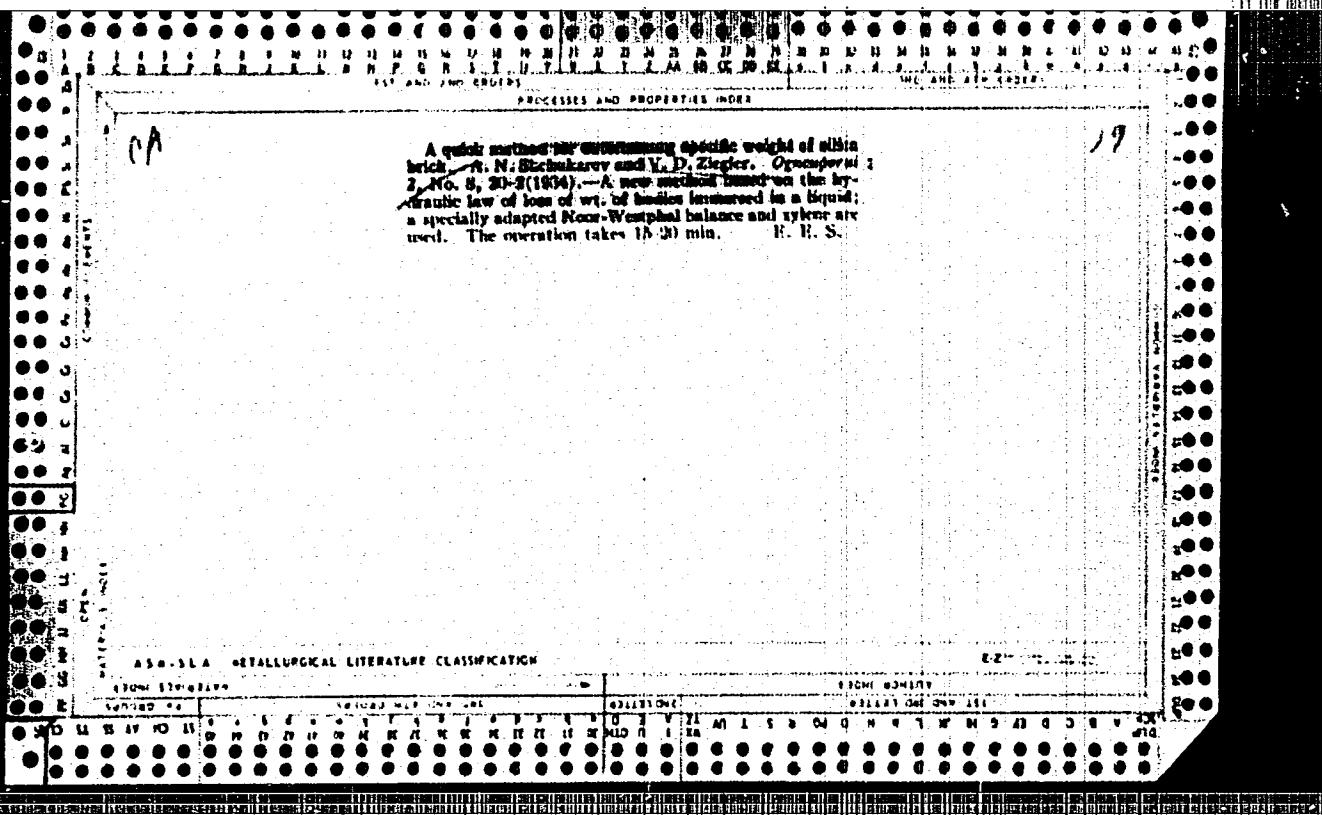
"The Words Nyomas, Szivas, Vakuum, Huzat", P. 142, (SZABVANYCSITAC,  
Vol. 5, No. 9, September 1953, Budapest, Hungary)

SC: Monthly List of East European Accessions (EEAL), LC, Vol. 4, No. 3,  
March 1955, Uncl.

VIEWEG, Reiner; ZIEGLER, Vilmos [translator]

Processing hydrochloric acid waste. Magy kem lap 18 no.2/3:  
85-89 F-Mr '63.

1. WEB KIR CHemie, Leipzig (for Vieweg). 2. Vegyimuveklet  
Tervezo Vallalat (for Ziegler).



JANOSSY, Lajos; ZIEGLER-NARAY, Maria

The hydromechanical model of wave mechanics. Pt. 2. Acta phys  
Hung 16 no. 4:345-353 '64.

1. Central Research Institute of Physics, Budapest. 2. Editorial  
board member, "Acta Physica Academiae Scientiarum Hungaricae" (for  
Janossy).

FEDELESOVA, M.; ZIEGLHOFFER, A.

Study of changes in adenosine triphosphoric-, adenosine di-phosphoric- and adenosine monophosphoric acids in the blood during restricted blood circulation. Bratisl. lek. listy 2 no.11:648-654 '63.

1. Oddelenie experimentalnej chirurgie Ustavu experimentalnej mediciny SAV; vedouci: akademik K.Siska.

\*

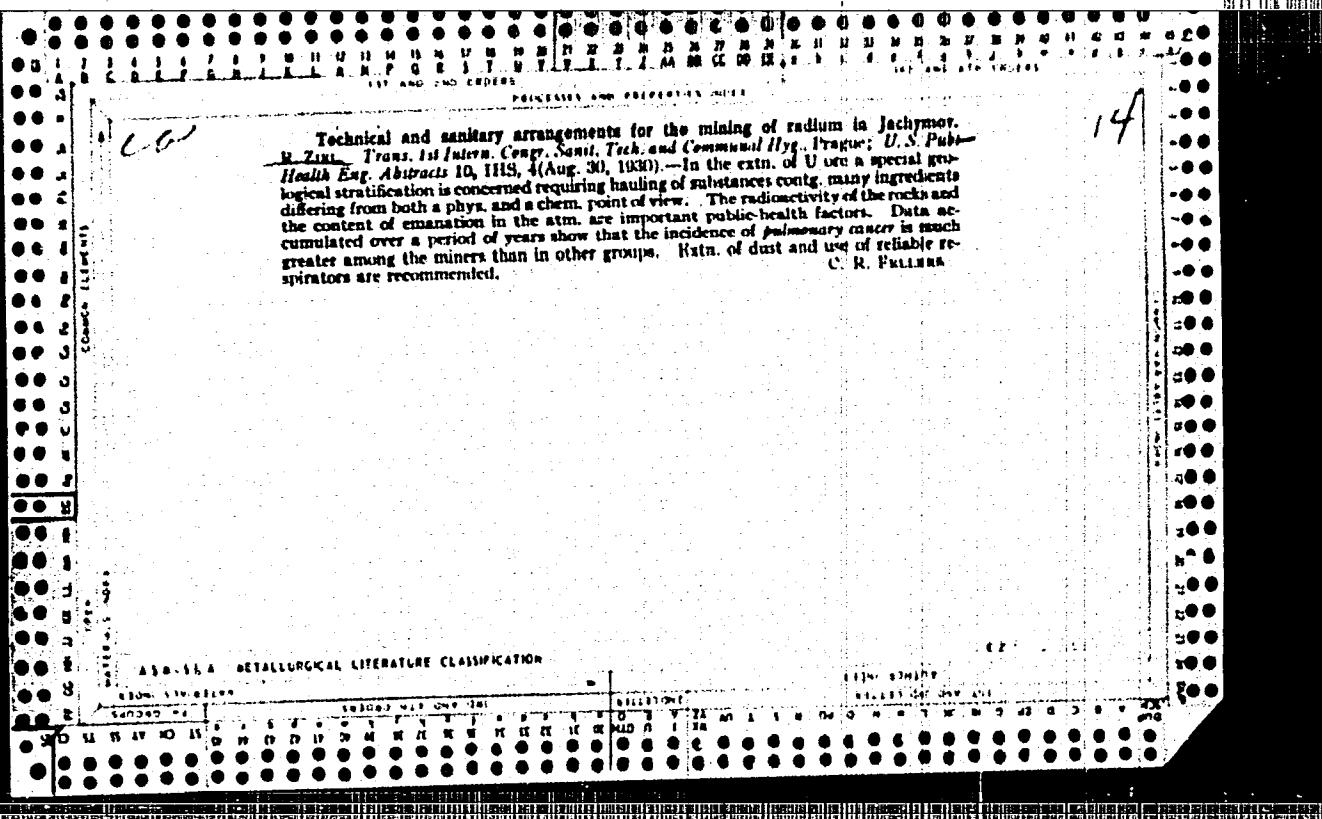
ZIEL.

Summer camps. p. 3.

Vol 6, no. 14, June 1953. ROLNIK SPOLDZIELCA. Warsaw, Poland

Vol 8, no. 35, Aug. 1955

So: Eastern European Accession. Vol 5, no. 4, April 1956



JAWORSKI, Marian; ZIELASKO, Antoni; GASICKA, Krystyna

Colorometric determination of ethylene oxide. Chem anal  
6 no.6:1005-1012 '61.

1. Institute of Organic Synthesis, Bialowina Slaska.

ZIELASKOWSKI.

"People's councils mobilize peasants for Road Deeds!" p. 10, (DROGWICTWO  
Vol. 10, No. 1, Jan. 1955. Warszawa, Poland)

SO: Monthly List of East European Accessions. (EEAL). LC, Vol. 4, No. 4.  
April 1955. Uncl.

"APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5

55188-65

WT(3)

LJF(6)

1970-1971

1970-1971

ABSTRACT: The capacity factor of paddles has been estimated experimentally for three

types of paddles. A method is given to estimate the depth capacity factor. The

APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5"

SUBMITTED: 30Apr64

ENCL# 00

SUB OUDIT: NP

L 33011-66

ACC NR: AP6024170

SOURCE CODE: FO/0046/65/C10/012/0791/0806

41

B

AUTHOR: Czerniewski, Michal--Chernevski, M.; Panta, Przemyslaw--Pan'ta, P.;  
Zielczynski, Mieczyslaw--Zel'chin'ski, M.; Zak, Wieslaw--Zhak, V.; Zarnowiecki,  
Krzysztof--Zharnovetski, K.ORG: Reactor Exploitation Department, Institute of Nuclear Research, Warsaw; Health  
Physics Department, Institute of Nuclear Research, WarsawTITLE: Bone tissue sterilization <sup>19</sup> using reactor fuel gamma radiation

SOURCE: Nukleonika, v. 10, no. 12, 1965, 791-806

TOPIC TAGS: bone, nuclear fuel, gamma radiation, radiation biologic effect,  
radiotherapyABSTRACT: An absolute ionization method of measurements of doses absorbed in bone  
tissue, and additional methods were developed. Measurements of spatial dose distri-  
bution in grafts were performed. From the detailed analysis it follows that each  
point of the graft absorbs in sterilization a dose of 3.3 Krad, with an accuracy of  
20%. In the two years of its application the sterilization method developed has  
proved satisfactory. This was evidenced in sterilization of more than one hundred  
lyophilized human bone grafts successfully used for therapeutical purposes. Theauthors thank Professor K. Ostrowski for his suggestion to use the facilities of the  
EWA APPROVED FOR RELEASE 09/19/2001 CIA RDP86-00513R002065110010-5  
and also for his valuable comments. The  
authors also thank Mr. J. Aleksandrowicz for over-all assistance in the project, Docent  
Z. Zagorski for discussion on the subject of chemical dosimeters and Mr. T. Berens for  
designing the containers, and general help. Orig. art. has: 12 figures and 14 formulas.  
Orig. art. in Eng. / 44A  
SUB-CODE: 06-18 / SUBM DATE: 14Oct65 / ORIG REF: 003 / SOV REF: 001 / OTH REF: 027

Sub-Cod

71 pl

09/19 1760

27.2400

39030  
P/046/62/007/003/004/008  
D256/D308

AUTHOR: Zieliński, Mieczysław

TITLE: Use of columnar ion recombination for determining  
the relative biological effectiveness of radiation

PERIODICAL: Nukleonika, v. 7, no. 3, 1962, 175-182

TEXT: A method of specific ionization measurements is proposed using a tissue-equivalent chamber with maintained conditions of columnar ion recombination; the specific ionization can then be measured directly in terms of the number of ions that recombine in the chamber. It is shown that if the conditions of recombination in the chamber are maintained in such a way that the efficiency of the ion collection is a linear function of the relative biological effectiveness (RBE), then the determined mean value of the specific ionization will correspond to the RBE of the radiation, independently of the composition and the character of the radiation. The method is suitable for RBE measurements and monitoring of compound radiations of an unknown composition, e.g. in conjunction with operation

Card 1/2

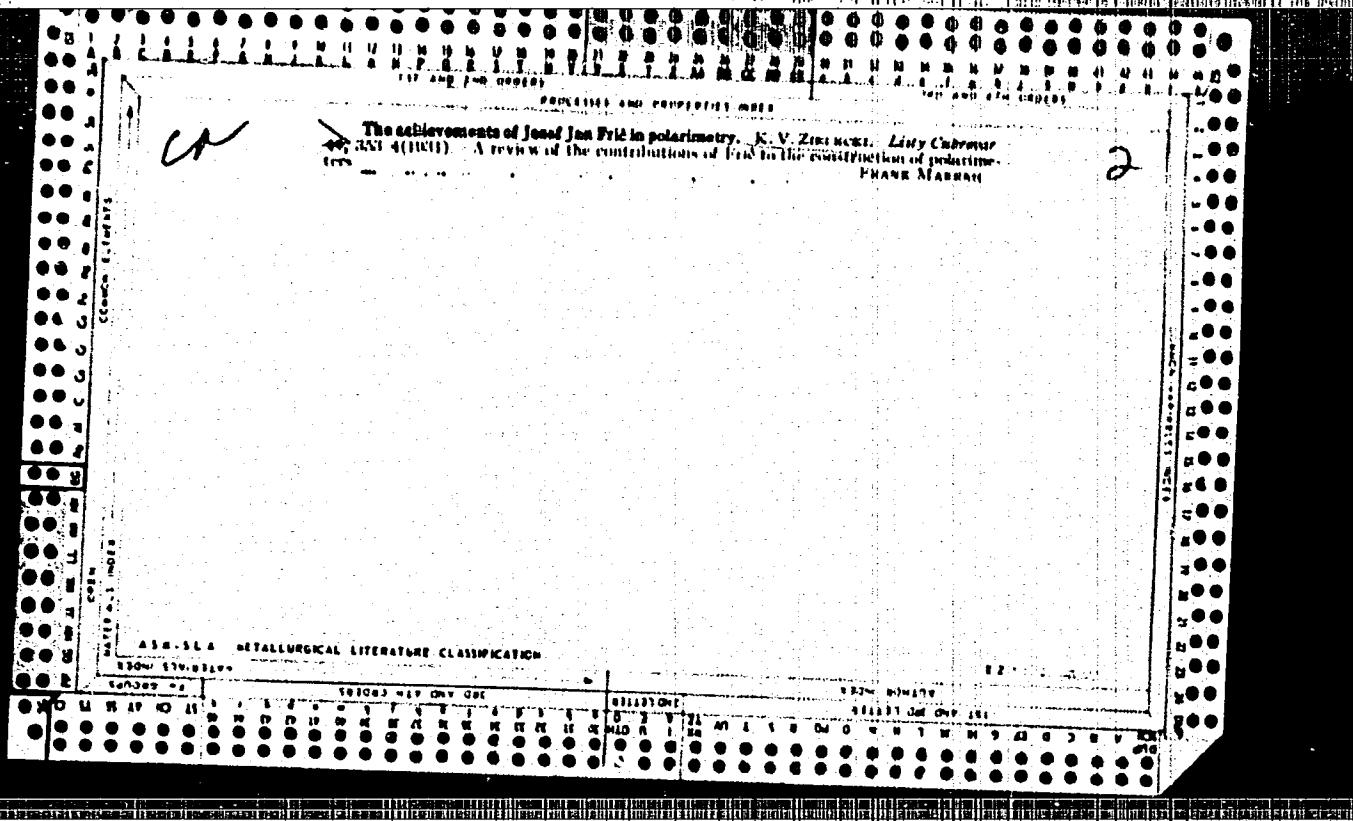
39030  
P/046/62/007/003/004/008  
D256/D308

Use of columnar ion ...

of particle accelerators. There are 3 figures. The most important English-language reference reads as follows: H.H. Rossi, Rad. Res., 10, no. 5, 522 (1959).

ASSOCIATION: Ob'yedinennyj institut yadernykh issledovanij  
Moskva (Joint Institute of Nuclear Research,  
Moscow)

Card 2/2



Distr: EG3d

Mean specific heats of some ternary azeotropes. W. Świetłowski and A. Zieleniewicz (Inst. Chem. Fiz. P.A.N., Warsaw). "Bull. Acad. polon. sci., Ser. sci., Chem. gal. et geogr.", 6, 365-8 (1958) (in English); cf. following abstract.—Mean specific heats for the temp. ranges between room temp. and the respective b.p.s were detd. for liquid mixts. of const. compn. equal to that of the azeotrope at atm. pressure. Weighed samples were transferred from a Świetłowski ebulliometer used as a thermostat to an isothermal calorimeter, water being used as a standard. The temp. ranges and specific heats for indicated compounds or azeotropes were: pyridine-(I)-AcOH-heptane (II) azeotrope, 60.3-21.8°, 0.689; I, 96.3-21.7°, 0.444; AcOH, 96.3-22.0°, 0.556; II, 96.3-21.8°, 0.590; I-AcOH-nonane (III) azeotrope, 120.0-22.1°, 0.002; I, 120.0-22.1°, 0.482; AcOH, 129.0-23.4°, 0.547; III, 129.0-22.1°, 0.000 cal./°C degree, resp.

Country : Poland  
Category : Physical Chemistry - Thermodynamics, Thermochemistry,  
Equilibria, Physicochemical analysis, Phase Transitions.  
Abs. Jour : RZhKhim., No 13, 1959  
Page No. : 45053  
Author : Swietoslawski, W. and Zielenkiewicz, A.  
Institut. : Not given  
Title : Heats of Evaporation in Homologous Series of Bi-  
nary Azeotropes  
Orig Pub. : Roczniki Chem., 32, No 4, 913-922 (1958)  
Abstract : The authors have measured the heat of evaporation  
of binary positive azeotropes ( $A_1B_1$ ) formed by:  
(1) aromatic hydrocarbons (benzene, toluene, p-  
xylene), acting as the azeotropic agent  $A_1$ , and  
homologous series of primary aliphatic alcohols  
( $H_1$ ) (methyl, ethyl, n-propyl, isopropyl, isobutyl,  
n-butyl, isoamyl, and n-hexyl alcohol) and (2)  
pyridine ( $A_2$ ) with hydrocarbons ( $H_1$ ) (heptane, n-  
octane, and n-nonane). The measurements were made  
at the bo of the respective azeotropes, using a  
method described earlier (RZhKhim., 1958, No 16,  
55592). When the heat of evaporation of the azeo-  
tropic agent  $A$  is higher than that of the series  
Card: 1/2

Country	: Poland	B.
Category	:	
Abs. Numr	:	45055
Author	: Swietorlaski, W. and Zieleniewicz, A.	
Institution	: Not given	
Title	Average Specific Heat Capacities of Positive Binary Azeotropes	
Orig Pub.	Roczniki Chem, 32, No 4, 923-928 (1958)	
Abstract	The authors have measured the average specific heat capacities ( $\bar{C}_p$ ) of the following positive binary azeotropes: toluene-isopropyl alcohol, toluene-isobutyl alcohol, p-xylene-isobutyl alcohol, p-xylene-isobutyl alcohol [sic!], p-xylene-isoamyl alc and o-xylene-isoamyl alc. The measurements were made over a temperature range extending from the bp of the respective azeotropes to a temperature of about 21°. The $\bar{C}_p$ of the pure components were also measured. It has been observed, in agreement with previously published work (R. Kremann, Die Eigenschaften der binaeren Fluessigkeitsgemische,	

Card: 1/2

CZARNOTA, J.; BARANOWSKI, B.; ZIELENKIEWICZ, W.

Characteristics of heat exchange in a differential microcalorimeter.  
Bul chim PAN 12 no.8:561-565 '64.

1. Institute of Physical Chemistry of the Polish Academy of  
Sciences, Warsaw. Submitted June 11, 1964.

L 05302-67

ACC NR: AP7000225

(A)

SOURCE CODE: PO/0099/66/040/002/0323/0326

MACZYNSKI, A., ZIELENKIEWICZ, A. and ZIELENKIEWICZ, W., of the Institute of Physical Chemistry, Polish Academy of Sciences (Instytut Chemii Fizycznej Polskiej Akademii Nauk) Warsaw.

"Ebulliometric Thermostat"

Warsaw, Roczniki Chemii, Vol 40, No 2, 1966, pp 323 - 326

Abstract (Authors' English summary): A thermostat was constructed which allows to maintain the temperature constant of within  $\pm 0.002^{\circ}\text{C}$  over extended periods of time. Orig. art. has: 2 figures and 2 tables. (JPRS: 36,002)

TOPIC TAGS: thermostat, laboratory instrument

SUB CODE: 13 / SUBM DATE: 20 Aug 65 / OTH REF: 003

KH

Card 1/1

0032 0740

POLAND/Organic Chemistry. Synthetic Organic Chemistry.

C

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64359.

Author : Malinowski Stanislaw, Basinski Stanislaw, Olszewska  
Maria, Zieleniewska Hanna.

Inst :

Title : Investigations into Aldol Reactions in the Gaseous  
Phase. III.

Orig Pub: Roczn. chem., 1957, 31, No 1, 123-129.

Abstract: By passing the mixed vapors of equimolar columns  
of formaldehyde and propionic, n-butyric or n-valeric  
aldehydes at temperatures of 250-325° over silica gel  
saturated with liquid glass consisting of Na<sub>2</sub>O : 3.18  
SiO<sub>2</sub> to the concentration of 7%, the corresponding  
alpha-methyl (I), alpha-ethyl (II) and alpha-(n propyl)-  
acroleins (III) are produced. The reactions are carried

Card : 1/2

4

APPROVED FOR RELEASE: 09/19/2001 CIA-RDP86-00513R002065110010-5"

Abs Jour: Ref. Zhur-Khimiya, No 19, 1958, 64359.

on as described earlier (see RZhKhim, 1958, 1204).  
The yields of I-III depend on temperature, volume  
of catalyst and the nature of the original alde-  
hydes. Under optimum conditions (275°, 40-45 g. of  
aldehyde per 1 g. of catalyst per hour), yields  
were: 45.5% for (I), 49.2% for (II), 59% for (III).  
I -III are characterized as semicarbazones and  
2,4 dinitrophenylhydrazones.

Card : 2/2

ZIELENIEWSKI, B.

The influence of spacing on the yield of sugar beets.

p. 57. ( GAZETA CUKRONICZA.) (Warszawa, Poland) Vol. 60. No. 2, Feb. 1958

SO: Monthly Index of East European Accession (EEAI) LC Vol. 7, No. 5, 1958

ZIELENIEWSKI, B.

TECHNOLOGY

Periodicals: GAZETA CUKROWNICZA. Vol. 60, no. 10, Oct 1958

ZIELENIEWSKI, B. Manuring sugar beets. p. 329.

Monthly List of East European Accessions (EEAI) LC, Vol. 8, no. 8,  
February 1959, Unclass.

"APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5

ZIELENIEWSKI, B.

Mikronawozy. Warszawa, Państwowe Wydawn. Rolnicze i Lense, 1950. 39 p. (Fertilizers containing trace elements)

DA Not in DLC

SO: Monthly List of East European Accessions (EEAL) 10, Vol. 6, no. 7, July 1957. Uncl.

APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5

"APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5

KOTARBINSKI, Tadeusz; ZIKLENIEWSKI, Jan

Labor productivity; some theoretical remarks. Review Pol Academy 5  
no.1:1-17 Ja-Mr '60.  
(Labor pruductivity)

(EKA 10:3)

APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5"

PAWLICKOWSKI, Tadeusz; ZIELENIEWSKI, Jerzy

From studies on regeneration in the Endocrinology  
Institute of the School of Medicine in Lodz. Zesz  
probl nauki pol no.18:72-76 pt.2 '59.

POLAND/General Biology - General Histology.

B.

Abs Jour : Ref Zhur - Biol., No 21, 1958, 94597

Author : Ber, Artur; Hochlinger, Helena; Zieleniowski, Jerzy  
Inst : -  
Title : Investigation of Melanophore Reaction. I. Influence

of Water Extracts of Postfermentative Penicillium Mycelium Chrysogenum Q-176 on the (Content of) Pigment in Tadpoles Xenopus laevis Daudin.

Orig Pub : Endokrynl. polska, 1956, 7, 188-194.

Abstract : The development of melanophores was shown in tadpoles treated with an extract of Penicillium. The authors exclude the possibility of the presence of melanophore hormone in the mycelium. -- From the authors' resume.

Card 1/1

BER, Artur; ZIELNIKOWSKI, Józef

Effect of prolonged starvation on limb regeneration in the frog.  
*Xenopus laevis.* Pat.Polska 9 no.1:35-38 Jan-Mar '58..

1. Z Zakładu Endokrynologii A.M. w Łodzi Kierownik: prof. dr A. Ber  
Adres autora: Łódź, ul. 22 Lipca 29/7.

(STARVATION, eff.

on limb regen. in frog (Pol))

(REGENERATION,

eff. of starvation on limb regen. in frog (Pol))

"APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5

KOTARBINSKI, T., prof.; ZIELENIEWSKI, J. doc.dr.

Labor productivity. Przegl techn 81 no.6:5-7 F '60.

APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5"

ZIELENIEWSKI, Jan, dr. (Warszawa)

Joseph Kunert's Transactions in maritime commerce; a book review.  
Tech gosp morska 11 no.7/8:229-230 Jl-Ag '61.

ZIELENIEWSKI, Jerzy; MUSIALOWA, Maria

Behavior of some morphological elements of the blood during  
adrenal regeneration in Sprague-Dawley rats. Endokr. Pol.  
16 no.4:425-430 Jl-Ag '65.

1. Katedra i Zaklad Endokrynologii AM w Lodzi (Kierownik:  
prof. dr. T. Pawlikowski).

PAWLICKOWSKI, Tadeusz, prof. dr.; ZIELINSKI, Jerzy

Influence of ACTH and cortisol on the adrenal regeneration in Sprague-Dawley rats. Endokr. Pol. 15 no. 6:629-636 - 1974.

1. Zaklad Endokrynologii Akademii Medycznej w Lublinie (Kierownik: prof. dr. T. Pawlikowski).

MIKOŁAJCZYK, Henryk; ZIELENIEWSKI, Jerzy

Effect of STH, TSH, ACTH, FSH, DOCA, cortisone and high-carbohydrate and high-protein diets on the urinary excretion of acid mucopolysaccharides and 17-ketosteroids in 2 normal males. Endokr. Pol. 14 no.6:581-586 N-D '63.

1. Zakład Endokrynologii Akademii Medycznej w Łodzi (Kierownik: Prof. dr T. Pawlikowski).

EXCERPTA MEDICA Sec.3 Vol.11/8 Endocrinology Aug 57

1494. BER A. and ZIELENIEWSKI J. Zakl. Endokrynol. AM, Łódź. \*Badania nad elektrotyrografią. Doniesienie 2. Badania chorych z prawidłową przemianą materii. Electrothyrography. 2. Investigations in patients with normal BMR. ENDOKR. POL. 1956, 7/1956 (259-262) Graphs 3 Examining patients with a small goitre and compensated thyroid function by the method of unipolar leads from both lobes of the thyroid gland, it was observed that besides the functional currents of heart and muscles, deviations of an indeterminate origin are present, probably dependent on the thyroid function. (III, 6)

POLAND/Human and Animal Physiology (Normal and Pathological).  
Internal Secretion. Thyroid Gland.

Abs Jour: Ref Zhur-Biol., No 17, 1958, 79755.

Author : Ber, Artur; Zieleniewski, Jerzy.

Inst :

Title : Electrothyreographic Investigations. II. Investigation  
of Patients with Unimpaired Metabolism of Substances.

Orig Pub: Endokrynol. polska, 1956, 7, 259-262.

Abstract: No abstract.

Card : 1/1

ZIELINSKI, Jerzy

A simplified method for breeding *Xenopus laevis*. Endokr. Pol. 15 no.2:227-228 Mr-Ap '64.

1. Zaklad Endokrynologii Akademii Medycznej w Lodz (Kierownik: prof. dr. T. Pawlikowski).

ZIELONOWSKI, R.

POL.

654.781.82.003 QZI.014.3

2132  
Józefik A., Zielonowski R. Technical and Economic Indices for Production and Use of Tipped Cutting Tools.  
"Wskazniki techniczne i ekonomiczne produkcji i uzytkowania narzędzi napawanych". Przegląd Mechaniczny No. 8, 1963, pp. 276-278, 5 figs.

On the basis of data obtained from experiments carried out by the Welding Institute over the production of cutting tools and - from the machine Tool and Machine Institute - of research carried out over the performance of tipped tools, the authors have computed technical and economic indices for the production and utilization of plain milling cutters. These indices are an aid to comparing tipped milling cutters arc-welded by means of ESW-18 electrodes with solid SW-9 high-speed steel milling cutters. A method is suggested for computing and classifying these indices. In summing up the details of this problem, the authors conclude that the mastering of the tipping process will materially contribute towards increasing the performance of tipped tools.

ZIELENIEWSKI, Ryszard, dr; KOZAKIEWICZ, Krystyna, mgr

Experiments concerning air injection in burners. Gaz woda techn  
sanit 37 no.12:401-402 D '63.

1. Central Gas Engineering Laboratory, Krakow Branch.

ZIELENIEWSKI, Ryszard, mgr; KOZAKIEWICZ, Krystyna, mgr

CO content in flue gases as an indicator of incomplete combustion. Gaz woda techn sanit 37 no.8:249-250 Ag '63.

1. Central Gas Engineering Laboratory, Krakow Branch.

ZIELENEWSKI, Ryszard, dr.

Combustion rate of gas mixtures. Gaz woda techn sanit 38  
no. 4:129-134 Ap '64

1. Central Gas Engineering Laboratory, Krakow Branch.

The vivianites of the Polish lowlands. S. Ziekrzynski  
*Compt. rend. soc. sci. Varsovie*, Classe III, 31, 103-6  
(1938); *Chem. Zentr.* 1939, I, 4900.—The vivianites are found as veins or pockets in peat and ore deposits. They have been formed by reactions at low temps. in an anhyd. environment under the influence of microorganisms. The cryst. vivianite (44-80%) is contaminated by plant residues,  $MnCO_3$ ,  $CaCO_3$ ,  $SiO_2$ , colloidal Fe phosphates, silicates and oxides. The crystals show strong pleochroism, the optic axial angle is  $73^{\circ}40'$ , the double refraction is 0.04-0.059. Chem. investigation of a  $H_2SO_4$  soln. of the mineral showed a content in  $Fe_3(PO_4)_2$ , illonite, C and water. When vivianite is heated, water is given off rapidly between 40 and  $80^{\circ}$  (up to 14% of the water). Above this temp. it is evolved slowly, the rate of evolution increasing again only at  $160^{\circ}$ . At  $200^{\circ}$  about 31% of the water has been lost, dehydration still being incomplete even at  $220^{\circ}$ . The color and phys. properties of the vivianite change as it is dehydrated.

H. C. Moore

APPROVED FOR RELEASE: 09/19/2001

CIA-RDP86-00513R002065110010-5"

ZIELENKIEWICZ, A.

Enthalpy of evaporation of heteroazoetropes. Bul chim PAN 12  
no.7:487-490 '64.

1. Institute of Physical Chemistry of the Polish Academy of  
Sciences, Warsaw. Submitted April 30, 1964.

POLAND/Atomic and Molecular Physics - High Pressure Physics.

Abs Jour : Ref Zhur Fizika, No 3, 1960, 5683

D-

of calorimeter is that to measure the thermal effect it is not necessary to know the specific heat of the investigated object. In the LFC use is made of automatic devices for measuring V accurate to  $\pm 0.1\%$ , and for measuring  $\Delta t = t_2 - t_1$  with a maximum error of  $\pm 0.001^\circ$  C. The LFC is placed in a water thermostatic bath, surrounded by an air thermostatic bath. Constant temperature is maintained in the thermostatic bath, accurate to  $\pm 0.001^\circ$  C. The LFC makes it possible to determine the thermal effects on the order of 1 cal/hr under the condition that the rate of flow of the liquid is 500 ml/hr. The LFC is calibrated by means of an electric heater made of manganin. -- T.V. Zakharova

Card 2/2

- 45 -

APPROVED FOR RELEASE: 09/19/2001 CIA-RDP86-00513R002065110010-5  
B-8  
POLAND / Physical Chemistry--Thermodynamics  
Thermochemistry. Equilibrium. Physico-  
chemical analysis. Phase transitions.

Abs Jour : Referat Zhur--Khimiya, No. 11, 1959, 37827

Author : Swietoslawski, W.; and Zielenkiewicz, A.

Inst : Polish Academy of Sciences

Title : Mean Specific Heats of Some Ternary Azeotropes.

Orig Pub : Bull Acad Polon Sci, Ser Sci Chim, Geol et  
Geograph, 6, No. 6, 365-366, XXIX (1958) (in  
English with a Russian summary)

Abstract : The authors have determined the specific heats of the starting components and of the following two mixtures: (1) 10.6 wt% pyridine, 3.4% acetic acid, and 86.0% n-heptane, and (2) 29.3% pyridine, 20.9% acetic acid, and 49.8% n-nonane, corresponding in composition to the positive-negative

Card 1/2

POLAND / Physical Chemistry--Thermodynamics.  
Thermochemistry. Equilibrium. Physico-  
chemical analysis. Phase transitions.

B-8

Abs Jour : Referat Zhur--Khimiya, No. 11, 1959, 37826

*o*-xylene-isoamyl alcohol; the mean specific heats of the pure components were also measured. The measurements were made at 20° intervals up to the bp of the azeotrope (at P = 1 atm) with an error of + 0.3%, using the mixing method. A description of the modified Swietoslawski ebulliometer used in the heating of the specimens investigated is given. The specific heat values calculated by the additivity rule were found to be smaller than the experimental values (the difference, in cal per gm per deg, varied within the limits of +0.017 to +0.058 for the various mixtures). The formation of binary positive azeotropes is related to an increase in the

Card 2/3

POLAND / Physical Chemistry--Thermodynamics.  
Thermochemistry. Equilibrium. Physico-

B-8

chemical analysis. Phase transitions.

APPROVED FOR RELEASE: 09/19/2001 CIA-RDP86-00513R002065110010-5"

Abs Jour : Referat Zhur--Khimiya, No. 11, 1959, 37826

average specific heat, in agreement with earlier published observations (A. K. Zhdanov, Zhur obshchui Khim, 11, 471 (1941)). -- S. Byk

Card 3/3

ACC NR: AP6018255

$$(3) \quad H_{C_9H_{12}} = 13.25 - 8.349 \times 10^{-3} T \text{ (at } 390.7\text{--}437.4^\circ\text{K)}$$

O

These formulas enable one to calculate the enthalpy of vaporization within the indicated temperature intervals with an accuracy equal to that of the calorimetric measurement. The specific heat values used in the calculation of the enthalpy were determined experimentally with an accuracy of  $\pm 0.2\%$ . Orig. art. has: 4 figures, 3 tables, and 3 formulas.

SUB CODE: 07/ SUBM DATE: 03Jun65/ ORIG REF: 013/ OTH REF: 003

Card 2/2 eph

SWIETOSLAWSKI, W.; ZIELINSKIEWICZ, A.

Evaporation enthalpies and entropies of several series of  
azeotropes. Bul Ac Pol chim 6 no.2:111-114 '58. (EEAI 9:6)

1. Institute of Physical Chemistry, Polish Academy of Sciences.  
Presented by W.Swietoslawski.  
(Azeotropes)

SWIETOSLAWSKI, W.; ZIELINSKIEWICZ, A.

Mean specific heat of some ternary azeotropes. Bul Ac Pol chim.  
6 no.6:367-369 '58. (EEAI 9:6)

1. Institute of Physical Chemistry, Polish Academy of Sciences.  
Presented by W. Swietoslawski.  
(Azeotropes) (Specific heat)

HANAKI AZOTROPIE WIEZ

Distr: 4E2c(j)/4E3d

7

Vaporization enthalpy of a homologous series of binary  
azeotropes. Wojciech Świątostawski and Anna Zielen-  
kiewicz (Univ. Warsaw). Roczniki Chem. 32, 143-22  
(1958) [English summary].—The vaporization enthalpy of  
binary pos. azeotropes ( $A_1, H_2$ ) composed of aromatic hydro-  
carbons as azeotropic agents  $A$  (benzene, toluene,  $\beta$ -xylene)  
with primary aliphatic ales, as series of homologs ( $H_1$ ) and  
of pyridine with  $\alpha$ -paraffinic hydrocarbons were determined.  
If the vaporization enthalpy of  $A_1$  is higher than those of  $H_1$ ,  
the gram-vaporization enthalpies of azeotropes increase with  
the normal b.p. of  $H_1$ , whereas if that of  $A_1$  is lower than  
those of  $H_1$ , a reverse phenomenon takes place. The heat  
of mixing at the b.p. of the azeotrope increases with rising  
b.p. of the homolog. The curves of mol. vaporization en-  
thalpies of the azeotropes plotted w/ their compn. show a  
max., the position of which depends on the vaporization  
entropy of  $A_1$ . A. Krajlewski

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J. J.

ANNA ZIELENKIEWICZ

Distr: 4E3d/4E2c(j)

The mean specific heats of binary positive azeotropes. Wojciech Świątowski and Anna Zielenkiewicz (Univ. Warsaw). Roczniki Chem. 34, 623-6 (1960) (long title summary).—A thermostat functioning on the principle of Świątowski's ebulliometer was used to det. mean sp. heats of liquids in the range from room temp. to the b.p. for tercylene (I) (0.433 and 0.441 cal./g. degree), *p*-xylene/(II) (0.447, 0.455), *m*-xylene (III) (0.448), *o*-xylene (IV) (0.408), iso-propyl alc. (V) (0.718), isobutyl alc. (VI) (0.693, 0.704), and isoamyl alc. (VII) (0.690), and of the azeotropes I-V (0.653), I-VI (0.687), II-VI (0.681), III-VI (0.683), II-VII (0.601), and IV-VII (0.630). The sp. heats of azeotropes are higher than the additive values, in agreement with Kreeman and Zhdanov's observation (J. Gen. Chem. U.S.S.R., 11, 471 (1941)).

A. Kriegelwitz

6  
2-may  
2

ZIELENIEWSKI, Ryszard, mgr.

Research on the improvement of gas burning and utilization,  
Gaz woda techn sanit 36 no.5:182-185 My '62.

1. Centralne Laboratorium Gazownictwa, Oddzial Krakow.

ZIELENKIEWICZ, W.

Efforts to save wood. p. 23.

(BUDOWNICTWO WIEJSKIE. Vol. 9, No. 5, May 1957. Warszawa, Poland)

SO: Monthly List of East European Accessions (EEAL) LC, Vol. 6, No. 10, October 1957. Uncl.

POLAND/Atomic and Molecular Physics - High Pressure Physics.

D-

Abs Jour : Ref Zhur Fizika, No 3, 1960, 5681

Author : Sivietoslawski, W., Zielenkiewicz, W.

Inst : Institute of Chemical Physics, Polish Academy of Sciences

Title : Thermostats Used with the Labyrinth Flow Calorimeter

Orig Pub : Bull. Acad. polon. sci. Ser. sci. chem., geol. et geogr.,  
1959, 7, No 2, 107-110

Abstract : Description of a thermostat for a labyrinth-flow calorimeter (Abstract 5683). The thermostat contains 400 liters of water. To stir such an amount of water, four propeller stirrers are used. In addition to the automatic apparatus for the control of the temperature, the thermostat contains also a Beckman ultrathermometer for detecting very small water temperature fluctuations. The changes in temperature in the thermostat did not exceed  $\pm 0.001^\circ\text{C}$  in 14 days.  
-- T.V. Zakharova

Card 1/1

- 44 -

COUNTRY : POLAND  
 CATEGORY : Laboratory Equipment. Apparatus, Their  
 ABS. JOUR. : Construction and Application  
 RZKhim., No. 1 1960, No. 987  
 AUTHOR : Swietoslawski, W.; Zielekiewicz, W.  
 INST. : Polish AS  
 TITLE : On a New Labyrinth Flow Calorimeter  
 ORIG. PUB. : Bull. Acad. polon. sci. Soc. sci. chim., Sec. I,  
 pt geogr., 1959, 7, No 2, 101-105  
 ABSTRACT : A flow calorimeter, in which water washing the  
 reaction vessel (volume 45 ml) passes through  
 a jacket in the form of a labyrinth, has been  
 constructed. The difference of the temperatures  
 of flow water at the inlet and outlet of the  
 labyrinth is measured correct to  $\pm 0.001^\circ$  by  
 means of a battery consisting of 25 copper-  
 constantan thermocouples. To measure the volume  
 of the flowing water, a special apparatus was

CARD:

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F-11

TITLE : RZKhim., No. 1  
 : 1960  
 APPROVED FOR RELEASE: 09/19/2001 No. CIA-RDP86-00513R002065110010-5

ORIG. PUB. :

ABSTRACT  
cont'd

: constructed which permits the measurement of  
 this value correct to  $\pm 0.1\%$ . The calorimeter  
 is located in a 700-liter thermostat whose  
 temperature is maintained constant, correct  
 to  $\pm 0.001^\circ$ , by means of a mercury-toluene  
 thermometer. The water thermostat is lo-  
 cated inside an air thermostat. The calori-  
 meter permits the measurement of small thermal  
 effects of protracted processes of the order  
 of 1 cal/hour. -- A. Vorob'yev

CARD:

2/2

water

SWIETOSLAWSKI, W.; ZIELENKIEWICZ, W.

Thermostats used with the labyrinth flow calorimeter. Bul Ac Pol  
chim ? no.2:107-110 '59. (EEAI 9:7)  
(Calorimeters and calorimetry)  
(Thermostat)

COUNTRY	:	POLAND	F
CATEGORY	:	Laboratory Equipment. Apparatus, Their Theory, Construction and Application	
ABS. JCUR.	:	RZKhim., No. 1 1960, No. 988	
AUTHOR	:	Swietoslawski, W.; Zielenkiewicz, W.	
INST.	:	Polish AS	
TITLE	:	Thermostats Used with the Labyrinth Flow Calo- rimeter	
ORIG. PUB.	:	Bull. Acad. polon. sci. Ser. sci. chim., geol. et geogr., 1959, 7, No 2, 107-110	
ABSTRACT	:	A description of the thermostats used in working with the labyrinth flow calorimeter (see abstr. 987) is given. The calorimeter is located in a water thermostat whose temperature is maintained constant, correct to $\pm 0.001^\circ$ , by means of a mercury-toluene thermoregulator and electronic relay. Diagrams of the electronic relay and of the voltage stabilizer [stabilivolt] of the power supply of the electronic lamps are given.	
CARD:	:	1/2	

F-12

ZIELENIEWSKI, Jan

Training conference in praxiology, Jablonna near Warsaw November  
6-11, 1962. Nauka polska 11 no.2:114-116 Mr-Ap '63.

1. Pracownia Ogolnych Problemow Organizacji Pracy, Polska Akademia  
Nauk, Warszawa.

ZIELENIEWSKI, Jan, Assistant professor

The Research Centre for General Theory of Organization, Review Pol  
Academy 6 no.1:49-52 Ja-Mr '61.

1. Polish Academy of Sciences, Research Center for General Theory of  
Organization, Warsaw, Palac Kultury; nauki.

(Polish Academy of Sciences) (Poland--Research)  
(Poland--Organization)

ZIELENIEWSKI, Jerzy

Effect of adrenal regeneration on ovarian function in white rats.  
Endokr.pol. 14 no.5:467-471 '63.

1. Zaklad Endokrynologii Akademii Medycznej w Lodz. Kierownik:  
prof. dr T. Pawlikowski.

ZISLENIEWSKI, R.; KOZAKIEWICZ, K.

Thermostatic safety devices for gas-buring apparatus. p. 104.

GAX, WODA I TECHNIKA SANITARNA. (Stowarzyszenie Naukowo-Techniczne Inżynierów i Techników Sanitarnych, Ogrzewnictwa i Gazownictwa) Warszawa, Poland.  
Vol. 33, no. 3, March 1959.

Monthly List of East European Accessions EEAI LC, Vol. 8, no. 7, July 1959.

Uncl.

ZIELENIEWSKI, R.

Adaptation of laboratory burners to the combustion of liquid and natural gas.  
p. 153

GAZ, WODA I TECHNIKA SANITARNA. (Stowarzyszenie Naukowo-Techniczne Inżynierów  
i Techników Sanitarnych, Ogrzewnictwa i Gazownictwa) Warszawa, Poland.  
Vol. 33, No. 3, March 1959.

Monthly List of East European Accession' (EEAI) LC, Vol. 8, no. 7, July 1959

Uncl.