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in the presence of tartaric acid, borate reacts with barium to produce a precipitate of constant composition and very low water-solubility. The analysis of this presumably complex barium borotartrate yields to empirical formula $Ba_5B_2C_{12}H_8O_{24} \cdot 4H_2O$. The (Chemistry--Analytical, TT, v. 9, no. 7) (over)

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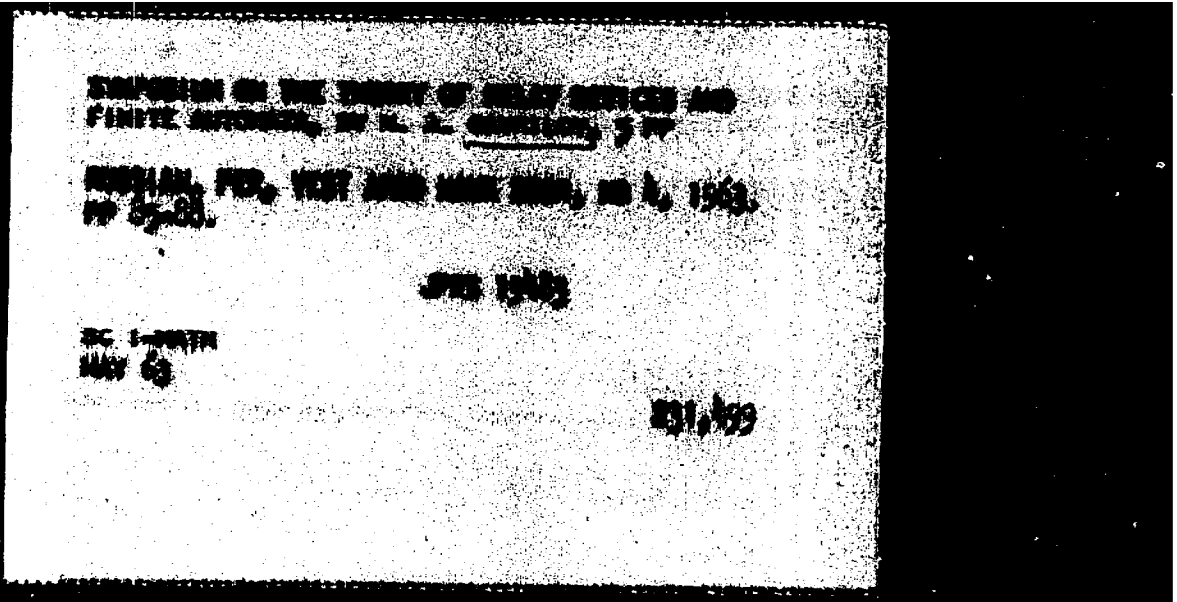
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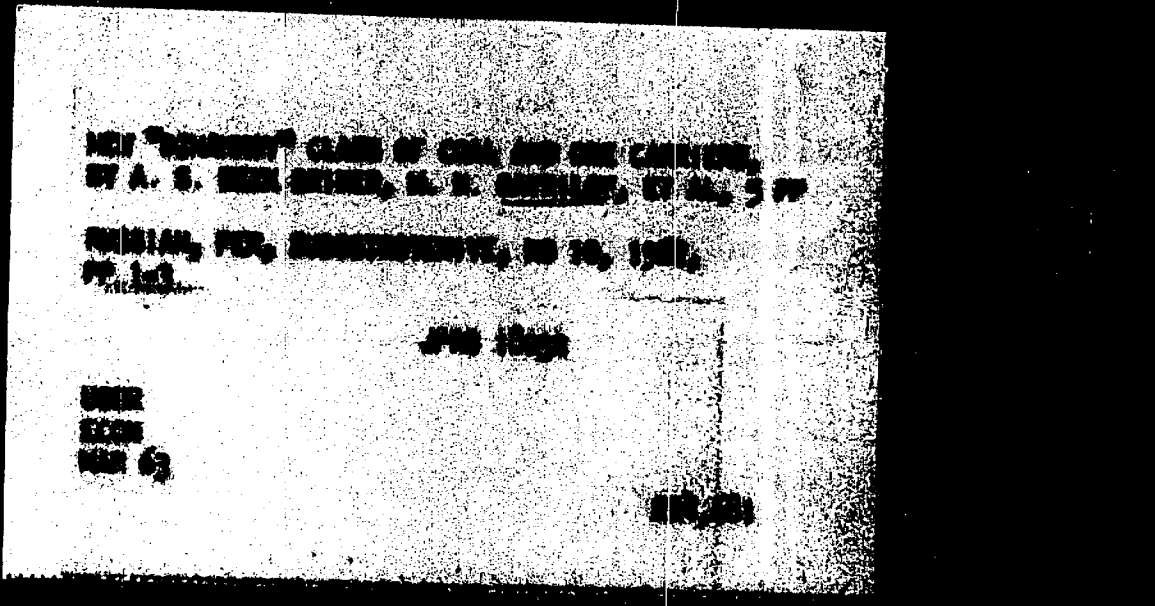
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